

Geological Society, London, Special Publications Online First

## **Mercury fluxes from volcanic and geothermal sources: an update**

E. Bagnato, G. Tamburello, G. Avard, M. Martinez-Cruz, M. Enrico, X. Fu, M. Sprovieri and J. E. Sonke

*Geological Society, London, Special Publications*, first published August 1, 2014; doi 10.1144/SP410.2

---

<b>Email alerting service</b>	click <a href="#">here</a> to receive free e-mail alerts when new articles cite this article
<b>Permission request</b>	click <a href="#">here</a> to seek permission to re-use all or part of this article
<b>Subscribe</b>	click <a href="#">here</a> to subscribe to Geological Society, London, Special Publications or the Lyell Collection
<b>How to cite</b>	click <a href="#">here</a> for further information about Online First and how to cite articles

---

### **Notes**

## Mercury fluxes from volcanic and geothermal sources: an update

E. BAGNATO<sup>1\*</sup>, G. TAMBURELLO<sup>1</sup>, G. AVARD<sup>2</sup>, M. MARTINEZ-CRUZ<sup>2</sup>, M. ENRICO<sup>3</sup>,  
X. FU<sup>3</sup>, M. SPROVIERI<sup>4</sup> & J. E. SONKE<sup>3</sup>

<sup>1</sup>*DiSTeM, University of Palermo, Via Archirafi, 36, 90123 Palermo, Italy*

<sup>2</sup>*Observatorio Vulcanológico y Sismológico de Costa Rica, Universidad Nacional,  
2346-3000 Heredia, Costa Rica*

<sup>3</sup>*Observatoire Midi-Pyrénées, CNRS-GET, 14, avenue Edouard Belin, 31400 Toulouse, France*

<sup>4</sup>*IAMC-CNR, Via del Mare 3, 91021 Torretta Granitola, Mazara del Vallo (TP), Italy*

\*Corresponding author (e-mail: [emanuelabagnato@virgilio.it](mailto:emanuelabagnato@virgilio.it))

**Abstract:** We review the state of knowledge on global volcanogenic Hg emissions to the atmosphere and present new data from seven active volcanoes (Poás, Rincón de la Vieja, Turrialba, Aso, Mutnovsky, Gorely and Etna) and two geothermal fields (Las Pailas and Las Hornillas). The variability of Hg contents ( $c. 4\text{--}125 \text{ ng m}^{-3}$ ) measured in gaseous emissions reflects the dynamic nature of volcanic plumes, where the abundances of volatiles are determined by the physical nature of degassing and variable air dilution. Based on our dataset and previous work, we propose that an average Hg/SO<sub>2</sub> plume mass ratio of  $c. 7.8 \times 10^{-6}$  ( $\pm 1.5 \times 10^{-6}$ ; 1 SE,  $n = 13$ ) is best representative of open-conduit quiescent degassing. Taking into account the uncertainty in global SO<sub>2</sub> emissions, we infer a global volcanic Hg flux from persistent degassing of  $c. 76 \pm 30 \text{ t a}^{-1}$ . Our data are derived from active volcanoes during non-eruptive periods and we do not have any direct constraint on the Hg flux during periods of elevated SO<sub>2</sub> flux associated with large-scale effusive or explosive eruptions. This suggests that the time-averaged Hg flux from these volcanoes is even larger if the eruptive contribution is considered. Conversely, closed-conduit degassing and geothermal emissions contribute modest amounts of Hg.

A significant amount of heavy metals is emitted into the atmosphere from natural (volcanoes, soil, seawater, etc.) and anthropogenic (industrial and mining activities, fossil fuel combustion, etc.) sources. Volatile emissions in passively or eruptive magmatic degassing volcanoes, fumaroles and geothermal fields influence the composition of the atmosphere and global climate, and impact the biogeochemical cycles of a number of trace elements (e.g. Robock & Oppenheimer 2003; Wallace 2005). Knowledge of the contribution of volcanic degassing systems and geothermal fields to the global budgets of heavy metals in the Earth's atmosphere is essential to evaluate the role that the volcanoes play on the reservoir cycling of these elements and the potential fates they could represent once they reach the ecosystems. Volcanic emissions are the major natural source of several trace metals (e.g. As, Cd, Cu, Pb, Zn, Tl, Sb and Sn) to the atmosphere by contributing 20–40% of volatile elements such as Bi, Pb, As or Sb and up to 40–50% of Cd annually (Hinkley *et al.* 1999; Martínez 2008; Mather *et al.* 2012; Henley & Berger 2013). However, current research on trace element emissions from active volcanism shows that the magnitude of some element fluxes is still highly uncertain. This is the case for atmospheric mercury (Hg)

which is a key contaminant in biochemical processes as well as in air, water and soils, for which volcanism may be an important natural contributor (Varekamp & Buseck 1986; Nriagu & Becker 2003; Pyle & Mather 2003; Bagnato *et al.* 2007, 2011; Martin *et al.* 2012). In volcanic plume gaseous elemental mercury (Hg<sup>0</sup> or GEM) is the dominant species ( $\geq 90\%$ ; Slemr *et al.* 1985; Schroeder & Munthe 1998; Bagnato *et al.* 2007; von Glasgow 2010; Martin *et al.* 2011) with an atmospheric lifetime of  $c. 0.5\text{--}1 \text{ a}$  (Lindqvist and Rodhe 1985; Slemr *et al.* 1985; Lindberg *et al.* 2007; Aryia *et al.* 2008), allowing it to be transported to great distances from the source. The relative proportions of gaseous Hg<sub>(g)</sub><sup>0</sup> and Hg(II) species in volcanic gas plumes, and the ratio of gaseous to particulate Hg forms, have rarely been determined in a systematic way (Bagnato *et al.* 2007; Witt *et al.* 2008a, b). Once deposited in surface waters, Hg is assimilated and processed by the biosphere that converts the relatively inert inorganic Hg into toxic methylated forms which can bioaccumulate in the food chain (Morel *et al.* 1998).

The current estimate of Hg emissions from natural processes (primary mercury emissions + re-emissions), is estimated to be  $5207 \text{ t a}^{-1}$  which represents nearly 70% of the global mercury

emission budget (Pirrone *et al.* 2010). Overall, the relative contribution of terrestrial surfaces is  $2429 \text{ t a}^{-1}$  (47% of the total natural emissions) and that from surface waters is  $2778 \text{ t a}^{-1}$  (53% of the total natural emissions) (Pirrone *et al.* 2010). Significant research has been published over the last 10 years quantifying and demonstrating the importance of emissions from volcanoes and Hg-enriched substrates ( $>100$  ppb Hg) (Friedli *et al.* 2003; Gustin 2003; Pyle & Mather 2003; Bagnato *et al.* 2007, 2011; Gustin *et al.* 2008). Earlier and more recent studies proposed that cataclysmic volcanic eruptions would have the potential to inject volatile Hg into the stratosphere (Delmas *et al.* 1992; Langway *et al.* 1995; Robock & Free 1995; Schuster *et al.* 2002) to change its global and regional cycle for a few years. In contrast, quiescent degassing and moderate eruptions exhale directly into the troposphere and can have long-term effects on local environments.

Regionally, a single persistently active volcano may act as the most important local point source (e.g.  $\text{SO}_2$  emissions from Etna volcano, Sicily, to the Mediterranean and Western Europe; Simpson *et al.* 1999; Bagnato *et al.* 2007). However important, there are currently many difficulties in quantifying the Hg flux from volcanic emissions due to the spatial and temporal variability in the activity from one volcano to another (Ferrara *et al.* 2000; Bagnato *et al.* 2011), or from different emission points on the volcano being studied (Dedeurwaerder *et al.* 1982; Bagnato *et al.* 2007). Due to the logistical restrictions of acquiring measurements at volcanoes (difficulty of access, limited power supply, etc.), Hg measurements often only comprise few data points. Two recent reviews narrow down previous estimates of volcanic Hg emission fluxes of  $0.6\text{--}830 \text{ t a}^{-1}$  (Varekamp & Buseck 1986; Nriagu & Pacyna 1988; Ferrara *et al.* 2000) to  $90 \text{ t a}^{-1}$  and  $700 \text{ t a}^{-1}$  (Pyle & Mather 2003; Pirrone *et al.* 2010). Volcanic Hg emissions may therefore represent c. 4–28% of natural primary emissions (Pirrone *et al.* 2010).

The large variability of these estimates partly reflects the complexity of the chemical and physical processes that Hg undergoes while interacting with the atmosphere. Most of the Hg observations used in previous inventories are derived from spot measurements taken from a few volcanoes, and there is much concern over the different methods in which different authors have attempted to assess Hg fluxes. All of these limitations may explain the large uncertainties in extrapolating data to global Hg emissions. Finally, the knowledge of Hg emissions from geothermal sources is still poorly investigated; they are therefore often considered as a negligible contributor to the atmosphere.

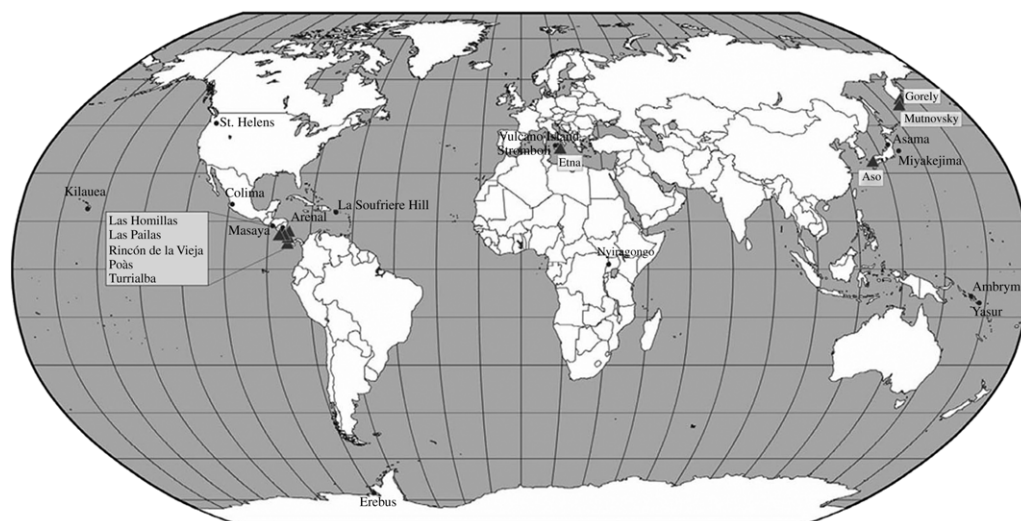
The objective of this work is to extend and refine the currently limited volcanic Hg dataset. We present here new Hg data for seven active volcanoes (Aso, Turrialba, Poás, Rincón de la Vieja, Etna, Gorely and Mutnovsky) and two important geothermal systems (Las Paillas and Las Hornillas, Costa Rica). We evaluate previous work on volcanic Hg fluxes, and combine the new data with the few relevant estimates of Hg fluxes from previous direct measurements on volcanic plumes to develop a realistic assessment of time-averaged volcanic Hg fluxes. Conventional gold-trap sampling and Lumex<sup>®</sup>/MultiGAS GPS-synchronized surveys were carried out to obtain real-time measurements of Hg and major volatiles in the fumes of the investigated volcanoes. These measurements, the first carried out at such locations, extend previous combined Lumex<sup>®</sup>/MultiGAS surveys reported elsewhere (Aiuppa *et al.* 2007; Witt *et al.* 2008a, b; Bagnato *et al.* 2013), and aim at improving our understanding of the magmatic systems and its degassing budget.

## Materials and methods

### Study areas

Measurements were carried out on the crater rim of seven quiescently degassing volcanoes and in two active geothermal areas with various surface manifestations. The study sites that are reported on here are located together with other locations discussed in the text in Figure 1. Aso volcano, in the central part of Kyushu Island ( $32^\circ 53'6.36''\text{N}$ ,  $131^\circ 5'7.44''\text{E}$ ), Japan is an active volcano with a c. 20 km diameter caldera that has some post-caldera central cones, including Mt Nakadake, where the current activity is mainly concentrated. The crater contains one of the hottest and most acidic volcanic lakes, with water temperature in the range  $40\text{--}80^\circ\text{C}$  and acidity levels of pH  $-1$  to  $+1$  during calm periods (Ohsawa *et al.* 2003, 2010; Terada & Sudo 2012). Low-temperature gases are emitted from the lake surface, and intense active high-temperature gas emissions (approximately  $800^\circ\text{C}$ ) are produced by fumaroles on the southern crater wall (Shinohara *et al.* 2010, 2013).

Gorely, Russia ( $52^\circ 33'28.8''\text{N}$ ,  $158^\circ 1'48''\text{E}$ ) is a Holocene caldera volcano, located at the northern end of the southern segment of the eastern volcanic belt of Kamchatka, c. 75 km SW of Petropavlovsk–Kamchatsky. It is one of the most active volcanoes in southern Kamchatka and currently contains an acidic crater lake which often displays persistent fumarolic activity with a steam plume up to several kilometres high on occasion (Kamchatka Volcanic Eruption Response Team, or KVERT). At the time of our campaign (September 2011),



**Fig. 1.** Locations of the studied volcanoes and geothermal systems for which we reported data on Hg emission rate (black filled triangles), and for which literature data are summarized in Table 2 (black filled circles).

vigorous jet-like degassing was taking place from a hot open vent and hot fractures in the northern wall of the crater.

Mutnovsky volcano ( $52^{\circ} 27' 20.88''\text{N}$ ,  $158^{\circ} 11' 32.32''\text{E}$ ) is located in the southern part of the Eastern Volcanic Front of the Kamchatka Peninsula, and has an associated hydrothermal system with magmatic activity (Shishkina *et al.* 2010). There are three individual groups of fumaroles in the Mutnovsky crater which show stable activity for a long period of time: the 'Active Funnel' (temperatures exceed  $600^{\circ}\text{C}$ ); the 'Bottom Field' (from 100 to  $150^{\circ}\text{C}$ ); and the 'Upper Field' (up to  $320^{\circ}\text{C}$ ) (Taran *et al.* 1992). Our investigation focused on the degassing from the first two fumarolic groups.

Mt Etna is the largest volcano in Europe (NE coast of Sicily, south of Italy;  $37^{\circ} 45' 18''\text{N}$ ,  $14^{\circ} 59' 42''\text{E}$ ) and one of the most active volcanoes in the world. Its present-day activity consists of persistent passive degassing of magmatic volatiles mostly from the four open vents on the volcano's summit (North East, Voragine, Bocca Nuova and South East) and diffuse soil emanations (Allard *et al.* 1991), occasionally interrupted by lava emissions associated with paroxysmal terminal to lateral eruptions (Madonia *et al.* 2013; Allard *et al.* 2006).

Poás ( $10^{\circ} 12' 0''\text{N}$ ,  $84^{\circ} 12' 0''\text{W}$ ) and Turrialba ( $10^{\circ} 1' 12.23''\text{N}$ ,  $83^{\circ} 45' 46.92''\text{W}$ ) volcanoes, Costa Rica, lie on the Central America Volcanic Front (Carr 1984; Carr *et al.* 2003). Turrialba is a strato-volcano that has seen increased fumarolic activity since 2002, with new fumarolic vents and fractures opening in and outside of the West Crater area in January 2010 and 2012, respectively. From 2007

onwards, after the hydrothermal stage which characterized the fumarolic discharge since at least the 1980s until fall 2001 (e.g. Cheminée *et al.* 1983; Vaselli *et al.* 2010), both the composition and high temperature (up to  $800^{\circ}\text{C}$ ) of the fumaroles indicate that the gases are directly supplied from the magma chamber with only limited interaction with the hydrothermal system.

Poás is an andesitic-basaltic composite volcano about 50 km from the capital San José. Its upper part consists of three distinct eruptive centres roughly aligned NW–SE, including the Main Crater in the middle that hosts a hyper-acidic lake (Laguna Caliente, pH between 1.8 and *c.* 0) (Martínez *et al.* 2000; Melián *et al.* 2007; Martínez 2008). The activity of the lake (Laguna Caliente) is characterized by persistent fumarolic emissions and a long history of sporadic phreatic eruptions, especially since 2006 (Lopes 2005; Martínez 2008).

The Rincón de la Vieja volcanic complex ( $10^{\circ} 49' 48''\text{N}$ ,  $85^{\circ} 19' 26''\text{W}$ ) is a composite andesitic stratovolcano located in NW Costa Rica, *c.* 25 km from Liberia in the province of Guanacaste and 25 km NW of the Miravalles volcano. It shows a wide range of fluid manifestations (a hyper-acidic crater lake, boiling- and mud-pools, thermal springs and associated gas discharges) (Tassi *et al.* 2005). The fluid vents located inside the Active Crater are likely to represent the most appropriate sampling sites for geochemical surveillance purposes; however, access is fairly difficult due to the steepness of the crater walls.

Las Hornillas and Las Pailas are two important geothermal systems which are part of the Miravalles

and Rincón de la Vieja volcanic complexes, respectively (Guanacaste Geothermal Province, Costa Rica). Both show surface manifestations such as vapour discharges from fumaroles and steam-heated pools. Las Hornillas geothermal field is a high-temperature reservoir, located in the NW part of Costa Rica in the Guanacaste Volcanic Range, c. 150 km from the capital, San José. Temperatures generally range between 230 and 240 °C and gas discharges are dominated by CO<sub>2</sub> with minor amounts of H<sub>2</sub>S and N<sub>2</sub> (Gherardi *et al.* 2002). The Las Pailas is a liquid-dominated two-phase reservoir with temperatures of 220–255 °C (at a depth of about 600–700 m) (Castro 2002), located at the southern flank of the volcano Rincón de la Vieja, Guanacaste Volcanic Cordillera. The heat source is located in the west part of the area and fluid flows are from west to northeast. Fracture permeability enables fluid and heat transport in this reservoir.

### *Major species in the plume*

Major volcanogenic species (H<sub>2</sub>O, CO<sub>2</sub> and SO<sub>2</sub>) in volcanic plumes and fumaroles in tandem with Hg quantities have been measured at some locations (where logistics and weather conditions allowed) by using the MultiGAS analyser, a portable instrument previously described in Aiuppa *et al.* (2010, 2011). The plume gases are actively pumped into the sampler (using an air pump) at 1.2 L min<sup>-1</sup> through a 1 µm Teflon membrane particle filter and pumped through a CO<sub>2</sub> gas detector (Licor LI-840 NDIR closed-path spectrometer; measurement range 0–3000 ppm; accuracy ±1.5%) (see Shinohara *et al.* 2008 for details) and a suite of electrochemical sensors for SO<sub>2</sub> (model 3ST/F City Tech. Ltd., calibration range 0–200 ppm), H<sub>2</sub>S (model EZ3H City Tech. Ltd., calibration range 0–50 ppm) and H<sub>2</sub> (model EZT3HYT City Tech., calibration range 0–50 ppm) detection. A sensor for relative humidity and *T* (model Galltec; 0–100%, –30 to 70 °C) was used in order to measure water content in the gaseous emissions. The sensors were housed in a weatherproof box mounted on a backpack frame and were calibrated, before and after fieldwork, with standard calibration gases (200 ppm SO<sub>2</sub>, 50 ppm H<sub>2</sub>S, 20 ppm H<sub>2</sub> and 3014 ppm CO<sub>2</sub>) mixed with ultrapure nitrogen to provide a range of desired concentrations (Aiuppa *et al.* 2011). The calibration gases also enabled cross sensitivity of the H<sub>2</sub>S sensor to SO<sub>2</sub> to be evaluated. Signals from sensors are simultaneously captured every 3 s by a data-logger board, which also enables data storage. The raw data are then processed by dedicated software (RatioCalc, by Tamburello, pers. comm. 2013) that allows a derivation of plume mass ratios of various compounds (e.g. CO<sub>2</sub>/SO<sub>2</sub>, H<sub>2</sub>O/CO<sub>2</sub>, H<sub>2</sub>O/SO<sub>2</sub>). Simultaneously, relative concentrations of S and

Cl were measured using the alkali-filter technique (Aiuppa *et al.* 2005; Shinohara & Witter 2005), where air was pumped at a known flow rate (from 4 to 6 L min<sup>-1</sup>) through filter packs containing three cellulose impregnated filters stacked in series for c. 60–100 min. After exposure to the plume, NaHCO<sub>3</sub> impregnated filters were leached with bi-distilled water and H<sub>2</sub>O<sub>2</sub> for 2 hours in order to extract SO<sub>4</sub><sup>2-</sup> and Cl<sup>-</sup> (Aiuppa *et al.* 2005). The leaching solution was then analysed for SO<sub>4</sub><sup>2-</sup> and Cl<sup>-</sup> by ion chromatography (IC) at DiSTeM, University of Palermo and INGV sez. Palermo (Italy). Blank filters and field blanks (i.e. filters from filter packs transported into the field and back in an identical manner to samples but not attached to a pump) were also analysed and found to be negligible compared to the measured values.

### *Hg detection in volcanic emissions*

Gaseous mercury concentrations in volcanic plume have been measured mostly by using the conventional gold-trap technique (Ferrara *et al.* 2000; Aiuppa *et al.* 2007; Bagnato *et al.* 2007, 2009a, b; Witt *et al.* 2008a; Mather *et al.* 2012) and, more recently, by a portable mercury analyser LUMEX-RA 915+<sup>®</sup> (Sholupov *et al.* 2004; Aiuppa *et al.* 2007; Witt *et al.* 2008b; Bagnato *et al.* 2013). The Lumex-RA 915+<sup>®</sup> analyser monitors gaseous elemental mercury (GEM) concentrations using differential atomic absorption spectrometry with correction for background absorption via the Zeeman Effect (Zeeman atomic absorption spectrometry using high-frequency modulation of light polarization or ZAAS-HFM; Sholupov *et al.* 2004). The Hg analysing system operates by pumping the air at 20 L min<sup>-1</sup> through a c. 3 cm diameter inlet to the multi-path detection cell which has an effective path length of 10 m. A zero correction resets the baseline every 5 min during sampling. The detection limit is c. 2 ng m<sup>-3</sup>, and the instrument has an accuracy of 20%. Lumex<sup>®</sup> allows for measurements at the presence of high levels of humidity (not more than 95%) and H<sub>2</sub>S (not more than 100 mg m<sup>-3</sup>), unlike the conventional gold-trap technique which is instead strongly affected by inhibition of Hg adsorption at high H<sub>2</sub>S levels (Schroeder *et al.* 1995). During the surveys, we measured an averaged relative humidity of c. 56% and 46% at Poàs and Turrialba volcanoes, respectively. The accuracy and precision of Lumex<sup>®</sup> has also been assessed through comparison with the traditional gold-trap/cold vapour atomic fluorescence spectrometry (CVAFS) system used in this work and at remote sites elsewhere (Kim *et al.* 2006; Aiuppa *et al.* 2007; Witt *et al.* 2008b).

In this study, Lumex<sup>®</sup> has been employed in a stationary mode to collect gaseous elemental



mercury (GEM) in the plumes of Turrialba (near-vent plume aging of a few seconds to a few minutes) and Poás volcanoes and by walking across the fumarolic areas of Las Pailas and Las Hornillas geothermal systems (at about 60 cm from the emitting vents). Gold-coated quartz sand traps (Au-traps; Brooks Rand Labs<sup>®</sup>) were simultaneously used to collect total gaseous mercury (TGM) in the plumes (and/or fumaroles) of the studied volcanoes, where gas was pumped into the trap at a rate of *c.* 0.5 L min<sup>-1</sup> for sampling periods of *c.* 30 min. An inline filter pack containing a filter impregnated with 1 M NaHCO<sub>3</sub> was used in series with the gold traps to measure the corresponding SO<sub>2</sub> concentration in the gas sampled (Aiuppa *et al.* 2007; Bagnato *et al.* 2007, 2011), and a 45 µm Teflon pre-filter has been located on the head of the gold-trap assembly to filter atmospheric air. On a few occasions a second gold trap was connected in series in order to monitor potential GEM breakthrough on the first trap, but this proved negligible. After sampling, the traps were sealed and analysed at the Midi-Pyrenees Observatory (Toulouse, France) by using dual-stage gold-trap amalgamation with atomic fluorescence detection (AFS, Brook Rand Model III) which employs high-purity argon as a carrier gas. The sampling train consists of the sampling gold trap followed by a new soda lime trap (Tekran Inc.), an analytical gold-coated sand trap (Brooks Rand) and the AFS detector. The soda lime trap was used to neutralize water vapour, residual acid gases and other impurities that may be co-collected on sampling traps under volcanic conditions. The AFS was calibrated using 40–200 µL Hg vapour injections from a thermostated liquid/vapour Hg reservoir. Calibration was repeated every three samples and showed a drift of the signal of <10%. More details on the operating conditions are given in Sonke *et al.* (2008).

Some studies address the question of whether the collection of gaseous Hg using a gold trap quantifies TGM (i.e. GEM plus RGM) or GEM; see Gustin *et al.* (1999) as an example. For the sake of simplicity, GEM and Hg are used without distinction in this article unless otherwise specified since in volcanic emissions GEM is the dominant form ( $\geq 90\%$ ; Bagnato *et al.* 2007; Zambardi *et al.* 2009; von Glasow 2010; Martin *et al.* 2011).

### *Volcanic Hg flux estimate*

Estimating the emission flux of Hg from volcanoes (plume and fumaroles) has generally taken the approach of scaling relationships between averaged Hg/SO<sub>2</sub> ratios in gas emissions and SO<sub>2</sub> fluxes (Varekamp & Buseck 1986; Pyle & Mather 2003; Aiuppa *et al.* 2007; Bagnato *et al.* 2007; Witt *et al.* 2008a, b; Bagnato *et al.* 2011) since volcanic

SO<sub>2</sub> is well monitored and there is a degree of consensus over the global volcanic SO<sub>2</sub> flux (Graf *et al.* 1997; Andres & Kasgnoc 1998; Halmer *et al.* 2002; Schmidt *et al.* 2012). If the ratio Hg/SO<sub>2</sub> in volcanic plumes can be reliably constrained, the emission flux of Hg can be estimated. The former ratio is usually determined by connecting alkali-impregnated filters (to trap plume SO<sub>2</sub>; accuracy  $\pm 10\%$  at 1 s; Aiuppa *et al.* 2004, 2005) in series or in parallel with the gold trap. An additional effective method we used to assess the averaged Hg/SO<sub>2</sub> ratio in volcanic emissions is to combine the Lumex analyser in parallel with the MultiGAS to obtain real-time measurements of Hg and major volatiles in the volcanic fumes (Aiuppa *et al.* 2007; Witt *et al.* 2008a, b; Bagnato *et al.* 2013). In this study, data on sulphur dioxide flux in the volcanic plumes have been measured directly and acquired from literature reports. At Mt Etna, Gorely and Mutnovsky volcanoes a dual-UV camera set-up was used to obtain time series of the SO<sub>2</sub> flux at a resolution of 0.33 Hz (Kantzas *et al.* 2010; Tamburello *et al.* 2011; Aiuppa *et al.* 2012). This technique allows the whole plume to be captured in a single image, allowing the possibility of exploring spatial variations in SO<sub>2</sub> emissions over a timescale of *c.* 1 s (Mori & Burton 2006; Bluth *et al.* 2007). For the other reported volcanoes, SO<sub>2</sub> flux estimates have been obtained by traversing beneath the plume with an ultraviolet Ocean Optics USB2000 spectrometer coupled to a vertically pointing telescope, or by downloading data from an installed permanent station close to the volcano. Evaluation methods of the acquired spectra are described in McGonigle *et al.* (2002) and Galle *et al.* (2003). All the SO<sub>2</sub> fluxes used in our calculations correspond to the period of our fieldwork. In these calculations, we assume that GEM and SO<sub>2</sub> are the main forms of Hg (Bagnato *et al.* 2007, 2011) and S in the plumes, respectively. Under certain circumstances however, such as those encountered in the geothermal fields (i.e. the S fluxes are very low), the Hg/SO<sub>2</sub> approach for estimating Hg flux is not readily applicable since there is no reliable method for measuring the flux of the dominant form of sulphur in these systems. In those areas, we used an alternative method to estimate Hg flux by combining the kriging contouring technique with the plume speed information (Bagnato *et al.* 2013).

## **Results**

### *Mercury and major species in volcanic emissions*

The simultaneous acquisition of CO<sub>2</sub>, H<sub>2</sub>S, SO<sub>2</sub> and H<sub>2</sub>O by the MultiGAS (Aiuppa *et al.* 2005, 2007,

2011) combined with Lumex Hg analyser allowed us to characterize the abundance of major species in air-diluted volcanic emissions, and therefore examine the relationship between atmospheric Hg concentrations and volcanogenic volatiles (see Tables 1 & 2). More specifically, data from the permanent MultiGAS station installed on the western rim of Turrialba volcano's West Crater showed CO<sub>2</sub>/SO<sub>2</sub> molar ratio time series almost constant over a period of two days, ranging from 1.15 to 2.55 and averaging 2.2. Instead, the H<sub>2</sub>O/CO<sub>2</sub> molar ratio varies over a wide range from 32 to 230 (mean 109) due to water condensation that removes water from the gas of the low-temperature sources (see Table 1). Here, simultaneous collection of gases by Lumex and gold-trap technique yielded Hg concentrations of 99 and 81 ng m<sup>-3</sup>, respectively (Table 2), which are comparable to those obtained during previous fieldwork performed in 2010 (averaged GEM *c.* 115 ng m<sup>-3</sup>, Table 2; Bagnato *et al.* unpublished data). Higher averaged GEM values were measured in the volcanic plumes of Mt Etna's craters (Bocca Nuova and North East craters; 125 and 100 ng Hg m<sup>-3</sup>, respectively), where the continuous measurements by MultiGAS gave H<sub>2</sub>O/CO<sub>2</sub> and H<sub>2</sub>O/SO<sub>2</sub> values of *c.* 93 and 46.5, respectively (Tables 1 and 2). We found that Hg data from the North East crater (99 ng Hg m<sup>-3</sup>; this work) is comparable to the lower value reported by Bagnato *et al.* (2007) (range 114–328 ng Hg m<sup>-3</sup>; Table 2). A very recent MultiGAS survey on Gorely's volcanic plume (Aiuppa *et al.* 2012) reports H<sub>2</sub>O/SO<sub>2</sub> = 43 ± 13, CO<sub>2</sub>/SO<sub>2</sub> = 1.2 ± 0.1 and H<sub>2</sub>/SO<sub>2</sub> = 0.11 ± 0.02 as a representative mean end-member for arc magmatism in the NW Pacific region (Table 1). These data have been acquired simultaneously to our Hg measurements inside the dilute plume of Gorely (at *c.* 300 m downwind from the source), where we found the lowest GEM concentrations (*c.* 4 ng m<sup>-3</sup>; Table 2) of those reported in this study.

The four transects performed with the portable MultiGAS on the rim of the dome close to the acidic lake at Poás volcano did not show relevant variability of the chemical composition; the CO<sub>2</sub>/SO<sub>2</sub> molar ratio ranged between 0.23 and 0.45 (mean 0.4; Table 1), suggesting a strong homogeneity of the fumarolic field. The H<sub>2</sub>O/CO<sub>2</sub> ratio is affected by condensation of water and is therefore more variable (170–215; mean 183; Table 1). The SO<sub>2</sub>/H<sub>2</sub>S ratio fell within the range 2–6.8, supporting a more extensive re-equilibration of the volcanic gas within the more reducing and colder hydrothermal system. A further short survey was carried out around the edge of the acid lake of Poás volcano, aimed at detecting the chemistry of the steam released by the lake. The results show

higher values of both CO<sub>2</sub>/SO<sub>2</sub> and H<sub>2</sub>O/CO<sub>2</sub>, 1.2 and 344 respectively, reflecting the contribution of water evaporation and SO<sub>2</sub> scrubbing processes (Symonds *et al.* 2001). Simultaneous Hg measurements by both gold traps and Lumex system in the fumarolic emissions collected on the top of the dome (hundreds of metres from the emissions) gave very low concentrations (6–7.4 ng m<sup>-3</sup>; Table 2). In general, given the uncertainties associated with gold-trap techniques, our data exhibit a good agreement by comparing Lumex® and gold-trap results (see Table 2), as also reported elsewhere by other authors (Kim *et al.* 2006; Aiuppa *et al.* 2007).

A first assessment of the chemical composition of gases released by the fumarolic field inside and near the bottom of Rincón de la Vieja's main crater has also been accomplished. Despite the great distance from which the gas was detected by the MultiGAS (hundreds of metres from the fumaroles), the concentrations of volcanic SO<sub>2</sub> and CO<sub>2</sub> were high enough (*c.* 1 and 10 ppm, respectively) to calculate molar ratios. In fact, the 0.8 km<sup>2</sup> wide hyper-acid lake at the bottom of the crater exerts a strong gas scrubbing on the volcanic gas emissions, as demonstrated by the high CO<sub>2</sub>/SO<sub>2</sub> molar ratios in the range 15–69, averaging 27 (see Table 1). The measured H<sub>2</sub>O/CO<sub>2</sub> ratios are variable (range 84–354) but still comparable to the other volcanoes. As at Poás, we detected very small Hg concentrations in the emissions from Rincón de la Vieja volcano (in the range 4–5 ng m<sup>-3</sup>; Table 2). The quite low Hg concentrations found in the fumes of these latter two volcanoes mainly reflect the long distance from which the gas was detected (hundreds of metres from the fumarolic emissions), and probably the gas scrubbing on gaseous emissions exerted by the acid volcanic lakes hosted inside (Symonds *et al.* 2001). However, more research is needed to improve knowledge of the implications of magmatic gas scrubbing on Hg distribution in volcanic emanations.

The fumarolic gas of Aso volcano is a mixture of the original magmatic gas and hydrothermal fluids (Shinohara *et al.* 2010, 2013). The authors suggest that gases deriving from the lake and fumarolic emissions have distinct compositions, with higher CO<sub>2</sub>/SO<sub>2</sub> and HCl/SO<sub>2</sub> (3.1 and 0.1, respectively) and lower SO<sub>2</sub>/H<sub>2</sub>S (19.2) ratios in fumaroles than the lake gases (1.6, 420 and 0.01, respectively), but they have similar H<sub>2</sub>/CO<sub>2</sub> ratios (0.06 and 0.03, respectively) (Table 1). Here we attempted separate quantification of Hg in the gases emitted from the Yudamari crater lake (lake gas) and the fumarolic area at the southern wall (fumarolic gas) by using the gold-trap method at two distinct sampling sites. Since the gas source is not a single vent for both the lake gases and the fumarolic gases,

## MERCURY FLUXES FROM VOLCANIC AND GEOTHERMAL SOURCES

**Table 1.** Molar ratios measured in the plume of the investigated volcanoes and in atmospheric emissions of geothermal systems

Volcano	H <sub>2</sub> O/ CO <sub>2</sub>	H <sub>2</sub> O/ SO <sub>2</sub>	CO <sub>2</sub> / SO <sub>2</sub>
Turrialba	109	240	2.2
Poás	183	73	0.4
Rincón de la Vieja	201	5424	27
Aso fumarole*	17	52	0.0012
Aso lake gas <sup>†</sup>	58	94	0.0002
Gorely <sup>‡</sup>	32	43	1.3
Mutnovsky AF	23	46	0.001
Mutnovsky BF	83	288	0.00004
Etna NEC	93	46	0.5
Etna BN	21	126	6
<i>Geothermal fields</i>			
Las Pailas	378	15 031	41
Las Hornillas	29	3220	113

\*Shinohara *et al.* (2010)<sup>†</sup>Shinohara *et al.* (2013)<sup>‡</sup>Aiuppa *et al.* (2012)

there can be some heterogeneity in composition as well as composition fluctuation. It is therefore quite difficult to estimate the real error in composition determination. Mercury measurements at Aso volcano gave values of *c.* 119 and 8 ng Hg m<sup>-3</sup> in the fumaroles and fumes generated by the crater lake evasion, respectively (Table 2). The mismatch exhibited by these data mainly reflects the weak gaseous emission from the surface of Yudamari crater lake during the survey compared to the strong degassing exhibited by the fumaroles sampled along the wall of the crater.

Finally, MultiGAS survey performed at the Bottom Field and the Active Crater of the Mutnovsky volcano yielded H<sub>2</sub>O/S<sub>tot</sub> molar ratios in the range *c.* 46.4–288.3, while the measured H<sub>2</sub>O/CO<sub>2</sub> and CO<sub>2</sub>/S<sub>tot</sub> ratios were 23.4 and 2 for the former and 83 and 3.4 for the fumaroles of the Bottom Field (Table 1). Simultaneously, we also measured *c.* 40 and 8 ng Hg m<sup>-3</sup> in the atmospheric emissions from the Active Crater and the Bottom Field, respectively (Table 2).

In general, our results are higher than the background atmospheric Hg values generally found in the unpolluted areas of the Northern Hemisphere (range 1.5–1.7 ng m<sup>-3</sup>; Lindberg *et al.* 2007; Sprovieri *et al.* 2010), as well in the urban sites (range 2.5–9.8 ng m<sup>-3</sup>; Liu *et al.* 2002; Rutter *et al.* 2008, 2009; Kim *et al.* 2011). The large variability of GEM concentrations (4–125 ng m<sup>-3</sup>) we measured in the gaseous emissions of the investigated volcanoes strongly reflects the dynamic nature of the plume, where the abundances of volatiles are determined by variations in Hg emission fluxes

from magma and dilution with background air. Overall, our data support the relevance of persistent volcanic activity from open-conduit degassing as a significant source of Hg for the atmosphere (see Table 2), and fall within the range proposed by many authors worldwide (Witt *et al.* 2008*a, b*; Bagnato *et al.* 2011; Mather *et al.* 2012; Table 2).

*Hg emissions from geothermal fields*

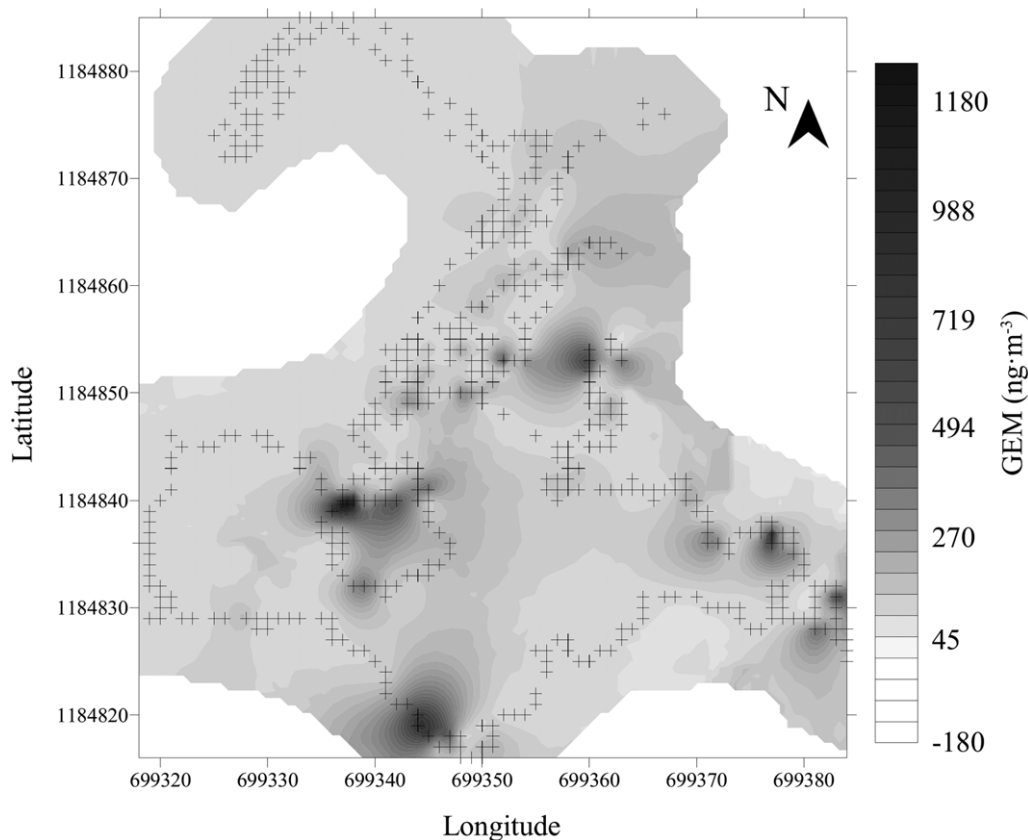
There are relatively few published datasets of Hg concentrations measured at or near geothermal features even if many areas characterized by geothermal activity have long been associated with elevated levels of Hg in soils and air (Varekamp & Buseck 1981, 1986; Cox 1983; Siegel & Siegel 1984). Previous studies by Robertson *et al.* (1977) and Christenson and Mroczek (2003) showed that essentially all the Hg in geothermal steam and gases is in the gaseous elemental form (GEM). A Lumex<sup>®</sup>/MultiGAS GPS-synchronized survey was carried out at Las Hornillas and Las Pailas geothermal fields (Costa Rica) with the aim of quantifying volatile species in the fumarolic emissions distributed in these areas. At each fumarole, measurements were performed by placing the inlet of the instruments close (within a few centimetres) to the emitting vent to reduce mixing with air. The combined Lumex<sup>®</sup>/MultiGAS technique allowed a large number of simultaneous GEM and SO<sub>2</sub> measurements to be made over short (half an hour) observation periods, and the time variability of the GEM/SO<sub>2</sub> ratio to be evaluated real-time. The MultiGAS survey revealed a spatial variability in terms of gas compositions and emission rate for both investigated geothermal areas (Table 1; Figs 2 & 3). We estimated an averaged CO<sub>2</sub>/H<sub>2</sub>S ratio ranging between 53 and 179 (with the higher values measured at the northern part of the field) and between 16.3 and 73 at Las Hornillas and Las Pailas, respectively. At Las Hornillas we also estimated an averaged H<sub>2</sub>O/CO<sub>2</sub> ratio ranging from 23 to 53, while at Las Pailas H<sub>2</sub>O/H<sub>2</sub>S and H<sub>2</sub>O/CO<sub>2</sub> ratios were estimated to vary from 8835 to 29 994 and from *c.* 230 to 542, respectively (Table 1). By walking around the fumarolic emissions, we estimated GEM concentrations of 140–530 ng m<sup>-3</sup> (averaged 260 ng m<sup>-3</sup>) and *c.* 145–1329 ng m<sup>-3</sup> (averaged 822 ng m<sup>-3</sup>) for Las Pailas and Las Hornillas emissions, respectively (Table 2; Figs 2 & 3). Measurements showed that both systems are releasing Hg at elevated levels well above those measured in the atmosphere outside the geothermal areas (*c.* 3.6 and 3.2 ng m<sup>-3</sup> for Las Pailas and Las Hornillas, respectively). Measurements performed by a FLIR SC660 infrared camera (located near the main vents) showed a very homogeneous distribution of



**Table 2.** Volcanic mean GEM concentrations (range; SD and number of measurements), mean GEM/SO<sub>2</sub> mass ratios and Hg fluxes for the volcanic systems used in this study

Volcano	Location	Date	Volcanism	Emission type	GEM (ng m <sup>-3</sup> (range; SD; n))	Method	GEM/SO <sub>2</sub>	ΦSO <sub>2</sub> (t a <sup>-1</sup> )	ΦHg (t a <sup>-1</sup> )	References
Gorely	Kamchatka	06/09/2011	Arc	Plume	4 ( <i>I</i> )	Au-trap/AFS	$3.3 \times 10^{-6}$	802 <sup>1</sup>	0.95	This study
Mutnovsky AF	Kamchatka	05/09/2011	Arc	Fumaroles	8 ( <i>I</i> )	Au-trap/AFS	$1.9 \times 10^{-7}$	150 <sup>2</sup>	0.01	This study
Mutnovsky BF(both plumes)	Kamchatka	05/09/2011	Arc	Fumaroles	40 ( <i>I</i> )	Au-trap/AFS	$1.4 \times 10^{-5}$	150 <sup>2</sup>	0.7 (0.71)	This study
Aso fumarole	Japan	2010	Arc	Fumaroles	119 ( <i>I</i> )	Au-trap/AFS	$5.2 \times 10^{-7}$	500 <sup>3</sup>	0.09	This study
Aso crater lake (both plumes)	Japan	2010	Arc	Lake fumes	8 ( <i>I</i> )	Au-trap/AFS	$1.3 \times 10^{-6}$	500 <sup>3</sup>	0.24 (0.33)	This study
Etna BN	Italy	24/06/2011	Non-arc	Plume	125 ( <i>I</i> )	Au-trap/AFS	$7.1 \times 10^{-6}$	1130 <sup>4</sup>	2.92	This study
Etna NEC (both plumes)	Italy	24/06/2011	Non-arc	Plume	99 ( <i>I</i> )	Au-trap/AFS	$3.4 \times 10^{-6}$	1000 <sup>4</sup>	1.24 (4.16)	This study
Poás	Costa Rica	16–18/04/2013	Arc	Fumaroles	7.4 (5–10)	Lumex	$4.9 \times 10^{-8}$	120 <sup>5</sup>	0.002	This study
Poás	Costa Rica	16–18/04/2013	Arc	Fumaroles	6 (4–7; 1.2; 5)	Au-trap/AFS	$3.7 \times 10^{-8}$ ( $2.5-3.6 \times 10^{-8}$ )	120 <sup>5</sup>	0.0016	This study
Turrialba	Costa Rica	07/03/2011	Arc	Plume	115 ( <i>I</i> )	Au-trap/AFS	$3.9 \times 10^{-6}$	1000 <sup>6</sup>	1.45	This study
Turrialba	Costa Rica	02–04/04/2013	Arc	Plume	100 (68–174)	Lumex	$5.6 \times 10^{-6}$ ( $3-9.8 \times 10^{-6}$ )	671 <sup>7</sup>	1.4	This study
Turrialba	Costa Rica	02–04/04/2013	Arc	Plume	81 (76–85; 4; 3)	Au-trap/AFS	$8.1 \times 10^{-6}$ ( $7.6-8.2 \times 10^{-6}$ )	671 <sup>7</sup>	1.9	This study
Rincón de la Vieja	Costa Rica	07/04/2013	Arc	Fumaroles	5 (4–5; 0.6; 2)	Au-trap/AFS	$3.5 \times 10^{-6}$ ( $4.3-2.7 \times 10^{-6}$ )	n.d.	n.d.	This study
Las Pailas	Costa Rica	08/04/2013	Arc	Geothermal	260 (140–530)	Lumex	$1.68 \times 10^{-5}$	n.d.	0.0008–0.002	This study
Las Hornillas	Costa Rica	09/04/2013	Arc	Geothermal	822 (145–1329)	Lumex	$4.01 \times 10^{-6}$	n.d.	0.004–0.01	This study
Nyiragongo	Africa	19–23/06/2010	Non-arc	Plume	373	Au-trap/CVAFS	$5.5 \times 10^{-6}$	2180	4.4	Bagnato <i>et al.</i> (2011)
Yasur	Vanuatu	21/10/2010	Arc	Plume	84	Au-trap/CVAFS	$1.7 \times 10^{-5}$	940	6	Bagnato <i>et al.</i> (2011)
Ambrym	Vanuatu	21/10/2010	Arc	Plume	68	Au-trap/CVAFS	$6.4 \times 10^{-6}$	7600	18	Bagnato <i>et al.</i> (2011)
Stromboli	Italy	24/07/2008	Arc	Plume	28	Au-trap/CVAFS	$4.7 \times 10^{-6}$	120	0.2	Bagnato <i>et al.</i> (2011)
Miyakejima	Japan	02–12/09/2006	Arc	Plume	95	Au-trap/CVAFS	$9 \times 10^{-6}$	2000	6.6	Bagnato <i>et al.</i> (2011)
Asama	Japan	25–29/09/2008	Arc	Plume	41	Au-trap/CVAFS	$5.5 \times 10^{-6}$	1850	3.7	Bagnato <i>et al.</i> (2011)
La Soufriere Hill	Less Antilles	01/07/2008	Arc	Plume	18	Au-trap/CVAFS	$1.2 \times 10^{-5}$	400	1.8	Bagnato <i>et al.</i> (2011)
Kilauea	Hawaii	2008	Arc	Plume	35	Au-trap/CVAFS	$1.6 \times 10^{-6}$	700	0.2	Mather <i>et al.</i> (2012)
Masaya	Nicaragua	24/02–04/03/2006	Arc	Plume	163.5	Au-trap/CVAFS	$2 \times 10^{-5}$	800	7.2	Witt <i>et al.</i> (2008a, b)
Cumulative flux from open-conduit volcanoes								16 590	56	This study
Global volcanism emissions								18 630–38 904 <sup>8</sup>	76 ± 30	This study

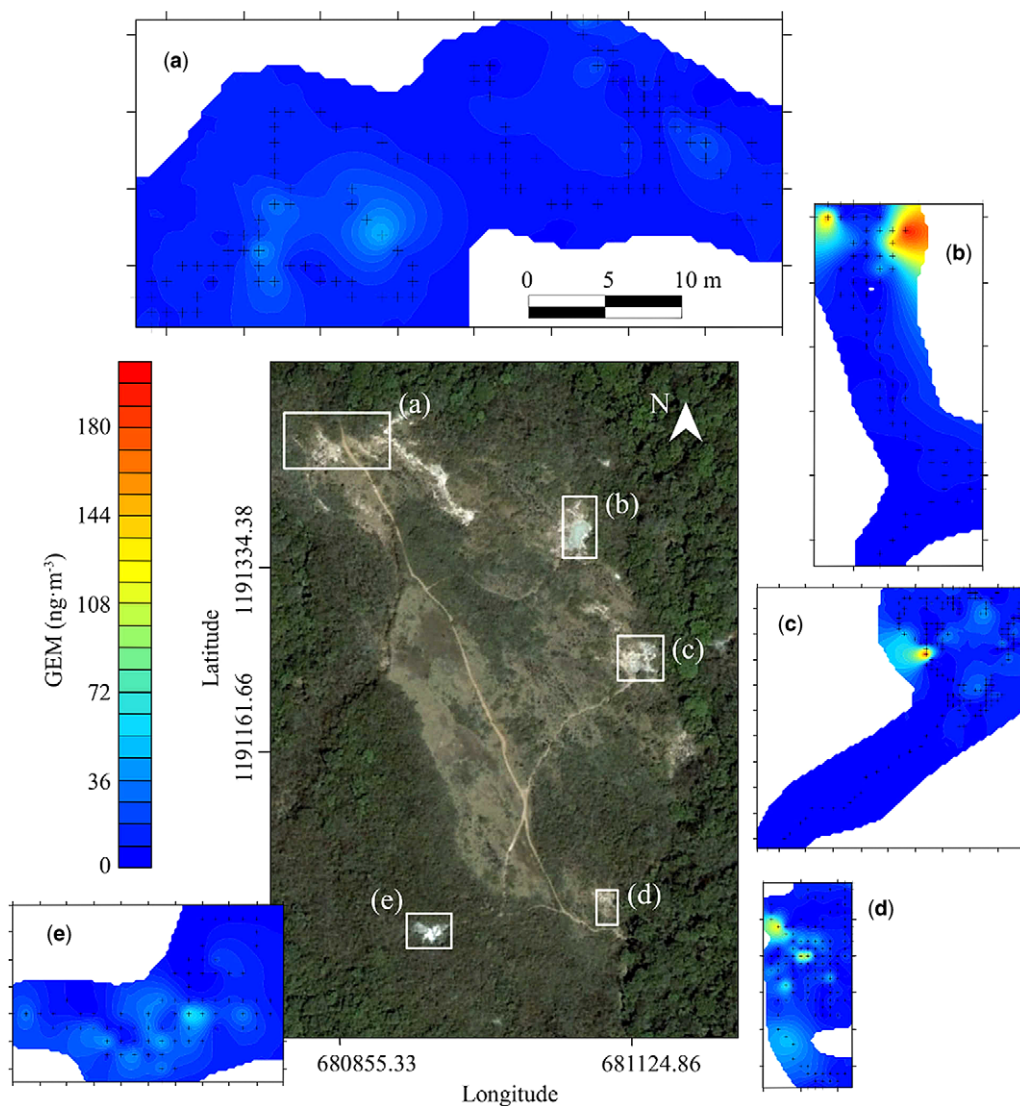
Literature Hg flux with relevant data sources and sampling/analytical methods are also reported: <sup>1</sup>Aiuppa *et al.* (2012); <sup>2</sup>G. Tamburello pers. comm., 2011; <sup>3</sup>Shinohara *et al.* (2010); <sup>4</sup>G. Tamburello pers. comm., 2013; <sup>5</sup>M. DeMoor pers. comm., 2013; <sup>6</sup>Campion *et al.* (2012); <sup>7</sup>G. Avard pers. comm., 2013; <sup>8</sup>Graf *et al.* (1997).



**Fig. 2.** Mercury distribution ( $\text{ng m}^{-3}$ ) in gaseous emissions at Las Hornillas geothermal field, Miravalles volcano, Costa Rica. The black crosses indicate the walking track across the fumarolic emission sites with the Lumex/MultiGAS synchronized survey. Coordinates are reported in UTM (Zone 16P).

gas temperatures at the emitting vents, generally close to the boiling point of water and ranging from 95 to 98 °C in both geothermal areas. Acquisition by the Lumex<sup>®</sup> system allowed us to detect pulsed sequences of GEM increases in the atmospheric plumes of the fumaroles of both geothermal systems (Fig. 4a, b). These GEM pulses were closely matched by parallel increases in CO<sub>2</sub> as illustrated in Figure 4a, b, supporting both a volcanic origin for the measured GEM concentrations and the role of early CO<sub>2</sub> escape from magmas as a carrier gas for Hg (Varekamp & Buseck 1981, 1986; Bagnato *et al.* 2013). At Las Pailas we estimated mean GEM/CO<sub>2</sub> mass ratios ranging from  $1.4 \times 10^{-9}$  to  $1.7 \times 10^{-8}$ , comparable with those measured in the gaseous manifestations at Las Hornillas field (from  $3.5 \times 10^{-9}$  to  $1 \times 10^{-7}$ ). In general, our estimated GEM/CO<sub>2</sub> values are similar to those obtained in other low-temperature volcanic-hydrothermal systems such as Tatun

volcanic field (GEM/CO<sub>2</sub> from  $4 \times 10^{-8}$  to  $40 \times 10^{-8}$ ; Witt *et al.* 2008a) and the low-*T* (85°–150 °C) CO<sub>2</sub>-rich fumaroles of Masaya volcano (GEM/CO<sub>2</sub> from  $0.1 \times 10^{-8}$  to  $1.9 \times 10^{-8}$ ; Witt *et al.* 2008b). Measurements at the higher-temperature (250 °C) fumaroles of La Fossa Crater, Vulcano island indicated GEM/CO<sub>2</sub> mass ratio of  $6.0 \times 10^{-8}$  (Aiuppa *et al.* 2007). Our data are also close to the values reported by Engle *et al.* (2006) for the fumaroles at Yellowstone Caldera, a hydrothermal system with temperatures ranging from 85 to 95 °C and in the presence of H<sub>2</sub>S ( $1.6 \times 10^{-9}$  to  $2.6 \times 10^{-9}$ ; Engle *et al.* 2006). The GEM concentrations derived from the real-time measurements at Las Pailas geothermal field also correlate well with the real-time H<sub>2</sub>O concentrations at all the active vents (Fig. 4b) with estimated mean GEM/H<sub>2</sub>O mass ratios ranging from  $4.3 \times 10^{-11}$  to  $3.6 \times 10^{-10}$ . These results provide new important information, improving the

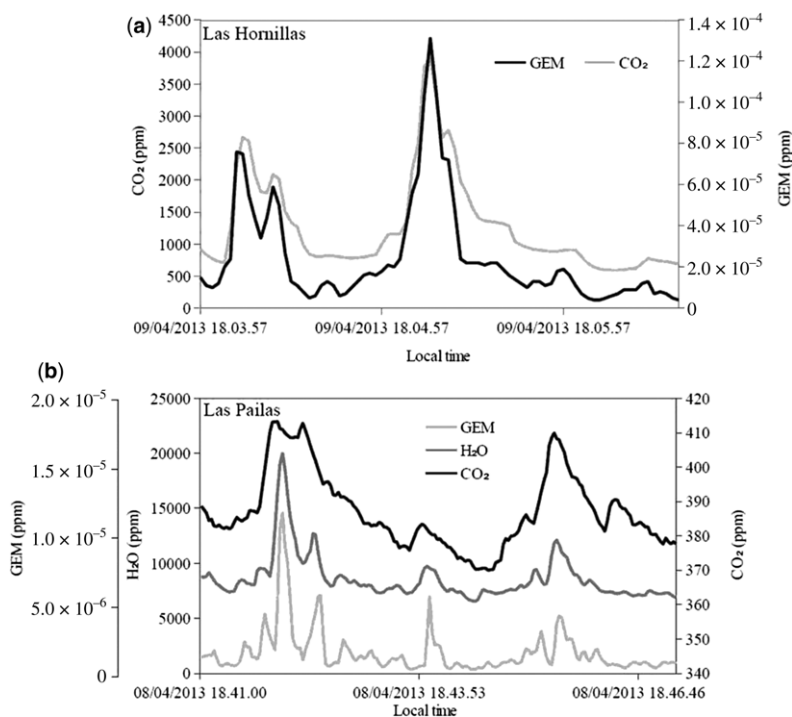


**Fig. 3.** Mercury distribution ( $\text{ng m}^{-3}$ ) in gaseous emissions at Las Pailas geothermal field, Rincón de la Vieja volcano, Costa Rica by the Lumex/MultiGAS acquisition technique. Here the surveyed areas are reported in five separated sectors (a–e) due to the extended surface area of the geothermal field. Coordinates are reported in UTM (Zone 16P).

currently sparse Hg database for gases from volcanogenic origin worldwide (Weissberg & Rohde 1978; Cox 1983; Siegel & Siegel 1984; Varekamp & Buseck 1986; Nakagawa 1999; Engle *et al.* 2006). Previous research suggested that steam condensation can affect GEM to other chemicals ratios (Varekamp & Buseck 1981, 1986). In this study, we did not attempt to measure Hg concentration in fumarolic steam condensates, so we cannot rule out the possibility that our derived Hg emissions are underestimating the real emissions to some extent

(Nakagawa 1999; Aiuppa *et al.* 2007). The release of Hg from previously enriched substrates as a result of geothermal processes has not been determined at Las Pailas and Las Hornillas, but it might contribute significantly in estimating the cumulative Hg flux from geothermal systems as recently demonstrated elsewhere (e.g. Nea Kameni fumarolic fields, Greece; Sulphur Banks, Hawaii, Varekamp & Buseck 1986; Miravalles volcano, Melián *et al.* 2004; United States, Engle *et al.* 2006; la Solfatara, Pozzuoli, Bagnato *et al.* 2009a, 2013).

## MERCURY FLUXES FROM VOLCANIC AND GEOTHERMAL SOURCES



**Fig. 4.** Temporal variations in CO<sub>2</sub> (grey line), GEM (black line) and H<sub>2</sub>O (light grey line) measured with MultiGAS and Lumex, respectively at (a) Las Hornillas and (b) Las Pailas geothermal systems. The positive temporal correlation exhibited by the two species may support CO<sub>2</sub> as potential carrier gas in transporting GEM in magmatic systems.

### The importance of Hg/S ratio

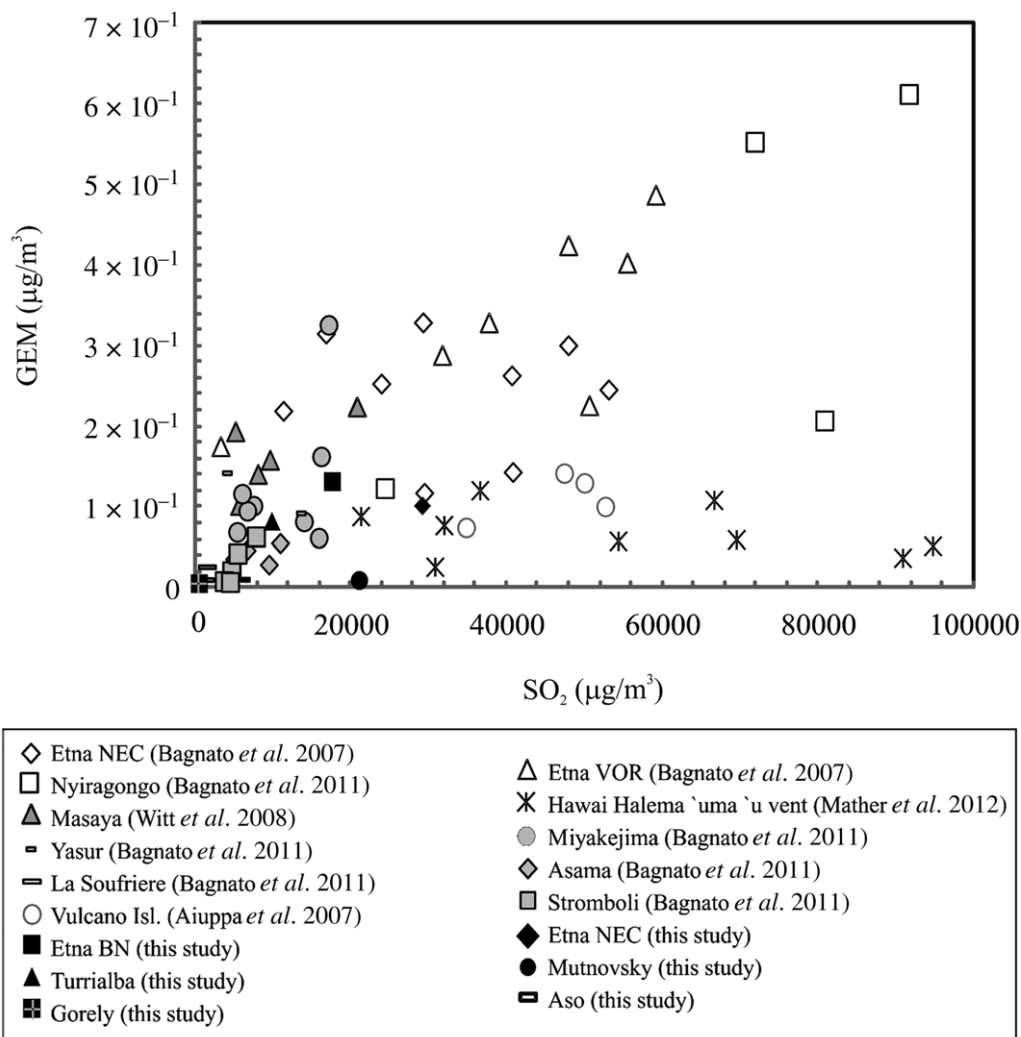
Concentrations of gaseous plume species at crater rims depend on both volcanic (gas flux) and non-volcanic factors (extent of mixing and dilution in the vent, wind speed and direction); comparing the concentrations of species between different volcanoes is therefore not necessarily meaningful and it is more useful to compare ratios of various species. A large variety of chemical methods have been used in the past to quantify Hg/SO<sub>2</sub> proportions in volcanic emissions (e.g. Varekamp & Buseck 1981; Nakagawa 1999; Ferrara *et al.* 2000; Bagnato *et al.* 2007), the majority of which require complex and time-consuming post-sampling analytical determinations in a laboratory, precluding real-time measurement and limiting the acquisition of robust datasets. The literature reported Hg/SO<sub>2</sub> mass ratios in volcanic emissions span four orders of magnitude (Pyle & Mather 2003), including fumarolic condensates and vapours which dominate the available data. Fumarolic emissions may differ in composition from the high-temperature gases released directly from magma (Symonds *et al.* 1992). The uncritical use of fumarolic Hg/SO<sub>2</sub> as a proxy for magmatic Hg/SO<sub>2</sub> will underestimate

volcanic Hg emission rates because SO<sub>2</sub> is mostly from high-temperature sources (magmatic), while low-temperature fumarolic environments are to some extent enriched in H<sub>2</sub>S (Pyle & Mather 2003; Aiuppa *et al.* 2007). This means that in estimating the Hg emission flux, it is more appropriate to consider the Hg/S<sub>tot</sub> ratio (where S<sub>tot</sub> = SO<sub>2</sub> + H<sub>2</sub>S) rather than the single S-species. Some recent assessments of global volcanic emissions of Hg have not properly accounted for this (Unni *et al.* 1978; Nriagu 1989; Ferrara *et al.* 2000). These considerations are especially important as the most significant factor that influences volcanic Hg flux estimates is the value of Hg/SO<sub>2</sub> that is adopted in the calculation. The variability we found in the averaged GEM concentrations (see Table 2) measured in gaseous emissions of the investigated volcanoes strongly highlights the importance of using Hg/S ratios to compare volcanic emissions. Given the uncertainties associated with the gold-trap technique, we found a good agreement by comparing atmospheric Lumex<sup>®</sup>/MultiGAS measurements and those made with gold-traps/filterpacks in determining Hg/SO<sub>2</sub> mass ratios (see Table 2, especially for Poás and Turrialba volcanoes). Most of the measured GEM/SO<sub>2</sub> range from 10<sup>-6</sup> to 10<sup>-5</sup>,

except for Aso fumaroles, Mutnovsky Active Funnel plume and Poás volcano which displays the lowest mass ratios ranging from  $10^{-8}$  to  $10^{-7}$  (see Table 2). At Rincón de la Vieja volcano, we measured a mean fumarolic GEM/SO<sub>2</sub> ratio of  $c. 3.5 \times 10^{-6}$  (range  $4.3\text{--}2.7 \times 10^{-6}$ ; Table 2). The GEM/SO<sub>2</sub> mass ratios estimated in the plumes of the North East Crater and Bocca Nuova at Mt Etna ( $3.4\text{--}7.1 \times 10^{-6}$  respectively; Table 2) are comparable to the mean value reported by Bagnato *et al.* (2007) ( $c. 8.8 \times 10^{-6}$ ). Finally, at

Las Pailas and Las Hornillas we measured GEM/SO<sub>2</sub> of  $1.68 \times 10^{-5}$  and  $4.1 \times 10^{-6}$ , respectively.

A broad correlation was found by plotting GEM versus SO<sub>2(g)</sub> concentrations (Hg/SO<sub>2</sub>  $c. 10^{-7}$  to  $10^{-5}$ ) in the plume of active degassing volcanoes from arc and non-arc setting (taken from both literature and this study; Fig. 5). The most likely reason for the differences observed in GEM/SO<sub>2</sub> mass ratios among the reported volcanoes from similar settings may simply be compositional variability (i.e. Hg and S contents; Fig. 5). Further, since Hg



**Fig. 5.** Scatter plot of gaseous elemental mercury (GEM) v. SO<sub>2(g)</sub> concentrations in the vent plumes of active volcanoes from arc and non-arc volcanism (grey filled symbols and black open symbols, respectively) (GEM/SO<sub>2</sub> ranging from  $10^{-7}$  to  $10^{-5}$ ; see Table 2). Our original data (black filled symbols) are set in the context of published volcanic gas compositions. Arc-volcanism setting: Masaya (Witt *et al.* 2008a), Miyakejima, Yasur, Stromboli, Asama, La Soufriere (Bagnato *et al.* 2011), Vulcano island (Aiuppa *et al.* 2007); non arc-volcanism setting: Etna NEC and Etna VOR (Bagnato *et al.* 2007), Nyiragongo (Bagnato *et al.* 2011), Hawai Halema'uma'u vent (Mather *et al.* 2012).

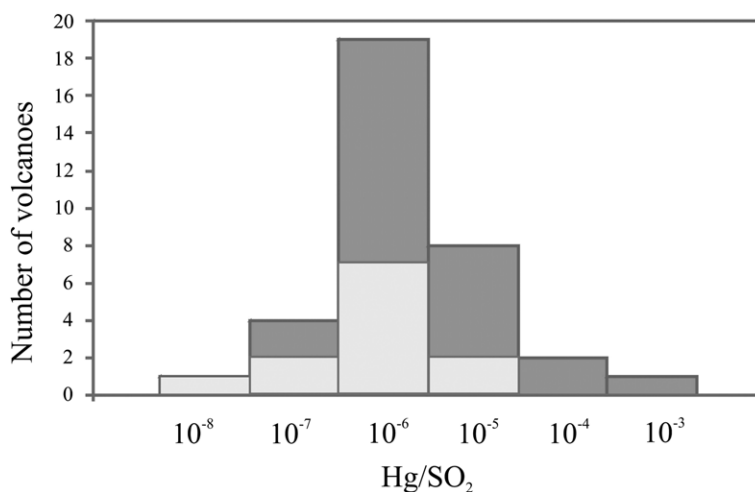


## MERCURY FLUXES FROM VOLCANIC AND GEOTHERMAL SOURCES

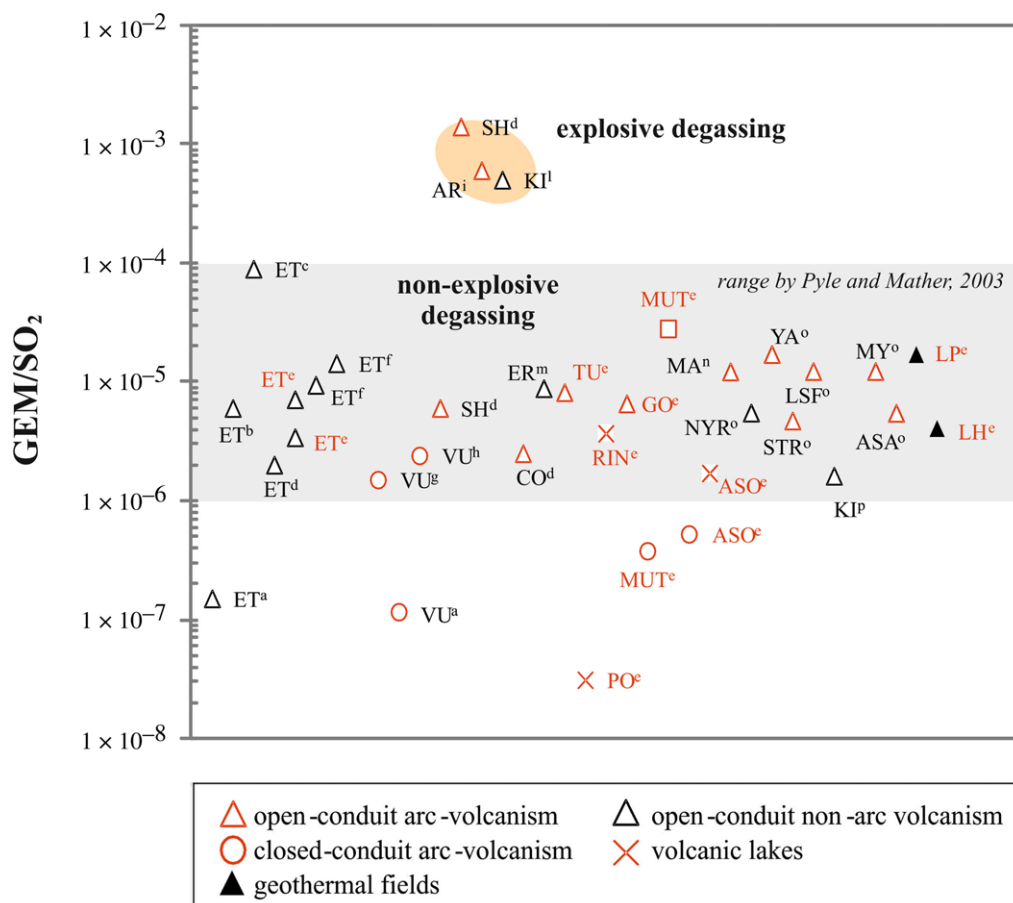
and  $\text{SO}_2$  have different solubility in silicate melts, they will be fractionated during magmatic degassing, implying that  $\text{Hg}/\text{SO}_2$  will be significantly changing in both time and space. Volcanoes with acidic lakes, such as Poàs and Rincón de la Vieja, can also fractionate the  $\text{Hg}/\text{SO}_2$  through scrubbing in the lakes of both  $\text{SO}_2$  and Hg as  $\text{H}_2\text{SO}_4$  and  $\text{HgCl}_2$ , respectively (Symonds *et al.* 2001; Shinohara *et al.* 2010). In such a case, the gaseous emanations from the lake will be fractionated in  $\text{Hg}/\text{SO}_2$  as a result of  $\text{SO}_2$  or Hg dissolution, or both. There is of course a need for continuing work to characterize both volcanic lake evasion fumes and low- and high-temperature volcanic emissions and to establish the factors which control the  $\text{Hg}/\text{S}$  ratios at different degassing volcanoes. Previous estimates of the global non-explosive volcanic Hg flux have used  $\text{Hg}/\text{SO}_2$  ratios between  $10^{-7}$  and  $10^{-5}$  (Varekamp & Buseck 1986; Nriagu 1989; Ferrara *et al.* 2000; Nriagu & Becker 2003) with  $10^{-6}$  as the best representative of gas emissions from quiescent open-conduit degassing volcanoes (Fig. 6). In general, most of our GEM/ $\text{SO}_2$  data from continuous degassing are in agreement with this ratio (Figs 6 & 7), and are similar to the value reported in Varekamp & Buseck's (1986) global estimate for passive emissions from non-erupting volcanoes ( $3.7 \times 10^{-6}$ ). Our data also fall within the range of best estimates ( $10^{-4}$  to  $10^{-6}$ ) proposed by Pyle & Mather (2003) for non-explosive volcanic degassing and are also in agreement with the ratios found at other passive degassing volcanoes

(see Tables 2 & 3; Fig. 7). However, our data together with most of the reported GEM/ $\text{SO}_2$  in the literature should probably be considered as the lower bounds because they mainly refer to passive emissions from non-erupting volcanoes; these are lower than the ratios measured during explosive eruptions which are also often accompanied by considerable  $\text{SO}_2$  emission (such as those documented for Kilauea, Arenal and Mt St Helens volcanoes; GEM/ $\text{SO}_2$  from  $10^{-4}$  to  $10^{-3}$ ; Varekamp & Buseck 1981; Ballantine *et al.* 1982; Fig. 7). Previous estimates of  $\text{Hg}/\text{SO}_2$  ratios from explosive volcanic eruptions gave values varying much more, between  $10^{-2}$  and  $10^{-5}$  (Buat-Menard & Arnold 1978; Varekamp & Buseck 1981; Ballantine *et al.* 1982; Shuster *et al.* 2002), suggesting that emissions from large explosive eruptions are rich in Hg.

By comparing plume GEM/ $\text{SO}_2$  mass ratios to the corresponding plume  $\text{H}_2\text{O}/\text{SO}_2$  ratios (Fig. 8; see Tables 1 & 2 for data sources), most of the open conduit 'arc' volcanoes (grey filled triangles in Fig. 8) show an overall positive correlation between  $\text{H}_2\text{O}/\text{SO}_2$  and GEM/ $\text{SO}_2$  ratios ( $r^2 = 0.8$ ). This may suggest that gas emissions in subduction zones contain more Hg (relative to  $\text{SO}_2$ ) than those from divergent (or within-plate) plate volcanism, as also previously suggested by some authors (Kim *et al.* 2000; Bagnato *et al.* 2011). Data from Rincón de la Vieja volcano as well as the two geothermal fields (Las Pailas and Las Hornillas) fall outside the main trend exhibited



**Fig. 6.** Summary of measured GEM/ $\text{SO}_2$  mass ratios in the volcanic bulk plume, compiled from this work (light grey bars) and published data (as reported in Fig. 5). The compilation strongly supports  $10^{-6}$  to  $10^{-5}$  for the  $\text{Hg}/\text{SO}_2$  plume ratio from quiescent degassing volcanoes as representative of gas emissions from the open-conduit degassing volcanoes.



**Fig. 7.** Summary of measured GEM/SO<sub>2</sub> mass ratios in worldwide volcanic emanations. The graph was constructed from results obtained in this study (red text), and previously published data (black text). A distinction was made between the emission type plotted in the graph to show open/closed conduit, arc/non-arc volcanism, volcanic lake evasion and geothermal fields. Most of the reported GEM/SO<sub>2</sub> values are lower than those measured for Kilauea, Arenal ( $10^{-4}$ ) and Mt St Helens volcanoes, during explosive eruptions (pink-coloured area). Reported data are taken from: <sup>a</sup>Ferrara *et al.* (2000); <sup>b</sup>Dedeurwaerder *et al.* (1982); <sup>c</sup>Buat-Menard & Arnold (1978); <sup>d</sup>Varekamp & Buseck (1986); <sup>e</sup>this work; <sup>f</sup>Bagnato *et al.* (2007); <sup>g</sup>Bichler *et al.* (1995); <sup>h</sup>Aiuppa *et al.* (2007); <sup>i</sup>Ballantine *et al.* (1982); <sup>j</sup>Siegel & Siegel (1984); <sup>k</sup>Kyle *et al.* (1990); <sup>l</sup>Witt *et al.* (2008a, b); <sup>m</sup>Bagnato *et al.* (2011); <sup>n</sup>Mather *et al.* (2012). ET, Etna; ER, Erebus; PO, Poás; ASO, Aso; MUT, Mutnovsky; GO, Gorely; MA, Masaya; KI, Kilauea; NY, Nyiragongo; YA, Yasur; TU, Turrialba; ASA, Asama; CO, Colima; VU, Vulcano Island; STR, Stromboli; RIN, Rincón de la Vieja; LSF, La Soufriere Hill; MY, Miyakejima; SH, St Helens; AR, Arenal; LP, Las Pailas; LH, Las Hornillas. The grey-coloured area shows the typical range of GEM/SO<sub>2</sub> ratios for continuous degassing volcanoes, as suggested by Pyle & Mather (2003).

by the other volcanoes (Fig. 8). We note that for the former, it is likely that either the sulphur scrubbing action in the acid lakes or the distance from which we sampled gases may have influenced the H<sub>2</sub>O/SO<sub>2</sub> and GEM/SO<sub>2</sub> values. To date however, we are not able to say which of the two factors dominates. The excess of water detected in the gaseous emissions at Las Hornillas and Las Pailas geothermal fields may explain the low estimated GEM/H<sub>2</sub>O mass ratios at these locations

(black filled triangles in Fig. 8). More data are required to explore this relationship further, however. A stronger correlation between GEM and SO<sub>2</sub> is evident for samples collected within plate/rift volcanism (Figs 5 & 8), in agreement with their similar degassing behaviour.

In arc-volcanism the origin of Hg can be manifold and fairly hard to understand. Metals released by gaseous discharges in volcanic arcs are ultimately sourced in the heterogeneous mantle wedge

**Table 3.** Literature Hg/SO<sub>2</sub> mass ratios for active volcanoes worldwide

Volcano	Hg/SO <sub>2</sub>	References
Etna's bulk degassing	$8.9 \times 10^{-5}$ to $1.5 \times 10^{-7}$	Dedeurwaerder <i>et al.</i> (1982), Varekamp & Buseck (1986), Ferrara <i>et al.</i> (2000)
Etna North East's fumaroles	$2 \times 10^{-6}$	Varekamp & Buseck (1986)
Colima	$2.5 \times 10^{-6}$	Varekamp & Buseck (1986)
Vulcano Island	$1.16 \times 10^{-7}$ to $2.4 \times 10^{-6}$	Ferrara <i>et al.</i> (2000), Aiuppa <i>et al.</i> (2007), Zambardi <i>et al.</i> (2009)
Mt Erebus	$8.8 \times 10^{-6}$	Kyle <i>et al.</i> (1990)
White Island	$2.9 \times 10^{-6}$	Wardell <i>et al.</i> (2008)
Mt Shasta	$39 \times 10^{-6}$	Varekamp & Buseck (1986)
Mt Hood	$7.1 \times 10^{-6}$	Varekamp & Buseck (1986)
Mt St Helens	$6.0 \times 10^{-6}$	Varekamp & Buseck (1986)

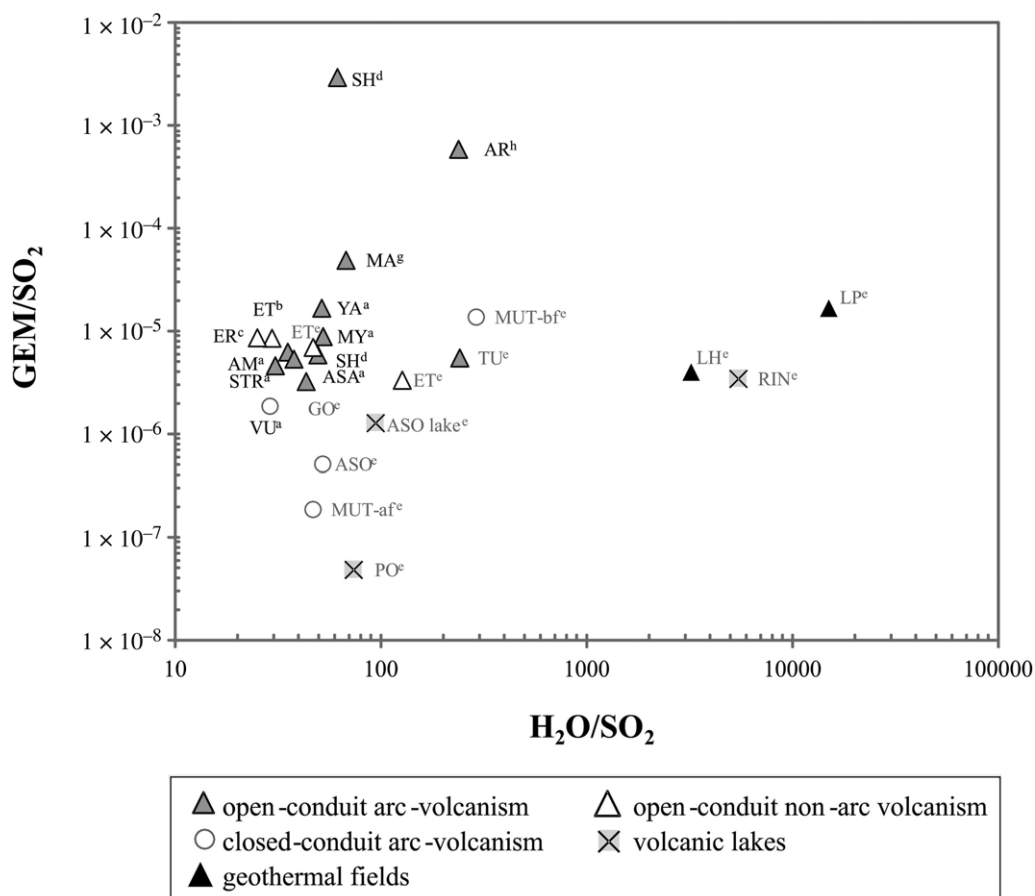
beneath volcanoes (Jenner *et al.* 2010; Stracke 2012) by processes involving volatile release from the underlying hydrated subducting slab and entrained oceanic sediment (Mungall 2002; Labanih *et al.* 2012; Henley & Berger 2013). More data are required to better explore Hg behaviour from volcanoes where a subduction component is involved in magmagenesis; recent studies (Pyle & Mather 2009) suggested that a significant component of Hg in volcanic emissions may originate from the subducting slab, as observed for other elements such as Br, Cl and I in volcanic emissions.

By putting our new data into the context of published volcanic gas compositions from various geodynamic settings (arc volcanism and within plate/rift volcanism; Fig. 9, data source in Tables 1 & 2), we find an extremely variable composition ranging from Hg to H<sub>2</sub>O-rich, respectively. The Hg-rich end-member of the triangular plot shown in Figure 9 corresponds to the 1980 Mt St Helens explosive eruption (SH), characterized by fairly high Hg to sulphur mass ratio ( $3.3 \times 10^{-3}$ ; Varekamp & Buseck 1986) which differs significantly from the ratio which usually characterizes the phase of passive degassing of the volcano ( $6 \times 10^{-6}$ ; Varekamp & Buseck 1986). Gaseous emissions from Rincón de la Vieja volcano (RIN) are H<sub>2</sub>O-rich and poor in sulphur and Hg, probably due to the strong gas scrubbing by the lake and/or the high air dilution of the collected fumes. Intermediate compositions identify no clear trend and emphasize the presence of a compositional gap for sulphur-poor, Hg- and H<sub>2</sub>O-rich volcanic gases. Water in volcanic gas represents an intrinsically highly variable parameter affected by condensation at numerous sites (fractures, hydrothermal systems, lakes and atmosphere), and its variability in this dataset is not surprising (as confirmed by the highly variable H<sub>2</sub>O/S<sub>tot</sub> ratios). Gases from closed-conduit volcanoes in a fumarolic stage of activity

and from volcanic lakes of arc-volcanism (i.e. Poás, Rincón de la Vieja, Mutnovsky and Aso) show very low Hg contents, probably due either to (1) the role played by the omnipresent wall-rock-gas and water-gas interactions which potentially favour Hg deposition in the hydrothermal envelope, or (2) the lake's gas scrubbing, respectively. Although limited, our data however suggest that there is no clear Hg/S<sub>tot</sub> and Hg/H<sub>2</sub>O dependence on volcanic setting, but further detailed investigations are required.

#### Hg emission fluxes from volcanic degassing

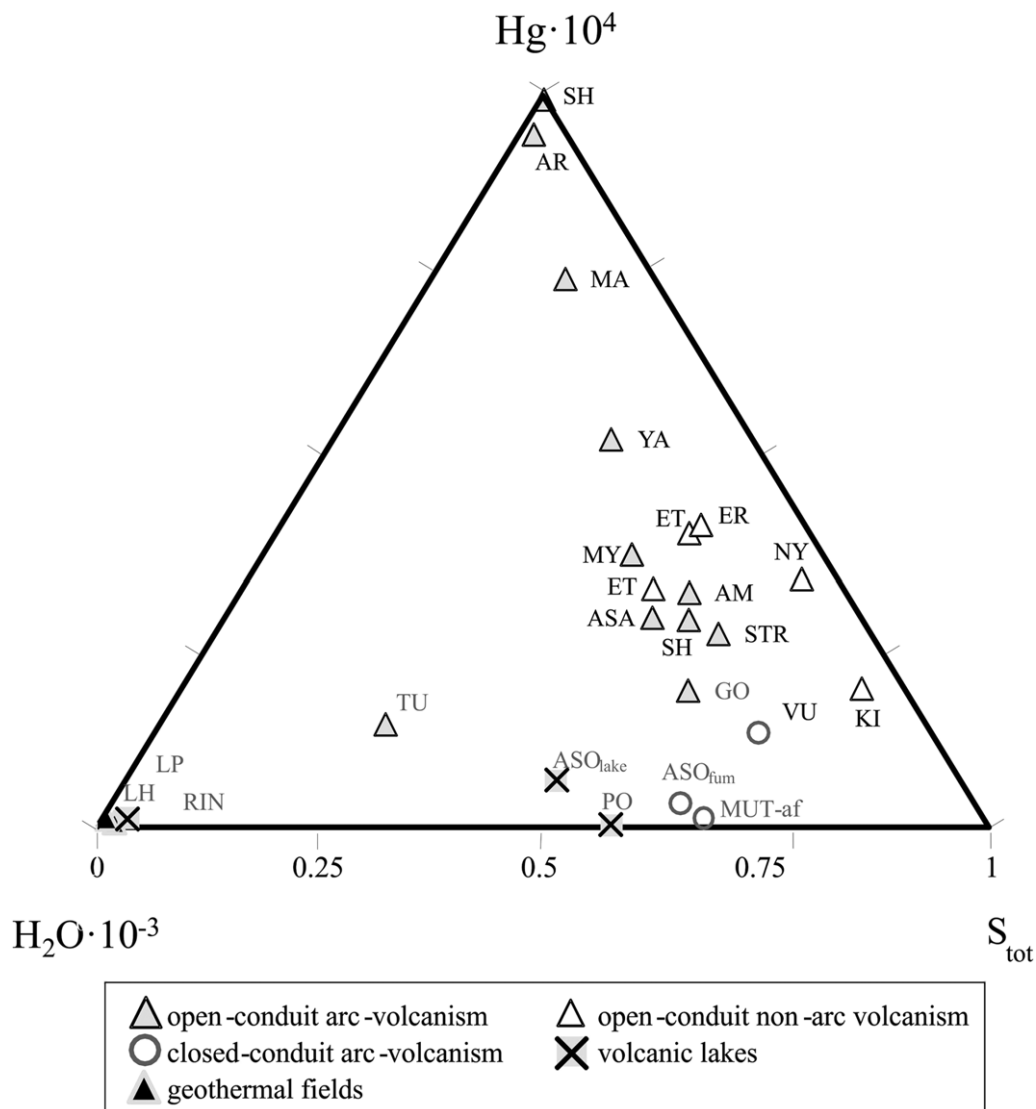
Mercury emission rates from volcanoes are usually assessed by scaling the determined GEM/SO<sub>2</sub> to the simultaneously determined SO<sub>2</sub> emission rates. The volcanic Hg emissions inventory is summarized in Table 2, where Hg fluxes from the literature with relevant data sources are also reported. The lowest Hg flux ( $0.002 \text{ t a}^{-1}$ ) in our dataset is derived for Poás volcano, followed by Aso (*c.*  $0.33 \text{ t a}^{-1}$ ), Mutnovsky ( $0.71 \text{ t a}^{-1}$ ), Gorely ( $0.9 \text{ t a}^{-1}$ ), Turrialba ( $2.4 \text{ t a}^{-1}$ ) and Etna ( $4.2 \text{ t a}^{-1}$ ). While our estimates of Hg flux in the plumes of Aso and Etna volcanoes are comparable to those reported in the literature by previous authors (*c.*  $0.3$  and  $5.4 \text{ t a}^{-1}$ , respectively; Andres & Kasgnoc 1998; Bagnato *et al.* 2007), our bulk Hg emission flux measured at Mutnovsky volcano ( $0.71 \text{ t a}^{-1}$ ) is about one order of magnitude higher than the value proposed by Taran *et al.* (1992) of  $0.05 \text{ t a}^{-1}$ . The difficult meteorological conditions encountered on the top of Rincón de la Vieja volcano in April 2013 did not allow us to estimate SO<sub>2</sub> emissions needed for calculating the Hg flux. No data on SO<sub>2</sub> flux related to Rincon's recent degassing activity are available; Hg emission rate from this volcano has not been included in our calculation of the cumulative flux from active volcanism.



**Fig. 8.** Scatter plots of GEM/SO<sub>2</sub> (mass) ratio v. H<sub>2</sub>O/SO<sub>2</sub> (molar) ratio of arc (grey symbols) and non-arc (black open triangles) volcanoes. The graph was constructed from results obtained in this study (grey text) and previously published data (black text). A distinction was made between the emission type plotted in the graph to show open/closed vent, arc/non-arc volcanism, volcanic lake evasion and geothermal fields. The diagram suggests that there is some difference between the GEM/SO<sub>2</sub> in arc and non-arc setting (see Table 2). ‘Arc’ volcanoes show an overall positive correlation between H<sub>2</sub>O content and GEM/SO<sub>2</sub> ratios, which might suggest that gas emissions in subduction zones may contain more Hg (relative to SO<sub>2</sub>) than those from divergent (or within-plate) plate volcanism. Literature H<sub>2</sub>O/SO<sub>2</sub> data and GEM/SO<sub>2</sub> mass ratios are from <sup>a</sup>Bagnato *et al.* (2011) (NY, Nyiragongo; YA, Yasur; AM, Ambrym; STR, Strombolim; ASA, Asama; MY, Miyakejima; VU, Vulcano); <sup>b</sup>Bagnato *et al.* (2007) (ET, Etna); <sup>c</sup>Kyle *et al.* (1990) (ER, Erebus); <sup>d</sup>Varekamp & Buseck (1986) (SH, St Helens); <sup>e</sup>this study (ET, Etna; PO, Poás; TU, Turrialba; RIN, Rincón de la Vieja; LH, Las Hornillas; LP, Las Pailas; ASO, Aso; MUT, Mutnovsky; GO, Gorely). Hg and H<sub>2</sub>O concentrations have been multiplied and divided, respectively, for a certain factor in order to improve the representation of data inside the graph.

The large spread of our estimated Hg fluxes reflects the variable GEM/SO<sub>2</sub> ratios (10<sup>-5</sup> to 10<sup>-8</sup>) exhibited by the investigated volcanoes and reveals variability in the SO<sub>2</sub> mass flux and hence in the underlying mode of degassing. We stress that data presented here refer to volcanoes in a state of passive degassing activity, and that the above-cited values have to be considered as a lower bound (Table 2). Our calculations confirm

that the continuous year-on-year emission from passively degassing volcanoes is relatively small if compared to explosive eruptions, which are usually accompanied by considerable SO<sub>2</sub> emission (e.g. 1991 Pinatubo eruption, Ansmann *et al.* 1997; 1980 Mt St Helens eruption, Varekamp & Buseck 1981). The most significant single-event emissions are those associated with rare, large explosive eruptions occurring once or twice per century



**Fig. 9.**  $\text{SO}_2$ – $\text{H}_2\text{O}$ – $\text{Hg}$  triangular plot of relative concentrations in the plumes of the investigated volcanoes. Our original data (grey text) are set in the context of published volcanic gas compositions from various geodynamic settings (black text). A distinction was made between the emission type plotted in the graph to show open/closed vent, arc/non-arc volcanism, volcanic lake surface and geothermal fields. Our dataset exhibits a very variable composition ranging from Hg to  $\text{H}_2\text{O}$ -rich, respectively. The most  $\text{H}_2\text{O}$ -poor composition is observed at Nyiragongo and Kilauea (NY and KI). Stromboli (ST) gases are S-enriched (and depleted in  $\text{H}_2\text{O}$ ) relative to ‘typical’ arc volcanic gases. The most  $\text{H}_2\text{O}$ -rich composition is observed at Rincón de la Vieja volcano while the most Hg-rich composition is observed for Mt St Helens explosive eruption in 1980 (SH) and Arenal (AR). Data source are taken from Tables 1 and 2 (this study) and from the literature: NY, Nyiragongo; YA, Yasur; AM, Ambrym; STR, Stromboli; ASA, Asama; VU, Vulcano, Bagnato *et al.* (2011); MA, Masaya, Witt *et al.* (2008a, b); ET<sub>bulk</sub>, Etna bulk, Bagnato *et al.* (2007); SH, St Helens, Varekamp & Buseck (1986); AR, Arenal, Zimmer *et al.* (2004); Ballantine *et al.* (1982); ER, Erebus, Kyle *et al.* (1990).

which may instantaneously release 10 to  $10^3$  Tg  $\text{SO}_2$  to the atmosphere (Pyle & Mather 2003). Such large eruptions, which may release  $>10^3$  to  $10^4$  Mg Hg

per event, are thought to significantly perturb the atmospheric Hg reservoir for *c.* 1 yr (Pyle & Mather 2003).



### Hg emission fluxes from geothermal areas

Mercury fluxes from geothermal areas are still poorly investigated and highly uncertain due to the complexity of making appropriate measurements of Hg emissions (e.g. Gustin 2003), but probably total  $\leq 60 \text{ t Hg a}^{-1}$  (Varekamp & Buseck 1986; Nriagu & Becker 2003; Pyle & Mather 2003). The Hg/SO<sub>2</sub> approach for estimating Hg flux from volcanoes is not readily applicable to geothermal sources since the hydrothermal fluids are efficient sulphur scrubbers and there is no reliable method for measuring the flux of the dominant form of sulphur in these systems. In order to define Hg emission rates from Las Pailas and Las Hornillas geothermal fields, we interpolated (using the kriging contouring technique) the site-specific emission data to obtain contour maps of the Hg deposition flux which, by integration over the entire area covered by the survey (Fig. 3), allowed calculation of the integrated amount of Hg on a horizontal surface above the fumarolic areas ( $\text{ng m}^{-1}$ ). In order to calculate the Hg fluxes through these contoured area, the gas rising speed was extrapolated from *in situ* observations, displaying similar values to other not over-pressurized fumarolic emissions (Vulcano island, Tamborello *et al.* 2011; Campi Flegrei, Aiuppa *et al.* 2013; Nea Kameni, Bagnato *et al.* 2013). By considering a highly variable gas speed of  $0.5\text{--}1.5 \text{ m s}^{-1}$ , we estimated a time-averaged total Hg flux ranging from  $0.0008$  to  $0.0024 \text{ t a}^{-1}$  for Las Pailas and from  $0.004$  to  $0.012 \text{ t a}^{-1}$  for Las Hornillas emissions (Table 2). The cumulative Hg flux released from both geothermal fields therefore ranges from  $0.005$  to  $0.014 \text{ t a}^{-1}$ , appearing less important than the volcanic flux from continuous passive degassing and accounting for a very small proportion of the global Hg flux for geothermal emissions (c.  $60 \text{ t a}^{-1}$ ; Varekamp & Buseck 1986).

### Discussion

Our measurements provide new information to the debate on Hg emission from active volcanism (Table 4). Worldwide volcanic Hg fluxes vary widely since volcanoes display a wide range of activity from continuous to short-lived events, matched perhaps by their diversity and geographic distribution. While it has long been known that volcanoes are important atmospheric sources of Hg (Varekamp & Buseck 1981, 1986; Bagnato *et al.* 2007, 2009a, b, 2011), considerable uncertainty remains about the global volcanic Hg flux as the available data on volcanogenic Hg emissions is still sparse and incomplete (Pyle & Mather 2003). Using the database on active and passively

**Table 4.** Previous estimates of global volcanic emission rates of mercury

Source	Emission rate, ton Hg a <sup>-1</sup>
Pyle & Mather (2003)	75 (passive degassing) – 700 (explosive activity)
Nriagu & Becker (2003)	37.6 (passive degassing) – 57 (explosive activity)
Varekamp & Buseck (1986)	30 (passive degassing) – 830 (explosive activity)
Ferrara <i>et al.</i> (2000)	0.6–1.3
Fitzgerald (1986)	20–90
Nriagu (1989)	500 (range 30–1000)
Unni <i>et al.</i> (1978)	3–9.0
Anderson (1975)	100

degassing volcanoes of the island arc type, Anderson (1975) came up with a value of  $100 \text{ t a}^{-1}$ . This estimate contrasts with a much lower flux of  $3\text{--}9 \text{ t a}^{-1}$  reported by Unni *et al.* (1978). From measurements of non-eruptive emissions at Kilauea (Hawaii), Colima (Mexico), Cascades volcanoes (Mt Shasta, Mt Hood and Mt St Helens) and Etna (Italy), Varekamp & Buseck (1986) estimated the global Hg emission to be  $830 \text{ t a}^{-1}$ . Based on measurements of Hg/S ratios at three volcanoes (Kilauea, Etna and White Island), Fitzgerald (1986) was able to derive a global Hg flux of  $20\text{--}90 \text{ t a}^{-1}$ . More recently, Ferrara *et al.* (2000) used the Hg/S data for three volcanoes in Italy and Sicily (Etna, Stromboli and Vulcano) to derive a global annual flux of only  $0.6\text{--}1.3 \text{ t a}^{-1}$ . Nriagu & Becker (2003) derived a global Hg fluxes ranging from  $37.6$  (from degassing plume) to  $57 \text{ t a}^{-1}$  (from explosive activity). More recently, Pyle & Mather (2003) suggested that Hg contributed annually by passively degassing volcanoes globally is of the order  $75 \text{ t a}^{-1}$  (range  $6\text{--}900 \text{ t a}^{-1}$ ), or  $700 \text{ t a}^{-1}$  (range  $80\text{--}4000 \text{ t a}^{-1}$ ) if considering the total time-averaged annual emission from degassing and erupting volcanoes. It is clear from Table 4 that there is a considerable range in estimates of the total Hg flux from volcanoes. The differences between previous Hg fluxes principally reflect the paucity of appropriate data on Hg in volcanic emissions and the ways that previous authors interpreted published data on volcanic Hg.

Table 2 summarizes our new observations on passive volcanic degassing, as well as previously published work. If considering the 12 volcanoes reported in Table 2 and referring to plume emission type, they account for a cumulative SO<sub>2</sub> flux of  $c. 6 \text{ Mt a}^{-1}$  or 62% of the global SO<sub>2</sub> volcanic emissions from continuously passive degassing volcanoes ( $9.7 \text{ Mt a}^{-1}$ ; Andres & Kasgnoc 1998). We can therefore estimate Hg emissions in two ways: (1) a cumulative Hg emission flux based on Hg/SO<sub>2</sub> ratios and SO<sub>2</sub> fluxes at these 12 individual volcanoes, which amounts to  $56 \text{ t a}^{-1}$ ; and (2) a

global Hg emission flux based on the globally averaged Hg/SO<sub>2</sub> ratio of  $7.8 \times 10^{-6}$  ( $\pm 1.5 \times 10^{-6}$ ; 1 SE,  $n = 13$ ) and the global volcanic SO<sub>2</sub> flux of  $9.7 \text{ Mt a}^{-1}$  (Andres & Kasgnoc 1998). While Andres & Kasgnoc (1998) did not present an uncertainty budget associated with their  $9.7 \text{ Mt a}^{-1}$  SO<sub>2</sub> flux estimate, a second study on volcanic SO<sub>2</sub> fluxes proposed an uncertainty of 35% relative standard deviation (Graf *et al.* 1997) and is widely considered as appropriate (Schmidt *et al.* 2012). By combining our updated average Hg/SO<sub>2</sub> plume mass ratio for quiescent volcanic activity of  $7.8 \times 10^{-6}$  ( $\pm 1.5 \times 10^{-6}$ ; 1 SE,  $n = 13$ ) (Table 2) and the uncertainties on SO<sub>2</sub> flux we therefore obtain a global volcanic Hg flux from continuous passive degassing of  $76 \pm 30 \text{ t a}^{-1}$  (Table 2). Our estimate is similar to that of Pyle & Mather (2003), but for the first time includes an estimate of uncertainty.

Since the global SO<sub>2</sub> emission rate used in this calculation ( $9.7 \text{ Mt a}^{-1}$ ; Andres & Kasgnoc 1998) represents open-conduit passive degassing, the total Hg flux from volcanoes during strong eruptive events may be even larger than that provided by the reported data in this study (Roos-Barraclough *et al.* 2002; Schuster *et al.* 2002).

The measurements of gas emissions from closed-conduit volcanoes in a fumarolic-stage of activity and geothermal emissions suggest a much smaller flux of Hg released from these activities. We estimated a cumulative Hg flux from three closed-conduit volcanoes (Aso, Mutnovsky and Poás; this study) and both geothermal fields (Las Pailas and Las Hornillas; this study) of  $c. 1 \text{ t a}^{-1}$  (Table 2). If we also consider Hg fluxes from various quiescent degassing closed-conduit volcanoes in a fumarolic stage of activity recently reported in literature ( $0.55\text{--}0.65 \text{ t a}^{-1}$ ; Bagnato *et al.* 2011), our estimated cumulative Hg flux from closed-conduit volcanoes would increase to  $c. 1.6 \text{ t a}^{-1}$  ( $c. 2\%$  of our estimate of the yearly global Hg emission from passive degassing volcanoes). Although comprising a marginal contribution to the global volcanic non-eruptive Hg emissions from closed-conduit degassing volcanoes, these new data represent the first available assessment of Hg emissions directly sampled at these locations. The minor Hg input from closed-conduit volcanoes in fumarolic-stage of activity compared to open-conduit volcanoes mainly reflects different SO<sub>2</sub> degassing rates and the role played by the omnipresent wall-rock–gas and water–gas interactions, which potentially favour Hg deposition in the hydrothermal envelope (Figs 5 & 6).

### Concluding remarks

This study provides new information to expand the limited database of volcanic Hg emissions

and allows some inferences to be made regarding the quality of previous estimates of global volcanic Hg inventories. A critical assessment of the published literature suggests that some global Hg emission estimates are likely to be significantly underestimated. We demonstrate that volcanic degassing is an important component of the global atmospheric Hg budget, even when only considering quiescent phases of volcanic activity. Using our dataset in tandem with literature information, we propose a mean Hg/SO<sub>2</sub> mass ratio of  $c. 7.8 \times 10^{-6}$  ( $\pm 1.5 \times 10^{-6}$ ; 1 SE,  $n = 13$ ) from quiescent degassing volcanoes as representative of gas emissions from the open-conduit degassing volcanoes used in this study. Taking into account the uncertainty in global SO<sub>2</sub> emissions, we derive a global volcanic Hg flux from quiescent volcanoes of  $c. 76$  ( $\pm 30$ )  $\text{t a}^{-1}$ , which is comparable to previous literature estimates. Our analysis suggests that volcanoes in a state of passive degassing represent an important contribution to global volcanic mercury emissions into the atmosphere. It is therefore likely that volcanic contributions to the global atmospheric Hg budget will be even more important during large eruptive events. For this reason, the measurement of Hg in eruptive gases is a priority for future research.

This work was supported by a research grant funded by the Universidad Nacional in Heredia, Costa Rica during March–April 2013. JES acknowledges funding from the European Research Council (ERC-2010-StG-20091028). The authors gratefully acknowledge the OVSICORI-UNA, Costa Rica for their logistic support in the field. The authors also wish to thank two anonymous reviewers for helpful comments on a previous version of the manuscript. Finally, EB thanks M. Liuzzo (INGV, sez. Palermo, Italy) and S. Calabrese (DiSTeM, Unipa) for their helpful contribution in sampling gases.

### References

- AIUPPA, A., FEDERICO, C., GIUDICE, G., GURRIERI, S., PAONITA, A. & VALENZA, M. 2004. Plume chemistry provides insights into the mechanisms of sulfur and halogen degassing at basaltic volcanoes. *Earth and Planetary Science Letters*, **222**, 469–483.
- AIUPPA, A., FEDERICO, C. *ET AL.* 2005. Emission of bromine and iodine from Mount Etna volcano. *Geochemistry, Geophysics, Geosystem*, **6**, Q08008, <http://dx.doi.org/10.1029/2005GC000965>
- AIUPPA, A., BAGNATO, E., WITT, M. L. I., MATHER, T. A., PARELLO, F., PYLE, D. M. & MARTIN, R. S. 2007. Real-time simultaneous detection of volcanic Hg and SO<sub>2</sub> at La Fossa Crater, Vulcano (Aeolian Islands, Sicily). *Geophysical Research Letters*, **34**, L21307, <http://dx.doi.org/10.1029/2007GL030762>
- AIUPPA, A., BERTAGNINI, A., MÉTRICH, N., MORETTI, R., DI MURO, A., LIUZZO, M. & TAMBURELLO, G. 2010. A

- model of degassing for Stromboli volcano. *Earth and Planetary Science Letters*, **295**, 195–204.
- AIUPPA, A., SHINOHARA, H., TAMBURELLO, G., GIUDICE, G., LIUZZO, M. & MORETTI, R. 2011. Hydrogen in the gas plume of an open-vent volcano, Mount Etna, Italy. *Journal of Geophysical Research*, **116**, B10204, <http://dx.doi.org/10.1029/2011JB008461>
- AIUPPA, A., GIUDICE, G. ET AL. 2012. First volatile inventory for Gorely volcano, Kamchatka. *Geophysical Research Letters*, **39**, L06307, <http://dx.doi.org/10.1029/2012GL051177>
- AIUPPA, A., TAMBURELLO, G. ET AL. 2013. First observations of the fumarolic gas output from a restless caldera: implications for the current period of unrest (2005–2013) at Campi Flegrei. *Geochemistry Geophysics Geosystems*, **14**, 4153–4169.
- ALLARD, P., CARBONNELLE, J. ET AL. 1991. Eruptive and diffuse emissions of CO<sub>2</sub> from Mount Etna. *Nature*, **351**, 387–391, <http://dx.doi.org/10.1038/351387a0>
- ALLARD, P., BEHNCKE, B., D'AMICO, S., NERI, M. & GAMBINO, S. 2006. Mount Etna 1993–2005: anatomy of an evolving eruptive cycle. *Earth-Science Reviews*, **78**, 85–114.
- ANDERSON, A. T. 1975. Some basaltic and andesitic gases. *Reviews of Geophysics and Space Physics*, **13**, 37–56.
- ANDRES, R. J. & KASGNOC, A. D. 1998. A time averaged inventory of subaerial volcanic sulfur emissions. *Journal of Geophysical Research*, **103**, 25 251–25 261.
- ANSMANN, A., MATTIS, I., WANDINGER, U., WAGNER, F., REICHARDT, J. & DESHLER, T. 1997. Evolution of the Pinatubo aerosol: Raman Lidar observations of particle optical depth, effective radius, mass and surface area over Central Europe at 53.4°N. *Journal of the Atmospheric Science*, **54**, 2630–42641.
- ARIYA, P. A., SKOV, H., GRAGE, M. M. L. & GOODSITE, M. E. 2008. Gaseous elemental mercury in the ambient atmosphere: review of the application of theoretical calculations and experimental studies for determination of reaction coefficients and mechanisms with halogens and other reactants. *Advances in Quantum Chemistry*, **55**, 43–55.
- BAGNATO, E., AIUPPA, A. ET AL. 2007. Degassing of gaseous (elemental and reactive) and particulate mercury from Mount Etna volcano (Southern Italy). *Atmospheric Environment*, **41**, 7377–7388, <http://dx.doi.org/10.1016/j.atmosenv.2007.05.060>
- BAGNATO, E., PARELLO, F., VALENZA, M. & CALIRO, S. 2009a. Mercury content and speciation in the Phlegrean Fields volcanic complex: evidences from hydrothermal system and fumaroles. *Journal of Volcanology and Geothermal Research*, **187**, 250–260.
- BAGNATO, E., ALLARD, P., PARELLO, F., AIUPPA, A., CALABRESE, S. & HAMMOUYA, G. 2009b. Mercury gas emissions from La Soufrière Volcano, Guadeloupe Island (Lesser Antilles). *Chemical Geology*, **266**, 276–282.
- BAGNATO, E., AIUPPA, A., PARELLO, F., ALLARD, P., LIUZZO, M., GIUDICE, G. & SHINOHARA, H. 2011. New clues on mercury contribution from Earth volcanism. *Bulletin of Volcanology*, **73**, 497–510, <http://dx.doi.org/10.1007/s00445-010-0419-y>
- BAGNATO, E., TAMBURELLO, G., AIUPPA, A., SPROVIERI, M., VOUGIOUKALAKIS, G. E. & PARKS, M. 2013. Atmospheric mercury emissions from substrate and fumaroles at Nea Kameni volcanic system, Santorini (Greece). *Geochemical Journal*, **47**, 437–450.
- BALLANTINE, D. S., FINNEGAN, D. L., PHELAN, J. M. & ZOLLER, W. H. 1982. Measurement of Hg/S ratios from five volcanoes. *EOS Transaction, American Geophysical Union*, **63**, 1152.
- BICHLER, M., POLJANK, K. & SORTINO, F. 1995. Determination and speciation of minor and trace elements in volcanic exhalation by NAA. *Journal of Radioanalytical and Nuclear Chemistry*, **92**, 183–194, <http://dx.doi.org/10.1007/BF02041722>
- BLUTH, G. J. S., SHANNON, J. M., WATSON, I. M., PRATA, A. J. & REALMUTO, V. J. 2007. Development of an ultra-violet digital camera for volcanic SO<sub>2</sub> imaging. *Journal of Volcanology and Geothermal Research*, **161**, 47–56, <http://dx.doi.org/10.1016/j.jvolgeores.2006.11.004>
- BUAT-MENARD, P. & ARNOLD, M. 1978. The heavy metal chemistry of particulate matter emitted by Mount Etna volcano. *Geophysical Research Letters*, **5**, 245–248.
- CAMPION, R., MARTINEZ-CRUZ, M. ET AL. 2012. Space- and ground-based measurements of sulphur dioxide emissions from Turrialba Volcano (Costa Rica). *Bulletin of Volcanology*, **74**, 1757–1770, <http://dx.doi.org/10.1007/s00445-012-0631-z>
- CARR, M. J. 1984. Symmetrical and segmented variation of physical and geochemical characteristics of the Central American volcanic front. *Journal of Volcanology and Geothermal Research*, **20**, 231–252.
- CARR, M. J., FEIGENSON, M. D., PATINO, L. C. & WALKER, J. A. 2003. Volcanism and geochemistry in Central America: progress and problems. In: EILER, J. (ed.) *Inside the Subduction Factory*. American Geophysical Union, Washington, 153–174.
- CASTRO, S. 2002. Reservoir engineering studies in the Las Pailas geothermal field, Costa Rica. Geothermal Training in Iceland 2002, Report 4. UNU-GTP, Iceland.
- CHEMINÉE, J. M., JAVOY, M. & DELORME, H. 1983. Temperature and gas data from Turrialba. SEAN 08.01, Smithsonian Institution, [http://www.volcano.si.edu/world/volcano.cfm?vnum01405-070&volpage0var#sean\\_0801](http://www.volcano.si.edu/world/volcano.cfm?vnum01405-070&volpage0var#sean_0801)
- CHRISTENSON, B. W. & MROCEK, E. K. 2003. Potential reaction pathways of Hg in some New Zealand hydrothermal environments. *Society of Economic Geology*, **10**, 111–132.
- COX, M. E. 1983. Summit outgassing as indicated by radon, mercury, and pH mapping, Kilauea volcano, Hawaii. *Journal of Volcanology and Geothermal Research*, **16**, 131–152.
- DEDUERWAERDER, H., DECADT, G. & BAEYENS, W. 1982. Estimations of mercury fluxes emitted by Mt Etna volcano. *Bulletin of Volcanology*, **45**, 191–196.
- DELMAS, R. J., KIRCHNER, S., PALAIS, J. M. & PETIT, J. R. 1992. 1000 years of explosive volcanism recorded at the South Pole. *Tellus*, **44B**, 335–350.
- ENGLE, M. A., GUSTIN, M. S., GOFF, F., COUNCE, D. A., JANIK, C. J., BERGFELD, D. & RYTUBA, J. J. 2006. Atmospheric mercury emissions from substrates and fumaroles associated with three hydrothermal systems in the western United States. *Journal of*

## MERCURY FLUXES FROM VOLCANIC AND GEOTHERMAL SOURCES

- Geophysical Research*, **111**, D17304, <http://dx.doi.org/10.1029/2005JD006563>
- FERRARA, R., MAZZOLAI, B., LANZILLOTTA, E., NUCARO, E. & PIRRONE, N. 2000. Volcanoes as emission sources of atmospheric mercury in the Mediterranean basin. *The Science of the Total Environment*, **259**, 115–121.
- FITZGERALD, W. F. 1986. Cycling of mercury between the atmosphere and oceans. In: BUAT-MÉNARD, P. (ed.) *The Role of Air-Sea Exchange in Geochemical Cycling*. D. Reider Publishing Co., Dordrecht, 363–408.
- FRIEDLI, H. R., RADKE, L. F., LU, J. Y., BANIC, C. M., LEITCH, W. R. & MACPHERSON, J. I. 2003. Mercury emissions from burning of biomass from temperate North American forests: laboratory and airborne measurements. *Atmospheric Environment*, **37**, 253–267.
- GALLE, B., OPPENHEIMER, C., GEYER, A., MCGONIGLE, A. J. S., EDMONDS, M. & HORROCKS, L. A. 2003. A miniaturised ultraviolet spectrometer for remote sensing of SO<sub>2</sub> fluxes: a new tool for volcano surveillance. *Journal of Volcanology and Geothermal Research*, **119**, 241–254.
- GHERARDI, F., PANICHA, C., YOCK, A. & GERARDO-ABAYA, J. 2002. Geochemistry of the surface and deep fluids of the Miravalles volcano geothermal system (Costa Rica). *Geothermics*, **31**, 91–128.
- GRAF, H. F., FEICHTER, J. & LANGMANN, B. 1997. Volcanic sulfur emissions: estimates of source strength and its contribution to the global sulfate distribution. *Journal of Geophysical Research-Atmospheres*, **102**, 10 727–10 738.
- GUSTIN, M. S. 2003. Are mercury emissions from geologic sources significant? A status report. *Science of the Total Environment*, **304**, 153–167.
- GUSTIN, M. S., LINDBERG, S. E. *ET AL.* 1999. Nevada STORMS project: measurement of mercury emissions from naturally enriched surfaces. *Journal of Geophysical Research*, **104**, 21 831–21 844.
- GUSTIN, M. S., LINDBERG, S. E. & WEISBERG, P. J. 2008. An update on the natural sources and sinks of atmospheric mercury. *Applied Geochemistry*, **23**, 482–493.
- HALMER, M. M., SCHMINCKE, H. U. & GRAF, H. F. 2002. The annual volcanic gas input into the atmosphere, in particular into the stratosphere: a global data set for the past 100 years. *Journal of Volcanology and Geothermal Research*, **115**, 511–528.
- HENLEY, R. W. & BERGER, B. R. 2013. Nature's refineries – Metals and metalloids in arc volcanoes. *Earth-Science Reviews*, **125**, 146–170.
- HINKLEY, T. K., LAMOTHE, P. J., WILSON, S. A., FINNEGAN, D. L. & GERLACH, T. M. 1999. Metal emissions from Kilauea, and a suggested revision of the estimated worldwide metal output by quiescent degassing of volcanoes. *Earth and Planetary Science Letters*, **170**, 315–325.
- JENNER, F. E., O'NEILL, H. C., ARCULUS, R. J. & MAVROGENES, J. A. 2010. The magnetite crisis in the evolution of arc-related magmas and the initial concentration of Au, Ag, and Cu. *Journal of Petrology*, **51**, 2445–2464, <http://dx.doi.org/10.1093/petrology/egq063>
- KANTZAS, E. P., MCGONIGLE, A. J. S., TAMBURELLO, G., AIUPPA, A. & BRYANT, R. G. 2010. Protocols for UV camera volcanic SO<sub>2</sub> measurements. *Journal of Volcanology and Geothermal Research*, **94**, 55–60, <http://dx.doi.org/10.1016/j.jvolgeores.2010.05.003>
- KIM, C. S., BROWN, G. E. & RYTUBA, J. J. 2000. Characterization and speciation of mercury-bearing mine wastes using X-ray absorption spectroscopy. *Science of the Total Environment*, **261**, 157–168.
- KIM, K. H., MISHRA, V. K. & HONG, S. 2006. The rapid and continuous monitoring of gaseous elemental mercury (GEM) behavior in ambient air. *Atmospheric Environment*, **40**, 3281–3293.
- KIM, K. H., SHON, Z. H., NGUYEN, H. T., JUNG, K., PARK, C. G. & BAE, G. N. 2011. The effect of man made source processes on the behavior of total gaseous mercury in air: a comparison between four urban monitoring sites in Seoul Korea. *Science of the Total Environment*, **409**, 3801–3811.
- KYLE, P. R., MEEKER, M. & FINNEGAN, D. 1990. Emission rates of sulfur dioxide, trace gases and metals from Mount Erebus, Antarctica. *Geophysical Research Letters*, **17**, 2125–2128.
- LABANIEH, S., CHAUVEL, C., GERMA, A. & QUIDELLEUR, X. 2012. Martinique: a clear case for sediment melting and slab dehydration as a function of distance to the trench. *Journal of Petrology*, **53**, 2441–2464.
- LANGWAY, C. C., OSADA, K., CLAUSEN, H. B., HAMMER, C. U. & SHOJI, H. 1995. A 10-century comparison of prominent bipolar volcanic events in ice cores. *Journal of Geophysical Research*, **100**, 16 241–16 247.
- LINDBERG, S., BULLOCK, R. *ET AL.* 2007. A synthesis of progress and uncertainties in attributing the sources of mercury in deposition. *Ambio*, **36**, 19–32.
- LINDQVIST, O. & RODHE, H. 1985. Atmospheric mercury – a review. *Tellus Series B-Chemical and Physical Meteorology*, **37**, 136–159.
- LIU, S. L., NADIM, F., PERKINS, C., CARLEY, R. J., HOAG, G. E., LIN, Y. H. & CHEN, L. T. 2002. Atmospheric mercury monitoring survey in Beijing, China. *Chemosphere*, **48**, 97–107.
- LOPES, R. (ed.) 2005. Volcanoes in Costa Rica. In: *The Volcano Adventure Guide*. Cambridge University Press, 286–292.
- MADONIA, P., RIZZO, A. L., DILIBERTO, I. S. & FAVARA, R. 2013. Continuous monitoring of fumarole temperatures at Mount Etna (Italy). *Journal of Volcanology and Geothermal Research*, **257**, 12–20, <http://dx.doi.org/10.1016/j.jvolgeores.2013.03.001>
- MARTIN, R., WITT, M. L. I., PYLE, D. M., MATHER, T. A., WATT, S. F. L., BAGNATO, E. & CALABRESE, S. 2011. Rapid oxidation of mercury (Hg) at volcanic vents: insights from high temperature thermodynamic models of Mt Etna's emissions. *Chemical Geology*, **283**, 279–286.
- MARTIN, R., WITT, M. L. I. *ET AL.* 2012. Bioindication of volcanic mercury (Hg) deposition around Mt. Etna (Sicily). *Chemical Geology*, **310–311**, 12–22.
- MARTÍNEZ, M. 2008. *Geochemical evolution of the acid crater lake of Poás Volcano (Costa Rica): insights into volcanic-hydrothermal processes*. Ph.D. thesis, University of Utrecht, the Netherlands.
- MARTÍNEZ, M., FERNÁNDEZ, E. *ET AL.* 2000. Chemical evolution and volcanic activity of the active crater lake of Poás volcano, Costa Rica, 1993–1997.



- Journal of Volcanology and Geothermal Research*, **97**, 127–141.
- MATHER, T. A., WITT, M. L. I. ET AL. 2012. Halogens and trace metal emissions from the ongoing 2008 summit eruption of Kilauea volcano, Hawaii. *Geochimica et Cosmochimica Acta*, **83**, 292–323.
- MCGONIGLE, A. J. S., OPPENHEIMER, C., GALLE, B., MATHER, T. A. & PYLE, D. M. 2002. Walking traverse and scanning DOAS measurements of volcanic gas emission rates. *Geophysical Research Letters*, **29**, 46-1–46-4, <http://dx.doi.org/10.1029/2002GL015827>
- MELIÁN, G. V., PÉREZ, N. M. ET AL. 2004. Emisión difusa de dióxido de carbono y vapor de mercurio en el volcán Miravalles, Costa Rica. *Revista Geológica de América Central*, **30**, 179–188.
- MELIÁN, G. V., GALINDO, I. ET AL. 2007. Diffuse emission of hydrogen from Poás Volcano, Costa Rica, América Central. *Pure and Applied Geophysics*, **164**, 2465–2487.
- MOREL, F. M. M., KRAEPIEL, A. M. L. & AMYOT, M. 1998. The chemical cycle and bioaccumulation of mercury. *Annual Review of Ecology, Evolution, and Systematics*, **29**, 543–566.
- MORI, T. & BURTON, M. 2006. The SO<sub>2</sub> camera: a simple, fast and cheap method for ground-based imaging of SO<sub>2</sub> in volcanic plumes. *Geophysical Research Letters*, **33**, L24804, <http://dx.doi.org/10.1029/2006GL027916>
- MUNGALL, J. E. 2002. Roasting the mantle: slab melting and the genesis of major Au and Au rich Cu deposits. *Geology*, **30**, 915–918.
- NAKAGAWA, R. 1999. Estimation of mercury emissions from geothermal activity in Japan. *Chemosphere*, **38**, 1867–1871.
- NRIAGU, J. O. 1989. A global assessment of natural sources of atmospheric trace metals. *Nature*, **338**, 47–49.
- NRIAGU, J. & PACYNA, J. M. 1988. Quantitative assessment of worldwide contamination of air, water and soil, by trace metals. *Nature*, **33**, 134–139.
- NRIAGU, J. & BECKER, C. 2003. Volcanic emissions of mercury to the atmosphere: global and regional inventories. *Science of the Total Environment*, **304**, 3–12.
- OHSAWA, S., SUDO, Y. ET AL. 2003. Some geochemical features of Yudamari Crater Lake, Aso volcano, Japan. *Geothermal Research Rep Kyushu University*, **12**, 62–65 (in Japanese with English abstract).
- OHSAWA, S., SAITO, T. ET AL. 2010. Color change of lake water at the active crater lake of Aso volcano, Yudamari, Japan: is it in response to change in water quality induced by volcanic activity? *Limnology*, **11**, 207–215.
- PIRRONE, N., CINNIRELLA, S. ET AL. 2010. Global mercury emissions to the atmosphere from anthropogenic and natural sources. *Atmospheric Chemistry and Physics Discussion*, **10**, 4719–4752.
- PYLE, D. M. & MATHER, T. A. 2003. The importance of volcanic emissions for the global atmospheric mercury cycle. *Atmospheric Environment*, **3**, 5115–5124.
- PYLE, D. M. & MATHER, T. A. 2009. Halogens in igneous processes and their fluxes to the atmosphere and oceans from volcanic activity: a review. *Chemical Geology*, **263**, 110–121.
- ROBERTSON, D. E., CRECELIUS, E. A., FRUCHTER, J. S. & LUDWICK, J. D. 1977. Mercury emissions from geothermal power plants. *Science*, **196**, 1094–1097, <http://dx.doi.org/10.1126/science.860131>
- ROBOCK, A. & FREE, M. P. 1995. Ice core as an index of global volcanism from 1850 to the present. *Journal of Geophysical Research*, **100**, 11549–11567.
- ROBOCK, A. & OPPENHEIMER, C. (eds) 2003. *Volcanism and the Earth's Atmosphere*. American Geophysical Union, Washington, Geophysical Monograph Series, **137**.
- ROOS-BARRACLOUGH, F., MARTINEZ-CORTIZAS, A., GARCIA-RÓDEJA, E. & SHOTYK, W. 2002. A 14,500 year record of accumulation of atmospheric mercury in peat: volcanic signals, anthropogenic influences and a correlation to bromine accumulation. *Earth and Planetary Science Letters*, **202**, 435–451.
- RUTTER, A. P., SCHAUER, J. J. ET AL. 2008. A comparison of speciated atmospheric mercury at an urban center and an upwind rural location. *Journal of Environmental Monitoring*, **10**, 102–108.
- RUTTER, A. P., SNYDER, D. C. ET AL. 2009. In situ measurements of speciated atmospheric mercury and the identification of source regions in the Mexico City Metropolitan Area. *Atmospheric Chemistry and Physics*, **9**, 207–220.
- SCHMIDT, A., CARSLAW, K. S. ET AL. 2012. Importance of tropospheric volcanic aerosol for indirect radiative forcing of climate. *Atmospheric Chemistry and Physics*, **12**, 7321–7339, <http://dx.doi.org/10.5194/acp-12-7321-2012>
- SCHROEDER, W. H. & MUNTHE, J. 1998. Atmospheric mercury – an overview. *Atmospheric Environment*, **32**, 809–822.
- SCHROEDER, W. H., KEELER, G., KOCK, H., ROUSSEL, P., SCNEEBERGER, D. & SCHAEDELICH, F. 1995. International field intercomparison of atmospheric mercury measurement methods. *Water Air and Soil Pollution*, **80**, 611–620.
- SCHUSTER, P. F., KRABENHOFT, D. P. ET AL. 2002. Atmospheric mercury deposition during the last 270 years: a glacial ice core record of natural and anthropogenic sources. *Environmental Science and Technology*, **36**, 2303–2310, <http://dx.doi.org/10.1021/es0157503>
- SHINOHARA, H. & WITTER, J. B. 2005. Volcanic gases emitted during mild Strombolian activity of Villarica volcano, Chile. *Geophysical Research Letters*, **32**, L20308, <http://dx.doi.org/10.1029/2005GL024131>
- SHINOHARA, H., AIUPPA, A., GIUDICE, G., GURRIERI, S. & LIUZZO, M. 2008. Variation of H<sub>2</sub>O/CO<sub>2</sub> and CO<sub>2</sub>/SO<sub>2</sub> ratios of volcanic gases discharged by continuous degassing of Mt. Etna Volcano, Italy. *Journal of Geophysical Research*, **113**, B09203, <http://dx.doi.org/10.1029/2007JB005185>
- SHINOHARA, H., YOSHIKAWA, S. & MIYABUCHI, Y. 2010. Degassing of Aso Volcano, Japan through an Acid Crater Lake: Differentiation of Volcanic Gas-Hydrothermal Fluids Deduced from Volcanic Plume Chemistry. *American Geophysical Union, Fall Meeting 2010*, San Francisco, abstract #V23A-2387.
- SHINOHARA, H., YOSHIKAWA, S. & MIYABUCHI, Y. 2013. Degassing activity of a volcanic crater lake: volcanic plume measurements at the Yudamari crater lake,



## MERCURY FLUXES FROM VOLCANIC AND GEOTHERMAL SOURCES

- Aso volcano, Japan. In: ROUWET, D., TASSI, J. & VAN-DEMEULEBROUCK, J. (eds) *Volcanic Lakes, Advances in Volcanology*. Springer-Heidelberg, accepted.
- SHISHKINA, T., BOTCHARNIKOV, R. E., HOLTZ, F., ALMEEV, R. R. & PORTNYAGIN, M. 2010. Solubility of H<sub>2</sub>O and CO<sub>2</sub>-bearing fluids in tholeiitic basalts at pressures up to 500 MPa. *Chemical Geology*, **177**, 115–125.
- SHOLUPOV, S., POGAREV, S., RYZHOV, V., MASHYANOV, N. & STROGANOV, A. 2004. Zeeman atomic absorption spectrometer RA-915+ for direct determination of mercury in air and complex matrix samples. *Fuel Processing Technology*, **85**, 473–485.
- SIEGEL, S. M. & SIEGEL, B. Z. 1984. First estimate of annual mercury flux at the Kilauea main vent. *Nature*, **309**, 146–147.
- SIMPSON, D., WINIWARTER, W. ET AL. 1999. Inventorying emissions from nature in Europe. *Journal of Geophysical Research*, **104**, 8113–8152.
- SLEMR, F., SCHUSTER, G. & SEILER, W. 1985. Distribution, speciation, and budget of atmospheric mercury. *Journal of Atmospheric Chemistry*, **3**, 407–434.
- SONKE, J. E., ZAMBARDI, T. & TOUTAIN, J. P. 2008. Indirect gold trap MC-ICP-MS coupling for Hg stable isotope analysis using a syringe injection interface. *Journal of Analytical Atomic Spectrometry*, **23**, 569–573.
- SPROVIERI, F., PIRRONI, N., EBINGHAUS, R., KOCK, H. & DOMMIGUE, A. 2010. A review of worldwide atmospheric mercury measurements. *Atmospheric Chemistry and Physics*, **10**, 8245–8265.
- STRACKE, A. 2012. Earth's heterogeneous mantle: a product of convection-driven interaction between crust and mantle. *Chemical Geology*, **330–331**, 274–299.
- SYMONDS, R. B., REED, M. H. & ROSE, W. I. 1992. Origin, speciation, and fluxes of trace-element gases at Augustine volcano, Alaska: insights into magma degassing and fumarolic processes. *Geochimica et Cosmochimica Acta*, **56**, 633–657.
- SYMONDS, R. B., GERLACH, T. M. & REED, M. H. 2001. Magmatic gas scrubbing: implications for volcano monitoring. *Journal of Volcanology and Geothermal Research*, **108**, 303–341.
- TAMBURELLO, G., MCGONIGLE, A. J. S., KANTZAS, E. P. & AIUPPA, A. 2011. Recent advances in ground-based ultraviolet remote sensing of volcanic SO<sub>2</sub> fluxes. *Annals of Geophysics*, **54**, 199–208, <http://dx.doi.org/10.4401/ag-5181>
- TARAN, Y. A., PILIPENKO, V. P., ROZHKOV, A. M. & VAKIN, E. A. 1992. A geochemical model for fumaroles of the Mutnovsky volcano, Kamchatka. *USSR Journal of Volcanology and Geothermal Research*, **49**, 269–283.
- TASSI, F., VASELLI, O. ET AL. 2005. The hydrothermal-volcanic system of Rincon de la Vieja volcano Costa Rica: a combined (inorganic and organic) geochemical approach to understanding the origin of the fluid discharges and its possible application to volcanic surveillance. *Journal of Volcanology and Geothermal Research*, **148**, 315–333.
- TERADA, A. & SUDO, Y. 2012. Thermal activity within the western slope geothermal zone of Aso volcano, Japan: development of a new thermal area. *Geothermics*, **42**, 56–64.
- UNNI, C., FITZGERALD, W. F., SETTLE, D., GILL, G., RAY, B., PATTERSON, C. C. & DUCE, R. 1978. The impact of volcanic emissions on the global atmospheric cycles of sulfur, mercury and lead. *EOS Transaction, American Geophysical Union*, **59**, 1223.
- VAREKAMP, J. C. & BUSECK, P. R. 1981. Mercury emissions from Mount St Helens during September 1980. *Nature*, **293**, 555–556.
- VAREKAMP, J. C. & BUSECK, P. R. 1986. Global mercury flux from volcanic and geothermal sources. *Applied Geochemistry*, **1**, 65–73.
- VASELLI, O., TASSI, F., DUARTE, E., FERNANDEZ, E., POREDA, R. J. & DELGADO HUERTAS, A. 2010. Evolution of fluid geochemistry at the Turrialba volcano (Costa Rica) from 1998 to 2008. *Bulletin of Volcanology*, **72**, 397–410, <http://dx.doi.org/10.1007/s00445-009-0332-4>
- VON GLASGOW, R. 2010. Atmospheric chemistry in volcanic plumes. *PNAS*, **107**, 6594–6599.
- WALLACE, P. J. 2005. Volatiles in subduction zone magmas: concentrations and fluxes based on melt inclusions and volcanic gas data. *Journal of Volcanology and Geothermal Research*, **140**, 217–240.
- WARDELL, L. J., KYLE, P. R. & COUNCE, D. 2008. Volcanic emissions of metals and halogens from White Island (New Zealand) and Erebus volcano (Antarctica) determined with chemical traps. *Journal of Volcanology and Geothermal Research*, **177**, 734–742.
- WEISSBERG, B. G. & RODHE, A. G. 1978. Mercury in some New Zealand geothermal discharges. *New Zealand Journal of Science*, **21**, 365–369.
- WITT, M. L. I., MATHER, T. A., PYLE, D. M., AIUPPA, A., BAGNATO, E. & TSANEV, V. I. 2008a. Mercury and halogen emissions from Masaya and Telica volcanoes, Nicaragua. *Journal of Geophysical Research*, **113**, B06203, <http://dx.doi.org/10.1029/2007JB005401>
- WITT, M. L. I., FISCHER, T. P., PYLE, D. M., YANG, T. F. & ZELLMER, G. F. 2008b. Fumarole compositions and mercury emissions from the Tatun Volcanic field, Taiwan: results from multi-component gas analyser, portable mercury spectrometer and direct sampling techniques. *Journal of Volcanology and Geothermal Research*, **178**, 636–643.
- ZAMBARDI, T., SONKE, J. E., TOUTAIN, J. P., SORTINO, F. & SHINOHARA, H. 2009. Mercury emissions and stable isotopic compositions at Vulcano Island (Italy). *Earth and Planetary Science Letters*, **277**, 236–243, <http://dx.doi.org/10.1016/j.epsl.2008.10.023>
- ZIMMER, M. M., FISCHER, T. P., HILTON, D. R., ALVARADO, G. E., SHARP, Z. D. & WALKER, J. A. 2004. Nitrogen systematics and gas fluxes of subduction zones: insights from Costa Rica arc volatiles. *Geochemistry Geophysics Geosystems*, **5**, <http://dx.doi.org/10.1029/2003GC000651>