

Optimization of anti-proliferative activity using a screening approach with a series of bis-heterocyclic G-quadruplex ligands

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ABSTRACT

Using a phenotypic screening and SAR optimization approach, a phenyl-bis-oxazole derivative has been identified with anti-proliferative activity, optimized with the use of a panel of cancer cell lines. The lead compound was synthesized by means of a short and effective two-step synthesis using Pd-catalyzed direct arylation. The compound stabilizes several quadruplex DNA sequences including a human telomeric DNA and one from the promoter of the HSP90 gene, although the structure–activity relationships of the series are not obviously related to the quadruplex binding.

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The targeting of G-tracts in genomic DNA that can fold into higher-order quadruplex structures,^{1–3} with the aid of small molecules,⁴ is emerging as a novel approach to cancer therapy, and may well have applications in other diseases. This concept has recently been validated by the direct visualization of quadruplex DNA in human cells and by the finding that they can be stabilized by small molecules.⁵ Many first- and second-generation quadruplex-binding and stabilizing ligands share common features of large planar surface areas and pendant cationic side-chains. Some have high ($>10^3$) selectivity for quadruplex over duplex DNA and have shown selectivity for cancer cell types over normal cells. Few have progressed to *in vivo* evaluation in human cancers, in part because of the perception that their features are insufficiently drug-like. More recently, a number of families of non-polycyclic quadruplex ligands have also been described, some of which have selectivity for particular promoter quadruplexes,³ such as those encoded in the c-MYC,⁶ c-KIT⁷ and HSP90⁸ promoter sequences. A large number of such quadruplex DNA targets are likely to exist, very few of which have yet been validated. This raises the challenge that even if selectivity for a chosen quadruplex is achieved, there is currently no straightforward approach that can be taken to ensure that such a target is unique within quadruplex space, although it is highly likely that particular quadruplexes can have unique sequences and tertiary structures. An example of this is shown by the (two) quadruplexes encoded in the c-KIT promoter.⁹ Telomeric quadruplexes may also have unique features since the length of

the single-stranded telomeric DNA sequence is sufficient for several consecutive quadruplexes to be formed from it.¹⁰

Optimization of biological activity for a given library of compounds has for the most part utilized structure–activity data obtained from quadruplex affinity measurements. Structure-based molecular design using crystallographic or NMR information has mostly focused on telomeric¹¹ or c-MYC quadruplexes,¹² simply because of the lack of structural data on many other potential quadruplex targets. We present here an alternative approach that starts with a family of quadruplex binding compounds and optimizes them by means of several cycles of phenotypic screening for anti-proliferative activity in a panel of cancer cell lines. This does not rely on the assumption of any particular quadruplex target, and we cannot discount the possibility that there may be alternative non-quadruplex targets involved.

We have previously identified a series of *para*-disubstituted phenyl-bis-triazoles (Fig. 1, compound I) as well as a focused library of disubstituted naphthyl-bis-triazoles (of which compound CL67 (II) emerged as the lead quadruplex-binding compound targeting the HIF-1 α pathway,¹³ presumed to be via a HIF-1 α promoter quadruplex¹⁴). Separately we have shown that a series of disubstituted phenyl bis-oxazole compounds has selectivity for the (hitherto unreported) promoter quadruplexes in the HSP90 gene.⁸

This series of HSP90 promoter quadruplex-binding compounds have to date shown significant broad-spectrum anti-proliferative activity in a panel of human carcinoma cell lines, which includes the MCF-7 breast cancer line. In addition, a closely-related small library of *meta*-disubstituted phenyl-bis-triazoles (III) has yielded

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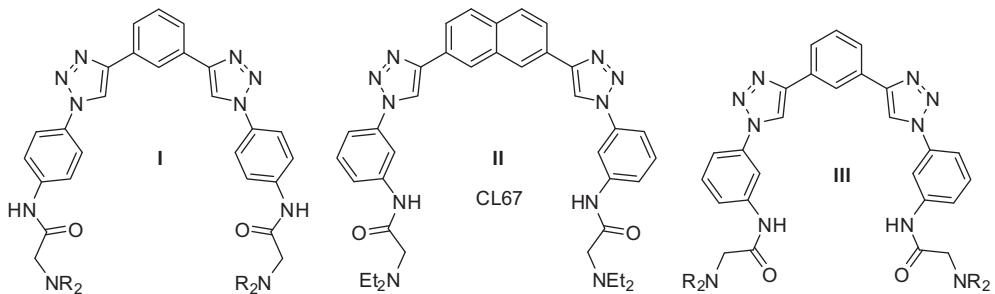
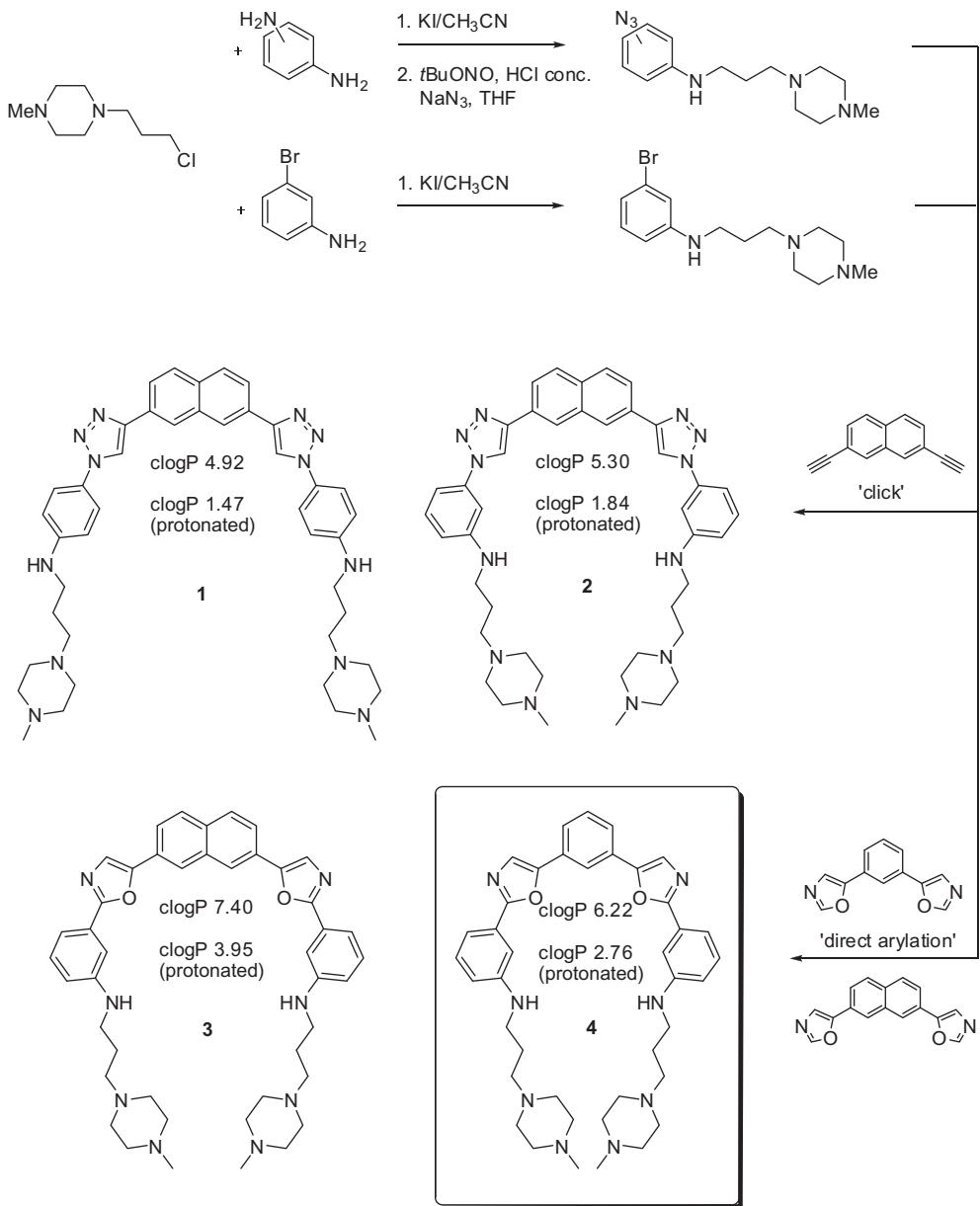


Figure 1. Three bis-triazole frameworks all of which lack low μM in vitro potency against MCF7 breast cancer cells.

no potent lead compound that fulfills the criteria of activity towards MCF7 cells. As part of ongoing optimization studies of compound CL67 (with an IC_{50} value of $>25 \mu\text{M}$ in the MCF-7 breast carcinoma cell line) we have recently discovered that small structural modifications influencing, among other factors, lipophilicity,

can lead to very large changes in biological profile. We describe here the design, synthesis and biological evaluation of this particular class of compounds and have found that a phenotypic screening approach as a tool for SAR optimization is superior to considerations of in vitro quadruplex binding.



Scheme 1. Synthetic pathway for the preparation of final bis-heterocyclic compounds **1–4**.

Our synthetic efforts initially focused on a Cu[I]-catalyzed 1,3-dipolar cycloaddition,¹⁵ which allowed us to gain access to compounds **1** and **2**. This approach was subsequently expanded to include a direct arylation reaction giving rapid access to compounds **3** and **4**. The Scheme highlights the synthetic approach in more detail. Commercially available 1-(3-chloropropyl)-4-methyl-piperazine was reacted in two separate reactions with *meta*-phenylenediamine and 3-bromo-aniline, respectively. The *meta*-halogenated analogue was isolated, purified over silica and directly employed in a direct arylation reaction using pivalic acid and a palladium catalyst to give final compounds **3** and **4**.¹⁶ The azide precursor utilized in the cycloaddition reaction was generated via the standard sodium azide/^tBuONO protocol.¹⁷ Finally, a click reaction was used to generate target compounds **1** and **2**. A number of end-groups and side-chain lengths have been explored, but here the focus is on the *N*-methyl-piperazine substituted compounds in view of earlier studies indicating that this is a favored end-group.¹⁸

A 96 h sulforhodamine B (SRB) assay¹⁹ to evaluate the anti-proliferative activity of **1** showed that IC₅₀ values were above 25 µM for all lines in a small panel of cancer cell lines (Table 2). A second round of cellular evaluation examined the effects of modifying the substitution pattern on the terminal phenyl ring system from *para* to *meta*. This did not enhance potency, as judged by lower IC₅₀ values in the MCF7 and A549 cell lines (compound **2**), but a significant decrease in IC₅₀ value for the two renal cancer cell lines was observed. This is in accord with previous findings for compound CL67,¹³ which has highly selective activity in renal compared to a number of other carcinoma cell lines and suggests that the *meta* substitution may be a critical factor in this selectivity. As the initial rationale for these compounds was to further develop a SAR study on CL67, it was decided to alter the linking heterocyclic moiety from a triazole to an oxazole ring, which would contribute to an overall increase in lipophilicity and possibly also stability in particular, as the oxazoles are aryl-substituted at the 2-position.

Compound **3** was generated using our previously-reported approach⁸ using a direct arylation/C–H activation step to facilitate the late-stage introduction of the polar side-chains (**Scheme 1**). The third round of cellular evaluation using compound **3** showed a significant improvement in anti-proliferative activity (**Table 2**), with IC₅₀ values in all the cell lines of between 1.2 and 2.1 μM, the sole structural difference being two oxazole rings replacing two triazoles. This is a conservative structural change yet results in a sig-

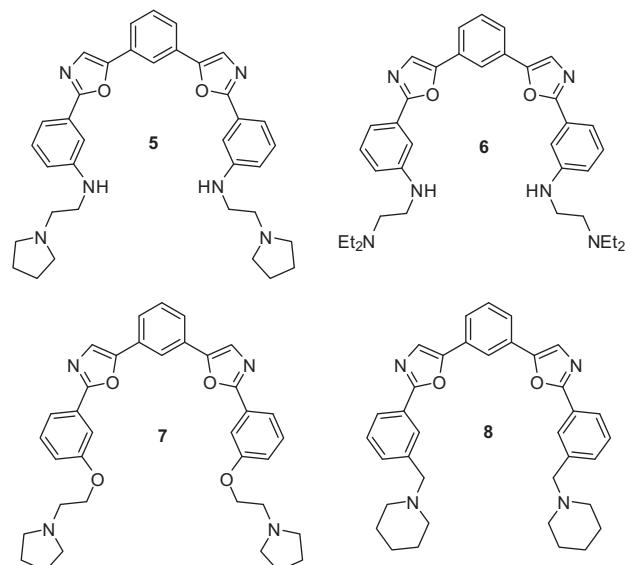
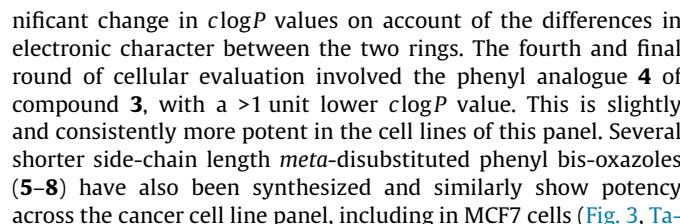


Figure 3. Expanded focused library of *meta*-disubstituted phenyl-bis-oxazoles (5–8) with broad spectrum anti-proliferative activities.

Table 1

ΔT_m Values ($^{\circ}\text{C}$) for FRET analyses of compounds **1–8** at 1 and 5 μM concentrations with a series of quadruplex-forming DNA sequences: F21T, c-KIT2, HSP90A

F21T		c-KIT2		HSP90A		dsDNA		
μM	1	5	1	5	1	5	1	5
1	<2	<10	<2	<10	<2	<10	0	0
2	2.9	<10	2.4	<10	5.7	<10	0	0
3	<2	<10	<2	<10	2.5	<10	0	0
4	<2	20.1	<2	18.3	<2	25.6	0	0
5	<2	24.7	<2	17.0	<2	n/a	0	6.8
6	<2	13.3	<2	13.6	<2	19.5	0	0
7	<2	25.9	<2	20.3	3.4	27.7	0	5.3
8	<2	n/a	<2	n/a	5.0	n/a	2.0	n/a

dsDNA represents duplex DNA.

Esd's are from triplicate measurements and average 0.3 °C.

Table 2

Short-term anti-proliferative activity (IC_{50} values in μM) for compounds **1–8**, in a cancer cell line panel, comprising MCF7 (breast), 786-O/RCC4 (renal), A549 (lung cancer) and WI38 (fibroblast) cell lines, determined using a 96 h SRB assay

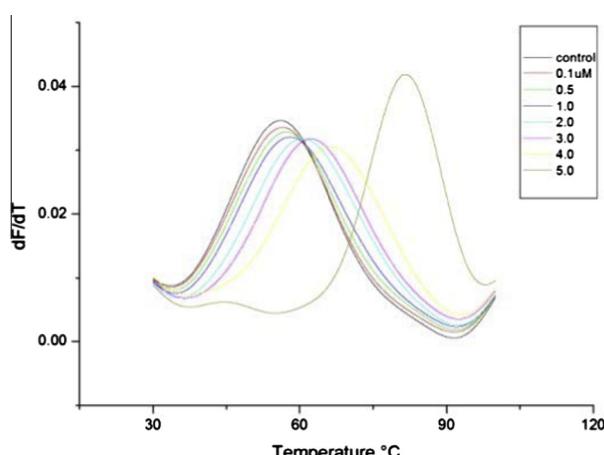


Figure 2. FRET melting curve for compound **4** with the 21-mer Hsp90A promoter quadruplex sequence at different ligand concentrations. There is a significant shift in melting temperature (T_m) for compound **4** at the 5 μ M concentration, whereas there is little shift at the lower concentrations.

Short term IC ₅₀					
	MCF-7	786-O	RCC4	A549	WI38
1	>25	>25	>25	>25	>25
2	>25	8.7 ± 0.1	8.4 ± 0.1	>25	15.3 ± 0.5
3	1.9 ± 0.2	1.2 ± 0.3	2.1 ± 0.1	1.6 ± 0.3	3.1 ± 0.2
4	1.0 ± 0.2	0.9 ± 0.1	1.2 ± 0.2	1.2 ± 0.4	1.1 ± 0.2
5	2.1 ± 0.5	1.1 ± 0.3	1.1 ± 0.1	0.6 ± 0.1	1.0 ± 0.1
6	3.1 ± 0.2	1.3 ± 0.1	1.1 ± 0.1	2.5 ± 0.4	2.8 ± 0.2
7	1.2 ± 0.3	0.7 ± 0.1	1.3 ± 0.1	1.0 ± 0.2	1.0 ± 0.3
8	1.4 ± 0.2	2.3 ± 0.2	1.9 ± 0.3	1.2 ± 0.1	4.4 ± 0.2

Esds are from triplicate (or more) measurements. IC₅₀ values shown as >25 µM are all substantially above this value and were not determined with equivalent accuracy.

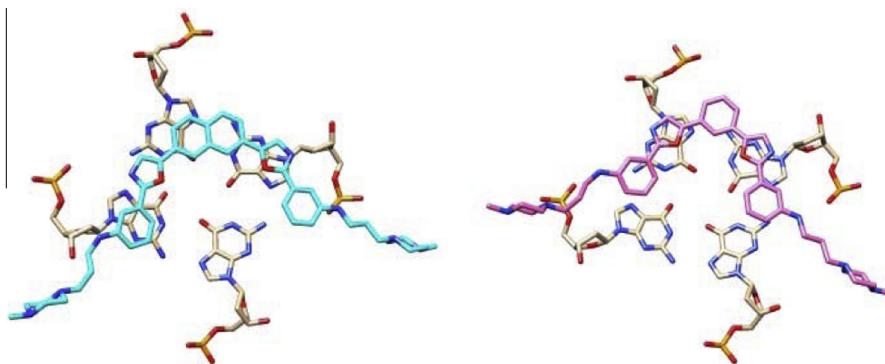


Figure 4. Representative low-energy molecular models for complexes of compounds **3** (left, colored cyan) and **4** (right, colored purple) with the parallel human telomeric quadruplex.²¹ Models were obtained by initial manual optimization based on previous crystal structures of bound complexes, followed by intermolecular energy calculations.

ble 2). This strongly suggests that side chain length is not critical for enhanced cellular activity.

An initial rationale for the development of these and other related ligands has been that they act by interacting with cellular G-quadruplexes such as those in human telomeres or in various promoter sequences of oncogenic genes. We have previously shown that the *para*-substituted phenyl-bis-oxazole framework has selectivity for a quadruplex encoded in the promoter of the HSP90 gene. Data from a FRET assay²⁰ (Table 1) shows that compounds **1–3** do not stabilize a human telomeric quadruplex sequence (F21T). However compound **4** (as well as **5** and **7**, although some duplex DNA binding is evident for these two compounds) does have a significant stabilizing effect on this quadruplex as well as on the HSP90 and c-KIT2 quadruplexes, albeit at higher concentrations (Table 1). Compound **4** does not stabilize double-stranded DNA in the range of concentrations used.

An initial molecular modeling examination of the potential interactions between compounds **3** and **4** and the parallel human telomeric quadruplex crystal structure²¹ has been undertaken (Fig. 4). The parallel topology was chosen since it has been observed in all ligand–telomeric quadruplex crystal structures determined to date.¹¹ The study suggests that there are unlikely to be major differences between the two complexes, although the smaller size of the core groups in compound **4** may result in improved stacking of the two oxazole groups with adjacent guanine rings. This is in accord with the slightly superior stabilization observed in the FRET assays with compounds **4–8** at high ligand concentrations (Fig. 2). We cannot rule out, and at this stage, have not examined, alternative binding modes involving groove interactions, although it is possible that they might explain the large differences in biological responses between compounds **1, 2** and **3, 4**. No relevant experimental data on these alternative binding modes is currently available.

In summary this study has shown that in this series of compounds the determinants of optimal anti-proliferative potency are:

1. The possession of a central phenyl ring.
2. Two attached phenyl-oxazole rings.
3. These should be meta attached to the phenyl rings.
4. The length of the linker and nature of the terminal group is of lesser importance.

It is premature to propose that the cellular mode of action of these compounds involves G-quadruplex stabilization since by comparison previously-reported phenyl-bis-oxazoles and triazoles have shown superior stabilization ability with several quadruplex sequences (HSP90A and the human telomeric quadruplexes) yet with comparable anti-proliferative abilities. Few of the bis-triazole and oxazole derivatives presented here had any significant

quadruplex stabilization in FRET assays. Only at high concentrations (5 μM of compounds **4–7**) was there significant stabilization in the FRET assay (accurate curve fitting for compound **8** was not possible at 5 μM, so no FRET data is available at this concentration). This suggests that either the compounds act at other quadruplex loci that are yet to be determined, or that the compounds act by non-quadruplex mechanisms. Regardless of quadruplex target, we suggest that cell-based phenotypic screening can speedily optimize anti-proliferative activity in these series. This may be an useful approach to obtaining poly-targeted quadruplex-binding compounds with high biological activity and good pharmacological profiles. The compounds presented here have some structural resemblance to kinase inhibitors, although there is no indication that they interact with kinases—the related compound CL67 shows no activity in a large kinase panel (Welsh et al., unpublished observations).

Acknowledgments

S.A.O. and C.C. contributed equally to this work. We are grateful to Cancer Research UK (Programme Grant No. C129/A4489 to S.N.). The Novartis Institute of BioMedical Research, Boston, is thanked for obtaining CL67 data in their kinase panel.

References and notes

1. (a) Patel, D. J.; Phan, A. T.; Kuryavyi, V. *Nucleic Acids Res.* **2007**, *35*, 7429; (b) Burge, S.; Parkinson, G. N.; Hazel, P.; Todd, A. K.; Neidle, S. *Nucleic Acids Res.* **2006**, *34*, 5402; (c) Neidle, S. *Curr. Opin. Struct. Biol.* **2009**, *19*, 1; (d) Oganesian, L.; Bryan, T. M. *BioEssays* **2007**, *25*, 155.
2. (a) Huppert, J. L.; Balasubramanian, S. *Nucleic Acids Res.* **2005**, *33*, 2908; (b) Todd, A. K.; Johnston, M.; Neidle, S. *Nucleic Acids Res.* **2005**, *33*, 2901.
3. (a) Huppert, J. L.; Balasubramanian, S. *Nucleic Acids Res.* **2007**, *35*, 406; (b) Siddiqui-Jain, A.; Grand, C. L.; Bearss, D. J.; Hurley, L. H. *Proc. Natl. Acad. Sci. U.S.A.* **2002**, *99*, 11593; (c) Balasubramanian, S.; Hurley, L. H.; Neidle, S. *Nat. Rev. Drug Disc.* **2011**, *10*, 261.
4. See for example: (a) Sun, D.; Thompson, B.; Cathers, B. E.; Salazar, M.; Kerwin, S. M.; Trent, J. O.; Jenkins, T. C.; Neidle, S.; Hurley, L. H. *J. Med. Chem.* **1997**, *40*, 2113; (b) Monchaud, D.; Teulade-Fichou, M.-P. *Org. Biomol. Chem.* **2008**, *6*, 627; (c) Ou, T.-M.; Lu, Y.-J.; Tan, J.-H.; Huang, Z.-S.; Wong, K.-Y.; Gu, L.-Q. *ChemMedChem* **2008**, *3*, 690; (d) Franceschin, M. *Eur. J. Org. Chem.* **2009**, 2225; (e) Yang, D. Z.; Okamoto, K. *Future Med. Chem.* **2010**, *2*, 619; (f) Lee, H.-M.; Chan, D. S.-H.; Yang, F.; Yan, S.-C.; Che, C.-M.; Ma, D.-L.; Leung, C.-H. *Chem. Commun.* **2010**, 4680; (g) Neidle, S. *Therapeutic Aspects of Quadruplex Nucleic Acids*; Academic Press: San Diego, 2012.
5. Biffi, C.; Tannahill, D.; McCafferty, J.; Balasubramanian, S. *Nat. Chem.* **2013**, *5*, 182–186.
6. (a) Ma, Y.; Ou, T.-M.; Tan, J.-H.; Hou, J.-Q.; Huang, S.-L.; Gu, L.-Q.; Huang, Z.-S. *Eur. J. Med. Chem.* **2011**, *46*, 1906; (b) Boddupally, P. V.; Hahn, S.; Beman, C.; De, B.; Brooks, T. A.; Gokhale, V.; Hurley, L. H. *J. Med. Chem.* **2012**, *55*, 6076.
7. (a) Gunaratnam, M.; Swank, S.; Haider, S. M.; Galeska, K.; Reszka, A. P.; Beltran, M.; Cuenga, F.; Fletcher, J. A.; Neidle, S. *J. Med. Chem.* **2009**, *52*, 3774; (b) Bejugam, M.; Sewitz, S.; Shirude, P. S.; Rodriguez, R.; Shahid, R.; Balasubramanian, S. *J. Am. Chem. Soc.* **2007**, *129*, 12926.
8. Ohnmacht, S. A.; Micco, M.; Petrucci, V.; Todd, A. K.; Reszka, A. P.; Gunaratnam, M.; Carvalho, M. A.; Zloh, M.; Neidle, S. *Bioorg. Med. Chem. Lett.* **2012**, *22*, 5930.

9. (a) Phan, A. T.; Kuryavyi, V.; Burge, S.; Neidle, S.; Patel, D. J. *J. Am. Chem. Soc.* **2007**, *129*, 4386; (b) Hsu, S. T.; Varnai, P.; Bugaut, A.; Reszka, A. P.; Neidle, S.; Balasubramanian, S. *J. Am. Chem. Soc.* **2009**, *131*, 13399; (c) Wei, D.; Parkinson, G. N.; Reszka, A. P.; Neidle, S. *Nucleic Acids Res.* **2012**, *40*, 4691; (d) Todd, A. K.; Haider, S. M.; Parkinson, G. N.; Neidle, S. *Nucleic Acids Res.* **2007**, *35*, 5799.
10. Yu, H.; Gu, X.; Nakano, S. I.; Miyoshi, D.; Sugimoto, N. *J. Am. Chem. Soc.* **2012**, *134*, 20060.
11. Micco, M.; Collie, G. W.; Dale, A. G.; Ohnmacht, S. A.; Pazitna, I.; Gunaratnam, M.; Reszka, A. P.; Neidle, S. *J. Med. Chem.* **2013**, *56*, 2959.
12. Dai, J.; Carver, M.; Hurley, L. H.; Yang, D. *J. Am. Chem. Soc.* **2011**, *133*, 17673.
13. Lombardo, C. M.; Welsh, S. J.; Strauss, S. J.; Dale, A. G.; Todd, A. K.; Nanjunda, R.; Wilson, W. D.; Neidle, S. *Bioorg. Med. Chem. Lett.* **2012**, *22*, 5984.
14. De Armond, R.; Wood, S.; Sun, D.; Hurley, L. H.; Ebbinghaus, S. W. *Biochemistry* **2005**, *44*, 16341.
15. Moorhouse, A. D.; Santos, A. M.; Gunaratnam, M.; Moore, M.; Neidle, S.; Moses, J. E. *J. Am. Chem. Soc.* **2006**, *128*, 15972.
16. Strotman, N. A.; Chobanian, H. R.; Gou, Y.; He, J.; Wilson, J. E. *Org. Lett.* **2010**, *16*, 3578.
17. Direct arylation: 3,3'-(5,5'-(1,3-phenylene)bis(oxazole-5,2-diyl))bis(N-(3-(4-methylpiperazine-1-yl)propyl)aniline) (**3**, **4**). An oven dried microwave vial was charged with Pd(OAc)₂ (0.1 equiv), K₂CO₃ (3 equiv), ligand (2-dicyclohexylphosphino-2',6'-diisopropoxybiphenyl, 0.2 equiv), oxazole precursor (1 equiv) and arylbromide (2.2 equiv). The microwave vial was capped, evacuated and backfilled with argon and then pivalic acid (0.4 equiv) and toluene were added. The vial was heated at 110 °C for 72 h. After cooling to room temperature, the reaction mixture was concentrated and purified using silica gel. (400 MHz, CDCl₃) δ 1.26 (4H, s, 2 × CH₂), 1.65 (4H, br s, 2 × CH₂), 1.85 (4H, quintet, *J* = 6.3 Hz, 2 × CH₂), 2.31 (6H, s, 2 × CH₃), 2.51–2.55 (12H, m, 6 × CH₂), 3.30 (4H, t, *J* = 6.3 Hz, 2 × CH₂), 6.72 (2H, dd, *J* = 1.6, 8.1 Hz, 2 × Ar-H), 7.29 (2H, t, *J* = 7.8 Hz, 2 × Ar-H), 7.35 (2H, t, *J* = 1.6 Hz, 2 × Ar-H), 7.46 (2H, d, *J* = 7.8 Hz, 2 × Ar-H), 7.49–7.53 (2H, m, 2 × Ar-H), 7.69 (2H, dd, *J* = 1.5, 7.8 Hz, 2 × Ar-H), 8.01 (1H, br s, Ar-H). ¹³C NMR (100 MHz, CDCl₃) δ 162.1 (2 × quat), 150.4 (CH), 149.1 (2 × CH), 129.7 (2 × CH), 129.6 (2 × CH), 129.0 (2 × CH), 128.1 (2 × CH), 124.1 (2 × quat), 124.0 (2 × quat), 119.7 (2 × quat), 115.3 (CH), 115.2 (2 × CH), 109.8 (2 × CH), 57.2 (2 × CH₂), 55.3 (4 × CH₂), 53.2 (4 × CH₂), 46.0 (2 × CH₂), 43.5 (2 × CH₂), 25.6 (2 × CH₃). HRMS (ES⁺) calculated for C₄₀H₅₀N₈O₂ [M+1]⁺ 675.4135, found 675.4125.
18. Click reaction: 3,3'-(4,4'-(naphthalene-2,7-diyl)bis(1H-1,2,3-triazole-4,1-diyl))bis(N-(3-(4-methylpiperazin-1-yl)-propyl) aniline) (**1**, **2**). Dialkyne (1 equiv) was dissolved in the appropriate volume of solvent (H₂O:BuOH 1:1), followed by the addition of required azide (3 equiv) and the catalytic mixture of CuSO₄ × 5H₂O (0.05 equiv), sodium ascorbate (0.5 equiv) and bathophenanthroline disulfonic acid disodium salt hydrate (0.1 equiv). The reaction was performed with excess of the required amine (6 equiv). The mixture was heated under microwave irradiation for 15 min at 120 °C. Crude product was obtained following evaporation of solvents and subsequent C18 reversed phase semi-prep purification. ¹H NMR (400 MHz, CDCl₃) δ 1.81 (4H, quintet, *J* = 6.6 Hz, 2 × CH₂), 2.45 (4H, t, *J* = 6.6 Hz, 2 × CH₂), 2.58 (6H, s, 2 × CH₃), 2.67 (8H, br s, 4 × CH₂), 2.94 (8H, br s, 4 × CH₂), 4.16 (4H, t, *J* = 7.3 Hz, 2 × CH₂), 7.64–7.68 (2H, m, 2 × Ar-H), 7.75–7.77 (4H, m, 4 × Ar-H), 7.93–8.03 (4H, m, 4 × Ar-H), 8.19 (2H, t, *J* = 1.8 Hz, 2 × Ar-H), 8.46–8.48 (4H, m, 4 × Ar-H). ¹³C NMR (100 MHz, CDCl₃) δ 148.2 (2 × CH), 143.2 (2 × CH), 138.0 (2 × CH), 133.6 (quat), 133.3 (quat), 131.0 (2 × CH), 128.8 (2 × quat), 128.1 (2 × quat), 125.0 (2 × CH), 124.3 (2 × quat), 118.6 (2 × quat), 118.2 (2 × CH), 118.1 (2 × CH), 110.7 (2 × CH), 54.4 (2 × CH₂), 53.3 (4 × CH₂), 50.7 (4 × CH₂), 43.8 (2 × CH₂), 41.7 (2 × CH₂), 23.3 (2 × CH₃). HRMS (ES⁺) calculated for C₄₂H₅₂N₁₂ [M+1]⁺ 725.4516, found 725.4482.
19. Sulforhodamine B assay (SRB). Cells were counted and diluted to the required concentration in 20 mL medium. For cell lines MCF7, A549, RCC4 and 786-O, 1000–4000 cells with 160 µL media (WI38: 6000/well) were seeded into each well of a 96 well plate (Nunc, Denmark). After incubation for 24 h, the compound to be tested was dissolved in 40 µL of medium and was added in a range of concentrations, and the cells incubated for 96 h. The medium was then removed and the cells fixed by incubation with TCA (10%, Sigma-Aldrich, UK) for 30 min at 4 °C. After removal of the TCA, the cells were washed with deionised water five times and dried at 60 °C for 1 h. The cells were then incubated with sulforhodamine B (80 µL, 0.4% in 1% acetic acid, Acros Organics, UK) for 15 min at rt. The SRB was removed, the wells washed with 1% acetic acid (200 µL), and dried at 60 °C for 1 h. Tris-base (100 µL, 10 mM, Acros Organics, UK) solution was added to each well, and the plates were gently shaken for 5 min. The absorbance at 540 nm was measured with a plate reader (Spectrostar Omega, BMG Labtech, Germany). The data were normalized to the value of 100 for the control experiment (untreated cells), and the IC₅₀ values were obtained by interpolation from a plot with Origin (Version 7.0, OriginLab Corp.), as the concentration leading to an absorbance intensity of 50%.
20. FRET DNA melting assays on compounds **1**–**8** were performed as previously described (Guyen, B. Schultes, C. M. Hazel, P. Mann, J. Neidle, S. *Org. Biomol. Chem.* **2004**, *2*, 981) using a fluorescence resonance energy transfer (FRET) assay modified as a high-throughput screen in a 96-well format. The labelled oligonucleotides had attached the donor fluorophore FAM: 6-carboxyfluorescein and the acceptor fluorophore TAMRA: 6-carboxytetramethyl-rhodamine. The FRET probe sequences were diluted from stock to the correct concentration (400 nM) in a 60 mM potassium cacodylate buffer (pH 7.4) and then annealed by heating to 95 °C for 10 min, followed by cooling to room temperature in the heating block (3–3.5 h). The compounds were stored as a 1 mM stock solution in 10% DMSO/90% 1 mM HCl; final solutions (at 2 × concentration) were prepared using 60 mM potassium cacodylate buffer (pH 7.4). Relevant controls using BRACO-19 (in addition to blank runs) were also performed to check for quality of DNA samples (e.g., F21T). 96-Well plates (MJ Research, Waltham, MA) were prepared by aliquoting 50 µL of the annealed DNA into each well, followed by 50 µL of the compound solutions. Measurements were made on a DNA Engine Opticon (MJ Research) with excitation at 450–495 nm and detection at 515–545 nm. Fluorescence readings were taken at intervals of 0.5 °C in the range 30–100 °C, with a constant temperature being maintained for 30 s prior to each reading to ensure a stable value. Final analysis of the data was carried out using a script written in the program Origin 7.0 (OriginLab Corp., Northampton, MA). The advanced curve-fitting function in Origin 7.0 was used for calculation of ΔT_m values. Esds in ΔT_m are ±0.1 °C.
21. Parkinson, G. N.; Lee, M. P. H.; Neidle, S. *Nature* **2002**, *417*, 876.