

## Manuscript Details

<b>Manuscript number</b>	MICROC_2019_2385
<b>Title</b>	A multi-analytical approach for the characterization of black crusts on the facade of an historical Cathedral
<b>Article type</b>	News
<b>Keywords</b>	Black crusts, marble stone, cultural heritage, conservation, black carbonaceous particles, urban air pollution
<b>Manuscript category</b>	General
<b>Corresponding Author</b>	Valeria Comite
<b>Corresponding Author's Institution</b>	Università degli Studi di Milano
<b>Order of Authors</b>	Valeria Comite, Jose Santiago Pozo-Antonio, Carolina Cardell, Luciana Randazzo, Mauro La Russa, PAOLA FERMO

## Submission Files Included in this PDF

### File Name [File Type]

Dear Editor.docx [Cover Letter]

Manuscript.docx [Manuscript File]

Fig. 1.tif [Figure]

Fig. 2.tif [Figure]

Fig. 3.tif [Figure]

Fig. 4.tif [Figure]

CAPTION.docx [Figure]

Table1.docx [Table]

Table 2.docx [Table]

Table 3.docx [Table]

Table 4.docx [Table]

Table 5.docx [Table]

Supplementary material.pptx [Supplementary Material]

To view all the submission files, including those not included in the PDF, click on the manuscript title on your EVISE Homepage, then click 'Download zip file'.

## Research Data Related to this Submission

There are no linked research data sets for this submission. The following reason is given:  
Data will be made available on request

Dear Editor,

This research study is focused on the characterization of black crusts collected from the Dome of Monza located in the homonymous city (N Italy), a hot spot from the point of view of the atmospheric pollution. These samples have been studied for the first time and the work fits into a broader context based on the analysis of black crusts on stone monuments in urban areas affected by high pollution. The aim was to study the black crusts and substrate specimens by a multi-analytical approach including the analysis of the carbonaceous fraction with an innovative methodology, X-ray Diffraction (XRD), FT-IR spectroscopy, Ion Chromatography (IC), Stereomicroscopy (SM), Polarized Light Microscopy (PLM), High Resolution Scanning Electron Microscopy coupled with Energy Dispersive X-ray spectroscopy (HRSEM-EDX) and Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP/MS).

This integrated approach has allowed identifying the sources of pollution responsible for the decay of the building carbonatic materials of the Cathedral.

This Cathedral represents an interesting case study for the scientific community since to date the process of black crusts formation on natural stones has not yet been understood in depth.

On behalf of the coauthors  
**Valeria Comite**

# A multi-analytical approach for the characterization of black crusts on the facade of an historical Cathedral

Valeria Comite<sup>a</sup>, Jose Santiago Pozo-Antonio<sup>b</sup>, Carolina Cardell<sup>c</sup>, Luciana Randazzo<sup>d</sup>, Mauro Francesco

La Russa<sup>d</sup>, Paola Fermo<sup>a</sup>

<sup>a</sup> *Dipartimento di Chimica, Università degli Studi di Milano, Via Golgi 19, Milan, Italy;*

[valeria.comite@unimi.it](mailto:valeria.comite@unimi.it) , [paola.fermo@unimi.it](mailto:paola.fermo@unimi.it)

<sup>b</sup> *Departamento de Enxeñaría de Recursos Naturais e Medio Ambiente, Escola de Enxeñaría de Minas e Enerxía, University of Vigo, Campus Lagoas-Marcosende s/n, 36310 Vigo, Spain;*

[ipozo@uvigo.es](mailto:ipozo@uvigo.es)

<sup>c</sup> *Department of Mineralogy and Petrology, Faculty of Science, University of Granada, Av. Fuentenueva S/N, 18002 Granada, Spain;* [cardell@ugr.es](mailto:cardell@ugr.es)

<sup>d</sup> *Dipartimento di Biologia, Ecologia e Scienze della Terra (DiBEST), Università della Calabria, Via Pietro Bucci, 87036 Arcavacata di Rende, CS, Italy;* [luciana.randazzo@unical.it](mailto:luciana.randazzo@unical.it) , [mlarussa@unical.it](mailto:mlarussa@unical.it)

## Abstract

This study focuses on the characterization of black crusts collected from the Dome of Monza located in the homonymous city (N Italy), a hot spot from the point of view of the atmospheric pollution. Black crusts and substrate specimens were analysed by a multi-analytical approach including X-ray Diffraction (XRD), FT-IR spectroscopy, Ion Chromatography (IC), Stereomicroscopy (SM), Polarized Light Microscopy (PLM), High Resolution Scanning Electron Microscopy coupled with Energy Dispersive X-ray spectroscopy (HRSEM-EDX) and Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP/MS). The characterization of the carbon fraction (organic carbon, OC, and elemental carbon, EC) was performed using a new approach based on the use of Carbon Hydrogen Nitrogen (CHN) analysis and Thermogravimetric Analysis (TGA). The integrated approach allowed the identification pollution sources responsible for the black crusts forming process. The precise identification of the main substances responsible for the surface degradation phenomena, in particular those leading to the blackening and disintegration of the carbonate substrates, is essential for the definition of conservative intervention and maintenance strategies, as well as for the development of emission reduction policies on a local scale.

**Keywords:** Black crusts; marble stone; cultural heritage; conservation; black carbonaceous particles, urban air pollution

## 1. Introduction

Atmospheric pollution (such as gases and aerosol particulate matter) interacts with stone materials and generates various typologies of decay [1-5]. Among those we can list, for instance, salt crystallization that can lead to physical stress [6-8] or soiling that may evolve to black crusts when in presence of humidity and absence of water washout processes [9-11]. The formation of black crusts is one of the most dangerous phenomenon in architectural heritage [12-14]. Currently, emissions from mobile combustion sources [15-17] are the main agents responsible for air pollution in cities, although a significant decrease is expected in the next years, due to reinforced pollution prevention law and policies. Especially the monuments located in the historic centres of large cities are subjected to typical anthropogenic emissions. Sulphur dioxide, despite its decreasing concentration emitted in the atmosphere, is the main specie responsible for sulphation process that occurs on the stone [18-25]. Up to now, black crusts have been the subject of numerous studies [13,26-28] which highlighted the importance of investigating this topic in order to study physical-chemical decay of stone surfaces [29-34]. Black crusts formation is not only a form of alteration of carbonate stones but also of other types of rock such as granite [25,35-38].

In this work the results obtained from the analysis of black crusts taken from a historical cathedral, Dome of Monza (dating to 15<sup>th</sup> century), placed in a polluted urban centre in the North of Italy, will be discussed. The cathedral is located in the homonymous city placed ca. 20 km NE of Milan. Monza covers an area of 33.09 km<sup>2</sup> and has 123.776 inhabitants and is the third largest city in the Lombardy region (Northern Italy). The city suffers from high pollution produced by the intense vehicular traffic, the high use of domestic heating, as well as the pollution produced by industrial and agricultural activities typical of the Po Valley. In Monza NO<sub>x</sub> emissions are due mainly to traffic (69%) but are also due to domestic heating (12%), industries (12%) and incinerators (3%) while anthropogenic VOC emissions are caused by the use of solvents (69%) and by traffic (9%). NH<sub>3</sub> emissions on a regional scale are due to agriculture and breeding activities. The percentage values listed are higher than those recorded for the city of Milan. Moreover, in spite of SO<sub>2</sub> reduction recorded in atmosphere emissions during the last years (for the city of Monza this pollutant is more attributable to industrial activity (82%) (INEMAR, INventario EMISSIONI Aria, ARPA Lombardia 2014), the presence of SO<sub>2</sub> still dominates the chemical composition of the deposits in urban environments

Black crusts sampled from the Dome façade studied in the present work, have developed on the different marbles used for the monument construction. In order to completely characterize the samples, several analytical techniques were used such as: XRD, FT-IR/ATR, SM, MLP, HRSEM-

119  
120  
121 EDX, IC; LA-ICP/MS, CHN and TGA. This integrated approach allowed us to gain information  
122  
123 about the composition (chemical and mineralogical) and texture of the crusts making it possible to  
124  
125 identify the pollution sources causing the stone decay, as well as the variability in BC composition  
126  
127 depending both on the exposure conditions of the analysed crust surfaces and their age (older crusts  
128  
129 are present on original marbles and crusts of more recent formation have evolved on integration  
130  
131 marbles used during the cathedral restoration).

## 132 **2. Materials and analytical methods**

### 133 **2.1 Dome of Monza**

134  
135  
136 The samples analysed in this work were taken from the facade of the Dome of Monza (Basilica of  
137  
138 San Giovanni Battista) located in Monza in the homonym square (Fig. S1 in Supplementary material).  
139  
140 The façade, a remarkable example of 15<sup>th</sup>-century architecture, and divided into three parts, was  
141  
142 designed and finished by Matteo da Campione between 1300-1350; it is characterized by alternating  
143  
144 dark and white-colored rows of stone blocks. From the 18<sup>th</sup> to the 20<sup>th</sup> centuries several and complex  
145  
146 restoration interventions followed that involved some areas of the façade [39]. The dark and white-  
147  
148 colored degraded rows were replaced with new ones, all the missing columns of white marble  
149  
150 integrated, and the rose window was consolidated. Furthermore, some elements of the three-light  
151  
152 windows and the quadrangular compartments on the black background of the rosettes in the central  
153  
154 area were restored. In the final phase, a cleaning restoration was performed in the lower central and  
155  
156 upper part of the façade. In 2017 the façade was restored once again.

### 157 **2.2 Sampling**

158  
159 During the last restoration carried out in 2017, 9 samples of black crusts developed on marble stones,  
160  
161 were taken from the Dome façade at different heights (Fig. S2) following the indications of the  
162  
163 restorers. The samples show variable color, morphology and thickness, depending on the exposure  
164  
165 position and on textural features of the underlying substrate.

166  
167 Two groups of samples were identified, group A and group B (Fig. 2S), according to the information  
168  
169 available on restoration interventions in the specific areas from where they were taken:

170  
171 - group A is formed by samples 1MD, 3MS and 4MS located at ca. 16m high, and taken from the  
172  
173 area restored by Conca (original marbles replaced in 1735); therefore, these crusts are about 280  
174  
175 years old, i.e. a pollutants accumulation of about 280 years was present;

176  
177 -group B is formed by samples taken from a height of ca. 12m, i.e. 5MC, 6MD, 7MD, 8MD and  
178  
179 9MD, and samples 11MD and 12MD taken at ca. 5 m height (lower part of the façade) in areas that

178  
179  
180 have never been restored (Fig. S2), therefore, these crusts are about 650 years old, i.e. 650 years of  
181 accumulation.  
182

183 A brief sample description accompanied by information regarding sampling point, height and  
184 information on eventual previous restorations is summarized in Table 1.  
185  
186

## 187 188 **2.3 Analytical methods**

### 189 190 2.3.1. Stereomicroscope (SM)

191  
192 A stereomicroscope (SMZ 1000, Nikon, Japan) equipped for microphotography was used to look at  
193 the chromatic and textural features, as well as the conservation state of the BCs and their marble  
194 substrates.  
195  
196  
197

### 198 199 2.3.2. Polarized Light Microscopy (PLM)

200  
201 Mineralogical composition, texture, microstructure and conservation state of the BCs and the marble  
202 substrates were studied by Polarized Light Microscopy (PLM) in transmitted and reflected light using  
203 a Carl Zeiss Jenapol U instrument (Germany) with a digital camera (Nikon D-7000). To this end  
204 samples were prepared as polished thin sections.  
205  
206  
207

### 208 209 2.3.3 X-ray diffraction (XRD)

210  
211 Mineralogical composition of the BCs was obtained by means of XRD (Siemens D5000) according  
212 to the random powder method. In the diffractograms obtained, relative abundance (semi-quantitative  
213 estimation) was determined for each mineral phase using the area of highest-intensity diffraction  
214 peaks and the intensity ratios established from artificial mixtures of standard minerals [40].  
215  
216  
217

### 218 219 2.3.4 Fourier transform infrared spectroscopy (FT-IR)

220  
221 Chemical composition of the BCs was characterized using FT-IR spectroscopy (Thermo Nicolet  
222 6700) in Attenuated Total Reflectance (ATR) mode. Functional groups in the absorbance spectra for  
223 each sample were identified.  
224  
225

### 226 227 2.3.5 High Resolution Scanning Electron Microscopy coupled with Energy Dispersive X-ray 228 spectroscopy (HRSEM-EDX)

229  
230 The micro-texture and chemical composition of the BCs and marbles substrates were analyzed using  
231 a HRSEM-EDX. The instrument was a Supra 40Vp Carl Zeiss (Germany) furnished with BSE  
232 (backscattered electrons) and SE (secondary electrons) detectors (InLens) that deliver chemical and  
233  
234  
235  
236

237  
238  
239 morphological images respectively, as well as a microanalysis system (Aztec 3) to provide elemental  
240 analyses by EDX (X-Max 50 mm detector). Bulk samples and polished thin sections were mounted  
241 on Al stubs with double-sided adhesive C tape and carbon coated. Samples were analyzed under  
242 high vacuum level. Single-point analyses were acquired at 10 kV (bulk samples) and 20 kV beam  
243 energy (thin sections). High-resolution X-ray maps (1024 x 768 pixels, 10 ms dwell time, 2.5 h  
244 acquisition, 20 eV/ch resolution, and 43 to 1000 frames) were obtained from selected areas (thin  
245 sections) by compiling elemental distribution maps through the use of the *Find Phases* tool  
246 implemented in the Aztec 3 EDX. This allowed us to highlight the amount, location and morphology  
247 of crystalline/amorphous phases present in the samples.  
248  
249  
250  
251  
252  
253  
254

### 255 2.3.6 Ion chromatography (IC)

256  
257 Ion Chromatography has been employed to quantify the main ions present in the BCs. Three  
258 milligrams of powder, withdrawn from the sample surface, were placed in a test tube and treated  
259 with 3mL of MilliQ water. The test tubes were put in an ultrasonic bath for 1 h, then the solutions  
260 centrifuged and injected for IC analyses by means of an auto-sampler. Measurements of cationic  
261 ( $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$  and  $\text{NH}_4^+$ ) and anionic ( $\text{NO}_2^-$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Cl}^-$ ) species were carried out by  
262 using an ICS-1000 HPLC system equipped with a conductivity system detector. More details on the  
263 analytical procedure are reported elsewhere [28,41-42].  
264  
265  
266  
267  
268  
269

### 270 2.3.7 Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP-MS)

271 BCs as well chemical analyses in terms of trace elements were performed by LA-ICP-MS. This  
272 method allows to investigate a great number of elements with spot resolutions of about 40–50  $\mu\text{m}$ ,  
273 also allowing the determination of micrometric compositional variations [43-45]. Analyses were  
274 carried out using an Elan DRCE instrument (Perkin Elmer/SCIEX) connected to a New Wave UP213  
275 solid-state Nd-YAG laser probe (213nm). The analytical procedure to characterize the BCs is well-  
276 established in varied studies in the literature [27, 46-48].  
277  
278  
279  
280  
281

### 282 2.3.8 Carbonaceous fraction analysis

283 For quantifying the carbonaceous components present in the BCs, CHN analyses were performed  
284 by a CHN analyser (CHNS/O Perkin Elmer 2400 Series II Elemental Analyzer using an accessory  
285 for the analysis of solids). TG analyses were carried out by a Mettler Toledo TGA/DSC 3+  
286 instrument which allows simultaneous TG and DSC (Differential Scanning Calorimetry) analyses.  
287 The analyses were conducted in the range 30°- 800° C, increasing the temperature with a rate of  
288 20° C/minute. The carbonaceous components were estimated in temperature ranges defined by  
289  
290  
291  
292  
293  
294  
295

296  
297  
298 previously studied standards and using two different atmospheres, i.e. the inert and the oxidant  
299  
300 one.  
301  
302

### 303 **3 Results**

#### 304 **3.1 SM, PLM, XRD and FT-IR analysis**

305  
306  
307 The stereomicroscope (SM) study revealed that all BCs samples exhibit intense black color on surface  
308 (except 1MD), though their thickness and surface features vary depending on location and height of  
309 sampling in the building. For instance, 12MD sample taken at 5m high is made of 3 layers as clearly  
310 seen in Fig. S3a in supplementary material (indicated in the red rectangle); the one in contact with  
311 the substrate is orange-brown, on top there is a white layer and above a black layer). Some crusts  
312 clearly show particles of different color (white, red and grey) on their surface such as 7MD, 1MD and  
313 11MD samples, these two last displaying uneven surface continuity (Fig. S3 b,c,d).  
314  
315

316  
317  
318 The PLM study confirmed that all crust substrates are marbles, however exhibiting diverse  
319 petrographic characteristics in that marbles are of different types. Fig. 1 shows the marked  
320 microstructural differences in terms of crystal size, shape and grain boundaries. Note that all marbles  
321 are physically degraded presenting varied damage degree. 1MD and 12MD marbles exhibit  
322 granoblastic-heteroblastic fabric and mosaic type distribution of polygonal cleavage crystals forming  
323 triple junctions at 120°. 1MD displays the largest crystal size ranging from ca. 0.2 to 1.4mm (Fig. 1a),  
324 while crystals size range from ca. 0.2 to 0.5mm in 12 MD (Fig. 1b). Fissures and cracks are abundant  
325 in both marbles. 4MS marble exhibits heteroblastic fabric. Crystals have sub-angular to rounded  
326 shapes and lobate boundaries with sizes ranging from ca. 50µm to 0.3mm (Fig. 1c). Note the intense  
327 cracking of the substrate and the calcite crystals from the marble substrate embedded in the gypsum  
328 matrix of the BC (Fig. 1d). The highly degraded 7MD marble (Fig. 1e) has a microstructure made of  
329 rounded crystals with a bimodal size distribution (<40 µm to 0.3mm). A porfido-granoblastic fabric  
330 characterizes 9MS marble, composed of non-equidimensional rounded shaped-grains ranging in size  
331 from few microns to 0.4mm (Fig. 1f). Marble is severely degraded as to develop granular  
332 disaggregation.  
333  
334

335  
336  
337 Gypsum ( $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ ) was detected by XRD in all BCs samples as the main mineral phase (Table  
338 2). Moreover, quartz ( $\text{SiO}_2$ ) was detected in all BCs in different amounts, except in 4MS. Other  
339 minerals were detected in low amounts: i) weddellite ( $\text{CaC}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ ) was identified in 4MS, 5MC  
340 and 9MS (mainly in the latter), and ii) calcite ( $\text{CaCO}_3$ ) in 11 MD.  
341  
342

343  
344  
345 FT-IR analysis confirmed the XRD results; gypsum is the main mineral phase in BCs (Fig. 2). The  
346 presence of this mineral is confirmed through the bands assigned to 1) OH functional group related  
347  
348  
349  
350  
351  
352  
353  
354



355  
356  
357 to the hydration state of the mineral and 2) the S-O vibrations. In the first case, the characteristic OH-  
358 (H<sub>2</sub>O) bands were confirmed by means of the intense doublet at 3520/ 3401 cm<sup>-1</sup>, a slight shoulder at  
359 3244 cm<sup>-1</sup>, two weak bands at 2640 and 2510 cm<sup>-1</sup> and medium intense peaks at 1682/1618 cm<sup>-1</sup>.  
360  
361 Regarding S-O vibration, the sharpened band centred at 1102 cm<sup>-1</sup> assigned to S-O stretching and the  
362 lower intense bands at 667, 596 and 460 cm<sup>-1</sup> corresponding to S-O bending vibration [49-52]. The  
363 detection of quartz by XRD is confirmed by FT-IR through a slight shoulder at 1004 cm<sup>-1</sup>, typically  
364 assigned to the asymmetrical stretching vibration of Si-O of quartz [79,53]. With respect to calcium  
365 oxalates, note that the discrimination of weddellite and whewellite is made from the peaks assigned to  
366 the symmetric O-C-O stretching vibrations at 1317 cm<sup>-1</sup> and 1324 cm<sup>-1</sup> for whewellite and weddellite  
367 respectively. In addition, the vibration corresponding to H-O-H vibration of lattice water is detected  
368 at 1630 cm<sup>-1</sup> and 781 cm<sup>-1</sup> for both oxalates and at 669 cm<sup>-1</sup> and at 603 cm<sup>-1</sup> for whewellite and  
369 weddellite respectively [46,54]. However, confirmation of the presence of calcium oxalate was only  
370 possible in some samples due to two difficulties: 1) the coincidence between the oxalate FT-IR bands  
371 between 600 and 700 cm<sup>-1</sup> with those of gypsum (note that gypsum is the main mineral of all the  
372 crusts); and 2) the detection in all samples of a broad shoulder-shaped band around 1370-1400 cm<sup>-1</sup>  
373 that frames the 1314 or 1327 cm<sup>-1</sup> band of the oxalates. Despite these difficulties, XRD results allowed  
374 us to identified this oxalate as weddellite. Therefore, identification of oxalates (weddellite) could be  
375 evidenced by FT-IR through the disappearance or smoothing of the doublet at 1681-1619 cm<sup>-1</sup> of  
376 gypsum and its transformation into a wide band nearby these wavenumbers, possibly as a result of  
377 the contribution of the 1630 cm<sup>-1</sup> band (intermediate to double) of weddellite. By means of this  
378 doublet modification, the presence of wedellite would be confirmed only in 9MS BC where this  
379 oxalate was found in a proportion greater than 3% by XRD. On the other hand, the FT-IR spectrum  
380 of the 8MD BC is striking, since here in this sample only gypsum and quartz were detected by DRX.  
381 However, the band detected at 1321 cm<sup>-1</sup> could indicate the existence of calcium oxalate together  
382 with the bands at 788 cm<sup>-1</sup> and 596 cm<sup>-1</sup>. Calcite, present in trace amounts only in 11MD BC, was  
383 identified by means of FT-IR through the FT-IR bands at 1419 and 877, and 713 cm<sup>-1</sup>. In all the BCs  
384 samples, a band around 1400-1370 cm<sup>-1</sup> is detected; in some cases (1MD, 9MS, 3MS, 7MD, 5MC  
385 and 11MD), showing considerable high intensity. This band could be assigned to the symmetric CH<sub>3</sub>  
386 bending vibrations and, together with those at 711 and 795 cm<sup>-1</sup> could indicate the existence of  
387 organic matter. Also, the band around 1400-1370 cm<sup>-1</sup> could be assigned to the asymmetric NO<sub>3</sub>  
388 stretching vibrations [49]; this band together with those at 870 and 711 cm<sup>-1</sup> (characteristic of the  
389 bending vibration of NO group) would indicate the presence of nitrate salts.  
390  
391  
392  
393  
394  
395  
396  
397  
398  
399  
400  
401  
402  
403  
404  
405  
406  
407  
408  
409  
410  
411  
412  
413

### 3.2 HRSEM-EDX and LA-ICP-MS analysis

HRSEM-EDX allowed us to obtain detailed information on crusts' morphologies and their interaction with the substrate, as well as information about the chemical composition of the particles present BCs. Only a few representative samples of Group A (1MD and 4MS) and B (7MD, 9MS and 12MD) were analyzed in this study.

LA-ICP-MS analysis allowed us to determine the trace elements (average concentrations in ppm) related to representative BCs samples (Table 3), in particular, the criteria chosen were: different accumulation time of pollutants, height and location of the specimens on the façade of the Dome (right and left).

Results about trace elements suggest concentration especially for some metals such as: As, Ba, Cd, Cu, Fe, Mn, Ni, Pb, Sb, Sn, Sr, V, and Zn, in agreement with the HRSEM-EDX results. In particular obtained data underline that specimens placed at a lower height (12m and 5m) and with a greater period of accumulation of pollutants (650 years) show higher contents of heavy metals. In particular, 12 MD BC, located at 5 m from the floor level, shows the highest concentration values.

### 3.3 Ion Chromatography

Table 4 reveals that all samples show high sulphate concentrations due to the conversion of calcium carbonate of the marble substrate [52,55-26] into gypsum. Furthermore, low concentrations of chlorides, nitrites and nitrates were also detected. Contrary to what observed in previous works [28,34] sodium, potassium and ammonium were not detected. It is worth to notice how average sulphate concentration is slightly lower in group B, i.e. the older crusts (Table 4) indicating that the crusts growth is not linear with time.

### 3.4 Carbonaceous fraction analysis (OC and EC)

The analysis of BCs has shown that sulphur is the most abundant element of anthropogenic origin followed by carbon. The in-depth analysis of carbon could suggest possible pollution sources that cause black crusts formation. Total carbon (TC) is composed of OC (Organic Carbon), EC (Elemental Carbon) and CC (Carbonatic Carbon). In this work CC derives mainly from the marble substrate, although deposition of carbonate-rich soil dust is also possible. EC is responsible for the black colour of the crusts; it has a primary origin and is emitted by combustion processes [57-58]. OC is both of primary origin, emitted by traffic or biomass burning [41,59-60], and of secondary origin; in this case it is formed by gaseous precursors such as VOC [57,61-62]. Elemental analysis by CHN allowed the determination of TC within the BCs. Furthermore TGA has been successfully applied in the past to discriminate OC from EC [63]. The analysis by TGA/DCS allowed to quantify carbonate carbon

473  
474  
475 percentage (CC%) while elemental carbon percentage (EC%) has been calculated as described in  
476  
477 details in La Russa et al., 2017. The organic carbon percentage (OC%) was obtained by difference  
478  
479 (OC= TC – (CC+EC). Table 5 shows the obtained results of the carbon fraction for the two groups  
480  
481 of BCs A and B.

482  
483 The values obtained are different from those obtained from samples collected from other monuments  
484  
485 previously analyzed [28,34]. In this study, we have observed higher OC values (on average 3.22 for  
486  
487 group A and 4.12 for group B) with respect to EC (on average 0.83 for group A and 1.02 for group  
488  
489 B), in accordance with what happens in the aerosol particulate matter in Milan where OC is the main  
490  
491 constituent of TC [42,62-64].

#### 492 **4 Discussion**

493  
494 Samples cross-sections examined with PLM allowed the evaluation of the crust-substrate interface.  
495  
496 Regardless of the marble substrate, crusts sampled from the highest altitude (16m) are made of a  
497  
498 single gypsum layer. 1MD BC (thickness ca. 60-130 $\mu$ m) is composed of well-developed needle-like  
499  
500 gypsum crystals which also fill the marble fissures and cracks near the surface. As Fig. 1a shows, the  
501  
502 borders of crystals near the surface are intensely fractured into micrometric grains. Instead, gypsum  
503  
504 crystals in 4MS crust display both prismatic and granular habits yielding a thick crust (thickness ca.  
505  
506 145 $\mu$ m) with copious embedded BCP (Black Carbonaceous Particles) (Fig. 1c,d). BCs are clearly  
507  
508 stratified in several layers of different nature, color and thickness in the BCs samples from group B  
509  
510 (see Fig. 1s). Here, an orange-brown layer appears at the contact between the substrate and the crust.  
511  
512 This layer was identified as weddellite by Capittelli et al. [54]. This layer is called patina noble [65].  
513  
514 At 12m high in the building, 9MS (thickness ca. 100-200 $\mu$ m) and 7MD (thickness ca. 330  $\mu$ m) BCs  
515  
516 exhibit, above the orange-brown layer, a gypsum layer embedding (sub)-spherical BCP in addition to  
517  
518 red, yellow and grey particles (Fig. 1g,h). At 5m height, 12MD BC (thickness ca. 270  $\mu$ m) show the  
519  
520 orange-brown layer which is discontinuous since it is fissured and filled by acicular microcrystalline  
521  
522 gypsum crystals (Fig. 1b). The surface of the crust shows intense black color due to the abundant  
523  
524 spheroidal BCP (Fig. 1b).

525  
526 SEM images of bulk 1MD and 4MS (belonging to group A) samples show that crusts are made of an  
527  
528 interlocked structure of lenticular and/or hexagonal plate-like crystals arranged as rose-like clusters  
529  
530 composed of Ca and S, attributed to gypsum. Fig. 3a is a false-color map based on SEM-EDX  
531  
532 elemental mapping of 4MS thin section showing the many fissures of the calcite crystals (from the  
533  
534 marble substrate) which are filled by microcrystalline gypsum. Note the copious particles of different  
535  
536 nature (e.g. Fe, Si, Mg, Al and Pb) inserted into the gypsum crust matrix. Moreover, as revealed by  
537  
538 SEM-SE/BSE analyses, crust contains spherical spongy-like carbonaceous particles (Fig. 3b,c),

532  
533  
534 rounded Fe-rich particles (Fig. 3d,e), Si-rich particles, and cluster of particles where some of the  
535 following components were identified, i.e. Pb-chloride, Ca-phosphate, Zn, V, Cr, Co and Br (Fig. 3e,  
536 3f). Analysis of bulk 7MD, 9MS and 12MD samples (group B) by SEM show that crusts have a  
537 powdered texture mostly made of granular and (hexagonal) plate-like gypsum crystals as well as  
538 tabular crystals in lesser amount Spheroidal spongy and/or smooth BCP as well as metal particles are  
539 more abundant than in crusts from group A (1MD and 4MS). Analysis of BCs cross sections reveals  
540 the intense crack network developed in the marble area in contact with the crusts, particularly severe  
541 in samples of group B. Though cracks do not show parallel orientation to the crust surface, they have  
542 the potential for physical disintegration (Fig. 4a,b,c,d), as proved during handling for microscopic  
543 analyses. Fig. 4c shows the dissolution patterns in the calcite crystals and the crystallization of  
544 gypsum filling the calcite fissures and cracks (Fig. 4b). SEM-EDX study has also revealed the  
545 numerous metal particles embedded in the gypsum crust matrix in BCs from group B, their  
546 morphology and distribution (Fig. 4). In particular, the X-ray map of a selected area in the 7MD  
547 sample (Fig. 4b) proves the presence of gypsum, Si-rich, Fe-rich, P-rich and Pb-K-rich particles, these  
548 last preferentially located at the marble crust interface. Pb enriched level in the inner zone of crusts  
549 has been reported elsewhere [66], probably due to the affinity of Pb with calcite from the carbonate  
550 substrate. The BC having the most complex structure of all studied crusts is 12MD, sampled at the  
551 lowest height in the Cathedral (5m altitude). Note in Fig. 4c and 4d the well-defined layered structure  
552 and the abundant and varied particles enclosed within the 12MD BC. Gypsum is present in the  
553 threefold layered crust (see Fig. S4). The second non-continuous layer, which is the thicker one, is  
554 enriched in Mg and Zn. Si-rich, Si/Na-rich and Fe-rich particles are evenly distributed in the two  
555 uppermost layers. Minor amount of Pb were detected as compared to 7MD crust (Fig. 4b,d)

556  
557  
558 XRD and FTIR results allowed to identify that BCs are composed mainly by gypsum as result of the  
559 wet or dry deposition of atmospheric sulphur [67] and the subsequent sulphur reaction with the  
560 calcium from the marble, regardless of the age (groups A and B). Despite the most important source  
561 of sulphur is SO<sub>2</sub> emissions from fossil fuels, also biological activities and construction materials  
562 (mortars and concrete) could contribute in the black crust formation [38, 68-72]. In addition, BCs  
563 belonging to group B (650 years old) showed higher quantities of calcium oxalates, since 9MS  
564 (mainly), 8MD and 5 MC showed weddellite. In samples from the group A (280 years old), only  
565 traces of this mineral were found in 4MS. Capittelli et al. [54] also identified a mixture of calcium  
566 oxalate (both weddellite and whewellite) and gypsum in yellow-ochre layer on marble of the Milan  
567 Cathedral (Italy). As was found in our samples, Capittelli et al. [54] reported the presence of organic  
568 matter and calcite in their patinas. As was indicated by other authors [73-74], calcium oxalates can  
569 be formed via biological activity of organisms and reactions of organic compounds in rain or aerosols  
570

591  
592  
593 with the stone. Despite sources of oxalate can be also the application of substances to preserve the  
594 rock surfaces [74], in the current research we do not know if conservation products such as  
595 consolidants have been used in these surfaces.  
596  
597

598 All the analysed samples showed low values of ions concentration can be attributed to the diminishing  
599 porosity inside the crusts.  
600

601 Higher values of TC, OC and EC (see Table 5) have been found for group B which is formed by  
602 crusts present on original substrates, because of these samples have more years of pollutant  
603 accumulation. It has been evidenced a preferential accumulation of OC in almost all BCs samples.  
604 OC/EC ratios show values between 4.5-8.2 (four-point-five and eight-point-two) indicating the  
605 formation of secondary organic compounds [75]. This result evidences, contrary to what we believed  
606 in the past, that vehicular traffic is not the only source of pollution that affects monuments exposed  
607 in polluted urban environment. As previously mentioned, organic substances have different origins  
608 also coming from VOC that can be embedded in the BC. We should point out that about 50% of the  
609 organic fraction present in powder deposits collected from marble surfaces from in Milan [42] is  
610 water-soluble and the presence of these substances can be worrying because contributing to accelerate  
611 stone decay. The higher values of OC observed for some of our BCs samples can be attributed to the  
612 effect of the biologic colonization that can also develop on the crusts; in fact, some studies indicate  
613 [76-79] that these processes could create an excess of organic material. In a recent study, the  
614 biological colonization present on the façade of the Dome of Monza [80] has been highlighted. For  
615 example, in sample 1MD (Fig. S2c), characterized by a higher OC value with respect to the other  
616 samples, this phenomenon was clearly visible. It is also interesting to compare EC/TC ratios with the  
617 results obtained for other European cities [34,42,75,81-83]. In fact, groups A and B show an average  
618 value of EC/TC (0.18 and 0.14 respectively, see Table 3) lower than those observed for the city of  
619 Seville (where the same ratio move in the range 0.22- 0.36 indicating a contribution due to traffic  
620 emissions) and Milan (0.5-0.9 indicating also in this case a contribution due to traffic emissions).  
621 This confirms that BCs collected from Dome of Monza are influenced by organic substances,  
622 probably of secondary origin (because of the high OC/EC ratio).  
623  
624  
625  
626  
627  
628  
629  
630  
631  
632  
633  
634  
635

636 High CC values detected in group B indicate the greater alteration of the marble substrates as  
637 compared to group A of the BCs. In fact, SEM analyses have shown in these BCs samples the  
638 presence of calcite crystals embedded in the gypsum matrix coming from the degraded marble  
639 substrate (see Fig. 3a). Calcite was also detected by HRSEM-EDX, XRD and FT-IR spectroscopy.  
640 TGA allowed the calculation of gypsum percentage in the BCs and the results were validated by IC  
641 (Fig. S5). It is worth noting that the amount of gypsum present in the BCs is variable (Fig. S5), in  
642 agreement with the results obtained with the PLM study. For example, BC of 4MS sample is made  
643  
644  
645  
646  
647  
648  
649

650  
651  
652 mostly by gypsum, while the BC of 12 MD sample is formed by two well defined layers, being the  
653 outermost layer the one made of gypsum. These differences can be caused by different parameters  
654 such as the height of sampling, the morphology of the sampled surfaces (vertical or horizontal), the  
655 exposure to atmospheric agents and the wash out. In fact, the dynamics that are put in motion during  
656 the sulphation process are very complex.  
657

658  
659 Since black crusts could be considered as a passive sampler of atmospheric pollution, a deeper  
660 investigation of the geochemical features was obtained by measuring the average concentrations of  
661 trace elements. The concentrations of the main elements measured within some BCs chosen as  
662 representative are show in table 4. In general, the results are characterized by a certain variability  
663 among BCs samples. Fe, Pb and Zn are the most abundant elements, while As, Ba, Cd, Cu, Mn, Ni,  
664 Sb, Sn and V were always present in a lower amount (in terms of average concentrations).  
665

666  
667 This difference among the analysed BCs samples can be due to several factors such as: different  
668 accumulation period, different height of sampling, different exposure to sources of and to wash out.  
669 Sample 12DM shows the highest concentrations of heavy metals, since this sample was taken from  
670 the lower part of the façade, most exposed to vehicular traffic and characterized by a longer pollutants  
671 accumulation period. On the contrary, samples 1MD and 4MS, show the lower concentration of heavy  
672 metals because were taken from the highest areas not directly exposed to vehicular traffic and also  
673 affected by a lower accumulation period. Specimens 9MS and 7MD BCs, display intermediate  
674 concentrations in heavy metals because taken from a medium height compared to the other samples,  
675 but whit a period of accumulation higher then samples 1MD and 4MS.  
676

677  
678 In general, all samples show high contents of Fe, related to emissions from the numerous industries  
679 (oil refining, storage, metal extraction and metallurgy, energy production and distribution, and waste  
680 incinerator) placed in the near industrial area of Po valley [84-85]. In addition, the highest  
681 concentrations of Pb, Fe, and Zn observed for BCs of group B are in agreement with the use of lead  
682 gasoline (15, 86-91Rodriguez-Navarro and Sebastian 1996), employed until about 25 years ago. At  
683 the same time, Cu, Ni, Cr, and V well match with the use of other combustibles such as: oil  
684 combustible, diesel, and gasoline [15, 86-91], widespread after the abolition of leaded gasoline.  
685

686  
687 The enrichment observed in group B is mainly due to the high vehicular traffic that characterizes the  
688 city of Monza and the metropolitan area of Milan, although the cathedral has been located in a  
689 pedestrian zone in the last decades. In fact, in the past years, numerous vehicles have been authorized  
690 to circulate both in the vicinity and inside the square where the monument is located (see Fig.S1). On  
691 the contrary, group A samples collected from 16m from the ground, often display lower  
692 concentrations of the same heavy metals.  
693  
694  
695  
696  
697  
698  
699  
700  
701  
702  
703  
704  
705  
706  
707  
708

709  
710  
711  
712  
713  
714  
715  
716 **5 Conclusions**  
717  
718

719 BC formed on marble stones have been collected from the facade of Duomo of Monza and analysed  
720 by several analytical techniques. Two groups of samples were identified, group A and group B,  
721 according to the information available on restoration interventions in the specific areas from where  
722 they were taken. Petrographic analyses have confirmed that those belonging to group B show a more  
723 complex stratigraphy attributable to the greater accumulation time of pollutants compared to BCs of  
724 group A which show a simpler morphology. Furthermore, XRD and FT-IR confirm that all BCs are  
725 mainly composed of gypsum, with some traces of oxalate and quartz was detected in different  
726 amounts for the different groups; weddellite was mainly found in samples belonging to the group B  
727 (older samples) The analysis of the main ions present in the BCs has shown low concentrations for  
728 these. The HRSEM-EDX analysis showed that samples of group B have accumulated more metal  
729 particles, in particular, Pb-rich particles that are particularly mixed with Cl, these located in the  
730 interface marble-crust because of the affinity of Pb for calcite. Conversely, the Pb-rich particles are  
731 present in the surface in group A although in much less amount. Since group B samples are closer to  
732 the bottom of the building the impact of vehicle pollution are evident.

733 As regards the carbonaceous species (OC and EC), it has been demonstrated that there is a preferential  
734 accumulation of organic substances, probably of the secondary origin or biological contamination,  
735 can be also accumulate.

736 Significant and different concentration amounts between group A and B by as regards iron, lead, zinc,  
737 copper and vanadium have been highlighted; this represents the fingerprint of the air pollution  
738 occurred in the different years of deposition in the analysed BCs. In fact, the city of Monza is affected  
739 by pollution from the industrial area, as well as from vehicular traffic. The high concentrations of the  
740 different species observed in group B are mainly due to vehicular traffic although the cathedral has  
741 been located in a pedestrian zone since a few decades. On the contrary, group A samples, collected  
742 from low heights, often display lower concentrations of the same heavy metals because they are less  
743 exposed to direct emissions.

744 In conclusion this study confirms the importance of BCs chemical characterization in order to  
745 evaluate the actual risk of degradation to which a monument is submitted when exposed to  
746 environmental pollution.  
747  
748  
749  
750  
751  
752  
753  
754  
755  
756  
757  
758  
759  
760  
761  
762  
763  
764  
765  
766  
767

768  
769  
770 **Acknowledgments**  
771

772 J. Santiago Pozo-Antonio thanks the Spanish Ministry of Economy and Competitiveness (MINECO)  
773 for his “Juan de la Cierva-incorporación” (IJCI-2017-3277) contract.  
774

775 Paola Fermo and Valeria Comite would like to thanks Ing. Benigno Mörlin Visconti for the precious  
776 collaboration and Dr. Francesco Piovani for the help during sampling and for the useful discussion.  
777

778 C. Cardell thanks the financial support provided by Spanish Research Projects AERIMPACT  
779 (CGL2012-30729) and EXPOAIR (P12-FQM-1889), the European Regional Development Fund  
780 (ERDF), and the Andalusian Research Group RNM-179. SEM-EDX analyses were performed in the  
781 Scientific Instrumentation Centre (CIC) of the University of Granada.  
782  
783  
784  
785

786 **Reference**  
787

- 788 [1] M. Del Monte, C. Sabbioni, O. Vittori, Airborne carbon particles and marble deterioration, *Atmos. Environ.*  
789 15 (1981) 645–652. DOI :10.1016/0004-6981(81)90269-9.  
790
- 791 [2] P. Brimblecombe, History of urban air pollution, in: J. Finger, O. Herter, F. Palmer (Eds). *Urban air*  
792 *pollution—European aspects*. Kluwer, Dordrecht, 1999, pp.7–20.  
793
- 794 [3] P. Brimblecombe, Air pollution and architecture, past, present and future, *J. Archit. Conserv.* 6 (2000) 30–  
795 46. <https://doi.org/10.1080/13556207.2000.10785268>.  
796
- 797 [4] A.V. Turkington, B.J. Smith & W.B. Whalley, Short-term stone surface modification; an example from  
798 Venice. *Proceedings of the 4th international symposium on the conservation of monuments in the*  
799 *Mediterranean basin*. Technical chamber of Greece, Rhodes 1, 1997, pp. 359–372.  
800
- 801 [5] G. Zappia, C. Sabbioni, C. Riontino, G. Gobbi, O. Favoni, Exposure tests of building materials in urban  
802 atmosphere, *Sci. Total. Environ.* 224 (1998) 235–244. doi:10.1016/S0048-9697(98)00359-3.  
803
- 804 [6] S. A. Ruffolo, M. F. La Russa, M. Ricca, C.M. Belfiore, A. Macchia, V. Comite, A. Pezzino, New insights  
805 on the consolidation of salt weathered limestone: the case study of Modica stone, *Bull. Eng. Geol. Environ.* 76  
806 (2017) 11–20. DOI 10.1007/s10064-015-0782-1.  
807
- 808 [7] M. F. La Russa, S. A. Ruffolo, M. A. de Buergo, M. Ricca, C.M. Belfiore, A. Pezzino, G. M. Crisci, The  
809 behaviour of consolidated Neapolitan yellow Tuff against salt weathering, *Bull. Eng. Geol. Environ.* 76 (2017)  
810 115–124. DOI 10.1007/s10064-016-0874-6.  
811
- 812 [8] M. Ricca, E. Le Pera, M. Licchelli, A. Macchia, M. Malagodi, L. Randazzo, N. Rovella, S. A. Ruffolo, M.  
813 L. Weththimuni, M. F. La Russa The CRATI Project: New Insights on the Consolidation of Salt Weathered  
814 Stone and the Case Study of San Domenico Church in Cosenza (South Calabria, Italy), *Coatings* 9, 330 (2019),  
815 1–15. DOI 10.3390/coatings9050330.  
816
- 817 [9] P. Brimblecombe, Environment and architectural stone, in: S. Sigismund, R. Snethlage (Eds.) *Stone in*  
818 *architecture*, 4th, 461st edn. Springer, Berlin, 2001, pp. 317–346.  
819  
820  
821  
822  
823  
824  
825  
826



827  
828  
829  
830  
831  
832  
833  
834  
835  
836  
837  
838  
839  
840  
841  
842  
843  
844  
845  
846  
847  
848  
849  
850  
851  
852  
853  
854  
855  
856  
857  
858  
859  
860  
861  
862  
863  
864  
865  
866  
867  
868  
869  
870  
871  
872  
873  
874  
875  
876  
877  
878  
879  
880  
881  
882  
883  
884  
885

- [10] N. Ghedini, C. Sabbioni, A. Bonazza, G. Gobbi, Chemical-thermal quantitative methodology for carbon speciation in damage layers on building surfaces. *Environ. Sci. Technol.* 40 (2006) 939–944. DOI 10.1021/es0501641.
- [11] N. Marinoni, M. P. Birelli, C. Rostagno, A. Pavede, The effects of atmospheric multi pollutants on modern concrete, *Atmos. Environ.* 37 (2003) 4701–4712. DOI 10.1016/j.atmosenv.2003.06.001.
- [12] A. Bonazza, C. Sabbioni, N. Ghedini Quantitative data on carbon fractions in interpretation of black crusts and soiling on European built heritage. *Atmos. Environ.* 39 (2005) 2607–2618. DOI 10.1016/j.atmosenv.2005.01.040.
- [13] C. M. Belfiore, D. Barca, A. Bonazza, V. Comite, M.F. La Russa, A. Pezzino, Application of spectrometric analysis to the identification of pollution sources causing cultural heritage damage. *Environ. Sci. Pollut. Res.* 20 (2013) 8848–59. DOI 10.1007/s11356-013-1810-y.
- [14] O. Farkas, S. Siegesmund, T. Licha, Á. Török, Geochemical and mineralogical composition of black weathering crusts on limestones from seven different European countries, *Environ. Earth Sci.* 77:211 (2018) DOI: 10.1007/s12665-018-7384-8.
- [15] C. Rodriguez-Navarro, E. Sebastian, Role of particulate matter from vehicle exhaust on porous building stone (limestone) sulfation, *Sci. Total. Environ.* 187 (1996) 79–91. DOI 10.1016/0048-9697(96)05124-8.
- [16] M. D. Geller, L. Ntziachristos, A. Athanasios Mamakos, Z. Zissis Samaras, D. A. Schmitz, J. R. Froines, C. Sioutas, Physicochemical and redox characteristics of particulate matter (PM) emitted from gasoline and diesel passenger cars, *Atmos. Environ.* 40 (2006) 6988–7004. DOI 10.1016/j.atmosenv.2006.06.018.
- [17] D. Contini, A. Gambaro, F. Belosi, S. De Pieri, W. R. L. Cairns, A. Donateo, E. Zanutto, M. Citron, The direct influence of ship traffic on atmospheric PM<sub>2.5</sub>, PM<sub>10</sub> and PAH in Venice. *J. Environ. Management* 92 (2011) 2119–2129. <https://doi.org/10.1016/j.jenvman.2011.01.016>.
- [18] R. Prikryl, J. Svabodová, K. Zák, D. Hradil, Anthropogenic origin of salt crusts on sandstone sculptures of Prague’s Charles Bridge (Czech Republic). Evidence of mineralogy and stable isotope geochemistry, *Eur. J. Mineral.* 16 (2004) 609–18. DOI: 10.1127/0935-1221/2004/0016-0609.
- [19] K. Torfs, R.E. Van Grieken, F. Buzek, Use of stable isotope measurements to evaluate the origin of sulfur in gypsum layers on limestone buildings, *Environ. Sci. Pollut. Res.* 31 (1997) 2650–5. <https://doi.org/10.1021/es970067v>.
- [20] J. M. Vallet, C. Gosselin, P. Bromblet, P. Rolland, V. Vergés-Belmin, W. Kloppmann, Origin of salts in stone monuments degradation using sulphur and oxygen isotopes: first results of the Bourges Cathedral (France), *J. Geochem. Explor.* 88 (2006) 358–62. DOI 10.1016/j.gexplo.2005.08.075.
- [21] J. Schweigstillová, R. Prikryl, M. Novotná, Isotopic composition of salt efflorescence from the sandstone castellated rocks of the Bohemian Cretaceous Basin (Czech Republic), *Environ. Geol.* 58 (2009) 217–25. DOI 10.1007/s00254-008-1510-y.
- [22] N. Schleicher, C. Recio, Source identification of sulphate forming salts on sandstones from monuments in Salamanca, Spain—a stable isotope approach, *Environ. Sci. Pollut. Res.* 17 (2010) 770–8. DOI 10.1007/s11356-009-0196-3.

886  
887  
888  
889  
890  
891  
892  
893  
894  
895  
896  
897  
898  
899  
900  
901  
902  
903  
904  
905  
906  
907  
908  
909  
910  
911  
912  
913  
914  
915  
916  
917  
918  
919  
920  
921  
922  
923  
924  
925  
926  
927  
928  
929  
930  
931  
932  
933  
934  
935  
936  
937  
938  
939  
940  
941  
942  
943  
944

- [23] W. Kloppmann, P. Bromblet, J.M. Vallet, V. Vergès-Belmin, O. Rolland, C. Guerrot, Building materials as intrinsic sources of sulphate: a hidden face of salt weathering of historical monuments investigated through multi-isotope tracing (B, O, S), *Sci. Total. Environ.* 409(9): (2011) 1658–69. DOI 10.1016/j.scitotenv.2011.01.008.
- [24] S. Kramar, B. Mirtič, K. Knöller, N. Rogan-Šmuc, Weathering of the black limestone of historical monuments (Ljubljana, Slovenia): oxygen and sulfur isotope composition of sulfate salts. *Appl Geochem* 26(9–10) (2011) 1632–1638. DOI: 10.1016/j.apgeochem.2011.04.020.
- [25] T. Rivas, S. Pozo, M. Paz, Sulphur and oxygen isotope analysis to identify sources of sulphur in gypsum-rich black crusts developed on granites. *Sci. Tot. Environ.* 482–483 (2014) 137–147. doi: 10.1016/j.scitotenv.2014.02.128.
- [26] P. Brimblecombe, C. M. Grossi, Aesthetic thresholds and blackening of stone buildings. *Sci. Total Environ.* 349 (2005) 175–189. DOI: 10.1016/j.scitotenv.2005.01.009.
- [27] D. Barca, V. Comite, C. M. Belfiore, A. Bonazza, M. F. La Russa, S. A. Ruffolo, G. M. Crisci, A. Pezzino, C. Sabbioni, Impact of air pollution in deterioration of carbonate building materials in Italian urban environments, *Appl. Geochem.* 48 (2014) 122–131. DOI: 10.1016/j.apgeochem.2014.07.002.
- [28] M.F. La Russa, P. Fermo, V. Comite, C.M. Belfiore, D. Barca, A. Cerioni, M. De Santis, L.F. Barbagallo, M. Ricca, S. A. Ruffolo, The Oceanus statue of the Fontana di Trevi (Rome): The analysis of black crust as a tool to investigate the urban air pollution and its impact on the stone degradation, *Sci. Tot. Environ.* 593-594 (2017) 297-309. DOI 10.1016/j.scitotenv.2017.03.185.
- [29] D. Barca, C. M. Belfiore, G. M. Crisci, M. F. La Russa, A. Pezzino, S A Ruffolo, Application of laser ablation ICP-MS and traditional techniques to the study of black crusts on building stones: a new methodological approach, *Environ. Sci. Pollut. Res.* 17 (2010) 1433–47. DOI 10.1007/s11356-010-0329-8.
- [30] N. Prieto-Taboada, M. Maguregui, I. Martinez-Arkarazo, M.A. Olazabal, G. Arana, J.M. Madariaga, Spectroscopic evaluation of the environmental impact on black crusted modern mortars in urban–industrial areas, *Anal. Bioanal. Chem.* 399 (2011) 2949–2959. DOI: 10.1007/s00216-010-4324-1.
- [31] N. Schiavon, G. Chiavari, D. Fabbri, Soiling of limestone in an urban environment characterized by heavy vehicular exhaust emissions, *Environ. Geol.* 46 (2004) 448–455. DOI: 10.1007/s00254-004-1046-8.
- [32] H. Morillas, M. Maguregui, C. García-Florentino, J.A. Carrero, J.M. Madariaga, The cauliflower-like black crusts on sandstones: a natural passive sampler to evaluate the surrounding environmental pollution, *Environ. Res.* 17 (2016) 218–232. DOI 10.1016/j.envres.2016.02.015.
- [33] V. Comite, M. Álvarez de Buergo, D. Barca, C.M. Belfiore, A., Bonazza, M.F. La Russa, A. Pezzino, L. Randazzo, S.A. Ruffolo, Damage monitoring on carbonate stones: Field exposure tests contributing to pollution impact evaluation in two Italian sites, *Constr. Buil. Mat.* 152 (2017) 907-922. DOI: 10.1016/j.conbuildmat.2017.07.048.
- [34] V. Comite, P. Fermo, The effects of air pollution on cultural heritage: the case study of Santa Maria delle Grazie al Naviglio Grande (Milan), *E.P.J. Plus.* 133 (12) (2018). 556 DOI: 10.1140/epjp/i2018-12365-6.

- 945  
946  
947 [35] J. Sanjurjo-Sánchez, J.R. Vidal Romani, C. Alves, Comparative analysis of coatings on granitic substrates  
948 from urban and natural settings (NW Spain), *Geomorphology* 138 (2012) 231–242. DOI:  
950 10.1016/j.geomorph.2011.09.008.
- 951 [36] J. S. Pozo-Antoniao, M.P. Fiorucci, A. Ramila, A.J. López, T. Rivas, Evaluation of the effectiveness of  
952 laser crust removal on granites by means of hyperspectral imaging techniques, *App. Surf. Sci.* 347 (2015) 832–  
954 838.
- 955 [37] J. S. Pozo-Antonio, A. Ramil, T. Rivas, A. J. López, M. P. Fiorucci, Effectiveness of chemical, mechanical  
956 and laser cleaning methods of sulphated black crusts developed on granite, *Const. Build. Mat.* 112 (2016) 682–  
958 690. <http://dx.doi.org/10.1016/j.conbuildmat.2016.02.195>.
- 959 [38] J. S. Pozo-Antonio, M. F. C. Pereira, C. S. A. Rocha Microscopic characterisation of black crusts on  
960 different substrates, *Sci. Tot. Environ.* 584–585 (2017) 291–306.  
962 <http://dx.doi.org/10.1016/j.scitotenv.2016.12.080>.
- 963 [39] R. Cassanelli, Monza anno 1300. La Basilica di S. Giovanni Battista e la sua facciata. Comune di Monza.  
964 1988.
- 965 [40] C. R. de Kimpe, Clay and silt analysis, in: M. R. Carter (Eds.), *Soil sampling and methods of analysis*.  
966 Canadian Society of Soil Science, Lewis Publisher; 1993. pp. 719–730.
- 967 [41] A. Piazzalunga, V. Bernardoni, P. Fermo, R. Vecchi, Optimisation of analytical procedures for the  
968 quantification of ionic and carbonaceous fractions in the atmospheric aerosol and applications to ambient  
969 samples. *Anal. Bioanal. Chem.* 405, (2013) 1123–1132. DOI 10.1007/s00216-012-6433-5.
- 970 [42] P. Fermo, R. Gonzalez Turrion, M. Rosa, A. Omega, A new approach to assess the chemical composition  
971 of powder deposits damaging the stone surfaces of historical monuments, *Environ. Sci. Pollut. Res.* 22 (2015)  
972 6262–6270. DOI 10.1007/s11356-014-3855-y.
- 973 [43] B. Gratuze, Obsidian characterization by laser ablation ICPMS and its application to prehistoric trade in  
974 the Mediterranean and the Near East: sources and distribution of obsidian within the Aegean and Anatolia, *J.*  
975 *Archaeol. Sci.* 26 (1999) 869–881. DOI:10.1006/jasc.1999.0459.
- 976 [44] E. Vander Putten, F. Dehairs, L. André, W. Baeyens, Quantitative in situ microanalysis of minor and trace  
977 elements in biogenic calcite using infrared laser ablation-inductively coupled plasma mass spectrometry: a  
978 critical evaluation, *Anal. Chim. Acta* 378 (1999) 261–272. DOI:10.1016/S0003-2670(98)00613-8.
- 979 [45] T. Wyndham, M. McCulloch, S. Fallon, C. Alibert, High resolution coral records of rare earth elements  
980 in coastal seawater: biogeochemical cycling and a new environmental proxy. *Geochim. Cosmochim. Acta* 68  
981 (2004) 2067–2080. DOI: 10.1016/j.gca.2003.11.004.
- 982 [46] D. Barca, C. M. Belfiore, G. M. Crisci, M. F. La Russa, A. Pezzino, S. A. Ruffolo, A new methodological  
983 approach for the chemical characterization of black crusts on building stones: a case study from the Catania  
984 city centre (Sicily, Italy), *J. Anal. At. Spectrom.* 26 (2011) 1000-1011. DOI 10.1039/C0JA00226G.
- 985 [47] M.F. La Russa, C.M. Belfiore, V. Comite, D. Barca, A. Bonazza, S.A. Ruffolo, G.M. Crisci, A. Pezzino,  
986 Geochemical study of black crusts as a diagnostic tool in cultural heritage, *Appl. Phys. A Mater. Sci. Process.*  
987 113 (2013) 1151–1162. DOI: 10.1007/s00339-013-7912-z.
- 988  
989  
990  
991  
992  
993  
994  
995  
996  
997  
998  
999  
1000  
1001  
1002  
1003

- 1004  
1005  
1006 [48] M.F. La Russa, V. Comite, N. Aly, D. Barca, P. Fermo, N. Rovella, F. Antonelli, E. Tesser, M. Aquino,  
1007 S.A. Ruffolo, Black crusts on Venetian built heritage, investigation on the impact of pollution sources on their  
1008 composition, *Eur. Phys. J. Plus* 133: 370 (2018). DOI: 10.1140/epjp/i2018-12230-8.  
1009  
1010 [49] G. Socrates, *Infrared and Raman characteristic group frequencies: Tables and Charts*. Chichester: Wiley  
1011 (Pub.), 2001.  
1012  
1013 [50] M. D. Lane. Mid-infrared emission spectroscopy of sulfate and sulfate bearing minerals. *Am. Miner.* 92  
1014 (2007) 1-18. DOI: 10.2138/am.2007.2170.  
1015  
1016 [51] N. Schiavon, Kaolinisation of granite in an urban environment. *Environ. Geol.* 52 (2007) 399-407. DOI  
1017 10.1007/s00254-006-0473-0.  
1018  
1019 [52] D. Gulotta, M. Bertoldi, S. Bortolotto, P. Fermo, A. Piazzalunga, L. Toniolo, The Angera stone: a  
1020 challenging conservation issue in the polluted environment of Milan (Italy). *Environ. Earth. Sci.* 69 (2013)  
1021 1085–1094. DOI 10.1007/s12665-012-2165-2.  
1022  
1023 [53] B. J. Saikia, G. Parthasarathy, Fourier Transform Infrared Spectroscopic Characterization of Kaolinite  
1024 from Assam and Meghalaya, North eastern India, *J. Mod. Phys.*, 1 (2010) 206-210.  
1025  
1026 [54] F. Cappitelli, L. Toniolo, A. Sansonetti, Advantages of using microbial technology over traditional  
1027 chemical technology in removal of black crusts from stone surfaces of historical monuments. *Appl. Environ.*  
1028 *Microbiol.* 73(17) (2007)5671–5675. DOI 10.1128/AEM.00394-07.  
1029  
1030 [55] P. Fermo, S. Goidanich, V. Comite, L. Toniolo, D. Gulotta, Study and Characterization of Environmental  
1031 Deposition on Marble and Surrogate Substrates at a Monumental Heritage Site *Geosciences* 8 (2018) 349–  
1032 266. DOI 10.3390/geosciences8090349.  
1033  
1034 [56] V. Sabatini, E. Pargoletti, V. Comite, M. A. Ortenzi, P. Fermo, D. Gulotta, G. Cappelletti, Towards Novel  
1035 Fluorinated Methacrylic Coatings fo Cultural Heritage: A Combined Polymers and Surfaces Chemistry Study,  
1036 *Polymers* 11 (2019) 1190–1207. DOI 10.3390/polym11071190.  
1037  
1038 [57] D. R. Gentner, G. Isaacman, D. R. Worton, A. W. H. Chan, T. R. Dallmann, L. Davisa, S. Liud, D. A.  
1039 Day, L. M. Russell, K. R. Wilson, R. Weber, A. Guha, R. A. Harley, A. H. Goldstein, Elucidating secondary  
1040 organic aerosol from diesel and gasoline vehicles through detailed characterization of organic carbon  
1041 emissions, *Proc. Natl. Acad. Sci. U.S.A.* 109 (2012) 18318–18323. DOI 10.1073/pnas.1212272109.  
1042  
1043 [58] S. Fuzzi, M. O. Andreae, B. J. Hueber, M. Kulmala, T. C. Bon, M. Boy, S. J. Doherty, A. Guenther,  
1044 M. Kanakidou, K. Kawamura, V.-M. Kerminen, U. Lohmann, L. M. Russell, U. Poschl, Critical assessment  
1045 of the current state of scientific knowledge, terminology, and research needs concerning the role of organic  
1046 aerosols in the atmosphere, climate, and global change, *Atmos. Chem. Phys.*, 6 (2006)2017–2038.  
1047 <https://doi.org/10.5194/acp-6-2017-2006>.  
1048  
1049 [59] I. Vassura, E. Venturini, S. Marchetti, A. Piazzalunga, E. Bernardi, P. Fermo, F. Passarini, Markers and  
1050 influence of open biomass burning on atmospheric particulate size and composition during a major bonfire  
1051 event, *Atmos. Environ.* 82 (2014) 218–225. DOI: 10.1016/j.atmosenv.2013.10.037.  
1052  
1053 [60] K.R. Daellenbach, C. Bozzetti, A. Křepelová, F. Canonaco, R. Wolf, P. Zotter, P. Fermo, M. Crippa, J.G.  
1054 Slowik, Y. Sosedova, Y. Zhang, R.-J. Huang, L. Poulain, S. Szidat, U. Baltensperger, I. El Haddad, A. S. H.  
1055  
1056  
1057  
1058  
1059  
1060  
1061  
1062

1063  
1064  
1065 Prévôt, Characterization and source apportionment of organic aerosol using offline aerosol mass spectrometry,  
1066 *Atmos. Meas. Tech.* 9 (1) (2016) 23–39. <https://doi.org/10.5194/amt-9-23-2016>.

1067  
1068 [61] A. L. Robinson, N. M. Donahue, M. K. Shrivastava, E. A. Weitkamp, A. M. Sage, A. P. Grieshop, T. E.  
1069 Lane, J. R. Pierce, S. N. Pandis, Rethinking Organic Aerosols: Semivolatile Emissions and Photochemical  
1070 Aging, *Science* 315 (2007) 1259–1262. DOI: 10.1126/science.1133061.

1071  
1072 [62] V. Bernardoni, R. Vecchi, G. Valli, A. Piazzalunga, P. Fermo PM10 source apportionment in Milan (Italy)  
1073 using time-resolved data, *Sci. Tot. Environ.* 409 (2011) 4788–4795. DOI 10.1016/j.scitotenv.2011.07.048.

1074  
1075 [63] P. Fermo, A. Piazzalunga, R. Vecchi, G. Valli, M. Ceriani, A TGA/FT-IR study for measuring OC and  
1076 EC in aerosol samples, *Atmos Chem Phys*, 6 (1) (2006) 255–266.

1077  
1078 [64] R. Vecchi, V. Bernardoni, P. Fermo, F. Lucarelli, F. Mazzei, S. Nava, P. Prati, A. Piazzalunga, G. Valli,  
1079 4-hours resolution data to study PM10 in a “hot spot” area in Europe, *Environ. Monit. Assess.* 154 (2009) 283–  
1080 300. DOI 10.1007/s10661-008-0396-1.

1081  
1082 [65] R. Rossi Manaresi, Oxalate patinas and conservation treatments, p. 113–127, in: M. Realini and L. Toniolo  
1083 (Eds.), *The oxalate films in the conservation of works of art. Proceedings of the 2nd International Symposium,*  
1084 *25 to 27 March 1996 EDITEAM. Bologna, Italy, 1996, pp. 113–127.*

1085  
1086 [66] S.A. Ruffolo, V. Comite, M.F. La Russa, C.M. Belfiore, D. Barca, A. Bonazza, G.M. Crisci, A. Pezzino,  
1087 C. Sabbioni, Analysis of black crusts from the Seville Cathedral: a challenge to deepen understanding the  
1088 relationship among microstructure, micro chemical features and pollution sources. *Sci. Total Environ.* 502  
1089 (2015) 157–166. DOI 10.1016/j.scitotenv.2014.09.023.

1090  
1091 [67] A.E. Charola, R. Ware, Acid deposition and the deterioration of stone: a brief review of a broad topic,  
1092 *Geol. Soc. Spec. Pub.* 205 (2002)393–406. <https://doi.org/10.1144/GSL.SP.2002.205.01.28>.

1093  
1094 [68] T. Hosono, E. Uchida, C. Suda, A. Ueno, T. Nakagawa. Salt weathering of sandstone at the Angkor  
1095 monuments, Cambodia: identification of the origins of salts using sulphur and strontium isotopes, *J. Archaeol.*  
1096 *Sci.* 33: (2006)1541–51. DOI: 10.2465/ganko.93.411.

1097  
1098 [69] N. Schleicher, C. Recio, Source identification of sulphate forming salts on sandstones from monuments  
1099 in Salamanca, Spain—a stable isotope approach, *Environ. Sci. Pollut. Res.* 17 (2010) 770–8. DOI:  
1100 10.1007/s11356-009-0196-3.

1101  
1102 [70] W. Kloppmann, P. Bromblet, J.M. Vallet, V. Vergès-Belmin, O. Rolland, C. Guerrot, Building materials  
1103 as intrinsic sources of sulphate: a hidden face of salt weathering of historical monuments investigated through  
1104 multi-isotope tracing (B, O, S), *Sci. Total. Environ.*;409(9) (2011) 1658–69. DOI:  
1105 10.1016/j.scitotenv.2011.01.008.

1106  
1107 [71] S. Kramar, B. Mirtič, K. Knöller, N. Rogan-Šmuc, Weathering of the black limestone of historical  
1108 monuments (Ljubljana, Slovenia): oxygen and sulfur isotope composition of sulfate salts, *Appl. Geochem.*  
1109 26(9–10) (2011)1632–8. <https://doi.org/10.1016/j.apgeochem.2011.04.020>.

1110  
1111 [72] J.S. Pozo-Antonio, M.F.C. Pereira, C.S.A. Rocha, I. Puente, C. Figueiredo, Comparative study of  
1112 deterioration forms on nearby granitic bridges from an urban setting in the NW Iberian Peninsula  
1113 *Geomorphology*, 274 (2016) 11-30. <https://doi.org/10.1016/j.geomorph.2016.09.009>.

1114  
1115  
1116  
1117  
1118  
1119  
1120  
1121

1122  
1123  
1124  
1125  
1126  
1127  
1128  
1129  
1130  
1131  
1132  
1133  
1134  
1135  
1136  
1137  
1138  
1139  
1140  
1141  
1142  
1143  
1144  
1145  
1146  
1147  
1148  
1149  
1150  
1151  
1152  
1153  
1154  
1155  
1156  
1157  
1158  
1159  
1160  
1161  
1162  
1163  
1164  
1165  
1166  
1167  
1168  
1169  
1170  
1171  
1172  
1173  
1174  
1175  
1176  
1177  
1178  
1179  
1180

- [73] M. Del Monte, C. Sabbioni, G. Zappia, The origin of calcium oxalates on historical buildings, monuments and natural outcrops, *Sci. Tot. Environ.* 67 (1987) 17-39. [https://doi.org/10.1016/0048-9697\(87\)90063-5](https://doi.org/10.1016/0048-9697(87)90063-5).
- [74] J. Russ, W.D. Kaluarachchi, L. Drummond, H.G.M. Edwards. 1999. The Nature of a Whewellite-Rich Rock Crust Associated with Pictographs in Southwestern Texas. *J. Stud. Conserv.* 44, Issue 2 (1999) 91-103. <https://doi.org/10.1179/sic.1999.44.2.91>.
- [75] R. Vecchi, M. Chiari, A. D'Alessandro, P. Fermo, F. Lucarelli, F. Mazzei, S. Nava, A. Piazzalunga, P. Prati, F. Silvani, G. Valli, A mass closure and PMF source apportionment study on the sub-micron sized aerosol fraction at urban sites in Italy. *Atmos. Environ.* 42 (2008) 2240–2253. DOI 10.1016/j.atmosenv.2007.11.039.
- [76] D. Camuffo, M. Del Monte, C. Sabbioni, Influenza delle precipitazioni e della condensazione sul degrado superficiale dei monumenti in marmo e calcare. *Materiali Lapidari. Volume Speciale del Bollettino*, Roma, p.15-36, 1987.
- [77] C. Sabbioni, G. Zappia, Oxalate patinas on ancient monument: the biological hypothesis. *Aerobiologia* Issue 7 (1991) 31–37. doi:10.1007/BF02450015.
- [78] C. Sabbioni, Contribution of atmospheric deposition to the formation of damage layers. *Sci. Total. Environ.* 167 (1995) 49–55. [https://doi.org/10.1016/0048-9697\(95\)04568](https://doi.org/10.1016/0048-9697(95)04568).
- [79] A. Bonazza, P. Brimblecombe, C. M. Grossi, C. Sabbioni, Carbon in black crust from the tower of London. *Environ. Sci. Technol.* 41 (2007) 4199–4204. <https://doi.org/10.1021/es062417w>.
- [80] D. Gulotta, F. Villa, F. Cappitelli, L. Toniolo, Biofilm colonization of metamorphic lithotypes of a renaissance cathedral exposed to urban atmosphere, *Sci. Tot. Environ.* 639 (2018) 1480–1490. <https://doi.org/10.1016/j.scitotenv.2018.05.277>.
- [81] C. Saiz-Jimenez, Air pollution and Cultural Heritage. *Proceedings of the International, Workshop on Air Pollution and Cultural Heritage*, Seville, Spain, 2003.
- [82] C. Perrino, S. Canepari, M. Catrambone, S. Dalla Torre, E. Rantica, T. Sargolini, Influence of natural events on the concentration and composition of atmospheric particulate matter, *Atmos. Environ.* 43 (2009) 4766–4779. DOI: 10.1016/j.atmosenv.2008.06.035.
- [83] S. Sandrini, S. Fuzzi, A. Piazzalunga, P. Prati, P. Bonasoni, F. Cavalli, M.C. Bove, M. Calvello, D. Cappelletti, C. Colombi, D. Contini, G. De Gennaro, A. Di Gilio, P. Fermo, L. Ferrero, V. Gianelle, M. Giugliano, P. Ielpo, G. Lonati, A. Marinoni, D. Massabò, U. Molteni, B. Moroni, G. Pavese, C. Perrino, M. G. Perrone, M. R. Perrone, Spatial and seasonal variability of carbonaceous aerosol across Italy, *Atmos. Environ.* 99 (2014) 587–598. DOI 10.1016/j.atmosenv.2014.10.032.
- [84] C. Colombi, *Relazione Campagna PoAIR febbraio 2014*. Available online: <http://www.arpalombardia.it/qariafiles/varie/Relazione%20campagna%20PoAIR%20febbraio%202014.pdf> (accessed on 5 August 2019).
- [85] EEA Report No 1/2017. *Climate change, impacts and vulnerability in Europe 2016 An indicator-based report*. European Environment Agency. Luxembourg: Publications Office of the European Union, 2017.

1181  
1182  
1183  
1184  
1185  
1186  
1187  
1188  
1189  
1190  
1191  
1192  
1193  
1194  
1195  
1196  
1197  
1198  
1199  
1200  
1201  
1202  
1203  
1204  
1205  
1206  
1207  
1208  
1209  
1210  
1211  
1212  
1213  
1214  
1215  
1216  
1217  
1218  
1219  
1220  
1221  
1222  
1223  
1224  
1225  
1226  
1227  
1228  
1229  
1230  
1231  
1232  
1233  
1234  
1235  
1236  
1237  
1238  
1239

[86] M. D. Geller, L. Ntziachristos, A. Athanasios Mamakos, Z. Zissis Samaras, D. A. Schmitz, J. R. Froines, C. Sioutas Physicochemical and redox characteristics of particulate matter (PM) emitted from gasoline and diesel passenger cars. *Atmos Environ* 40 (2006) 6988–7004. DOI 10.1016/j.atmosenv.2006.06.018.

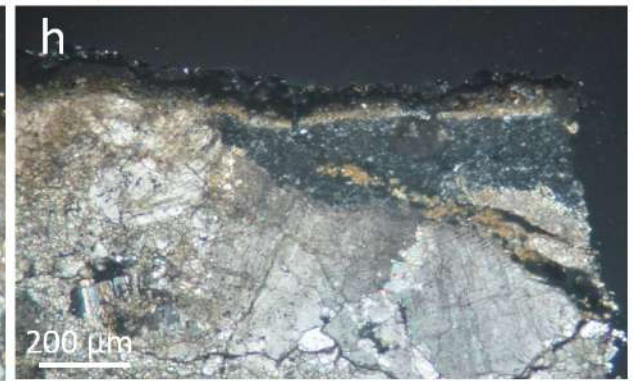
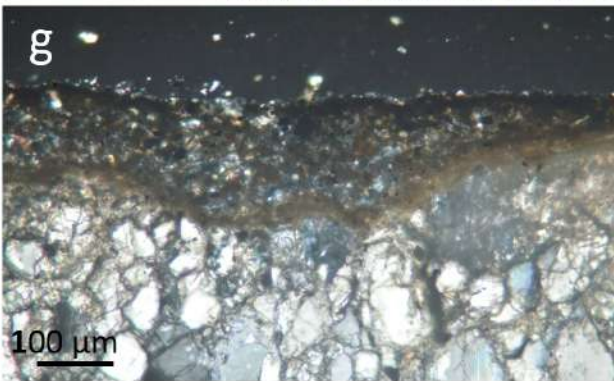
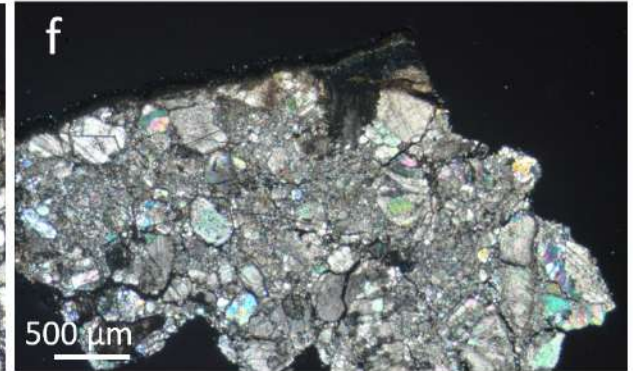
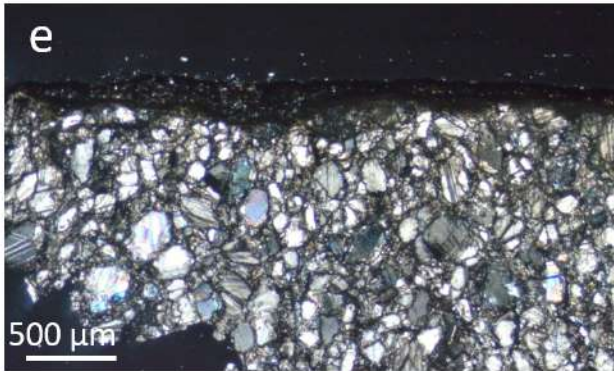
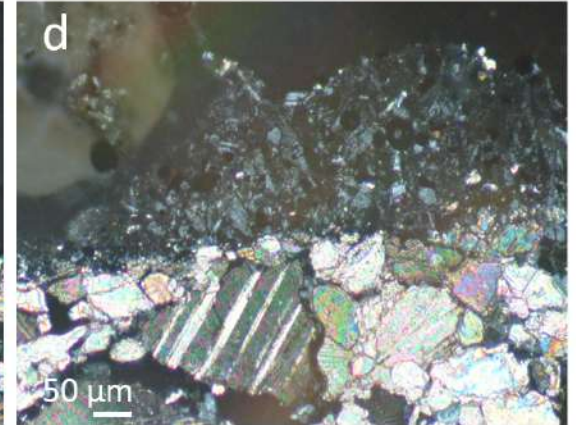
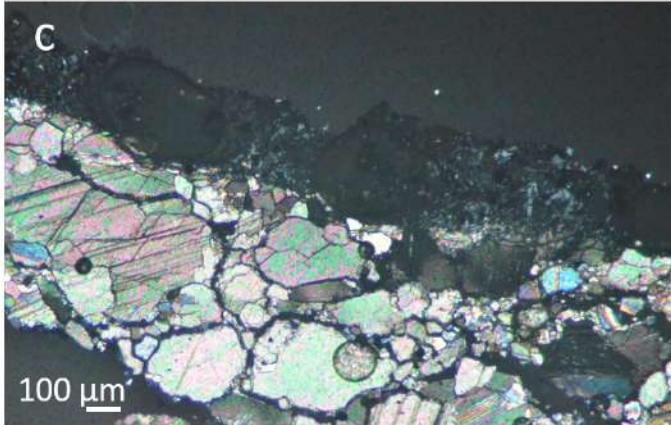
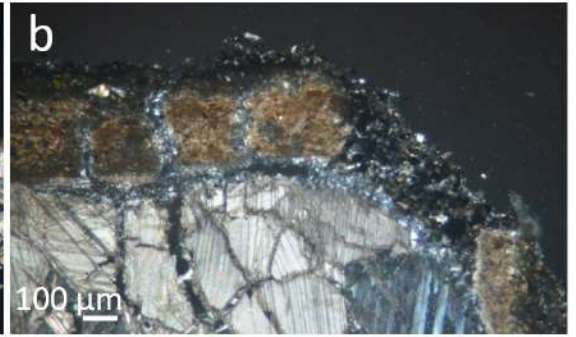
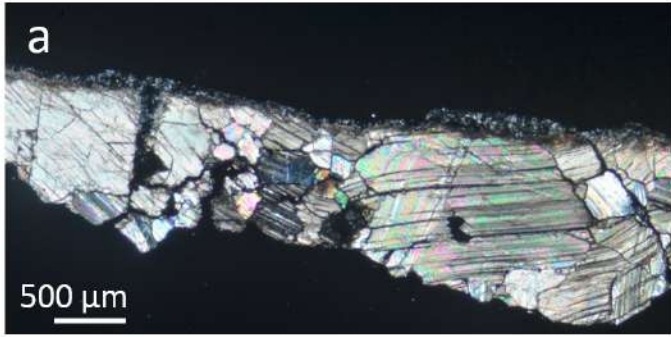
[87] J. Sternbeck, A. Sjödin, K. Andréasson, Metal emissions from road traffic and the influence of resuspension results from two tunnel studies, *Atmos. Environ.* 36 (2002) 4735–4744. DOI 10.1016/S1352-2310(02)00561-7.

[88] D.R. Gómez, M.F. Ginè, A. C. Sánchez Bellato, P. Smichowski, Antimony: a traffic related in the atmosphere of Buenos Aires, Argentina, *J. Environ. Monit.* 7 (2005) 1162–1168. DOI 10.1039/b508609d.

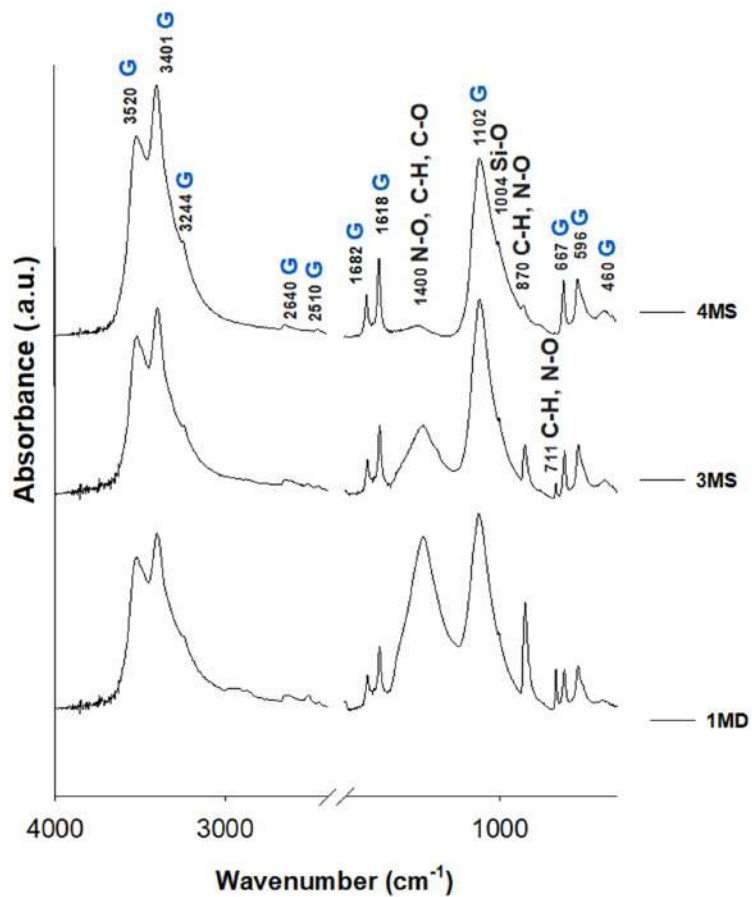
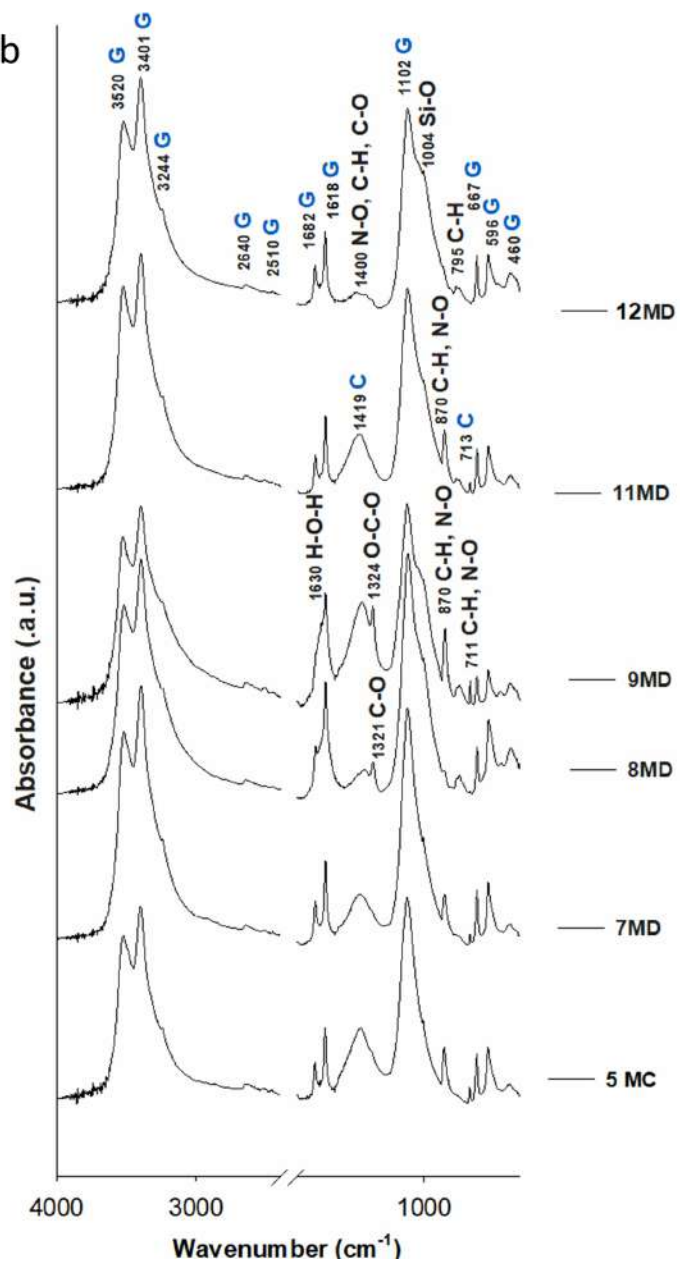
[89] G. Dongarrà, E. Manno, D. Varrica, Possible markers of traffic-related emissions, *Environ. Monit. Assess.* 154 (2009) 117–125. DOI 10.1007/s002549900071.

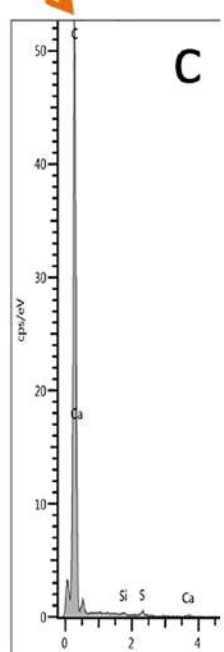
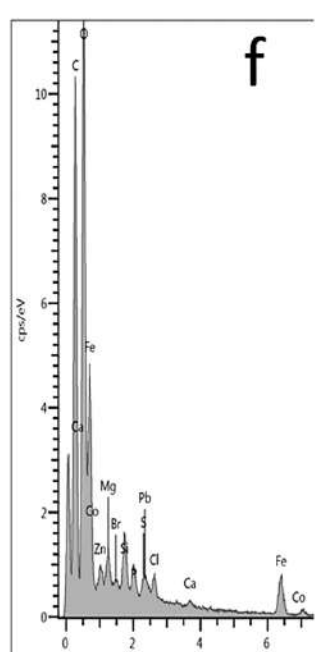
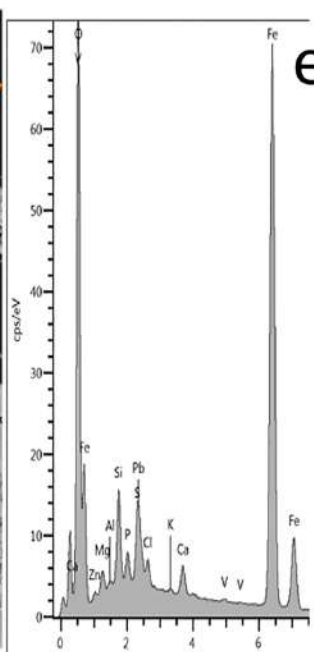
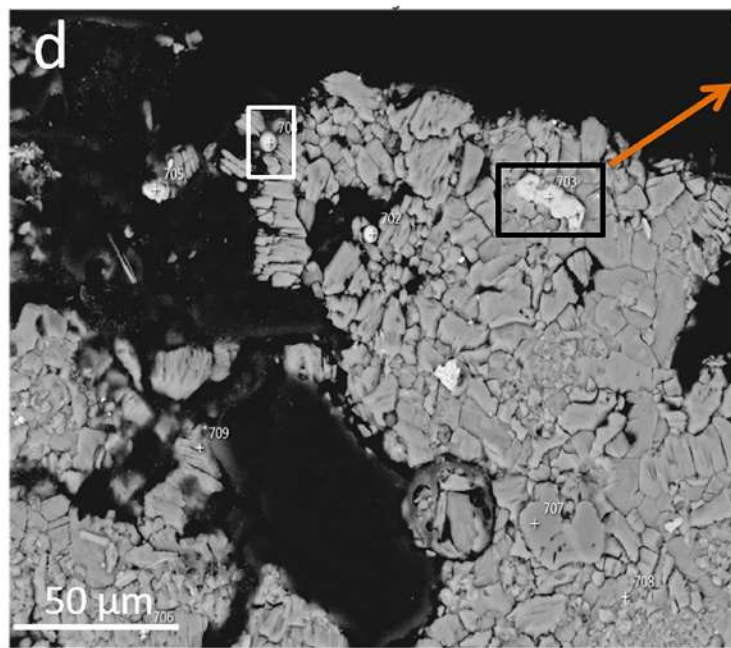
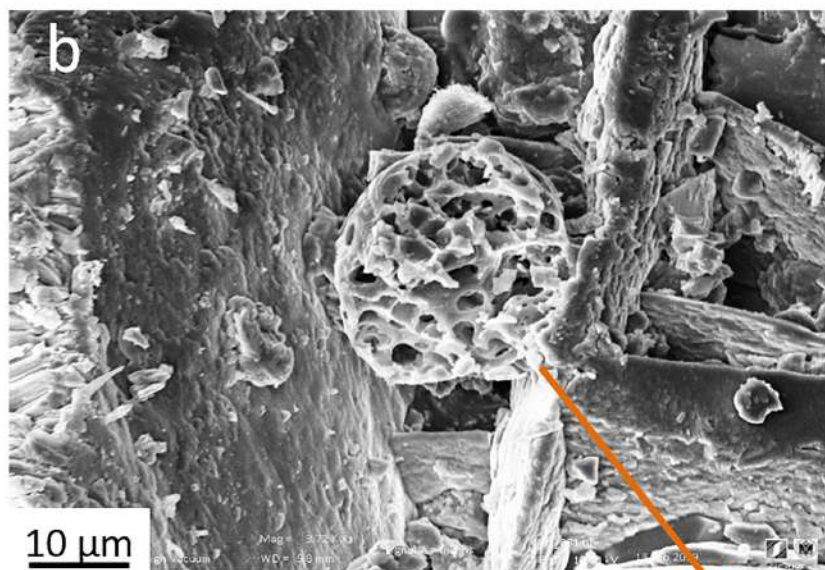
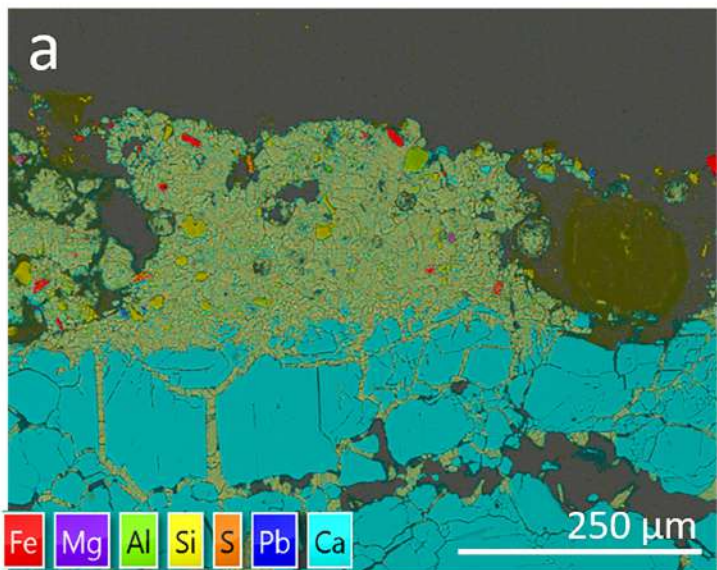
[91] H. Harmens, D.A. Norris, G.R. Koerber, A. Buse, E. Steinnes, A. Rühling, Temporal trends in the concentration of arsenic, chromium, copper, iron, nickel, vanadium and zinc in mosses across Europe between 1990 and 2000. *Atmos. Environ.* 31 (2007) 6673–6687. DOI 10.1016/j.atmosenv.2007.03.062.

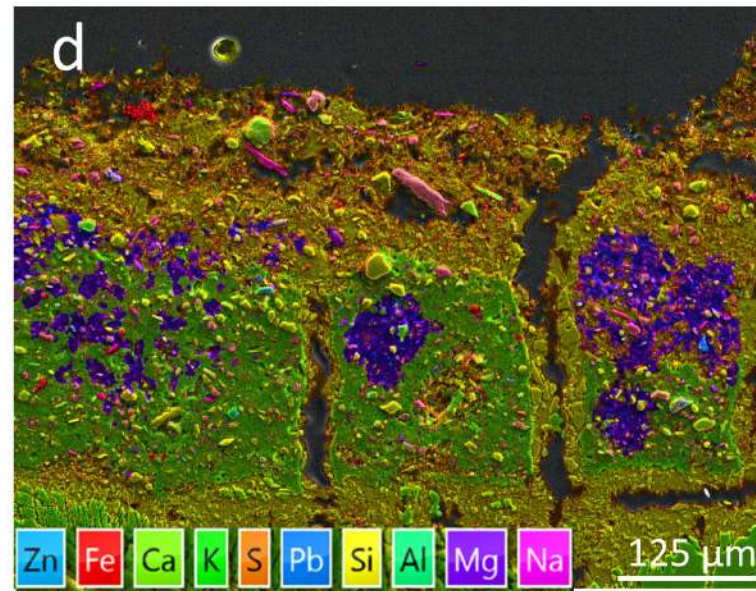
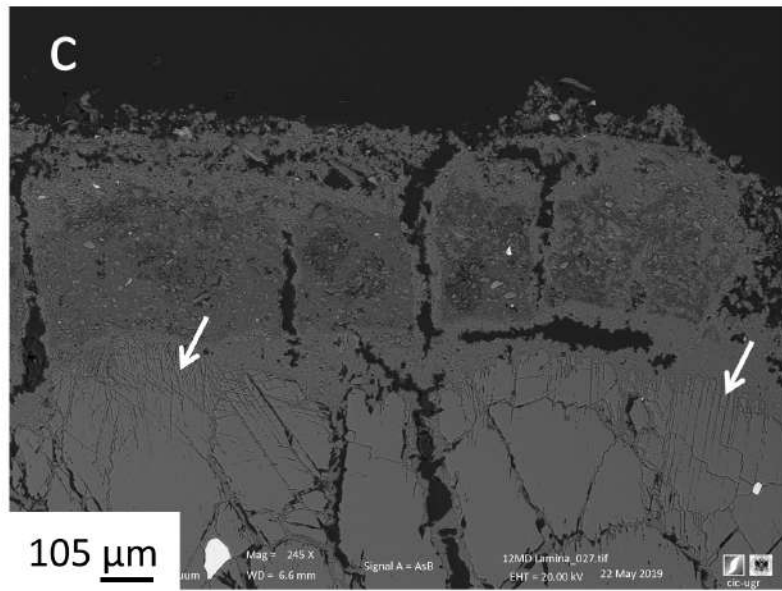
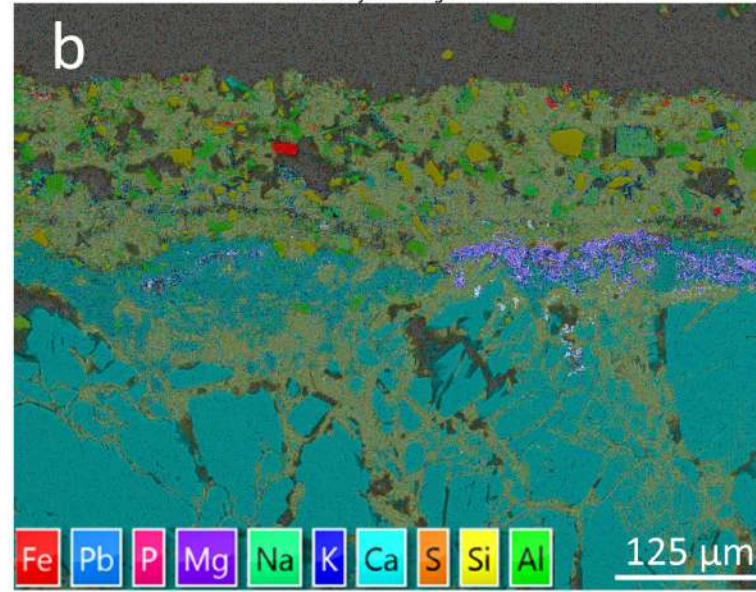
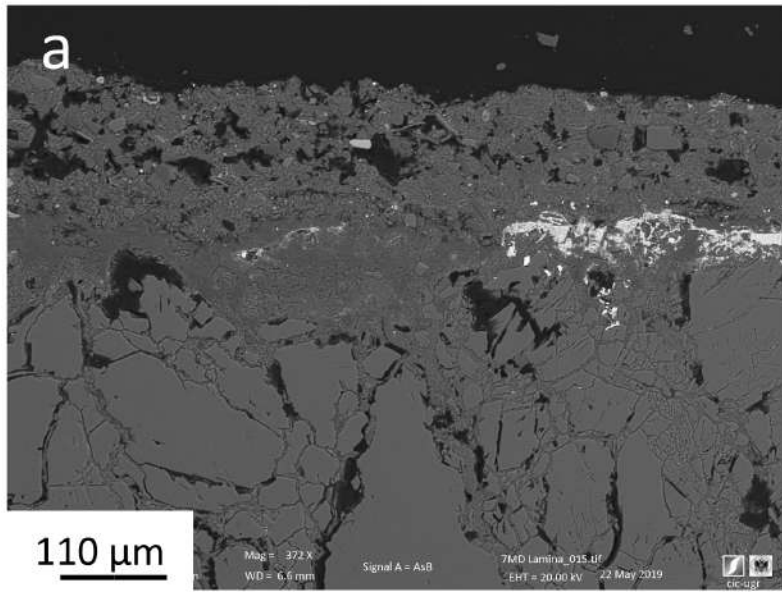
[91] H. Harmens, D.A. Norris, the participants of the moss survey, Spatial and temporal trends in heavy metal accumulation in mosses in Europe (1990–2005). Programme Coordination Centre for the ICP Vegetation, Centre for Ecology & Hydrology. Bangor, UK, Natural Environment Research Council, 2008.





**a****b**





**Fig. 1** Microphotographs of BCs samples (thin sections) from Monza Cathedral taken with PLM (crossed polars). a) 1MD (16m high), note the large crystal size and the gypsum crystals filling the calcite crystals fissures (arrow); b) 12MD (5m high), note the intense fissure system of the marble, the 3 layers of the crust and the abundant BCPs (arrow); c and d) 4MS (16m high), note the heteroblastic fabric of the marble and the small calcite crystals fragments being incorporated into the gypsum crust (arrow); e) 7MD and f) 9MS (12m high), note the dissimilar marble substrates highly fractured; g) 7MD and h) 9MS, note the orange-brown layer in contact with the marble.

**Fig. 2** ATR-FT-IR spectra of the black crusts sampled in Monza Cathedral. a) samples belonging to group A. b) samples belonging to group B. Effects assigned to the absorbance bands are depicted and the minerals detected by XRD are shown in blue (G: gypsum, C: calcite).

**Fig. 3.** SEM images showing the BC of 4MD sample: a) SEM-EDX false-color mineral map (thin section); note the significant fissure system of the marble -filled by gypsum- originating granular disaggregation, and the particles of varied nature enclosed into the BC, such as Fe-, Pb-, Si-, and Ca-rich particles; b) spherical spongy-like carbonaceous particle; c) EDX spectrum of carbonaceous particle shown in b); d) BSE-SEM image showing rounded Fe-rich particles (white box) and cluster of metal particles (black box); e) EDX spectrum of white box shown in d); f) average EDX spectrum of the black box shown in d).

**Fig. 4.** SEM microphotographs of 7MD and 12MD BCs (thin sections): a) BSE-SEM image of 7MD showing in brighter grey color the copious metal particles; note their accumulation in the interface between the marble and the BC; b) SEM-EDX false-color mineral map of 7MD; note the high amount of Pb and K in the interface marble-crust; c) BSE-SEM image of 12MD revealing the calcite crystals dissolution patterns (arrows) and the structure of the crust which is detaching from the substrate; d) SEM-EDX false-color mineral map of 12MD clearly showing the composition, morphology and distribution of the particles enclosed in the BC.

**Table 1** List of samples collected from the façade of the Dome of Monza.

<b>Sample</b>	<b>Height and location</b>	<b>Description</b>	<b>Stone</b>	<b>Years of Pollutants accumulation</b>
<b>1MD</b>	16 m rose window right side of the façade	Black crust on marble	The stone is a later integration (1735 year)	About 280
<b>3MS</b>	16 m rose window left side of the façade	Black crust on marble	The stone is a later integration (1735 year)	
<b>4MS</b>	16 m rose window left side of the façade	Black crust on marble	The stone is a later integration (1735 year)	
<b>5MC</b>	12 m rose window central part of the façade	Black crust on marble	Original	About 650
<b>7MD</b>	12 m rose window right side of the façade	Black crust on marble.	Original	
<b>8MD</b>	12 m rose window right side of the façade	Black crust and marble	Original	
<b>9MS</b>	12 m left double-arched window, original capital of the façade	Black crust on original marble	Original	
<b>11MD</b>	5 m double-arched window, right side of the façade	Black crust on original marble	Original	
<b>12MD</b>	5 m double-arched window, right side of the façade	Black crust and original	original	

**Table 2** Mineralogical composition of the crusts collected in Monza Cathedral. G: gypsum; Q: quartz; C: calcite; WD: weddellite; WW: whewellite. + + + +: > 50%; + + +: 30-50%; + +: 10-30%; +: 3-10%; tr: < 3%. -: not detected.

<b>Samples</b>	<b>G</b>	<b>Q</b>	<b>C</b>	<b>WD</b>	<b>WW</b>
<b>1MD</b>	++++	+	-	-	-
<b>3MS</b>	++++	+	-	-	-
<b>4MS</b>	++++	-	-	tr	-
<b>5MC</b>	++++	tr	-	tr	-
<b>7MD</b>	++++	++	-	-	-
<b>8MD</b>	+++	+++	-	-	-
<b>9MS</b>	++++	+	-	+	-
<b>11MD</b>	++++	+	tr	-	-
<b>12MD</b>	++++	++	-	-	-

**Table 3** Average concentrations (in ppm) of trace elements determined by LA-ICP-MS in the representative black crusts from Dome of Monza.

Element	Group A				Group B					
	1 MD 16m		4 MS 16m		7 MD 12m		9MS 12m		12 MD 5m	
	Average	St.dev	Average	St.dev	Average	St.dev	Average	St.dev	Average	St.dev
<b>As</b>	136	16	151	18	344	28	342	73	302	43
<b>Ba</b>	562	134	308	38	426	13	677	14	4405	109
<b>Cd</b>	13	4	15	6	41	15	223	33	131	56
<b>Cu</b>	118	16	570	168	578	38	548	168	5017	859
<b>Fe</b>	5584	770	36079	9897	69400	8774	84361	43077	255657	3477
<b>Mn</b>	278	27	425	91	907	4	721	268	1956	339
<b>Ni</b>	175	11	53	15	153	5	163	1	258	47
<b>Pb</b>	1464	151	2646	783	18015	2801	18973	3355	34554	1879
<b>Sb</b>	47	8	149	23	559	7	378	145	752	120
<b>Sn</b>	60	14	154	56	436	70	535	257	1033	603
<b>Sr</b>	870	73	1034	61	478	41	1488	796	607	51
<b>V</b>	55	5	118	35	351	33	210	15	550	38
<b>Zn</b>	542	34	462	74	2403	222	1750	430	3632	518

**Table 4** Concentration of ions (ppm) in the black crusts collected from the Dome of Monza analyzed with.

		<b>Ion concentrations ppm</b>				
	<b>Sample</b>	<b>Cl<sup>-</sup></b>	<b>NO<sub>2</sub><sup>-</sup></b>	<b>NO<sub>3</sub><sup>-</sup></b>	<b>SO<sub>4</sub><sup>2-</sup></b>	<b>Ca<sup>2+</sup></b>
<b>Group A</b>	1 MD	72	0	0	13508	15782
	3 MS	78	222	226	17019	10339
	4 MS	101	47	92	34295	12823
	<b>Average BC</b>	<b>84</b>	<b>90</b>	<b>106</b>	<b>21607</b>	<b>12981</b>
	<b>St.dev</b>	<b>9</b>	<b>68</b>	<b>66</b>	<b>6424</b>	<b>1573</b>
<b>Group B</b>	5 MC	53	0	628	12789	8545
	7 MD	91	201	215	9013	10863
	8 MD	278	35	372	18297	16017
	9 MS	141	45	315	6158	11007
	11 MD	0	42	0	10066	14142
	12 MD	32	69	181	12782	17485
	<b>Average BC</b>	<b>99</b>	<b>65</b>	<b>285</b>	<b>11518</b>	<b>13010</b>
	<b>St.dev</b>	<b>45</b>	<b>31</b>	<b>94</b>	<b>1858</b>	<b>1535</b>

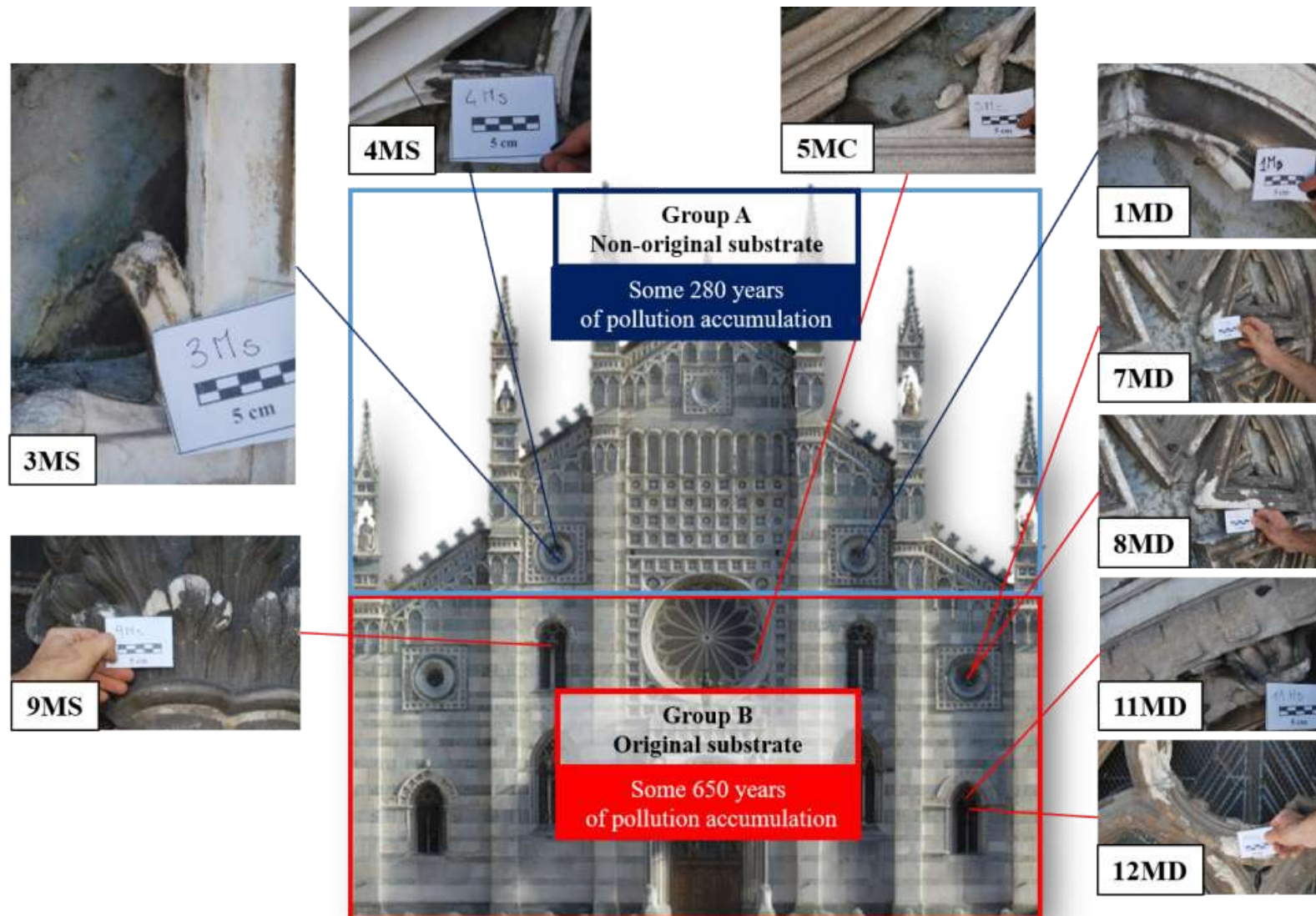


**Table 5.** CC (Carbonate Carbon), EC (Elemental Carbon), OC (Organic Carbon), OX (Oxalate) and TC (Total Carbon) concentrations (wt%) of the collected BCs. Some significant ratios for the analysed BCs are shown. Average values and standard deviations (wt%) were determined for group A and B.

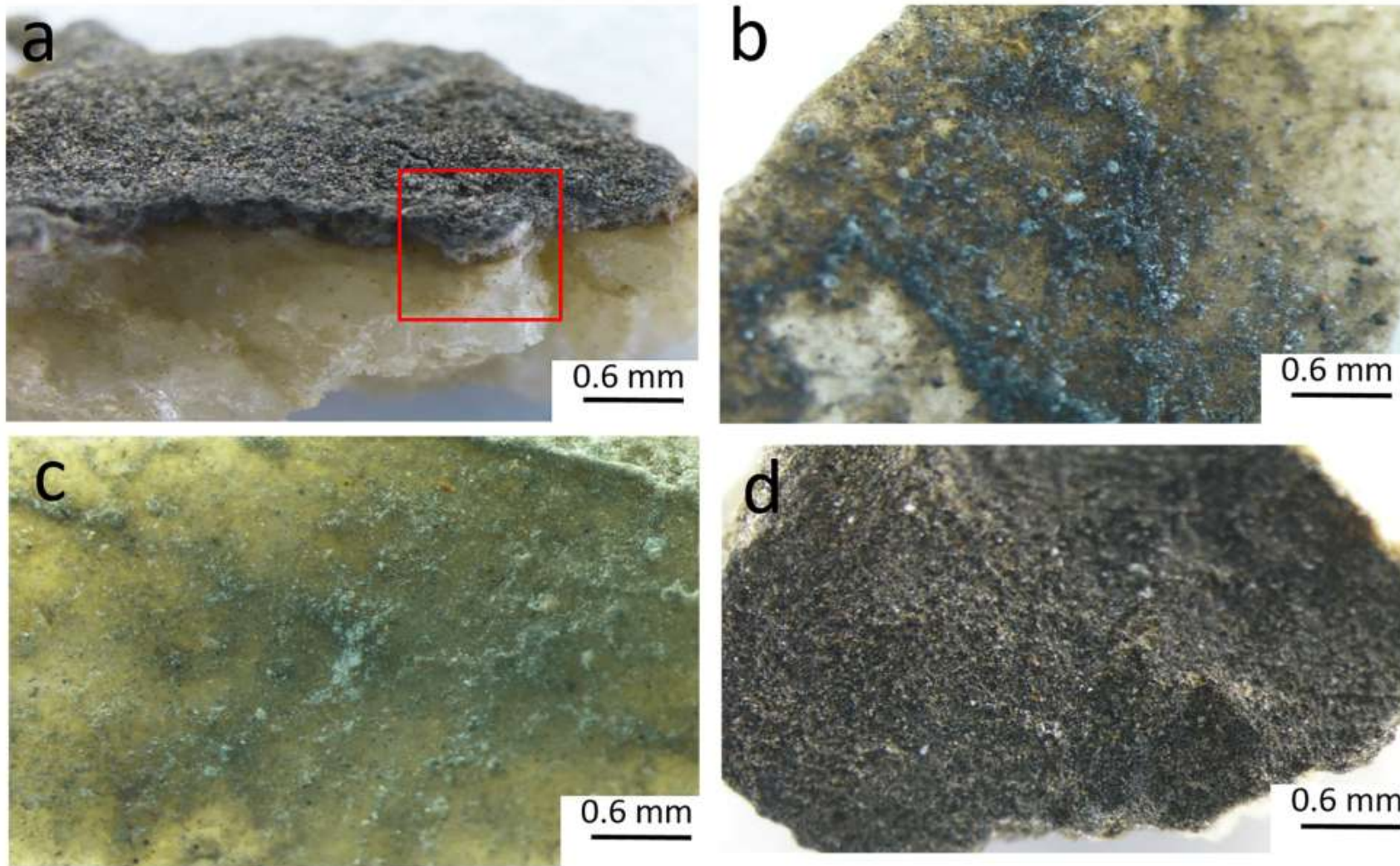
Sample	CC <sub>TGA</sub> %	EC <sub>TGA</sub> %	(EC+CC) <sub>TGA</sub> %	OC%	OX <sub>TGA</sub> %	TC <sub>CHN</sub> %	OC/EC	EC/TC
<b>1MD</b>	3,32	0,79	4,19	3,25	0,08	7,44	4,09	0,11
<b>3MS</b>	3,52	0,53	4,11	5,73	0,16	9,84	10,81	0,05
<b>4MS</b>	1,48	1,18	2,73	0,69	0,16	3,42	0,58	0,37
<b>Average BC group A</b>	2,77	0,83	3,68	3,22	0,13	6,90	5,16	0,18
<b>St,dev</b>	0,65	0,19	0,47	1,45	0,03	1,87	3,00	0,10
<b>5MC</b>	3,67	0,37	4,09	4,53	0,05	8,62	12,18	0,04
<b>7MD</b>	3,85	1,04	4,93	4,21	0,11	9,14	4,06	0,11
<b>8MD</b>	2,5	2,32	4,93	1,94	0,27	6,87	0,84	0,37
<b>9MS</b>	4,42	1,05	5,73	5,04	0,15	10,77	4,79	0,1
<b>11MD</b>	3,94	0,79	4,88	4,87	0,1	9,75	6,19	0,08
<b>12 MD</b>	3,57	0,53	4,87	4,1	0,17	8,97	3,43	0,13
<b>Average BC group B</b>	3,66	1,02	4,91	4,12	0,14	9,02	5,25	0,14
<b>St,dev</b>	0,26	0,28	0,21	0,46	0,03	0,53	1,15	0,05



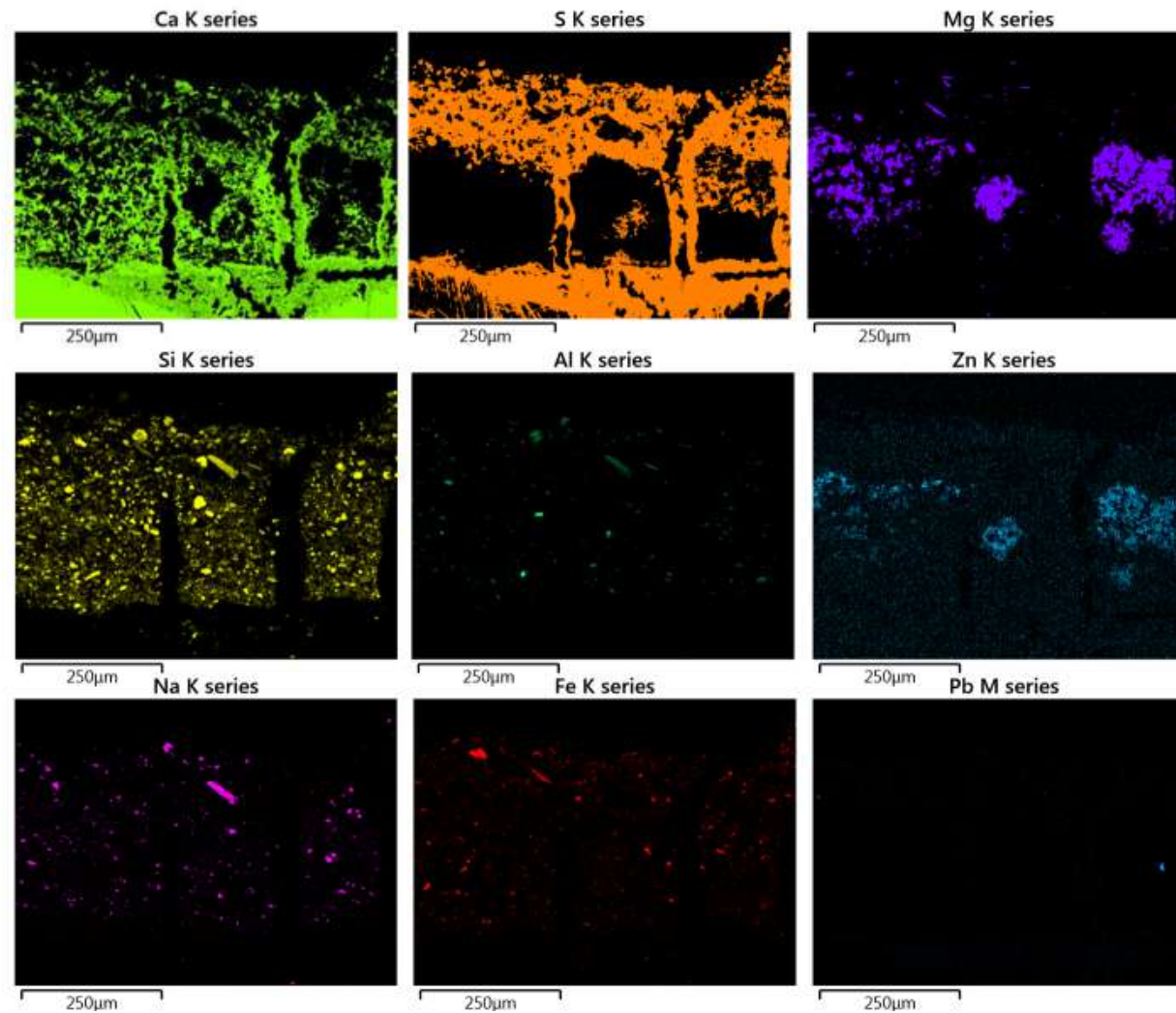
**Fig. S1.** Façade of the Dome of Monza (N Italy) located in the homonymous square.



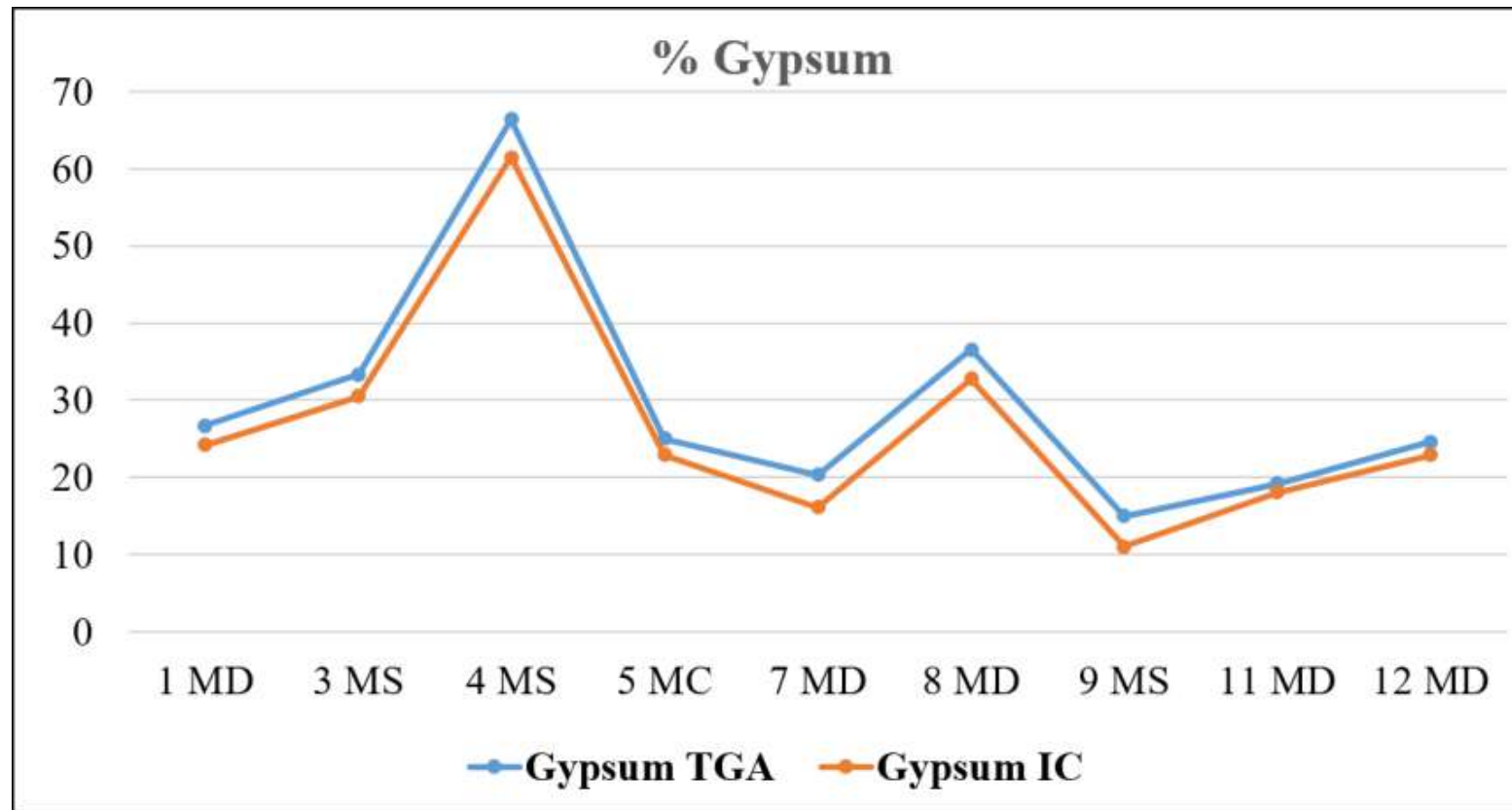
**Fig. S2.** The locations of sampling points, and photographs of the black crusts collected from the façade of the Dome of Monza.



**Fig. S3.** Stereomicroscope image of black crusts from Monza Cathedral. a and b) 12MD and 11MD samples (5m altitude), c and d) 1MD and 7MD samples (16m and 12 m altitude respectively)..



**Fig. S4.** SEM-EDX elemental mappings of 12MD black crust.



**Fig. S5.** Percentage of gypsum in the BCs obtained by IC and TGA.