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112	Abstract	issuing from the S to 2.5) in north-w surveys in 2017 a hydrous gas comp main sulfur specie evolved during ou one order of magn 3). This compositions is in volcano following reduced (backgrou was one order of r 20,200–30,200 t/ 900–10,167 t/day transport in March caused the volume absorbing and serr ratio to decrease. T salinity of the lake installation of a co- monitor any future	
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Electronic supplementary material

Table S1

Summary of all derived (molar) gas ratios in the Santa Ana crater lake plume.

For each ratio, the correlation coefficient of the best-fit regression line is indicated (R^2). SO₂ MAX is the peak SO₂ concentration measured in each measurement inteval where a ratio was calculated. (XLS 65 kb)

RESEARCH ARTICLE

The crater lake of llamatepec (Santa Ana) volcano, El Salvador: insights into lake gas composition and implications for monitoring

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13 Abstract

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We here present the first chemical characterization of the volcanic gas plume issuing from the Santa Ana crater lake, a hyper-14acidic crater lake (pH of -0.2 to 2.5) in north-western El Salvador. Our results, obtained during regular surveys in 2017 and 2018 1516using a Multi-GAS instrument, demonstrate a hydrous gas composition (H_2O/SO_2 ratios from 32 to 205) and SO₂ as the main sulfur species ($H_2S/SO_2 = 0.03-0.1$). We also find that gas composition evolved during our investigated period, with the $CO_2/$ 17SO₂ ratio decreasing by one order of magnitude from March 2017 (37.2 \pm 9.7) to November 2018 (< 3). This compositional 18evolution toward more magmatic (SO₂-rich) compositions is interpreted in the context of the long-term evolution of the volcano 1920following its 2005 and 2007 eruptions. We find that, in spite of reduced (background-level) seismicity, the magmatic gas supply into the lake was one order of magnitude higher in March 2017 (total volatile flux: 20,200-30,200 t/day) than in the following 2122periods (total volatile flux: 900–10.167 t/day). We propose that the elevated magmatic/hydrothermal transport in March 2017, combined with a 15% reduction in precipitation, caused the volume of the lake to decrease, ultimately reducing its sulfur 23absorbing and scrubbing capacity, and hence causing the gas plume CO₂/SO₂ ratio to decrease. The recently observed increases 24in temperature, acidity, and salinity of the lake are consistent with this hypothesis. We conclude that the installation of a 25continuous, fully-automated Multi-GAS is highly desirable to monitor any future change in lake plume chemistry, and hence 26the level of degassing activity. 27

28 Keywords Santa Ana volcano · Crater lakes · Volcanic gas plumes · Multi-GAS · Gas scrubbing · CO₂/SO₂ ratio · Wet volcano

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Introduction

The term wet volcano was introduced by Caudron et al. (2015) 31and is used to define a volcanic system characterized by the 32 presence of an active voluminous magmatic-hydrothermal 33 system. At such systems, the physical-chemical properties of 34 crater lakes are key to volcanic activity monitoring (e.g., 35Rowe et al. 1992; Takano et al. 1994; Ohba et al. 2008; 36 Christenson et al. 2010, 2015; Shinohara et al. 2015; Agusto 37 and Varekamp 2016; de Moor et al. 2016a, 2019; Caudron 38 et al. 2017). Temporal variations in the lake's physical-39 chemical state are thought to result from time-changing rates 40 of heat and fluid supply from the underlying magmatic-41 hydrothermal system (e.g., Rowe et al. 1992; Christenson 422000; Ohba et al. 2008). However, Rouwet et al. (2016) has 43recently postulated that classic monitoring techniques, involv-44 ing analysis of dissolved components in hyper-acidic crater 45lakes, are often of too low a temporal resolution to capture 46

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47 precursory signals to phreatic eruptions, the main hazard re-48 lated to peak-activity crater lakes (e.g., Christenson et al. 49 2010; de Moor et al. 2016a, 2019). This observation has mo-50 tivated further work to provide higher temporal resolution 51 time series to track long- and short-term changes at crater 52 lakes, to identify the parameters that need to be measured, 53 and the processes they can be used to track.

The magmatic gas species usually monitored in fuma-54roles and plumes of open-vent volcanoes (Fischer and 55Chiodini 2015) can also be detected in the gas plumes **Q3**56 released by hyper-acidic lakes. For hyper-acidic lake con-57ditions, CO₂ is not absorbed into lake water, but SO₂ 5859variably reacts with lake water to form H₂SO₄ (e.g., Tamburello et al. 2015; de Moor et al. 2016a, 2019; 60 Gunawan et al. 2016), and HCl degassing accelerates if 61 62pH < 0 conditions are met (Capaccioni et al. 2017). The Multi-GAS instrument (Aiuppa et al. 2005a; Shinohara 63 2005), while traditionally used to monitor volcanic gas 64 65 composition at "dry volcanoes" (e.g., Aiuppa et al. 2009, 2018; de Moor et al. 2016b), has recently proven 66 to be useful in measuring gas plumes from crater lakes 67 (Shinohara et al. 2015; Tamburello et al. 2015; Gunawan 68 69 et al. 2016; Hasselle et al. 2018) and in detecting precursory changes to phreatic eruptions (de Moor et al. 2016a, 702019). High-frequency, continuous observations of gas 7172compositions discharging from lakes can be of paramount 73importance in monitoring volcanic activity and in forecasting phreatic/phreatomagmatic eruptions (Stix and de 74Moor 2018; Battaglia et al. 2019). 75

Here, we characterize for the first time the composition of the lake gas plume released by Santa Ana crater
lake (March 2017–November 2018). We interpret the
temporal changes observed in tandem with lake level

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variations, SO_2 flux record, and seismicity, in an attempt to derive constrains on the current activity state 81 and to speculate on the potential changes that might 82 herald future reactivation of this restless volcano. 83

Geological and volcanological settings

Ilamatepec or Santa Ana volcano (13° 51' N, 89° 37.5' W; 85 2381 m asl) is located in western El Salvador (Fig. 1) and 86 is surrounded by two highly populated cities, Santa Ana 87 (pop. 522,000) and Sonsonate (pop. 420,000), both lying 88 within a radius of 25 km from the volcano (Pullinger 89 1998; Colvin 2008). It is one of the most active volcanoes 90 in El Salvador, with 13 VEI 2-3 eruptions reported since 91 AD 1500 (Mooser et al. 1958; GVP 2018), mostly phre-92atic to phreatomagmatic in nature (Pullinger 1998). The 93 last magmatic eruption occurred on October 1, 2005 94 (Scolamacchia et al. 2010) and was followed by small 95 phreatic eruptions on March 15 and April 27, 2007. The 96 youngest of the current four summit craters (0.5-km di-97 ameter) has hosted a small hyper-acidic crater lake 98 (Bernard et al. 2004; Colvin 2008; Colvin et al. 2013) 99 since 1904 (Carr and Pontier 1981). 100

The Santa Ana-Izalco-Coatepeque volcanic complex (< 101 200 ka; Pullinger 1998) includes two stratovolcanoes (Santa 102Ana and Izalco), the Coatepeque caldera (that is filled with a 103lake, Cabassi et al. 2019), and many parasitic cones, cinder 104cones, and explosion craters (Fig. 1; Pullinger 1998). The 105complex is part of the Central American Volcanic Arc, which 106results from subduction of the Cocos Plate below the 107 Caribbean Plate (Carr 1984; DeMets et al. 1990). 108

Fig. 1 Google Earth image (Image © 2019 Maxar Technologies) of the Santa Ana volcano and its surroundings. The location of the DOAS and seismic stations run by MARN are indicated. Inset: location of Santa Ana volcano in Central America



109 Temporal evolution of the crater lake

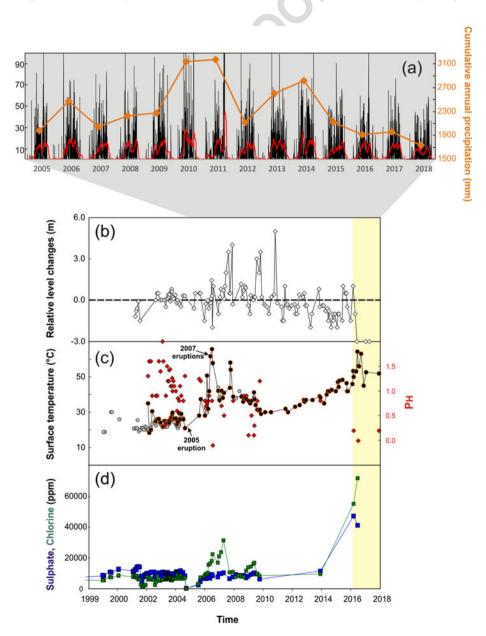
A large and permanent hydrothermal system beneath the vol-110 111 cano is implied by the many phreatic to phreatomagmatic 112eruptions of Santa Ana in the last thousand years (Pullinger 1998; Bernard et al. 2004). The hydrothermal system, topped 113114 by the hyper-acidic crater lake, manifests as hot springs along the lake shore, lake gas bubbling, and fumarolic emissions 115west of the lake. In 2000-2002, the lake floor was bowl-116 117shaped, with a diameter of 200 m and a maximum depth of 118 27 m (Bernard et al. 2004). Along the shoreline, bubbling hot 119 springs were observed (T~80 °C; Bernard et al. 2004). Prior to the October 2005 eruption, high-temperature fumaroles 120 discharged vigorously although showing a marked tempera-121 ture decline between 2002 and 2003 (being 532 °C in January 122

Daily precipitation (mm)

2000. 875 °C in June 2002. 264 °C in December 2003. 360 °C 123in January 2004; Bernard et al. 2004; Scolamacchia et al. 1242010; SNET Monthly Report). The 2005 and 2007 erup-125tions drastically modified lake geometry, temperature, and 126water chemistry, possibly due to changes in rate and com-127position of the volcanic gas input (Colvin 2008; Colvin 128129et al. 2013; Laiolo et al. 2017) (Fig. 2). The discharge rate and temperature of the fumaroles also decreased (to < 100130°C). However, the period 2010-2014 was poorly docu-131mented, inhibiting detailed evaluation of the activity of 132Santa Ana crater lake (Fig. 2). 133

Colvin (2008) proposed a physical model of the Santa Ana magmatic-hydrothermal system. According to Colvin (2008), a shallow degassing magma body (3–7 km depth, Carr and Pontier 1981; Halsor and Rose 1988) was overlain by a single-137

Fig. 2 Temporal variations in crater lake level, chemicophysical parameters and chemistry, 1999-2018 (the vellow-colored band identifies the temporal window covered by our Multi-GAS observations). a Daily precipitations (in mm) in the Santa Ana area (black line, left axis; the red solid line is a 30-day mobile average). The cumulative yearly precipitations. b Changes in lake level, in meters, expressed relative to a fixed benchmark position in the inner crater walls. c Temporal changes in lake surface temperature (left scale, grav circles, Bernard et al. 2004; Scolamacchia et al. 2010; black circles, MARN monitoring database) and pH (right scale; red diamonds). d Dissolved SO4 and Cl concentrations in the lake water are also shown (orange diamonds, right y-scale). Data from MARN monitoring database



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138phase vapor zone and acid-sulfate-chloride hot springs (before the 2005 eruption; Bernard et al. 2004). This vapor zone 139would be separated from a shallower (near-surface) two-140 141 phase (liquid + gas) region by a low-permeability mineral seal 142 (Pullinger 1998; Bernard et al. 2004). The fumarolic field, **Q4**143 present since at least the 1950s (Meyer-Abich 1956; Bernard 144et al. 2004), was ruptured during the 2005 eruption (Colvin et al. 2013). After the eruption, this part of the crater was 145flooded but the presence of sub-lacustrine fumaroles is 146 147highlighted by strong bubbling driving a vigorous convection 148 cell at the lake surface. This was still visible in 2018.

149 Periodical changes in activity are common in the recent 150history of Santa Ana crater lake. High activity levels were reported in 1920 and July 1992 (Gutiérrez and Escobar 1511994; Bernard et al. 2004). More recently, low-level activity 152periods (January-May 2000 and February 2002-June 2004) 153alternated with high activity periods (May 2000-February 1541552002 and June 2004-August 2005) and finally culminated in 156eruption on October 1, 2005 (Colvin 2008). Before (October 2005-March 2007) and after (May 2007-December 2007) the 157March-April 2007 phreatic eruptions, Colvin (2008) reported 158159a high level of activity.

160 The recent evolution of the lake system, illustrated in Fig. 2, is characterized thanks to monitoring results provided by 161MARN (Ministerio de Medio Ambiente e Recursos 162163Naturales), the Salvadoran environmental and natural resources research office. MARN regularly monitors fumarole 164165temperature, lake water temperature, and pH, as well as me-166teoric precipitation at the Los Naranjos station (approximately 3 km NNW of Santa Ana crater lake) (Fig. 2a). Variations in 167crater lake level (Fig. 2b) are also assessed by MARN com-168169paring the lake level from photographs taken at several dates with the lake level on a scaled reference photograph. 170171According to MARN database and previous literature infor-172mation (see figure caption for details), the pH of the lake water 173(Fig. 2c) showed strong fluctuations, between 2 and 0.5, be-174fore the 2005 and 2007 eruptions; after the March-April 2007 175phreatic eruptions, the lake water pH dropped to -0.2, the 176lowest pH measured so far. A sudden temperature rise of 10 177°C occurred prior to the 2007 eruptions (Fig. 2c). Afterwards, 178the lake level oscillated considerably, but mostly rose after the 2007 eruption and until 2010. By mid-2010, lake water tem-179perature had returned to pre-phreatic lake temperatures of 28-180181 32 °C. From 2011, two main periods of lake level drop were observed in 2014-2015 and in June 2017 to January 2018. 182Since February 2011, the lake water temperature started a 183184steady increasing trend, with a peak temperature registered in June 2017, yet below the pre-phreatic eruptive temperature 185of 65.6 °C. The most recent pH values are near zero, among 186187the highest on record. Sulfate and chloride concentrations in 188 the crater lake water both peaked in 2017, at respectively 18941,000–47,000 mg/L and 54,000–71,000 mg/L (Fig. 2d). No 190data on water chemistry are available for 2018.

Methods

We investigated the composition of gases emitted from the 192surface of Santa Ana crater lake. Gas compositions were mea-193sured in situ by Multi-GAS (multicomponent gas analyser 194 system) (Aiuppa et al. 2005a; Shinohara 2005). We used a 195compact sensor unit containing a non-dispersive infrared 19605 (NDIR) spectrometer (for CO_2 ; range = 0–3000 ppm); three 197electrochemical gas sensors for H_2S (range = 0–100 ppm), 198 SO₂ (range = 0-200 ppm), and H₂ (range = 0-200 ppm); 199and a relative humidity sensor (range = 0-100%) for indirectly 200 measuring H₂O. 201

An explorative Multi-GAS survey was conducted at Santa 202Ana in March 2017 to investigate the composition of the lake 203gas plume for the first time. Measurements were obtained 204using a mobile Multi-GAS from three distinct sites in the 205crater area, located at different distances from the lake (Fig. 2063): (i) on the S and SW outer crater rims, > 400 m from the 207lake; (ii) on the plateau, ~ 200 m NNE from the lake; and (iii) 208 on the eastern lake shore. The same operations, at the same 209measurement sites, were repeated in June 2017, and a contin-210uously recording Multi-GAS (measuring at 0.5 Hz rate in 4 211daily measurement cycles of 30 min each) was run on the 212plateau site between June 5 and 13 (Fig. 3). Based on the 213results of these initial surveys, the plateau site (Fig. 3) was 214selected as the best location for deployment of a semi-215permanent station, owing to relatively safe access compared 216to the lake shore and denser plume conditions compared to the 217outer rim. This semi-permanent station operated in April 2018 218for 3 days, while punctual measurements for periods of ~ 2 to 2193 h were also performed at the same site on May 3 and 220June 28, 2018. In November 2018, measurements were taken 221at both the plateau and the lake shore. Analytical data are 222summarized in Table S1. During the periods of observation, 223only a few low-temperature weakly degassing fumaroles (and 224a few hot springs) were visible, mostly on the SW shore and 225on the inner crater slope. 226

During our measurement period, both seismicity and 227SO₂ fluxes remained at background levels (Fig. 4), at least 228relative to records obtained during the 2005 eruptive un-229rest (Olmos et al. 2007). Seismicity was registered at the 230MARN seismic station located at San Blas, 1 km SE from 231the crater (Fig. 1). The SO_2 fluxes were measured with the 232permanent DOAS instrument of the NOVAC network 233(Galle et al. 2010), installed at 6 km SW from the crater 234(Fig. 1) and elaborated by the MARN monitoring service. 235During our specific Multi-GAS measurements in 2017-2362018, the SO₂ flux varied from 41 and 329 t/day and 237was slightly higher in 2017 (mean 165 ± 140 t/day) than 238in 2018 (mean 144 \pm 77 t/day) (Table 1). Considering the 239Q6 absence of high-temperature fumaroles during the studied 240period, we consider the degassing crater lake as the ex-241clusive source of the SO₂ detected by the DOAS. 242

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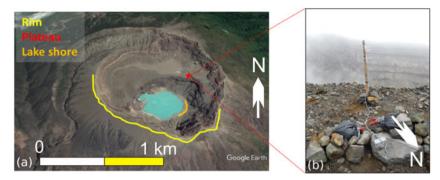
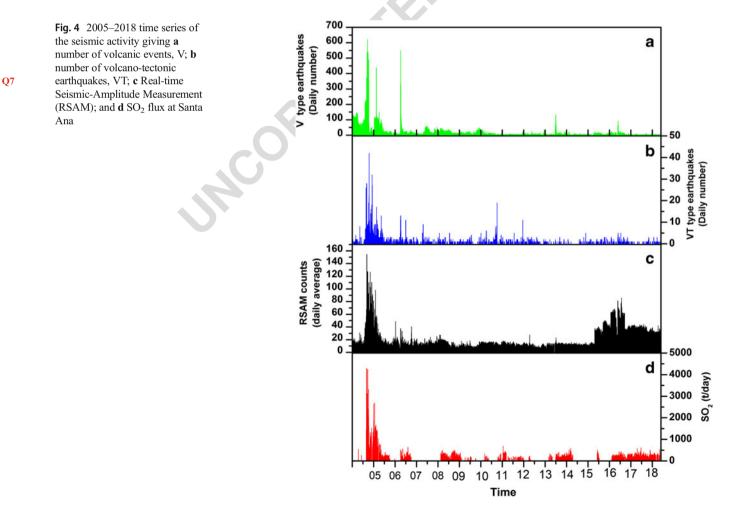


Fig. 3 a Google Earth image of Santa Ana volcano (Image © 2019 Maxar Technologies) with the location and tracks of Multi-GAS measurements in March 2017. The yellow and orange tracks correspond to measurements carried out by walking along the rim and

the lake shore, respectively. The red star is the plateau measurement site. It is also the location where the \mathbf{a} semi-permanent Multi-GAS was installed in June 2017 and in 2018 (b).

243 Results

Examples of Multi-GAS acquisitions at the three sites (rim, plateau, and lake shore), taken in the explorative March 2017 campaign, are given in Fig. 5. At the crater outer rim, in March and June 2017, we detected low amounts of SO₂ (\sim 1 ppm) and H₂O (\sim 1000s ppm above atmospheric background) (Fig. 5). H₂S concentrations were very low (\sim 0.1–0.4 ppm) and poorly correlated with SO₂ peaks, and the volcanic plume was 250typically too diluted for a volcanic CO₂ signal to be resolved 251over the atmospheric background (Fig. 5a). At the plateau and 252the lake shore, all the target volcanic gases (CO₂, SO₂, H₂, and 253H₂O) were detectable in the plume at concentrations higher 254than 1 ppm (see examples shown in Fig. 5b, c). At the plateau, 255gas concentrations varied considerably over time. In the ac-256quisition example of Fig. 5b, SO₂, CO₂, H₂, H₂O, and H₂S 257



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t1.2	Date Loca	Location Peak SO ₂ CO ₂ / SO ₂	ak SO ₂	CO ₂ / SO ₂	1 SD H ₂ / SO ₂	$_{2}^{\mathrm{H}_{2}^{/}}$ SO ₂	1 SD H ₂ O/ SO ₂		1 SD H ₂ S/ SO ₂	1 SD SO ₂ flux 1 SD CO ₂ flux 1 SD H ₂ flux 1 SD H ₂ O flux 1 SD H ₂ S flux 1 SD TV flux 1 SD	flux 1 SI	$O CO_2$ flu	IX 1 SD	H ₂ flux	1 SD	H ₂ O flux	1 SD H ₂ S	S flux 1	SD TV	flux 1 SI
t1.3	07/03/2017 Shore	re 16.5	6.5	31	13.7	0.42	0.11		2	240	96	5112	3049	3.1	1.5					
t1.4	07/03/2017 Plateau	eau 8.2	2	37.2	9.7	2.39	0.27 205	14		240	96	6135	2931	18	7.5	13,825	5615		20,217	17 8649
t1.5	08/03/2017 Shore	re 26.6		54.5	24.9	0.54	0.06 190	11	0.03	0.01 329	132	12,320	7486	5.5	2.3	17,565	7106 5.2	2.7	7 30,225	25 14,728
t1.6	05/06/2017 Plateau	eau 10.3		4.7	1.9	0.46	0.31		0.03	0.02 51	20	165	94	0.7	0.6		0.8	0.6	9	
t1.7	13/06/2017 Shore	re 20	~	5.4	0.1	0.06	0.02 77.4	4 3.9	-	41	16	152	61	0.1	0.04 8	892	360		1085	
t1.8	13/06/2017 Plateau	eau 15.6		4.2	1.4	0.84	0.75 75.8	8 48.2		0.03 41	16	118	62	1.1		873	657 1.3	0.8	8 1035	5 737
t1.9	06/04/2018 Plateau	eau 14.7	1.7	4.1	1.6	0.52	0.57 177	11		0.02 87	35	245	137	1.4	1.6 4	4327	1753 3.7	1.7	7 4664	
t1.10	12/04/2018 Plateau	eau 26.7	5.7	3.3	1.1	0.44	0.27 82	34	0.1	0.07 214	86	485	253	2.9	2.2	4931	2843 11	9.2	2 5644	
t1.11	13/04/2018 Plateau	eau 24.8	1.8	3.2	0.7	0.39	0.17 62	37		0.07 61	24	134	61	0.7	0.4	1063	764 2.6	2.5	5 1261	853
t1.12	14/04/2018 Plateau		17.6	3.3	0.5	0.44	0.21 32	18		81	32	184	79	1.1	0.7	728	503		994	
t1.13	03/05/2018 Plateau		29.5	2.9	0.8	0.39	0.15 75	26		117	47	233	113	1.4	0.8	2466	1306		2817	
t1.14	28/06/2018 Plateau		25.8	2.4	0.5	0.37	0.14 107	19		231	92	381	172	2.7	1.5 (6945	3042		7560	
t1.15	28/11/2018 Shore	re 41.5	.5	2.7	0.2		143		0.08	0.03 236	94	438	178		-	9483	3834 10	5.5	5 10,167	
t1.16	t1.16 28/11/2018 Plateau	eau 25	15	2.9	0.2	0.6	0.1 90	25		236	94	470	191	4.4	1.9	5968	2909		6679	3196

Table 1 Daily averaged gas ratios (molar) at the plateau and lake shore, calculated by averaging all the ratios from Table S1 where peak SO₂ concentrations is higher than 7 ppm (filtered dataset). The gas fluxes (in tons/day) were calculated by scaling the daily mean gas ratios to the daily averaged SO₂ fluxes from the MARN DOAS station. 1σ is the standard deviation

t1.1

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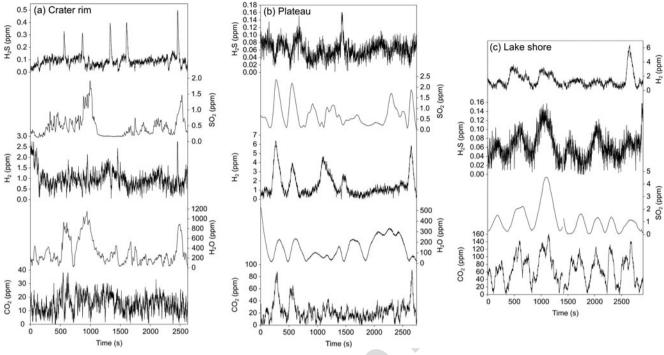


Fig. 5 Examples of Multi-GAS measurements at the crater rim (a), plateau (b), and lake shore (c) in March 2017

concentrations peaked at $\sim 2.5, 80, 6, 300, and 0.14$ ppm, 258respectively. Higher gas concentrations were observed episod-259260ically (in other acquisition windows), with peak concentrations of up to ~ 20 ppm for SO₂, ~ 150 ppm for CO₂ (above the 261atmospheric background of 403 ppm), ~ 20 ppm for H₂, \sim 2623000 ppm for H₂O (above background), and ~ 1.3 ppm for 263 H_2S (see Table S1). Measurements at the lake shore found the 264densest plume conditions (~ 30 ppm for SO_2) (Table S1). For 265266practicality and safety reasons, the plateau site was selected for observations in 2018, except in November when the plateau and 267268 the lake shore were both accessed for measurement (Table S1).

The acquired concentration time series were processed using the scatter plot technique described by Hasselle et al. (2018). To do this, sequences of scatter plots (e.g., Fig. 6) were

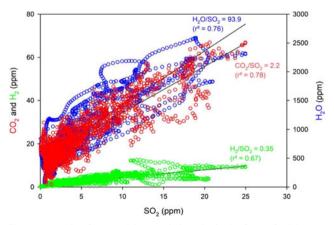


Fig. 6 Example of gas vs. SO_2 correlation plot in the form of a CO_2 , H_2 , H_2O vs. SO_2 scatter plot. Data taken at the plateau on June 28, 2018

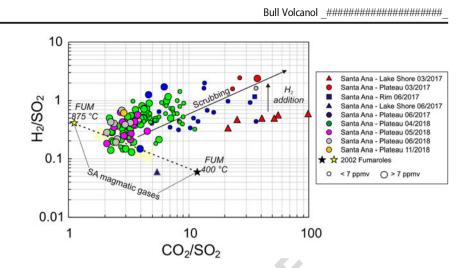
built for all sub-intervals where well-correlated concentration 272peaks were observed. Gas ratios were then obtained from the 273gradients of the best-fit regression lines. Gas ratios derived in 274this way are listed in the Appendix or Supplement (Table S1) 275and illustrated in the H₂/SO₂ vs. CO₂/SO₂ scatter plot of Fig. 2767. The plot shows that both ratios span more than one order of 277magnitude and define a general compositional trend from 278SO₂-rich to H₂-CO₂-rich compositions. Our results also show 279that, even at the plateau location where a dense gas plume was 280detected, the obtained H₂/SO₂ ratios and, to a minor extent, 281the CO₂/SO₂ ratios were anti-correlated with SO₂ concentra-282tions (Fig. 8), as observed elsewhere (e.g., at Masaya; Aiuppa 283et al. 2018). We cannot exclude that the high gas ratios at low 284SO₂ concentrations were (even partially) due to difficulties in 285resolving volcanic H₂ and CO₂ signals over the atmospheric 286backgrounds in dilute plume conditions-if so, the derived 287H₂/SO₂ and CO₂/SO₂ ratios would over-estimate the real vol-288canic signatures. Because of this possible concern, we find it 289 more prudent to analyze the temporal trends in gas composi-290tion (Fig. 9) concentrating on the ratios obtained for sub-291intervals with SO₂ above a 7-ppm concentration threshold 292 where ratios become independent of SO₂ concentrations 293(Fig. 8). The daily averaged gas ratios derived from the filtered 294 $(> 7 \text{ ppmv SO}_2)$ dataset are listed in Table 1. 295

Our filtered dataset highlights considerable changes in gas 296 composition during the investigated period (Table 1 and Fig. 297 9). In March 2017, we were only able to obtain CO_2/SO_2 and 298 H_2/SO_2 ratios at both the plateau and lake shore (Fig. 9 and 299 Table 1). The daily averaged CO_2/SO_2 ratios were similarly 300 high at both lake shore (31.0 ± 13.7) and plateau (37.2 ± 9.7). 301

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Fig. 7 H₂/SO₂ vs. CO₂/SO₂ ratios of Santa Ana lake gas emissions in 2017–2018 (data from Table S1). The compositions of high temperature fumaroles collected in June 2002 are also shown (sampled and analyzed by T.F.). From these, the yellowcolored area represents the inferred hypothetical composition of magmatic gases entering the Santa Ana crater lake in 2017– 2018. FUM = fumarole as sampled by T.F.



The H₂/SO₂ ratios were 0.42 ± 0.11 at the lake shore and higher (2.39 ± 0.27) at the plateau. In June 2017, the CO₂/ SO₂ ratios were drastically lower than in March (see Table 1), and again similar at both measurement sites (5.4 ± 0.1 at the lake shore and 4.2 ± 1.4 at the plateau; Fig. 9). The same

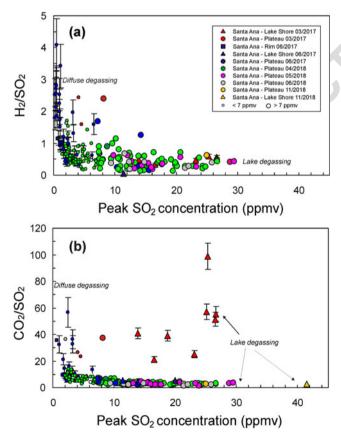


Fig. 8 Scatter plots of derived **a** H_2/SO_2 and **b** CO_2/SO_2 ratios vs. the peak SO₂ concentration in each integration interval (data from Table S1). At low (< 7 ppmv SO₂, small symbols) gas concentrations, the derived ratios are negatively correlated with SO₂ (taken as a proxy of plume density). At higher (> 7 ppm of SO₂, large symbols), gas ratios are independent on concentrations and are thus well representative of lake degassing only. The vertical error bars illustrate errors in the derived ratios (shown for a few selected data point only for clarity)

contrast in H_2/SO_2 ratios (already seen in March) between 307 lake shore (0.06 ± 0.02) and plateau (0.46 to 0.84 ± 0.75) 308 was observed, but at both sites, a H_2 -poorer (SO₂-richer) gas 309 than in March was detected. The H_2S/SO_2 ratios were similar 310 in March and June 2017 (0.03 to 0.06). The daily averaged 311 H_2O/SO_2 ratios were of 190 to 205 in March 2017 and of 75.8 312 to 77.4 in June (Table 1). 313

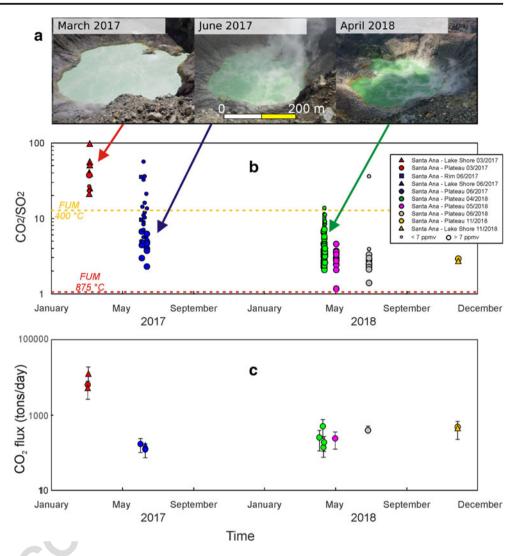
A further decrease in CO2/SO2 ratios was observed at the 314plateau in April, May, June, and November 2018, when the daily 315averaged ratio ranged between 4.1 ± 1.6 (April) and 2.4 ± 0.5 316 (June) (Table 1; Fig. 9). The H₂/SO₂ ratios in May–June 2018 317were also the lowest (daily averages from 0.37 ± 0.14 to $0.39 \pm$ 318 0.15) observed at the plateau since observations started at Santa 319Ana (Fig. 7). In 2018, the gas composition remained H₂O-rich 320 $(H_2O/SO_2$ between 32 ± 18 and 177 ± 11), and the H_2S/SO_2 ratio 321 varied between 0.08 ± 0.07 and 0.1 ± 0.07 (Table 1). H₂S was 322 below detection limit in May and June 2018. 323

324

Discussion

The Santa Ana crater lake has frequently witnessed phre-325 atic and phreatomagmatic eruptions in historical time, 326 most recently in 2005 and 2007, and thus poses a poten-327 tial threat to local inhabitants and numerous visitors. 328 Colvin et al. (2013) suggested that the crater lake entered 329a new period of quiescence in early 2008, though with a 330 higher steady-state mass/energy input than before the 3312005-2007 eruptions. This quiescent phase was con-332 firmed by the low levels of background seismicity and 333 SO₂ fluxes (Fig. 4) and by the relatively stable lake tem-334perature and chemistry (Fig. 2) (Colvin et al. 2013). The 335most recent data suggest, however, that a new cycle of 336 lake surface temperature increase, and pH decrease, 337 started sometime between late 2010 and early 2017 (Fig. 338 2). Although trends in lake water chemistry are difficult to 339 interpret, due to the discontinuous nature of the dataset, 340 sulfate and chloride concentrations were found to be 341

Fig. 9 Variations in a the lake level, b CO_2/SO_2 ratios, and c CO_2 fluxes, between March 2017 and June 2018 (data from Table S1 and Table 1). Dashed lines in b are the CO_2/SO_2 gas ratios of high-temperature fumaroles in 2002 (sampled by T.F.) Page 9 of 14 #####



higher in 2014 than in 2010 and peaked in 2017 (Fig. 2).
Overall, this evolution in lake water chemistry and temperature may potentially imply renewed unrest and therefore needs careful scrutiny.

346 Our lake gas plume results thus contribute to our understanding of the Santa Ana crater lake. Although the lack of 347 348 similar data prior to 2017 (and especially prior to the 2005-3492007 eruptions) partially hampers our interpretation, our gas data nevertheless contribute to assessing and hypothesizing on 350the current activity level at the crater lake. An important ob-351352servation in this study is that the compositional features of the 353Santa Ana plume are heterogeneous, in both space and time.

354 Spatial variability

Measurements taken in 2017 from three distinct locations (rim, plateau, and shore), at different distances from the gas emission source (i.e., mainly the crater lake), imply some spatial heterogeneity in plume composition. At the rim, gas observations are complicated by the highly diluted nature of the

plume detected; we consider the derived ratios strongly affect-360 ed by analytical uncertainty, e.g., due to the difficulty in re-361 solving volcanic gases over atmospheric background, espe-362 cially for H₂O, CO₂, and H₂ (Fig. 5a). We hence conclude that 363 the rim is not an ideal monitoring site, at least with the current 364 state of activity at Santa Ana. The plateau, instead, is a far 365 more promising monitoring site because, in addition to being 366 safer and more accessible than the crater lake shore (Fig. 3), it 367 is also systematically fumigated by a relatively dense plume 368(SO₂ at levels of tens of ppm). Importantly, our filtered 369 dataset, in which measurements taken at higher plume density 370 $(SO_2 > 7 \text{ ppm})$ are considered (Table 1; Fig. 9), confirms that 371CO₂/SO₂ ratios exhibit overlapping ranges at plateau and 372shore in all campaigns (March and June 2017 and 373 November 2018). The similarity of CO_2/SO_2 ratios (Fig. 10) 374at the plateau and shore confirms the utility of the latter site for 375monitoring. In contrast, H₂/SO₂ ratios at the plateau and at the 376 shore do not match in both the March and June 2017 datasets, 377 even in the filtered dataset (Table 1), and are systematically 378higher at the former, more distal site. We are confident that this 379

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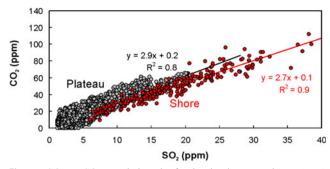


Fig. 10 CO_2 vs. SO_2 correlation plot for the simultaneous plateau (gray) and shore (red) Multi-GAS observations taken on November 28, 2018. The two derived CO_2/SO_2 ratios overlap within uncertainty

380 spatial change in composition (H₂/SO₂ ratio) cannot reflect analytical uncertainties, at least in the filtered dataset that in-381 382 cludes only measurements taken in dense plume conditions, where ratios are independent of SO₂ concentrations (see Fig. 383 3848). Also, SO₂ and H₂ are thought (Aiuppa et al. 2005b, 2011; 385 Ehhalt and Rohrer 2009) to behave conservatively (i.e., to be poorly reactive) over the short travel times of seconds to tens 386 of seconds associated with plume transport from the lake 387 shore to the plateau. As such, we consider it unlikely that the 388 389 H_2/SO_2 ratio difference between the shore and plateau is due to in-plume chemical processing. We find it, instead, more 390 likely that the compositional change between plateau and 391392 shore reflect some additional H₂ contributions from other sources, perhaps weakly degassing hydrothermal fumaroles 393 and steaming ground on the inner crater slope (Figs. 7 and 3948). We suggest that these additional, diffuse gases (see Fig. 3958) become mixed with the lake plume during plume transport 396 between emission from the lake surface and measurement at 397 the plateau, thus justifying the H₂ excess seen at the plateau 398(Figs. 7 and 8 and Table 1). This diffuse degassing source (see 399 Fig. 8) may also explain the H₂-CO₂-enriched (relative to lake 400 *degassing*; Fig. 8) compositions of dilute (SO₂ < 7 ppm) pla-401 402 teau and rim plumes.

403 Temporal trends

404 In addition to spatial heterogeneity, our measurements also 405 highlight important temporal changes in gas composition (Fig. 9). We observed that the CO_2/SO_2 ratio decreased by 406more than one order of magnitude in 1 year, from 407408 March 2017 (shore 31.0 ± 13.7 ; plateau 37.2 ± 9.7) to June– November 2018 (< 3.0 at both shore and plateau) (Fig. 9). The 409 H_2/SO_2 ratios at the plateau were also lower (and far less 410411 variable) in May-June 2018 than in March 2017, both in the total (Figs. 7 and 8) and filtered (Table 1) datasets. Overall, 412these observations imply a gas composition becoming more 413SO₂-rich over time. H₂S has remained a minor sulfur com-414 415pound throughout the entire period of observation (Table 1).

In lake gas plumes, the S composition and flux reflect acomplex and temporally variable balance between sulfur input

flux into the lake via the magmatic/hydrothermal gas supply at the lake-bottom, gas scavenging by lake water as dissolved the sulfur and/or mineral precipitates, and surface gas release (as either SO₂ or H₂S). Gas-water-rock reactions in the lake remove sulfur from the input gas via (Kusakabe et al. 2000; 422 Christenson and Tassi 2015; Delmelle and Bernard 2015; de Moor et al. 2016a): 424

 $3SO2 + 2H2O \rightarrow 2HSO4 - + S(e) + 2H +$ (1) 427

$$4SO2 + 4H2O \rightarrow 3HSO4 - + H2S + 3H +$$
 (2) 430

$$2H2S + SO2 \rightarrow 3S(e) + 2H2O$$
 (3) 428
433

in which S(e) is elemental sulfur. The increasingly 434 SO₂-rich compositions of the Santa Ana lake plume gas 435in 2018, relative to 2017, suggest a shift of the above 436reactions toward the left, i.e., they indicate a lower con-437 sumption of the reagents during gas-water-rock reactions 438into the lake. A reduced SO₂ dissolution into the lake 439 explains the tendency of our lake plume gas to become 440 increasingly SO₂-rich (i.e., more magmatic in nature). The 441 low H₂S contents in the Santa Ana lake plume imply that 442 reaction mechanisms (1) and/or (2) are most likely in-443 volved because the reversal of reaction (3) should lead 444to H₂S formation, which is not observed. 445

The composition of the Santa Ana input gas in 2017–2018 446 is unknown, in view of the lack of measurable fumaroles. 447 However, two high-temperature (400 to 875 °C) gas samples 448 were collected at Santa Ana by one of us (T.F.) in June 2002. 449Assuming these compositions as representative of the current 450 (2017–2018) magmatic gas input into the lake (Figs. 7 and 9) 451confirms that lake plume gas has become increasingly more 452magmatic in nature in 2018, relative to 2017. The triangular 453 plots of Fig. 11 additionally support a progressive evolution of 454 the Santa Ana plume gas toward more magmatic composi-455tions. These plots compare the Santa Ana lake gas composi-456tion of this study with (i) the magmatic fumaroles in 2002 and 457(ii) the compositions of lake gas plumes recently obtained at 458other quiescent and/or recently active crater lakes worldwide 459(see caption for data sources). The plots confirm that, over the 460 year of observations (2017-2018), the Santa Ana lake gas 461 evolved in composition from CO₂-H₂-rich and S-poor to more 462 S-rich, encompassing an intermediate position between the 463 CO2-H2-H2S-rich lake plumes as seen at "quiescent" crater 464 lakes (e.g., El Chichón and Viti; Hasselle et al. 2018), and 465the far more SO₂-rich plumes issuing from "recently erupting" 466 crater lakes (Yudamari, Copahue, Poás, and Rincón de la 467 Vieja; Shinohara et al. 2015; Tamburello et al. 2015; de 468 Moor et al. 2016a, 2019; Battaglia et al. 2019) (Fig. 11). The 4692018 lake plume is also approaching the "magmatic" compo-470sition of the high-T (875 °C) fumarole sampled in the restless 471 Santa Ana crater in 2002. 472

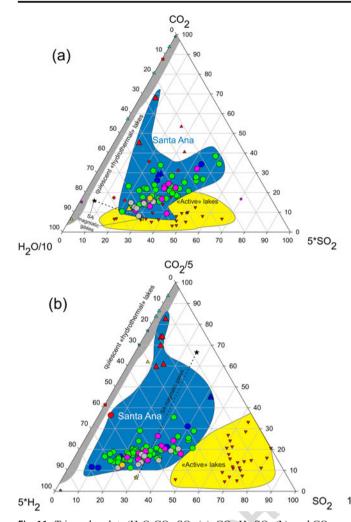
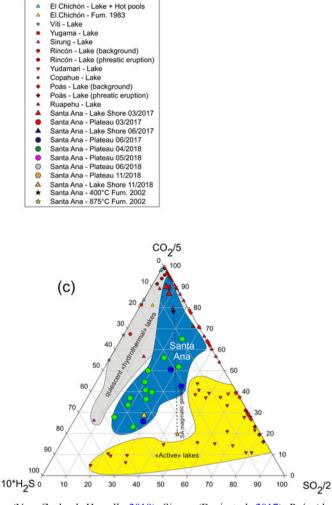


Fig. 11 Triangular plots (H₂O-CO₂-SO₂ (**a**), CO₂-H₂-SO₂ (**b**), and CO₂-H₂S-SO₂(**c**)) illustrating the temporal evolution of the chemical composition of the Santa Ana crater lake gas plume. The Santa Ana plume data (filtered dataset only; data from Table S1 filtered at the SO₂ threshold of \geq 7 ppmv) are compared with a selection of crater lake plume compositions from El Chichón (Mexico; Hasselle et al. 2018), Viti (Iceland; Hasselle et al. 2018), Yugama (Kusatsu-Shirane volcano, Japan; Hasselle 2019), Yudamari (Aso volcano, Japan; Shinohara et al. 2015), Copahue (Argentina-Chile; Tamburello et al. 2015), Ruapehu

473 Internal vs. external controls on lake gas evolution

The next obvious question is: what is the possible driver for the 474observed variations in gas compositions, and to what extent 475476 these signals represent potential hints for renewed volcanic un-477 rest? SO₂ dissolution into volcanic crater lakes involves nonequilibrium, kinetically controlled gas-water-solid reactions 478479(Kusakabe et al. 2000; Miyabuchi and Terada 2009; Christenson and Tassi 2015; de Moor et al. 2016a). Hence, since 480 timing of gas-water interaction is the key factor, reduced sulfur 481 absorption and scrubbing in 2018, as suggested by declining 482483 CO_2/SO_2 ratios, implies faster gas transit through the lake.

The gas transit time in a volcanic lake and, hence, the possible timescales of gas-water interactions scale to lake



(New Zealand; Hasselle 2019), Sirung (Bani et al. 2017), Poás (de Moor et al. 2016a), and Rincón de la Vieja (Battaglia et al. 2018). The 2018 Santa Ana plume gases have increasingly become more magmatic (more similar to the 2002 high-T fumaroles data of T.F.) with respect to the 2017 plume gases. Santa Ana crater lake gas occupies an intermediate position (blue-colored field) between "quiescent hydrothermal lakes" (El Chichón, Viti and Yugama; gray field) and "active lakes" (Copahue, Yudamari, Kawah Ijen; yellow field)

volume and depth (e.g., the deeper the lake, the longer the 486 gas residence time) (Christenson and Tassi 2015) and input 487 gas flux (de Moor et al. 2016a). Visual observations at the 488crater rim (Fig. 9) indicate that the Santa Ana crater lake level 489fell in 2018, relative to 2017. Our records (Fig. 2b) highlight 490that the crater lake level dropped by a maximum ~ 3 m be-491tween March and June 2017 and remained similarly low, or 492lower (Fig. 9), in April-May 2018. The dropping lake level in 493late 2017 to early 2018 is possibly part of a longer-term dry-494ing-out trend that started sometime in 2011 (Fig. 2b). Image-495based, semi-quantitative estimates of relative lake level chang-496es (Fig. 2b), based on relative level variations with respect to a 497reference level, show in fact that while the lake essentially 498 rose in level from 2007 to 2010, the trend reversed since 499

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5002011, when relative level changes have been either null (e.g., 2011-2013) or negative (e.g., 2015). The negative lake level 501fluctuations observed since June 2017 are, in particular, 502 503unique in the recent (post-2000) Santa Ana record and also 504 correspond to a phase of peaking lake temperature, salinity, and acidity (Fig. 2c, d). As a result, we propose that the post-505506June 2017 reduced lake volume caused a general decrease in the gas residence time in the lake, leading to less efficient 507sulfur reactions with lake water and ultimately to more SO₂-508 rich lake gas plume compositions. 509

510Precipitation records in the Santa Ana area (Fig. 2a) sug-511gest that the recent decrease in crater lake volume may have 512been caused, at least partially, by a decreased meteoric water supply to the lake itself. Precipitation in the Santa Ana area 513nearly halved in 2018 relative to 2010–2011 and was $\sim 15\%$ 514lower than in 2017 (Fig. 2a). It is therefore likely that the 515reduced meteoric water influx contributed to reducing the cra-516517ter lake volume. This conclusion is consistent with recent 518modelling work (Terada and Hashimoto 2017) demonstrating that low levels of precipitation can cause sizeable changes in 519crater lake temperature and composition, even at constant sub-520aqueous gas input into the lake. 521

522However, our gas flux measurements (Table 1) suggest that, in addition to reduced meteoric precipitations, an "inter-523nal" volcano-driven trigger was also likely at play. During our 524525Multi-GAS surveys, the SO₂ fluxes were the highest in March 2017 (with daily averages of 240 and 329 t/day on 526March 7 and 8, respectively; see Table 1). By scaling our 527 528 measured daily averaged lake plume gas ratios to the daily 529mean SO_2 fluxes, we can also calculate the H_2O , CO_2 , H_2 , and H₂S fluxes (Table 1). Results demonstrate that fluxes of 530531H₂O (13,825–17,565 t/day), CO₂ (5117 to 12,320 t/day), and total volatiles (TV 20,217-30,225 t/day; the total volatile flux 532is the sum of $H_2O + CO_2 + SO_2 + H_2$ fluxes in our specific 533case) were one order of magnitude higher in March 2017 than 534at any time since (H₂O 892-9483; CO₂ 118-485; TV 615-53510167 t/day; Table 1). These results thus suggest that an in-536537creased gas supply from the sub-limnic magmatic-hydrothermal system (as implied by the anomalously high March 2017 538fluxes) was a likely additional causal factor in driving the lake 539540toward dryness. We propose that the elevated gas supply and, hence, heat transfer into the lake caused more intense lake 541evaporation (resulting in decreasing lake volume; Fig. 2b) 542543and heating (where the lake warmed-up in late 2017; Fig 2c). We also argue that, because of the relatively high lake 544level in March 2017, the majority of the magmatic/ 545546hydrothermal S and Cl input was initially dissolved into the lake, thus justifying the anomalously elevated dissolved SO₄ 547and Cl (Fig. 2d) and the SO₂-poor lake plume gas (Fig. 9). In 548the following months, however, the lake level drop caused 549550more rapid fluxing of gas through the lake, reducing the timescales of gas-water interactions and, thus, the lake's ability to 551scrub magmatic sulfur, ultimately determining a more 552

magmatic (SO2-rich) lake gas plume in late 2017 and in5532018. The lack of any sizeable change in seismicity (Fig. 4)554perhaps suggests that the escalation in deep gas supply was555not elevated enough to cause pressurization/fracturing of the556sub-limnic hydrothermal-magmatic system.557

Conclusions and implications for monitoring 558

Our novel gas plume results highlight the dynamic nature of 559the Santa Ana crater lake and reveal rapid compositional evo-560 lution in only 2 years of observation (2017-2018). However, 561available information on gas plume chemistry is too restricted 562in time to allow firm conclusions to be made on the current 563state of activity of the volcano. In particular, we cannot deter-564mine when the phase of elevated total volatile fluxes we ob-565served in March 2017 actually started. Notwithstanding this, 566our results clearly show that the lake plume gas became in-567 creasingly more SO₂-rich, and therefore more magmatic in 568nature, in late 2017 and 2018. These gas variations have been 569paralleled by consistent variations in lake water chemistry and 570physical parameters, including increased lake temperature, 571acidity, and salinity, and a reduction in lake level and volume. 572

We propose these variations have been caused by a com-573bination of external and internal processes, such as a decrease 574in precipitation and increased mass/heat supply at the lake 575bottom in March 2017, or before. In our interpretation, a \sim 57615% drop in precipitation, and the elevated magmatic/ 577hydrothermal fluid supply in March 2017, combined to reduce 578the lake volume. In turn, this resulted in a shortened magmatic 579gas transit time through the lake water. This lead to a reduction 580of magmatic sulfur reacting with lake water and ultimately to a 581more SO₂-rich gas plume. 582

The dynamic evolution of degassing at Santa Ana volcano, 583highlighted in the present study, argues for the need of further 584observations and careful scrutiny of water/gas compositional 585features in the very near future. Comparison with gas plume 586data from other crater lakes worldwide demonstrates that the 587 2018 Santa Ana lake gas is intermediate in composition be-588 tween the CO₂-H₂-H₂S-rich lake plumes seen at "quiescent" 589crater lakes (e.g., El Chichón in Mexico), and the by-far more 590SO₂-rich plumes issuing from "recently erupting" lakes (e.g., 591Poás in Costa Rica). As such, any additional compositional 592 change toward the SO₂-rich magmatic gas end-member should 593seriously be considered as evidence of activity escalation. At 594Laguna Caliente (Poás, Costa Rica), increasing SO₂ typically 595peaks prior to phreatic/phreato-magmatic eruptions (de Moor 596et al. 2016a, 2019; Stix and de Moor 2018), reflecting increas-597ing magmatic gas influx into the lake. Although this critical 598situation seems not to have yet been reached at Santa Ana, 599our results underpin the need of reinforced volcano monitoring 600 at this potentially hazardous volcano. 601

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