

A NOVEL REVERSE ELECTRODIALYSIS APPLICATION TO GENERATE POWER FROM LOW-GRADE HEAT

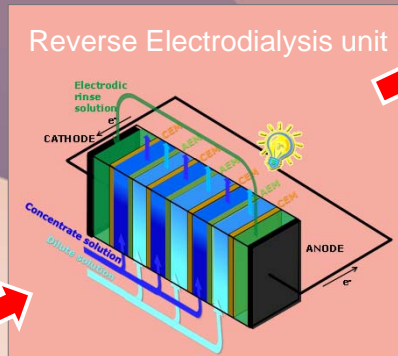
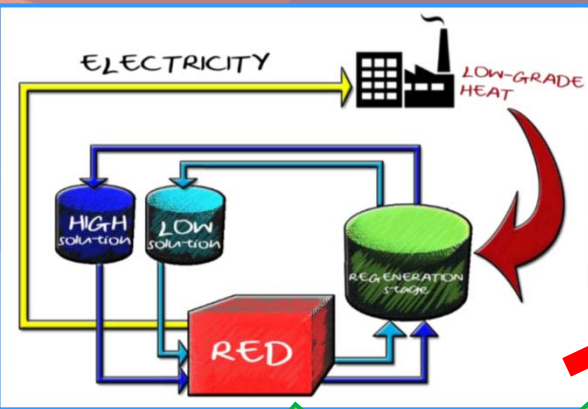
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INTRODUCTION: The urgent need to reduce dependency on fossil fuels and concerns of the negative impact of many dated and unsustainable industrial processes on the environment are topics of crucial importance nowadays. These have led the interest towards heat recovery technologies for the utilisation of waste heat from industrial processes to continuously grow. Only some heat engines are able to operate at low temperature level (<100°C), but none of them is able to achieve satisfactory value of efficiency. In this regard, a closed cycle based on the use of Reverse Electrolysis is theoretically able to operate at very low temperature will be presented in the following.



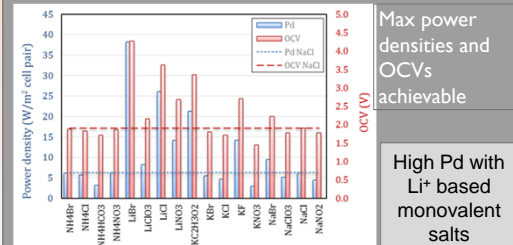
RED optimization via modeling

Performance of the RED unit with different salts

RED model assumptions:

- all solutions' and cell pairs' variables are evaluated at the average conditions between inlet-outlet of the feed channels;
- membrane properties (permselectivity, resistance) are kept constant (i.e. independent of the solute employed);
- solvent transport through membranes is neglected;
- polarization phenomena are neglected;
- ideal current distribution (i.e. no parasitic currents)

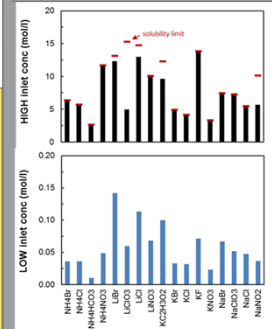
Results



Max power densities and OCVs achievable

High Pd with Li⁺ based monovalent salts

Optimal (i.e. for Pd,max) feed solutions concentrations



Optimal concentrate concentration equal to the solubility limit for most salts

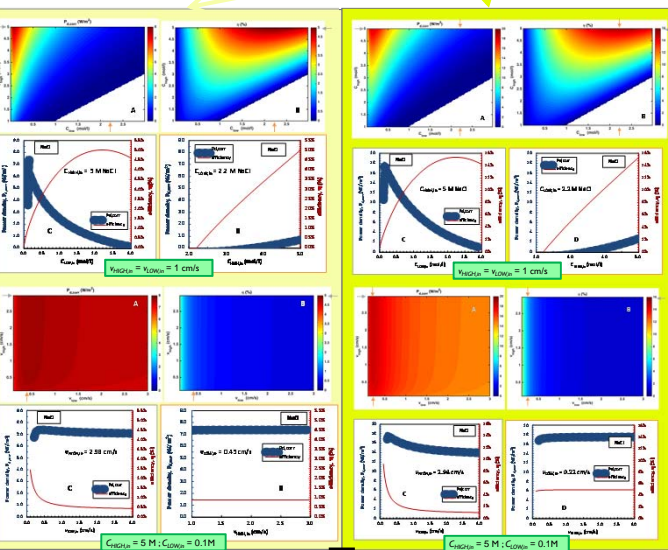
Optimal dilute concentration lies in a relatively narrow range (0.01 M and 0.15 M) for all the investigated salts.

CASE STUDIES

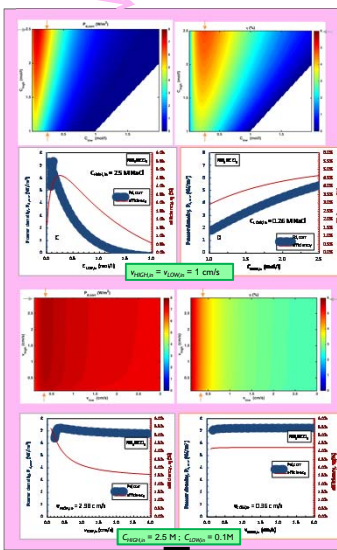
	Case study I	Case study II	Case study III
Case study typology	State of the art	Perspectives	Perspectives
Waste heat source temperature (°C)	90	90	60
RED stage			
Salt	NaCl	NaCl	NH ₄ HCO ₃
ABM permselectivity α_{ABM} (-)	0.65	0.90	0.90
CEM permselectivity α_{CEM} (-)	0.90	0.90	0.90
ABM electrical resistance R_{ABM} (Ω cm)	2.96	0.50	0.50
CEM electrical resistance R_{CEM} (Ω cm)	1.55	0.50	0.50
Temperature (°C)	25	25	25
Regeneration stage			
Typology	Solvent extraction	Solvent Extraction	Salt Extraction
Technology	MED	MED	Thermal degradation
Energy consumption	40 kWh/m ³ of extracted solvent	25 kWh/m ³ of extracted solvent	100 kJ/mol of extracted salt [1] and 4°C of thermal heating for F ₅
Sensitivity analysis	Parameters and variation range		
C_{Dilute} (mol/l)	1.0±5.0	1.0±5.0	1.0±2.5
C_{Conc} (mol/l)	0.01±3.0	0.01±3.0	0.01±2.0
α_{ABM} (cm/s)	0.1±3.0	0.1±3.0	0.1±3.0
α_{CEM} (cm/s)	0.1±3.0	0.1±3.0	0.1±3.0

Cycle Efficiencies

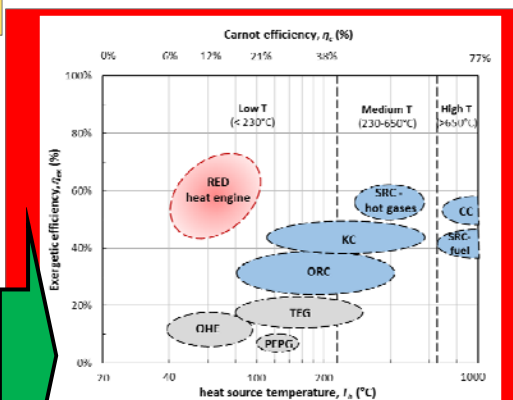
- Energetic efficiency $\eta = \frac{P_{d,corr}}{Q_{in}}$
- Carnot efficiency $\eta_c = 1 - \frac{T_{cold}}{T_{hot}}$
- Exergetic efficiency $\eta_{ex} = \frac{\eta}{\eta_c}$



With NaCl-water solutions efficiencies up to about 15% corresponding to an exergetic efficiency of about 85% can be achieved



With NH₄HCO₃-water solutions max $P_{d,corr}$ and η are achieved at very close values of the operating parameters



State of the art technologies for the conversion of heat into power.

Grey circles refer to technologies at very early stage of development. The Carnot efficiency (on the secondary horizontal axis) is evaluated assuming a cold sink temperature of 20°C. SRC-hot gases; Steam Rankine Cycle integrated with gas turbine/other topping cycles; SRC-fuel; Steam Rankine Cycle directly fuelled by oil, coal or other fuels; KC: Kalina Cycle; ORC: Organic Rankine Cycle; TEG: Thermoelectric Generation; PEPG: Piezoelectric Power Generation with waste heat-powered expansion/compression cycle; OHE: Osmotic Heat Engine.

RED could be promising as possible future converter of low-grade heat into electric energy