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# Irradiation-Resistant Heterostructures Based on Monolayer MoS<sub>2</sub>

Emanuele Sangiorgi<sup>1</sup>  | Francesca Migliore<sup>1</sup>  | Antonino Madonia<sup>1</sup>  | Antonino Alessi<sup>1,2</sup>  | Romain Grasset<sup>2</sup>  | Olivier Cavani<sup>2</sup>  | Salvatore Ethan Panasci<sup>3</sup>  | Emanuela Schilirò<sup>3</sup>  | Filippo Giannazzo<sup>3</sup>  | Fiorenza Esposito<sup>4,5</sup>  | Luca Seravalli<sup>4</sup>  | Marco Cannas<sup>1</sup>  | Simonpietro Agnello<sup>1,6</sup> 

<sup>1</sup>Department of Physics and Chemistry Emilio Segrè, University of Palermo, Palermo, Italy | <sup>2</sup>LSI, CEA/DRF/IRAMIS, CNRS, Ecole Polytechnique, Institut Polytechnique de Paris, Palaiseau, France | <sup>3</sup>Consiglio Nazionale delle Ricerche, Istituto per la Microelettronica e Microsistemi (CNR-IMM), Catania, Italy | <sup>4</sup>Institute of Materials for Electronics and Magnetism, National Research Council (IMEM-CNR), Parma, Italy | <sup>5</sup>Department of Chemical Science, Life and Environmental Sustainability, University of Parma, Parma, Italy | <sup>6</sup>ATEN Center, University of Palermo, Palermo, Italy

**Correspondence:** Antonino Madonia ([antonino.madonia@unipa.it](mailto:antonino.madonia@unipa.it))

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## ABSTRACT

Monolayer molybdenum disulfide is a semiconducting 2D material presenting very appealing characteristics for its incorporation into electronic flexible devices. At present, atomic vacancies in its structure are known to significantly affect its properties. Nonetheless, the effects of electron irradiation on it are still under investigation. In this work, the structural, electronic, and optical properties of monolayer molybdenum disulfide are studied before and after electron irradiation. By analyzing samples either obtained via a mechanical exfoliation process or grown via chemical vapor deposition, we show that following irradiation no significant change is present. This result indicates a strong resistance to the effect of electron irradiation at energies able to induce the formation of sulfur vacancies. By comparing our results to the current literature, we hypothesize that previously reported effects of electron irradiation are either due to the formation of molybdenum vacancies or to ionization of the electronic shell of this material. The presented results help to shed light on the role of atomic vacancies on the properties of this 2D material from a fundamental point of view and on its usage in extreme environment applications.

## 1 | Introduction

The interest in heterostructures based on Van der Waals materials has witnessed significant growth in recent years. Indeed, nowadays, a wide array of 2D materials presenting metallic, insulating, and semiconductor characteristics is available to researchers [1]. Such variety has led to the development of heterostructures composed of vertically stacked 2D covalent layers kept together by weak Van der Waals forces and interacting with their environment and substrate. Controllable stacking without the constraint of lattice matching has indeed allowed the development of prototype devices such as transistors [2], photodetectors [3], solar cells [4], and light-emitting diodes [5].

Among the many available 2D materials, transition metal dichalcogenides (TMDs) are renowned for their semiconducting electronic properties and tunable bandgap [6]. Molybdenum disulfide (MoS<sub>2</sub>) in particular is considered an especially appealing TMD, because of its abundance in nature as a mineral and the ability to easily process it down to single 2D monolayers [7].

Although bulk MoS<sub>2</sub> is a semiconductor presenting an indirect bandgap of 1.2 eV, its properties are significantly affected by thickness. Indeed, it has been found that the states at the  $\Gamma$  point of its band structure are subjected to a strong interlayer coupling. When the thickness of MoS<sub>2</sub> is reduced down to a single layer, this coupling leads to a shift in the position of the minimum of

Emanuele Sangiorgi and Francesca Migliore contributed equally to this work.

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the conduction band. In the monolayer case, such minimum is found at the K point of the Brillouin zone, so that direct transitions across the bandgap can be observed [8]. The recombination of photo-generated excitons thus leads to radiative transitions, absent in the case of multilayer MoS<sub>2</sub>, responsible for an intense photoluminescence (PL) peaking approximately at 1.8 eV. Nowadays, monolayer MoS<sub>2</sub> (1L-MoS<sub>2</sub>) is obtained following either top-down or bottom-up routes. Among the former, the gold-assisted exfoliation (GAE) method has been shown capable of yielding large-area flakes [9–11]. Although mechanical methods allow to obtain monolayers with unmatched structural and optical quality, the chemical vapor deposition (CVD) route is considered more appealing towards an industrial transfer [12, 13]. There is, indeed, a strong interest in incorporating the production of 1L-MoS<sub>2</sub> into the current fabrication procedures, facilitated by chemical synthetic pathways, with the intent of achieving flexible electronics integrating monolayer flakes. Such goal is realistically attainable considering the peculiar charge-transport properties of 1L-MoS<sub>2</sub>, with a room-temperature mobility higher than 200 cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup> [14], and surprising mechanical resistance and flexibility [15].

All of these characteristics have already led to the development of devices based on 1L-MoS<sub>2</sub> [16]. In such kind of devices, the monolayer flakes strongly interact with the underlying substrate, so that the final properties of the overall system are intimately linked to the characteristic of the whole heterostructure. Additionally, the performances of such devices are strongly connected to the quality of 1L-MoS<sub>2</sub> and the eventual presence of structural defects in the flakes can easily affect the final outcome [17]. Efforts have already been made towards tuning the properties of this material through mechanical [18] and chemical [19] modifications, thermal [20] and plasma [21] treatments, and by incorporating flakes into different dielectric environments [22, 23]. Although these strategies are viable during the production step, in view of using 1L-MoS<sub>2</sub> devices in real world applications, the effect of defect-inducing processes on this material has to be considered as well. Ionizing radiations are indeed well known to cause the formation of vacancies in 2D materials [24]. For devices used in extreme environments, such as solar cells in space, it is thus fundamental to know the effect of high-energy electrons on the structural properties of 1L-MoS<sub>2</sub> [25].

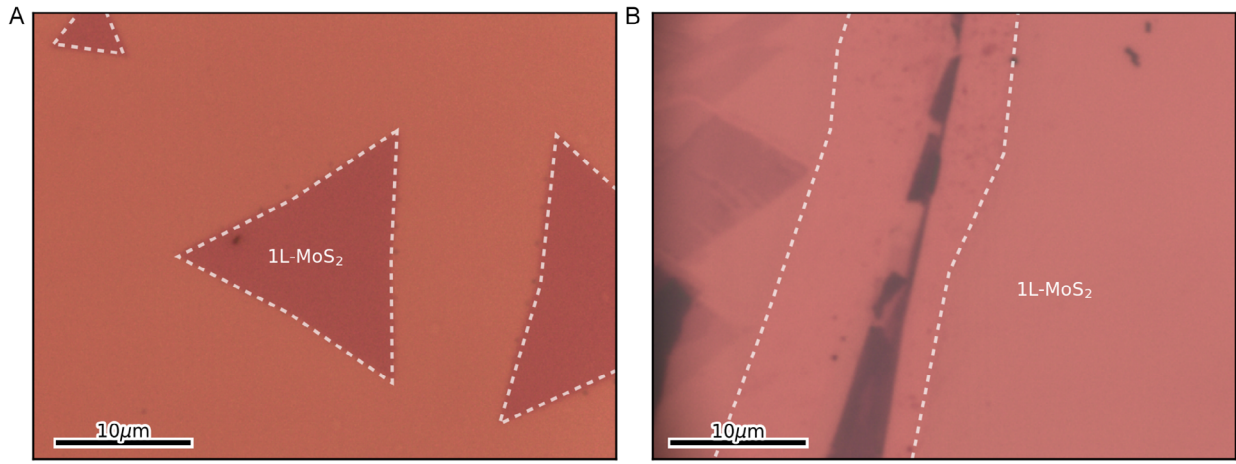
Electron irradiation was shown able to cause the formation of vacancies [26–28], alter the electronic properties of devices based on 1L-MoS<sub>2</sub> [29, 30], and change both the structural and optical properties of this material [31–33]. During this kind of irradiation, different processes can occur. In fact, the interactions between the electrons of the beam and those of the target induce excitation and ionization of the material's electronic shell, giving rise to the so-called ionizing dose effects [25]. Vacancy formation in such a case happens as a secondary process [24]. This process should be the only one present when the irradiation is performed with electrons having energies of few tens of keV. The exact cross-section depends on the irradiated sample's physical and chemical properties, so that its calculation often results challenging. On the other hand, at high energy (often in the hundreds of keV and MeV range), a direct knock-on causing atom displacement becomes probable too (displacement damage). Thresholds based on the formation energy of the different defects can in this case be used to selectively displace certain atoms. Depending on irradiation energy, the effects observed on the properties of

1L-MoS<sub>2</sub> can be selectively associated with sulfur vacancies, thus allowing us to understand their role in comparison to that of molybdenum vacancies. At present, the character of 1L-MoS<sub>2</sub> vacancies is still debated, with some works indicating that Mo and S ones introduce acceptor and donor levels in the energy band structure, respectively [34, 35]. Recent studies, on the other hand, indicate that S vacancies in 1L-MoS<sub>2</sub> cause a decrease in the concentration of negative charge carriers, thus giving them an acceptor character [36]. Finally, electron irradiation can also alter the properties of the substrate and other materials surrounding the 1L-MoS<sub>2</sub> flakes, either modifying their degree of interaction or directly worsening the quality of the electrical contacts in devices [29, 30]. Also, in such case, the characteristics of the whole heterostructure based on 1L-MoS<sub>2</sub> would significantly change due to the effect of irradiation.

In the present work, we explore the effect of high-energy electron irradiation on 2D MoS<sub>2</sub> trying at the same time to understand the role of atomic vacancies on its properties. By characterizing 1L-MoS<sub>2</sub> using micro-Raman and micro-PL spectroscopy, we intend to follow the changes to the structural, electronic, and optical properties of such material as a consequence of electron irradiation conducted at an energy of 0.5 MeV and fluences between  $8 \times 10^{12} \text{ e}^- \text{ cm}^{-2}$  and  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$ . In these conditions, our calculations indicate that S vacancies are the most likely to form. Because of this, we can compare our results with a previous work conducted at 2.5 MeV, an energy at which both Mo and S vacancies are directly formed by knock-on processes [33], and pinpoint the role of different vacancies. Our results indicate 1L-MoS<sub>2</sub> to be resistant to the effects of electron irradiation conducted at the explored energy and suggest that the effects witnessed at higher energies are mostly due to Mo vacancies.

## 2 | Results and Discussion

The method used for the production of 1L-MoS<sub>2</sub> flakes can influence the structure of the obtained 2D material. Representative images obtained via optical microscopy with a 100× objective are shown in Figure 1 for 1L-MoS<sub>2</sub> grown via CVD on a SiO<sub>2</sub>/Si substrate (Figure 1A) or obtained via GAE on Au/Ni/Si substrate (Figure 1B). The different contrast between the flakes and the substrate is due to the differences in refractive index between the monolayers and the material of the substrate, making 1L-MoS<sub>2</sub> almost transparent when deposited on the Au layer. Flakes grown by CVD result triangular in shape; indeed, the controlled 2D growth mostly proceeds in a diffusion-limited regime from the borders of 1L-MoS<sub>2</sub> flakes, leading to a distribution of crystalline monolayers sized between 5 μm and 25 μm (14 μm on average,  $\sigma = 4 \mu\text{m}$ ) [37, 38]. Only for some larger flakes (approximately 30 μm), the growth of a second layer of 1L-MoS<sub>2</sub> is observed starting from the center (Figure S1). When considering 1L-MoS<sub>2</sub> flakes produced via the GAE method, larger monolayer regions with sides up to hundreds of microns can be obtained. Although high-quality monolayer regions are obtained [39], this production method does not provide any control on the thickness of the flakes and the exfoliation often leads to residual additional MoS<sub>2</sub> layers on top of the regions of interest (Figure S2). Despite the differences, both methods allow the study of wide isolated monolayer MoS<sub>2</sub> portions found within each sample.



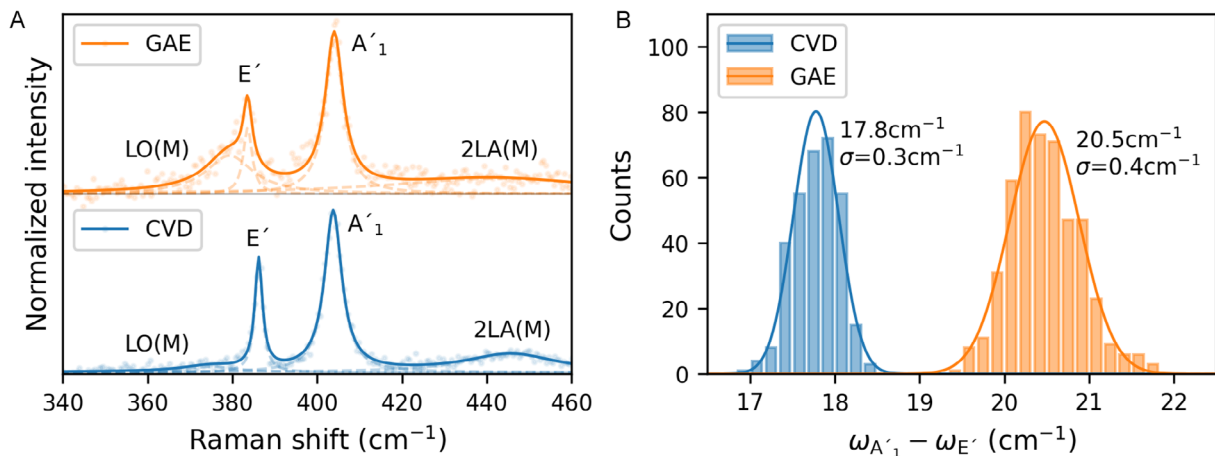
**FIGURE 1** | Optical microscopy images of 1L-MoS<sub>2</sub> flakes grown by CVD (A) or obtained by GAE (B). The dashed lines indicate the contour of the flakes.

In order to study the effect of electron irradiation as a function of fluence, ten different samples, five grown via CVD and five obtained through GAE, were selected. For each sample, a  $15\ \mu\text{m} \times 15\ \mu\text{m}$  1L-MoS<sub>2</sub> square region was characterized by Raman spectroscopy probing 100 points distributed across the surface at  $1.5\ \mu\text{m}$  steps. The results of the Raman characterization, which allow to study the electronic and structural properties of 1L-MoS<sub>2</sub> flakes, are shown in Figure 2. A typical 1L-MoS<sub>2</sub> Raman spectrum, as reported in Figure 2A, presents two main peaks at approximately  $385$  and  $405\ \text{cm}^{-1}$  associated with the in-plane E' and out-of-plane A'<sub>1</sub> vibration of the material's lattice [40]. Two additional less intense bands were observed as well, located at approximately  $375$  and  $450\ \text{cm}^{-1}$  and associated with the longitudinal optical LO(M) and with the second-order longitudinal acoustic 2LA(M) vibrations respectively [41, 42]. The frequency and amplitude of the observed peaks were obtained by fitting each experimental spectrum with Lorentzian bands.

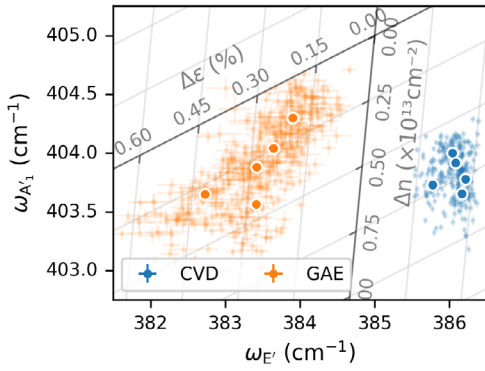
As shown in Figure 2B, the distance between the A'<sub>1</sub> and E' peaks was found to be  $17.8\ \text{cm}^{-1}$  ( $\sigma = 0.3\ \text{cm}^{-1}$ ) and  $20.5\ \text{cm}^{-1}$

( $\sigma = 0.4\ \text{cm}^{-1}$ ) for the flakes obtained by CVD and GAE, respectively. The distance between the two main 1L-MoS<sub>2</sub> Raman peaks confirms that the studied regions are indeed composed of monolayers; in fact, such frequency difference has been correlated to the thickness of MoS<sub>2</sub> and was found to be less than  $22\ \text{cm}^{-1}$  in the case of monolayers [43]. This difference is however found to be smaller in the case of flakes obtained via CVD and on insulating substrates [12, 44]. This effect, explaining the different distances found for CVD and GAE samples, stems from variations in the strain and doping of 1L-MoS<sub>2</sub>, in turn affected by the flakes' substrate, local environment, and production method.

The strain and negative charge-carrier concentration can be obtained from the frequency of the Raman E' and A'<sub>1</sub> peaks [45]. This method, detailed in the Supporting Information, allows to plot the strain-doping map shown in Figure 3. The map divides the plot into four quadrants related to 1L-MoS<sub>2</sub> under either compressive or tensile strain and presenting either negative or positive doping. From the plot, we can readily observe that all the studied flakes present an n-type doping, with a concentration of charge carriers on average equal to



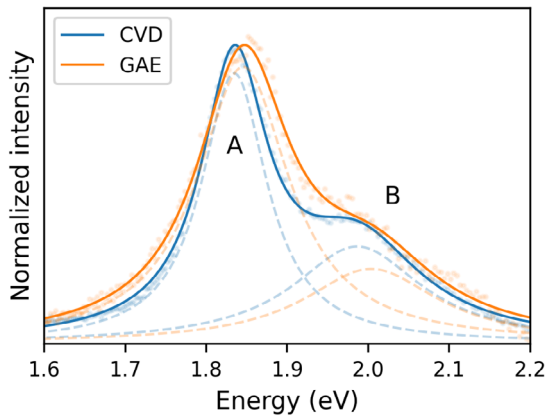
**FIGURE 2** | (A) Normalized Raman spectra of 1L-MoS<sub>2</sub>; experimental points are reported as semitransparent dots while the fitted spectra are reported as continuous lines; the Lorentzian peaks used to fit the experimental data are reported as dashed lines with labels indicating the corresponding vibration mode. (B) Histogram of the distances between the A'<sub>1</sub> and E' Raman peaks' frequencies of 1L-MoS<sub>2</sub>; the normal distribution fitting the data is reported as a continuous line whose center and standard deviation are reported as labels.



**FIGURE 3** | Strain-doping map of 1L-MoS<sub>2</sub> obtained from the frequency of the A<sub>1</sub>' and E' Raman peaks; semitransparent dots represent the peaks' frequencies measured across the surface of several flakes while the larger dots indicate the average for each single flake.

$0.71 \times 10^{13} \text{ cm}^{-2}$  ( $\sigma = 0.02 \times 10^{13} \text{ cm}^{-2}$ ) and to  $0.28 \times 10^{13} \text{ cm}^{-2}$  ( $\sigma = 0.04 \times 10^{13} \text{ cm}^{-2}$ ) for CVD and GAE samples, respectively. Additionally, while the CVD grown flakes present a compressive strain ( $\epsilon = -0.25\%$ ,  $\sigma = 0.01\%$ ), the GAE samples are under a tensile strain condition ( $\epsilon = 0.29\%$ ,  $\sigma = 0.03\%$ ). These results indicate that, compared to the GAE samples, the flakes grown via CVD show larger homogeneity in both strain condition and concentration of negative charge carriers.

The ten pristine samples were also characterized by means of photoluminescence (PL) spectroscopy to determine the characteristics of the photoemission of the studied 1L-MoS<sub>2</sub> flakes. For each sample, the same  $15 \mu\text{m} \times 15 \mu\text{m}$  1L-MoS<sub>2</sub> surface previously characterized by Raman spectroscopy was analyzed. PL spectra were obtained at 49 different locations inside the probed square region at steps of  $2 \mu\text{m}$ . The results of this characterization are shown in Figure 4, displaying normalized PL spectra representative of the two different kinds of samples. The observed photoemission is associated with the recombination of the A and B excitons. The two photo-excited excitons undergo a direct recombination from the conduction band to the valence band at the K point of the 1L-MoS<sub>2</sub>; the difference in energy of the emission



**FIGURE 4** | Normalized PL spectra of 1L-MoS<sub>2</sub>; experimental points are reported as semitransparent dots while the fitted spectra are reported as continuous lines; the Lorentzian peaks used to fit the experimental data are reported as dashed lines with labels indicating the corresponding excitonic recombination.

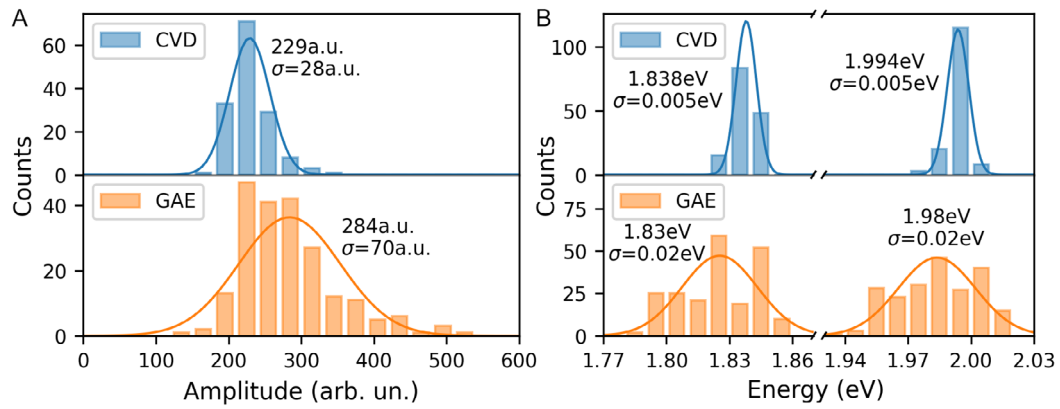
band associated to the two excitons is due to the valence band splitting caused by the spin-orbit interaction [8, 46].

The distribution of the PL amplitudes observed for the two different samples is shown in Figure 5A. The flakes obtained through the GAE method result in a more intense emission, with an amplitude on average equal to 284 a.u. ( $\sigma = 70$  a.u.), compared to those grown via CVD which display an average amplitude equal to 229 a.u. ( $\sigma = 28$  a.u.). Some studies have previously found that the presence of a Au substrate can quench the emission of 1L-MoS<sub>2</sub> due to an electron transfer mechanism [10, 39, 47]. On the other hand, similarly to our case, some studies observed that, by comparing the optical properties of 1L-MoS<sub>2</sub> flakes on different substrates, an increase in PL intensity can be witnessed on Au [48]. This result has been linked to the lower doping of the flakes deposited on Au compared to those on SiO<sub>2</sub>. Since the optical properties of 1L-MoS<sub>2</sub> are strongly affected by the concentration of negative charge carriers in the flake [49], we hypothesize that the contrasting reports are due to differences in the electronic properties of the studied material.

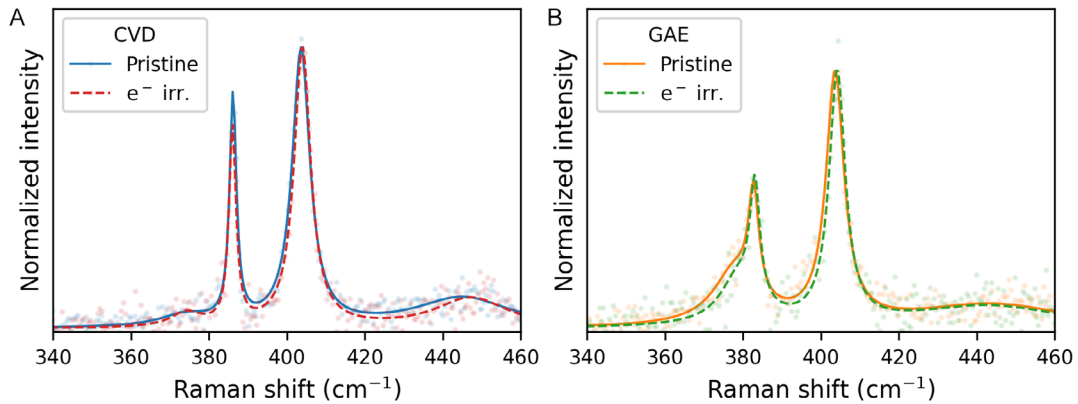
As shown in Figure 5B, both kinds of samples present emission peaks located on average at very similar energies, with the A exciton band of the CVD sample (GAE sample) peaking at 1.84 eV (1.83 eV) and the B exciton band at 1.99 eV (1.98 eV). Nonetheless, the flakes obtained with the GAE method present a larger variability in the positions of both peaks, as attested by the standard deviations of their energy distribution ( $\sigma = 0.02$  eV) which is four times larger than that obtained for CVD grown samples ( $\sigma = 0.005$  eV). The PL intensity and shape of 1L-MoS<sub>2</sub> flakes can be strongly affected by applied mechanical strain as well as by the concentration of negative charge carriers [40, 49]. We hypothesize that the large variability in the characteristics of the PL spectra observed for GAE flakes is indeed due to the inhomogeneity in the strain and doping of the studied flakes [10]. In fact, the large distribution of the PL peaks' amplitude and energy mirrors the variability in the structural and electronic properties of 1L-MoS<sub>2</sub>, found larger for GAE samples compared to the ones grown by CVD (Figure 3).

The effect of electron irradiation on 1L-MoS<sub>2</sub> was then studied at different fluences. The five characterized samples obtained with the two production methods were irradiated at the same electron energy, and the fluences were adjusted by exposing the samples to the beam for a longer duration. A beam energy of 0.5 MeV was selected by taking into consideration that the cross sections of the generation of Mo and S vacancies depend on it (Figure S3). At the chosen energy, the electrons can mostly cause displacement damage on S atoms, thus reducing the formation of Mo vacancies. This choice allows us to discern between the role of S and Mo vacancies and their effect on 1L-MoS<sub>2</sub> properties by comparing our study to experiments previously conducted at 2.5 MeV, since at this energy both atoms of the material are affected by displacement damage [33].

Following irradiation, the Raman and PL characterization were repeated on the same 1L-MoS<sub>2</sub> regions previously analyzed. Figure 6 shows the effects of irradiation on the properties as measured via Raman spectroscopy. Even after irradiation at the highest fluence of  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$ , the Raman spectra of both samples do not present any significant difference. Because of the variability of the Raman spectra within different portions of the same flake, the strain and concentration of negative charge carriers



**FIGURE 5** | Histograms of the total amplitude (A) and emission band energy (B) of the PL of 1L-MoS<sub>2</sub>; the normal distribution fitting the data is reported as a continuous line whose center and standard deviation are reported as labels.

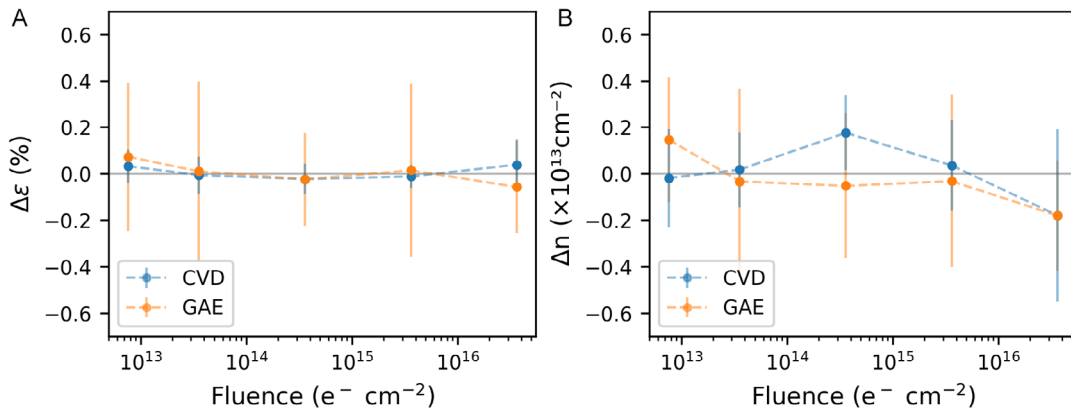


**FIGURE 6** | Normalized Raman spectra of 1L-MoS<sub>2</sub> grown by CVD (A) or obtained by GAE (B) before and after irradiation at a fluence of  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$ ; experimental points are reported as semitransparent dots while the fitted spectra are reported as continuous or dashed lines.

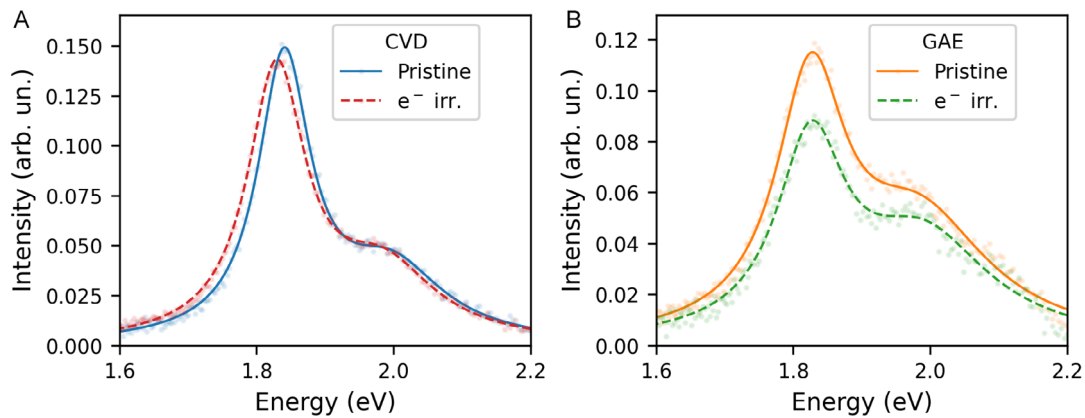
were calculated from the position of the E' and A'<sub>1</sub> peaks of each spectrum and their distributions were compared to those obtained for pristine samples.

The variations in strain and in negative charge-carrier concentration as a function of fluence are reported in Figure 7A,B, respectively. These results clearly indicate that the electron irradiation has negligible effects on the structural and electronic properties of 1L-MoS<sub>2</sub> at the explored fluences. The observed differences in strain and concentration of negative

charge carriers of the studied 1L-MoS<sub>2</sub> flakes are indeed found to be within  $3\sigma$  of their variability regardless of irradiation fluence. An additional indication about the lack of structural modification to the irradiated 1L-MoS<sub>2</sub> flakes can be found in the amplitude of the  $450 \text{ cm}^{-1}$  2LA(M) Raman peak. It has been reported that the amplitude of this band decreases as the concentration of defects in 1L-MoS<sub>2</sub> increases [41]. Nonetheless, as shown in Figure S4, the amplitude of this band does not vary with the fluence, suggesting that the performed irradiation does not lead to a significant defect formation in the studied material.



**FIGURE 7** | Strain variation  $\Delta\epsilon$  (A) and negative charge-carrier concentration variation  $\Delta n$  (B) of 1L-MoS<sub>2</sub> as a function of irradiation fluence.



**FIGURE 8** | PL spectra of 1L-MoS<sub>2</sub> grown by CVD (A) or obtained by GAE (B) before and after irradiation at a fluence of  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$ ; experimental points are reported as semitransparent dots while the fitted spectra are reported as continuous or dashed lines.

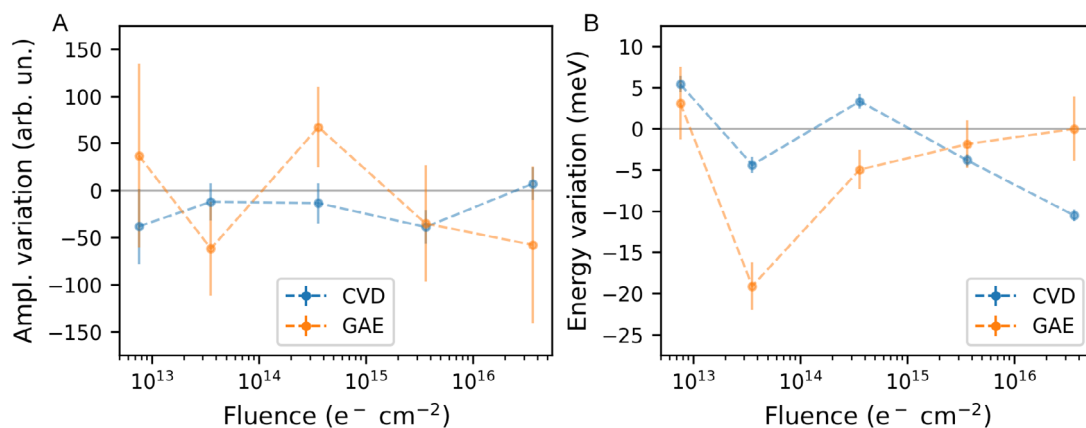
Finally, the effects of the irradiation on the optical properties of 1L-MoS<sub>2</sub> are shown in Figure 8, displaying representative PL spectra obtained before and after irradiation at the highest fluence of  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$ . A slight decrease in the intensity of the bands is observed, especially for the emission spectrum of the flake obtained through GAE. At the same time, in the case of the CVD sample, the A exciton band presents a small decrease in emission energy and results redshifted compared to the spectrum before irradiation.

To analyze in detail the effect of irradiation on the optical properties, the variation of the PL amplitude and of the A exciton peak energy is plotted as a function of fluence in Figure 9A,B. When comparing the emission amplitude of the irradiated samples with respect to that before irradiation, we observe that the differences fall within  $3\sigma$  of the inherent variability present across the probed 1L-MoS<sub>2</sub> flakes' surface. In fact, no obvious trend is found for both kinds of samples. When considering the energy of the emission bands, the same conclusion can be reached as their variations fall within the variability of the peaks' position found between different flakes. For instance, in the case of 1L-MoS<sub>2</sub> flakes grown by CVD at the highest fluence of  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$ , a redshift of 10 meV is observed. Despite this variation could appear to be significant, the A exciton peak position of CVD samples, as shown in Figure 5B, presents a standard deviation equal to 5 meV. Because of this, the apparent variation observed for the

flake irradiated at the highest fluence cannot be considered significant at the  $3\sigma$  level. This consideration can be applied to both bands of all the studied samples, as shown in Figure S5.

Overall, these results indicate that electron irradiation at an energy of 0.5 MeV and up to a fluence of  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$  does not affect the optical properties of the studied 1L-MoS<sub>2</sub> samples. Considering that the PL properties are strongly influenced by strain and doping of the flakes [10, 40, 49], this result is not surprising. Indeed, no variations were found in the structural and electronic properties of the probed monolayers after irradiation, as shown in Figures 6 and 7.

The performed experiments show that irradiation with electrons at an energy of 0.5 MeV and fluences comparable to those encountered by a solar cell during space missions does not induce any significant modification in the structure of 1L-MoS<sub>2</sub>, leaving its electronic and optical properties unvaried. Previous experiments conducted using electrons at lower energies were reported to strongly affect MoS<sub>2</sub> [27, 28]. The observed effects were associated with the displacement of sulfur atoms and with the formation of vacancies caused by the electron beam [26]. However, the irradiations were performed under flux and fluence values much higher than those used for our experiments. Moreover, the low beam energies imply higher stopping power values which further increase the dose rate and the total dose, so that the ionizing mechanism could still be dominant [27].



**FIGURE 9** | Total amplitude variation (A) and A emission band energy variation (B) of the PL of 1L-MoS<sub>2</sub> as a function of irradiation fluence.

Other studies demonstrated changes in the electronic performances of FETs and memory devices based on 1L-MoS<sub>2</sub>. Although modifications to the optical properties of 1L-MoS<sub>2</sub> were not always witnessed [30], the conducted studies were able to observe changes in the structural and electronic properties of the material, as witnessed by changes in the Raman spectra, and a decrease in the output current of the analyzed device [30, 32]. Also, in this case, sulfur vacancies were found and blamed to be the underlying cause for the decrease in electronic performance. Nonetheless, other studies indicate that electron irradiation could improve the performance of transistors incorporating 1L-MoS<sub>2</sub> [29] and improve its PL [31]. In these cases, it was hypothesized that the irradiation effects were caused by interfacial reactions such as the atomic diffusion at the metal contacts of the device and oxidation of MoS<sub>2</sub>.

The type of mechanism taking place during the interaction between the electrons and the atoms of 1L-MoS<sub>2</sub> depends on the energy of the irradiation beam. In order for a knock-on displacement of an atom to occur following a collision, the electron of the beam must be able to transfer to the atom an energy higher than a threshold value related to the vacancy formation energy of that atom in the material. Otherwise, only ionization processes due to interactions between the electrons of the beam with those of the shell of the material's atoms can occur [50]. As consequence, while the latter are always present, the displacement ones can be activated by changing the energy of the beam. The kinetic energy threshold of the beam electrons needed for a displacement damage event and the formation of a vacancy can be calculated by considering the energy transferred during an electron-nucleus collision [51]. In the case of 1L-MoS<sub>2</sub>, the formation energies of sulfur and molybdenum vacancies have been calculated to be equal to 6.1 and 13.9 eV, respectively [52]. By taking these values into consideration, as explained in the Supporting Information, thresholds below 100 keV and at about 400 keV are obtained for the formation of chalcogen and metal atom vacancies, respectively. Below these values, no vacancy can be formed by knock-on, while other mechanisms must be taken into account. Indeed, soft collisions can cause effects such as excitation or ionization of the atom, then leading to the generation of a vacancy as a secondary process [24].

The electron beam energy used in our experiments is high enough that the formation of vacancies can be considered due to knock-on collisions and that sulfur vacancies are the main type of defect expected by electron irradiation. Indeed, molybdenum vacancies would be generated in a very small number since the used energies are just above the threshold needed for their formation. At higher energies, instead, both sulfur and molybdenum vacancies are formed, as shown in Figure S3. Studies conducted with 1 MeV electrons in ambient atmosphere (while our irradiations were performed under He atmosphere) have shown the modification of the Raman and PL bands of 1L-MoS<sub>2</sub> associated with vacancy formation and MoS<sub>2</sub> oxidation [30, 31]. Another study conducted by our group at 2.5 MeV was able to witness a decrease in the intensity and a redshift in the position of the PL of 1L-MoS<sub>2</sub> [33]. By considering the higher energies involved in that experiment, we conclude that both sulfur and molybdenum vacancies must have formed and that the total concentration of defects is therefore higher. By using the data of the simulation reported in Figure S3, we can estimate that at 2.5 MeV, there are about 4 times more defects

related to S and Mo vacancies. In the 2.5 MeV experiments, a relevant difference in the PL intensity was observed already at a fluence of about  $4 \times 10^{14} \text{ e}^- \text{ cm}^{-2}$ , a value that is two orders of magnitude lower than the maximum fluence used for the irradiation at 0.5 MeV. Being this difference larger than the factor 4 in defect concentration, we can suggest that the origin of variations of the PL signal in the 2.5 MeV irradiated samples is related to the larger amount of Mo vacancies. By contrast, in the reported experiment, the explored conditions only allow for the formation of sulfur vacancies, and they did not yield any significant change in the structural, electronic, and optical properties of 1L-MoS<sub>2</sub>. This result suggests that, differently to what discussed in the literature, sulfur vacancies either have a smaller role on the properties of 1L-MoS<sub>2</sub> or they are less stable in our samples. Overall, the lack of effects on either the structural or optical properties indicates that the effect of irradiation on 1L-MoS<sub>2</sub> should be explored in more detail. Our analysis has shown a remarkable resistance of this monolayer material to the effects of electron irradiation at 0.5 MeV and a good resistance at 2.5 MeV [32], especially in terms of structural and electrical properties in a range of fluences up to  $4 \times 10^{16} \text{ e}^- \text{ cm}^{-2}$  (which roughly corresponds to about  $10^4$  kGy in both cases). Although measurements of the electrical properties of the studied 1L-MoS<sub>2</sub> would result in a more direct proof of its resistance to the effects of irradiation, performing such experiments requires the deposition of electrical contacts on top of small sized flakes. As discussed in the literature, electron irradiation can, in some cases, alter the quality of such contacts [29, 30], thus changing the properties of the device without, in line of principle, influencing the 1L-MoS<sub>2</sub> flakes. The performed statistical analysis of the spectroscopic characteristics of the studied monolayer flakes, on the other hand, is not affected by such aspect and has been used in previous investigations to characterize the sign and the concentrations of the free carriers [44, 45, 53]. Finally, further analysis conducted as a function of irradiation energy would give additional insight into the formation mechanisms and threshold energies of Mo and S vacancies, helping to clarify their role on the properties of 1L-MoS<sub>2</sub>.

### 3 | Conclusion

Monolayer MoS<sub>2</sub> flakes obtained via two different methods were characterized before and after high-energy electron irradiation. The methods used for preparing the studied flakes are representative of the most common approaches for the production of 1L-MoS<sub>2</sub>, consisting in the mechanical top-down exfoliation method assisted by gold substrate and the bottom-up chemical vapor deposition. By means of Raman spectroscopy, the structural and electronic properties of the analyzed 2D material were studied and statistically investigated for a set of five different flakes for each production method. It was found that mechanical exfoliation samples present a tensile strain in their structure, while samples obtained via the chemical route display a compressive strain. Additionally, a higher concentration of negative charge carriers was found in the case of CVD grown samples, compared to those obtained via GAE. In general, gold exfoliation resulted in MoS<sub>2</sub> flakes presenting more inhomogeneous properties. Moreover, the optical properties of our samples were also studied by means of PL spectroscopy, once again analyzing the amplitude and energy distributions of the photoemission generated by the same ten flakes. Although slightly, 1L-MoS<sub>2</sub>

produced via mechanical exfoliation displayed a stronger emission intensity. At the same time, the energy distribution of the observed PL bands resulted comparable for both kinds of samples. Once again, a larger inhomogeneity was displayed by the optical properties of GAE samples, in line with the Raman characterization results. After irradiating the studied samples with 0.5 MeV electrons, no significant changes were found in any of their properties regardless on the used substrate and preparation route. At the electron beam energy used for our experiments, sulfur vacancies are the defects most likely to form by displacement damage mechanism. This consideration allows us to hypothesize that, when the used irradiation beam consists of low-energy electrons, the effects of irradiation on monolayer MoS<sub>2</sub> discussed in the literature are caused by excitation and ionization mechanisms happening because of soft collisions. In the case of experiments conducted at higher energy, considering that a consistent number of molybdenum vacancies should be formed above 0.5 MeV, we can suggest that the witnessed effects should be due to Mo vacancies and that, as a consequence, limiting their generation is of primary relevance to improve radiation hardness. The lack of changes in samples irradiated at 0.5 MeV in which sulfur vacancies should be induced is in line with recent works indicating that the negative doping character of 1L-MoS<sub>2</sub> is not due to this kind of defect [36]. Finally, electron irradiation should be studied in more detail in order to reach a conclusive understanding of its effect. Such experiments would yield important knowledge regarding the usage of devices based on 1L-MoS<sub>2</sub> for extreme-environment applications. By testing that such material is resistant to the effects of electron irradiation, we are proving that 1L-MoS<sub>2</sub> could guarantee long-lasting energy production when used for satellite-mounted solar cells. Additionally, by carefully studying the electron energy dependence of such effect, the role of Mo and S vacancies on the properties of monolayer MoS<sub>2</sub> could be better understood from a fundamental point of view.

## 4 | Experimental Methods

### 4.1 | Preparation of GAE Samples

The samples' substrate are obtained by sputtering a film of Ni (10 nm) and of Au (10 nm) on top of SiO<sub>2</sub> (900 nm)/Si using a DC magnetron sputtering in a Q300T D PLUS system (Quorum Technologies, Laughton, United Kingdom), as described in previous publications [44]. Pressure was maintained at a maximum of 10<sup>-2</sup> mbar during the sputtering. A bulk MoS<sub>2</sub> stamp obtained from a freshly cleaved 2H-MoS<sub>2</sub> crystal was then pressed against the substrates. The pressing onto the substrate step was carried out immediately after the sputtering procedure to prevent contamination of the surface.

### 4.2 | Preparation of CVD Samples

The samples were prepared by Liquid-Precursors-Intermediated Chemical Vapour Deposition (LPI-CVD) at 820°C under sulfur vapor flux, using as Mo precursor a spin-coated solution of ammonium heptamolybdate tetrahydrate (AHT, Sigma-Aldrich, 99.98%) 0.003 mol/L, sodium idroxiide (NaOH) 0.060 mol/L, and OptiPrep™ (a solution of 60% (w/v) iodixanol in water, with a concentration of 0.5 ml for 5 ml). The CVD growth was carried out in an open tube with nitrogen as the carrier gas. Further details

can be found in references [38, 54, 55]. The used substrates were 290 nm thick SiO<sub>2</sub> coated silicon wafers.

### 4.3 | Sample Characterization

Structural, electronic, and optical properties of 1L-MoS<sub>2</sub> were studied via Micro-Raman and micro-PL spectroscopy using a LabRam HR-Evolution Spectrometer system (HORIBA France SAS, Lyon, France) coupled to a confocal microscope with a pin-hole of 200 μm and a 100× objective. A laser excitation source with a wavelength of 532 nm was used to record all spectra. Laser power was reduced to 1% or 5% of the 100 mW maximum using an ND filter, for CVD and GAE samples, respectively. 1800 lines/mm and 600 lines/mm gratings were used to acquire Raman data and PL spectra, respectively. All spectra were collected across a 15 × 15 μm<sup>2</sup> area, with a step size of 1.5 and 2 μm for Raman and PL measurements, respectively. Reported values calculated from the Raman and PL spectra correspond to the average of all the different measurements acquired across the relevant areas. The Raman peak at 520.70 cm<sup>-1</sup> arising from the Si present in the samples' substrate was used for Raman spectral calibration. The integrated Raman bands of 1L-MoS<sub>2</sub> have been used to normalize the PL bands intensity to take into account excitation instabilities and differences in focus position of the microscope objective. Spectral bands were fitted with Lorentzian-shaped peaks. All reported uncertainties correspond to three times the standard deviation.

### 4.4 | Irradiation Experiments

Electron irradiation was carried out with a Pelletron accelerator at Sirius facility (LSI, France) exploring the fluence range 8 × 10<sup>12</sup> e<sup>-</sup> cm<sup>-2</sup> up to 4 × 10<sup>16</sup> e<sup>-</sup> cm<sup>-2</sup> at room temperature (approximately 25°C) and under a 0.4 mbar He atmosphere. Electron beam energy was fixed at 0.5 MeV. Beam diameter was equal to 20 mm, and a current of 0.1 μA was used for the lowest fluence while 1 μA for all other fluences. Total fluence was changed by selecting the duration of irradiation.

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## Conflicts of Interest

The authors declare no conflicts of interest.

## Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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## Supporting Information

Additional supporting information can be found online in the Supporting Information section. **Supporting Fig. S1:** Optical microscopy image obtained with a 10× objective of 1L-MoS<sub>2</sub> flakes grown by CVD. Scale bar corresponds to 20 μm. **Supporting Fig. S2:** Optical microscopy image obtained with a 10× objective of 1L-MoS<sub>2</sub> flakes obtained by GAE. Scale bar corresponds to 20 μm. **Supporting Fig. S3:** SECTE<sup>[S4]</sup> simulated cross-sections for Mo and S vacancy production as a function of electron beam energy, displacement threshold energy 6.1 and 13.9 eV for S and Mo respectively. **Supporting Fig. S4:** Variation with respect to the not irradiated sample of the amplitude of the 2LA(M) Raman peak of 1L-MoS<sub>2</sub> grown by CVD (A) or obtained by GAE (B) as a function of irradiation fluence. **Supporting Fig. S5:** Variation of the B emission band energy of the PL of 1L-MoS<sub>2</sub> grown by CVD (A) or obtained by GAE (B) as a function of irradiation fluence.