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ABSTRACT

Thermal sulfurization of ultra-thin Mo-based films represents a promising approach for large-area growth of MoS₂. In this paper, we demonstrated that the crystalline quality (domains size and defects density), strain, doping, and light emission properties of monolayer (1L) MoS₂ obtained from sputter deposited MoO_x films on a c-sapphire substrate can be tailored by the sulfurization temperature (T_s) in the range from 700 to 800 °C. Starting from a continuous film with a nanocrystalline domains structure at T_s = 700 °C, a distribution of 1L MoS₂ triangular domains with 2.1 ± 0.6 and 2.6 ± 1.6 μm average sizes was obtained by increasing T_s to 750 and 800 °C, respectively. The increase in T_s was accompanied by a strong (25×) enhancement of the photoluminescence (PL) intensity. Furthermore, the average doping of MoS₂, evaluated from Raman analyses, evolved from a strong p-type doping (~1 × 10¹³ cm⁻²) after T_s = 700 °C, ascribed to residual MoO₃ in the film, to a low average n-type doping (~0.04 × 10¹³ cm⁻²) after T_s = 800 °C. The wide tunability of doping and PL of 1L MoS₂ by the sulfurization temperature can be exploited to tailor material properties for different specific applications.

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Molybdenum disulfide (MoS₂) belongs to the family of transition metal dichalcogenides (TMDs), which are layered materials formed by stacks of X–M–X trilayers (being M = Mo, W transition metals and X = S, Se, and Te chalcogens) with covalent in-plane bonds and weak van der Waals interaction between the layers.¹ The bandgap tunability as a function of the number of layers is one of the most appealing features of TMDs. In particular, MoS₂ exhibits an indirect bandgap (1.2 eV) for bilayer to multilayer films, and a direct bandgap of 1.8–1.9 eV for a monolayer (1L).^{2,3} This sizable energy bandgap, combined to good electron mobility values (up to ~200 cm²/V s),^{4,5} makes MoS₂ a promising channel material for future ultra-thin body field effect transistors (FETs) beyond silicon.⁶ In addition to electronics/optoelectronics,^{7,8} MoS₂ has been explored in several other fields, including sensing,⁹ photocatalysis,¹⁰ and photovoltaics.¹¹ This wide application potential currently motivates the search of scalable

approaches to produce MoS₂ films with high control in terms of thickness uniformity, crystalline quality, doping, and photoluminescence (PL) properties.¹² To this purpose, both top-down and bottom-up methods have been explored, which exhibit advantages and drawbacks. Among top-down approaches, mechanical exfoliation from bulk crystals still provides the highest electronic quality MoS₂, despite the limited size and poor thickness control of produced MoS₂ flakes.^{13,14} Recently, larger area (from mm² to cm²) 1L MoS₂ membranes with excellent crystalline quality have been obtained by mechanical exfoliation from MoS₂ stamps on metal surfaces (particularly gold), followed by transfer to the final insulating substrates.^{15,16} However, extending this method on wafer scale is currently not viable, especially due to the lack of MoS₂ bulk stamps of suitable size. Among bottom-up methods, chemical vapor deposition (CVD) is the most widely used and cost-effective approach for MoS₂ growth. To date, several variants of CVD

have been explored. The most common implementation is the vapor phase growth within a two-heating zone tube furnace, using S and MoO₃ powders as separate sources of the vapor precursors.¹⁷ This method proved to be effective to obtain 1L MoS₂ flakes, typically with triangular shape and variable size (from ~1 to ~100 μm), on different substrates, including SiO₂,¹⁷ sapphire,¹⁸ SiC,^{19,20} and GaN.^{21,22} However, the difficulty to simultaneously control S and MoO₃ vapor fluxes on the target substrate^{23,24} makes it challenging to achieve uniform MoS₂ coverage on wafer scale, whereas variations in the size, shape, and thickness of the flakes are typically observed in different wafer regions.^{17,25} In this context, metal organic chemical vapor deposition (MOCVD) emerged in the last few years as the most advanced approach for wafer scale growth of MoS₂ (and other TMDs), thanks to the excellent control of the process parameters, such as the gas precursors fluxes within the reaction chamber.²⁶ The excellent uniformity and electronic/optoelectronic quality of MOCVD grown TMDs recently enabled impressive advancements in integrated electronics/optoelectronics, like the demonstration of active pixel arrays of MoS₂ phototransistors,²⁷ and 3D integration of up to three levels (tiers) of 2D field effect transistors.²⁸ Proper optimization of MOCVD also allowed direct deposition of MoS₂ at low temperature (<400 °C) for back-end-of-line applications.²⁹ Based on these progresses, it is expected that MOCVD will represent the method of choice for 3D monolithic integration of TMDs and the development of More-than-Moore applications. However, due to their complexity and high maintenance cost, MOCVD equipment remain not affordable for most academic laboratories and small enterprises. Hence, alternative approaches enabling wafer scale MoS₂ growth using commonly available equipment are still highly desirable. In this respect, a viable alternative to achieve uniform MoS₂ coverage on large area is the annealing under a S flux of Mo-based thin films pre-deposited on the substrate either by physical deposition (e.g., Mo or MoO_x sputtering or evaporation^{30–32}) or by spin coating of liquid precursors [such as ammonium molybdate tetrahydrate (NH₄)₆Mo₇O₂₄].³³ In this case, the uniformity of MoS₂ on large area is controlled by the initial Mo-based film coverage and thickness. As an example, uniform coating of SiC and GaN surfaces with 1L MoS₂ has been recently achieved by sulfurization at a temperature of 700 °C of sputter deposited ultra-thin (1–2 nm) MoO_x films.^{34–36} The uniform 1L MoS₂ obtained under these sulfurization conditions was composed of domains with sizes from few tens to hundreds of nanometers, reminiscent of the nano-structure of the starting MoO_x film. Photoluminescence from these 1L MoS₂ films was significantly quenched, probably due to non-radiative excitonic recombination at domains boundaries. Furthermore, differently from the unintentional n-type doping typically observed in exfoliated or CVD grown MoS₂, these 1L-MoS₂ films showed a significant p-type doping explained by the presence of residual MoO₃ after the sulfurization process.^{34,35} On the other hand, a discontinuous coating by micrometer-size triangular 1L MoS₂ domains with intense photoluminescence and a low average n-type doping has been recently reported by sulfurization at 800 °C of ultra-thin MoO_x films on c-sapphire substrates.³⁷ These results indicate the possibility of tailoring the structural, electronic, and optical emission properties of 1L MoS₂ by the sulfurization approach. However, a deeper understanding of the impact of relevant process parameters on the mechanisms of MoS₂ formation and the final crystalline quality, photoluminescence, and doping behavior is still required.

In this paper, optical and atomic force microscopy (AFM), combined with micro-Raman and micro-PL spectroscopy, have been employed to systematically investigate the role played by the sulfurization temperature T_s, in the range from 700 to 800 °C, on the morphological properties (coverage, domain size, and thickness), PL, and doping/strain of ultra-thin MoS₂ grown on sapphire. Starting from MoO_x films with identical initial thickness, we observed the evolution from a continuous nanocrystalline film, showing low photoluminescence and p-type doping (at T_s = 700 °C) to a distribution of single crystalline micrometer-size triangular 1L MoS₂ domains, showing prominent photoluminescence and n-type doping (at T_s = 800 °C). Based on these results, MoS₂ formation in this temperature range was explained by the interplay of different physical/chemical mechanisms, i.e., (i) the vapor–solid reaction between S and MoO_x, leading to the conversion of the starting MoO_x into a uniform nanocrystalline MoS₂ layer at T_s = 700 °C, and (ii) MoO_x evaporation/diffusion from/on sapphire surface, playing a crucial role in the growth of micrometer MoS₂ domains at T_s = 750 and 800 °C.

The starting MoO_x films on the c-sapphire substrates were obtained by DC magnetron sputtering from a Mo target using a Quorum Q300-TD Plus equipment, followed by natural oxidation in air. Samples with identical ~1.8 nm thickness, evaluated by AFM step height measurement, were used for subsequent sulfurization experiments, carried out within a quartz tube with two different heating zones. The MoO_x/sapphire samples were placed in the central zone of the tube, heated at T_s = 700, 750, or 800 °C, whereas a crucible with sulfur powder was hosted in zone at a lower temperature of 150 °C. S vapors from the crucible were transported to the sample surface by a continuous Ar flux of 100 sccm for the entire duration of the process. The uniformity of MoS₂ coverage and the domains size were evaluated by optical microscopy and by AFM (operated in the tapping and peak-force tapping modes) using a Bruker Dimension Icon equipment with Si tips. Micro-PL spectra on the MoS₂ layers were acquired with a Horiba HR-Evolution system with a laser excitation wavelength of 532 nm (nominal maximum power 100 mW) in confocal configuration (100× objective). Furthermore, micro-Raman maps were collected using a WiTec Alpha equipment with a laser excitation at 532 nm, 1.5 mW power, and 100× objective.

The AFM morphology of the bare sapphire substrate and of the as-deposited MoO_x film are reported in Figs. S1(a) and S1(b) of the [supplementary material](#). The c-sapphire surface is nearly atomically flat with a root mean square (RMS) roughness of 0.1 nm, while as-deposited MoO_x exhibits a nano-grains morphology, resulting in a higher RMS = 0.25 nm than the bare substrate. [Figure 1](#) illustrates the evolution of MoS₂ coverage on sapphire surface, evaluated by optical microscopy and AFM, as a function of the sulfurization temperature. After MoO_x sulfurization at T_s = 700 °C, the presence of continuous MoS₂ film was deduced from the uniform optical contrast [see [Fig. 1\(a\)](#)]. This film also exhibits a nano-grains structure, as observed by the AFM image and the high resolution insert in [Fig. 1\(b\)](#). As compared to the morphology of as-deposited MoO_x, the formation of larger grains, resulting in an increased RMS = 0.35 nm, were observed after sulfurization at T_s = 700 °C, as illustrated in [Fig. S1\(c\)](#) of the [supplementary material](#). To evaluate the size distribution of MoS₂ nano-grains, maps of the tip-surface adhesion force were acquired in the peak-force tapping mode. The adhesion force proved to be very sensitive in discriminating between the inner part of nano-domains and

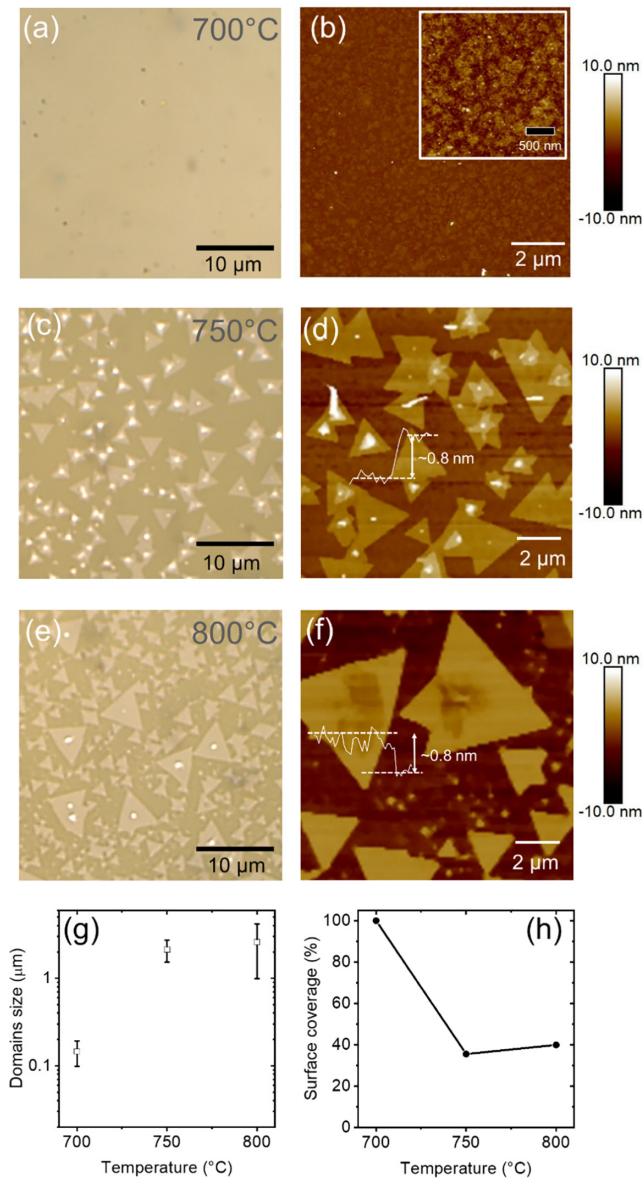


FIG. 1. Optical and morphological (AFM) images obtained for the MoS₂ on c-sapphire grown at 700 °C (a) and (b), 750 °C (c) and (d), and 800 °C (e) and (f). (g) Average MoS₂ domains size and (h) percentage of surface coverage on c-sapphire as a function of the sulfurization temperature.

their grain boundaries, as illustrated in Fig. S2(a) of the [supplementary material](#). Figure S2(b) shows a histogram of lateral grains size distribution, obtained by statistical analysis on the adhesion force map, from which a MoS₂ domain size $d = 150 \pm 65$ nm was evaluated. The continuous nanograin morphology of the film after sulfurization at 700 °C indicates a direct vapor–solid reaction between the incoming S and MoO_x rules MoS₂ formation without a significant redistribution of Mo on the sapphire surface by diffusion processes. An average thickness of the MoS₂ film of ~ 0.8 nm was evaluated by AFM step height measurements carried out on intentionally scratched regions of the film.

A very different morphology was observed by increasing the sulfurization temperature to $T_s = 750$ °C, i.e., the partial coverage of sapphire surface by triangular MoS₂ domains with a nearly uniform size distributions and average dimension of 2.1 ± 0.6 μm [as deduced from the optical microscopy in Fig. 1(c)]. The domains exhibit mainly 1L thickness, evaluated by step height AFM analyses in Fig. 1(d), except for a multilayer region present at the center of some triangles. Finally, as illustrated in Figs. 1(e) and 1(f), the sulfurization process at $T_s = 800$ °C results in a more inhomogeneous surface distribution of MoS₂ flakes, including both larger triangles (average size of 6.5 μm) and small ones (average size of 2 μm). The formation of these distributions of separated triangular domains indicates that the sulfurization processes at $T_s = 750$ and 800 °C are accompanied by partial MoO_x loss due to evaporation and a significant MoO_x surface diffusion. Such a scenario is supported by micro-Raman analyses performed inside and outside triangular domains (see Fig. S5 of the [supplementary material](#)), from which a bare sapphire crystal without any intermediate MoS_xO_y species in the region between MoS₂ domains can be also deduced.

Figures 1(g) and 1(h) summarize the changes in the MoS₂ domains size and the percentage of surface coverage as a function of T_s .

After this initial morphological analysis, the light emission properties of the MoS₂ layers produced at the different sulfurization temperatures were investigated by micro-photoluminescence (PL) spectroscopy. Figure 2(a) shows three representative PL spectra for MoS₂ grown at $T_s = 700$ °C (black line), 750 °C (red line), and 800 °C (blue line). The sapphire second order Raman peak has been subtracted from all the spectra. Furthermore, they were normalized with respect to the respective MoS₂ Raman peaks in order to make them comparable in terms of intensity. Each spectrum exhibits a maximum intensity, associated with the neutral MoS₂ exciton peak A^0 , and typical asymmetric shape, due to the convolution of two smaller components, i.e., the trion component at lower energy and the exciton component B at higher energy, respectively.³⁸ Noteworthy, the A^0 peak intensity showed a strong increase with increasing the sulfurization temperature. Specifically, starting from a quenched PL signal for the MoS₂ film grown at $T_s = 700$ °C (multiplied by a factor of 5 to make it clearer), an enhancement of the peak's intensity by a factor of ~ 10 was detected for MoS₂ domains grown at $T_s = 750$ °C, and a further $\sim 2.5 \times$ increase was observed for domains grown at $T_s = 800$ °C, as summarized in Fig. 2(b), left scale. Furthermore, the increase in peaks intensity is accompanied by a slight increase in the A^0 peak energy by ~ 20 meV with increasing T_s from 700 to 800 °C, as shown in Fig. 2(b), right scale. The observed differences in the light emission intensity represent strong evidence of the different crystalline quality of the MoS₂ samples. In fact, structural defects (including impurities or grain boundaries) are known to act as non-radiative recombination centers for excitons, causing a significant quenching of light emission from 1L MoS₂.^{2,39}

In the following, a detailed micro-Raman spectroscopy investigation of the samples is reported to provide further insights on the thickness uniformity and crystalline quality of the MoS₂ films obtained at the different sulfurization temperatures. Figure 3(a) shows three representative Raman spectra collected after sulfurization at $T_s = 700$ °C (bottom), 750 °C (middle), and 800 °C (top). The characteristic in-plane (E_{2g}) and out-of-plane (A_{1g}) first-order peaks of MoS₂ are clearly observed in the 370 – 420 cm^{-1} wavenumber range. The wavenumber difference ($\Delta\omega = \omega_{A_{1g}} - \omega_{E_{2g}}$) between these two main vibrational

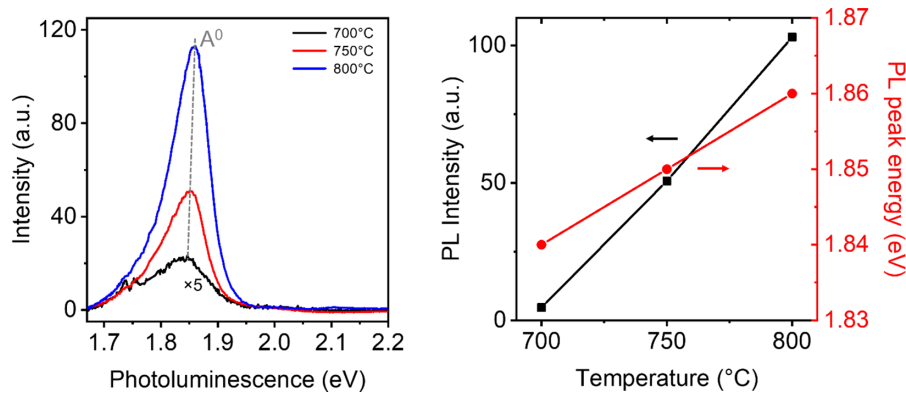


FIG. 2. (a) PL spectra for 1L-MoS₂ on c-sapphire at 700 °C (black line), 750 °C (red line), and 800 °C (blue line), where all the spectra are normalized with respect to the Raman MoS₂ peak. (b) Corresponding evolution of the PL exciton peak intensity and position as a function of the temperature, represented by black squares and red circles.

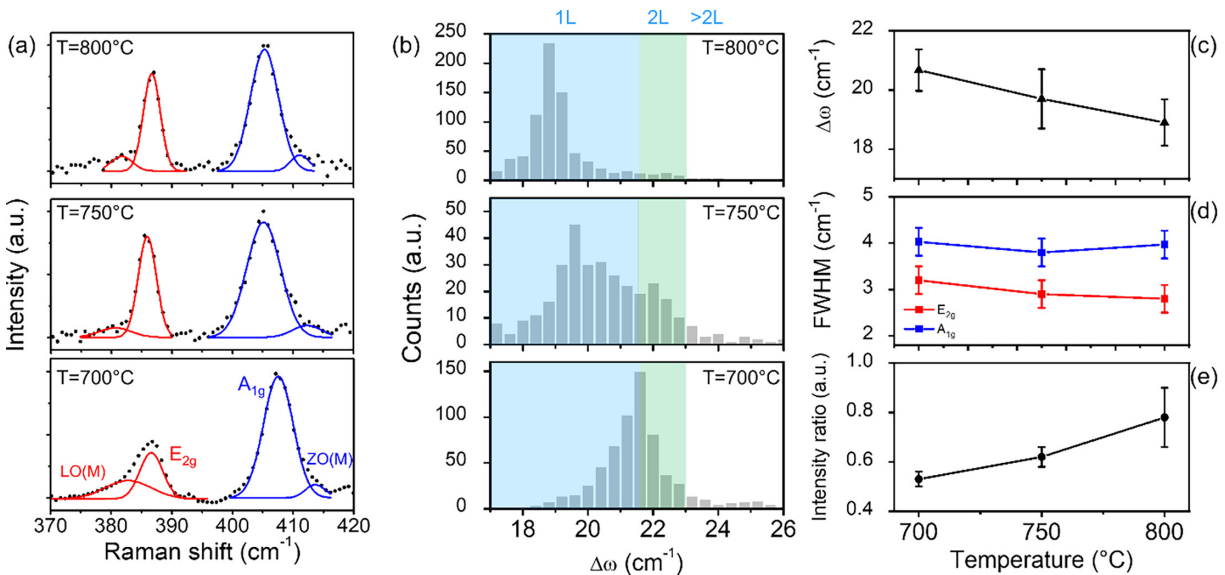


FIG. 3. (a) Raman spectra of MoS₂ grown on c-sapphire at 700 °C (bottom), 750 °C (middle), and 800 °C (top). (b) Histogram of the MoS₂ vibrational modes difference $\Delta\omega$ ($\omega_{A_{1g}} - \omega_{E_{2g}}$) distributions at 700 °C (bottom), 750 °C (middle), and 800 °C (top), where the blue, green, and white regions indicate the 1L, 2L, and multilayers MoS₂ ranges, respectively. Trend of (c) $\Delta\omega$, (d) FWHM, and (e) intensity ratio (E_{2g}/A_{1g}) as a function of the temperature.

peaks is commonly correlated with the number of layers in a MoS₂ film, with $\Delta\omega$ from ~ 18 to ~ 21.5 cm⁻¹ corresponding to 1L and larger $\Delta\omega$ values associated with 2L and multilayers.⁴⁰ Figure 3(b) reports the histograms of $\Delta\omega$ values in the range between 18 and 26 cm⁻¹, obtained by performing a statistical analysis on a large number of Raman spectra collected on the three samples. After sulfurization at $T_s = 700$ °C, the $\Delta\omega$ distribution was centered at 21.5 ± 1.2 cm⁻¹, indicating that the obtained MoS₂ film was mainly composed of 1L/2L with a small fraction corresponding to multilayers regions. Increasing the temperature to $T_s = 750$ °C, the maximum of $\Delta\omega$ distribution undergoes a shift to a lower wavenumber value (19.7 cm⁻¹) associated with 1L MoS₂, while a multilayers contribution is still present, due to the thicker central region of some triangular domains [as shown by the optical and AFM images in Figs. 1(c) and 1(d)]. Finally, sulfurization at $T_s = 800$ °C results in a narrow $\Delta\omega$

distribution centered at ~ 19 cm⁻¹, corresponding to 1L MoS₂, with negligible multilayer contributions. The shift of the maximum value of the $\Delta\omega$ distributions from ~ 21.5 to ~ 19 cm⁻¹ with increasing T_s , as illustrated in Fig. 3(c), can be correlated with differences in the strain and doping of 1L MoS₂ membranes grown at the different temperatures^{16,41,42} as discussed later on in this paper.

In addition to statistical information on the MoS₂ thickness uniformity, Raman analyses provided information on the crystalline quality of layers grown at the different sulfurization temperatures. The deconvolution analysis of the representative Raman spectra in Fig. 3(a) using Gaussian contributions allowed to identify, in addition to the E_{2g} and A_{1g} modes, two further components corresponding to the longitudinal [LO(M)] and out-of-plane optical modes [ZO(M)]. These vibrational modes, typically absent in the Raman spectra of a defect-free MoS₂, are activated by the presence of disorder or grain boundaries in

the MoS₂ crystalline structure, as explained by the phonon confinement model.^{43,44} In particular, due to its nanocrystalline structure, the uniform MoS₂ film grown at T_s = 700 °C exhibits a significantly larger LO(M) contribution as compared to the micrometer-size single crystalline 1L MoS₂ domains obtained at higher T_s. On the other hand, the ZO(M) contribution was not particularly affected by the sulfurization temperature. In addition to the LO(M) and ZO(M) contributions, directly related to defects, the crystalline quality of 1L MoS₂ can be also evaluated by the FWHM and the intensity ratio of the two main vibrational modes E_{2g} and A_{1g}. These parameters were extracted by repeating the fitting procedure illustrated in Fig. 3(a) on the entire sets of Raman spectra for the three different samples. As summarized in Figs. 3(d) and 3(e), the FWHM of the E_{2g} and A_{1g} modes exhibits a decrease with increasing T_s,⁴⁵ whereas their intensity ratio (I_{E2g}/I_{A1g}) shows a significant increase.⁴⁵ Specifically, it was observed an I_{E2g}/I_{A1g} enhancement of ~68% when the MoS₂ was grown at 800 °C compared to that obtained at 700 °C. These aspects confirmed the higher MoS₂ crystalline quality obtained for T_s > 700 °C.

As anticipated, the shift of the Δω distributions maximum toward lower values with increasing T_s [as reported in Fig. 3(c)] can be ascribed to changes in the strain and doping of 1L MoS₂. In particular, the local values of the strain (ε) and doping (n) can be quantitatively extracted from micro-Raman spectra by a correlative plot of the A_{1g} vs E_{2g} peak positions.¹⁶ Figure 4(a) shows such a correlative plot, where the black, red, and blue points are the experimental (ω_{E2g}, ω_{A1g}) peak positions for MoS₂ obtained by sulfurization at T_s = 700, 750, and 800 °C, respectively. The red and black straight lines represent the ideal ω_{E2g} vs ω_{A1g} behavior for a purely strained (strain line) and a purely doped (doping line) 1L MoS₂ membrane, respectively. The crossing point (ω_{A1g}⁰, ω_{E2g}⁰) of the strain and doping lines corresponds to the ideal case of unstrained and undoped MoS₂. Here, the Raman peak positions for a suspended 1L-MoS₂ (ω_E⁰ = 385 cm⁻¹, ω_A⁰ = 405 cm⁻¹) have been taken as the best approximations for this ideal case.⁴⁶ Four regions with different doping type (p and n type) and strain (tensile and compressive) are separated in the (ω_{E2g}, ω_{A1g}) area by the

intersection of the strain and doping lines, and the specific doping values for each of the data point in Fig. 4(a) can be extracted by simple geometrical calculations.

By performing this analysis on all the data points, the average value and standard deviation of the biaxial strain (ε) and doping (n) for 1L MoS₂ grown at the different temperatures were evaluated, as reported in Figs. 4(b) and 4(c). Compressive strain values in the range between -0.2% and -0.3% were obtained for the three MoS₂ samples independently on the sulfurization temperature [Fig. 4(b)]. Differently, significant variations in the doping values and type were observed by increasing T_s, going from a high p-type doping of (1.1 ± 0.2) × 10¹³ cm⁻² for T_s = 700 °C, to a smaller p-type doping of (0.17 ± 0.50) × 10¹³ cm⁻² for T_s = 750 °C, until reaching a small n-type doping of (0.04 ± 0.33) × 10¹³ cm⁻² for T_s = 800 °C, as shown in Fig. 4(c). The origin of the significant p-type doping in the MoS₂ sample obtained at the lowest sulfurization temperature can be ascribed to the presence of MoO₃ residues in the film, as demonstrated by XPS on samples grown under similar conditions.³⁴ In fact, MoO₃ is known to be a high work function oxide, leading to an overall Fermi level shift of MoS₂ to the valence band.⁴⁷ Thus, a reduction of MoO₃ residues in the MoS₂ domains obtained by sulfurization at higher temperature can explain the gradual reduction of the p-type doping and a transition to a low n-type average doping after T_s = 800 °C.

In conclusion, we demonstrated that the crystalline quality (domains size and defects density), strain, doping, and light emission properties of monolayer (1L) MoS₂ obtained starting from sputter deposited MoO_x films on a c-sapphire substrate can be tailored by the sulfurization temperature (T_s) in the range from 700 to 800 °C. Starting from a continuous film with a nanocrystalline domains structure at T_s = 700 °C, a distribution of 1L MoS₂ triangular domains with 2.1 ± 0.6 and 2.6 ± 1.6 μm average sizes was obtained by increasing T_s to 750 and 800 °C, respectively. Based on these results, MoS₂ formation in this temperature range was explained by the interplay of different physical/chemical mechanisms, i.e., (i) the vapor-solid reaction between S and MoO_x, leading to the conversion of the starting MoO_x

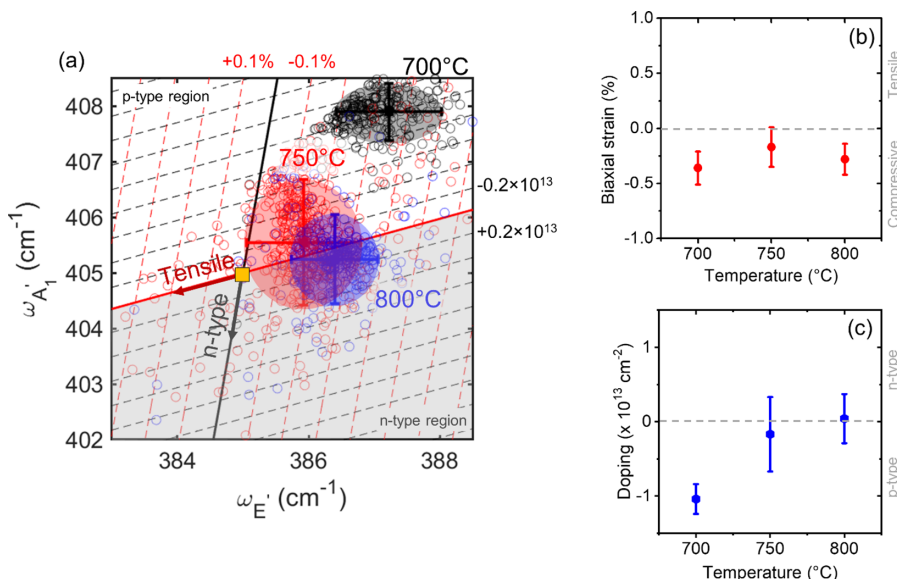


FIG. 4. (a) Correlative plot of the (ω_{E2g}, ω_{A1g}) Raman peak positions measured on MoS₂ obtained by sulfurization at T_s = 700 (black points), 750 (red points), and 800 °C (blue points). The red and black lines represent the ideal behavior for a purely strained and a purely doped 1L MoS₂ membrane, respectively, and their crossing point corresponds to the ideal case of unstrained and undoped MoS₂. (b) Biaxial strain as a function of the temperature, where the tensile and compressive contributions are separated by a gray dashed line. (c) Doping as a function of the temperature, where the n-type and p-type contributions are separated by a gray dashed line.

into a uniform nanocrystalline MoS₂ layer at T_s = 700 °C, and (ii) MoO_x evaporation/diffusion from/on sapphire surface, playing a crucial role in the growth of micrometer MoS₂ domains at T_s = 750 and 800 °C. The evolution of morphology with T_s was accompanied by a strong (25×) enhancement of the photoluminescence (PL) intensity, indicating a significant improvement of crystalline quality of the domains. Furthermore, the average doping of MoS₂, evaluated from Raman analyses, evolved from a strong p-type doping ($\sim 1 \times 10^{13} \text{ cm}^{-2}$) at T_s = 700 °C, ascribed to residual MoO₃ in the film, to a low average n-type doping ($\sim 0.04 \times 10^{13} \text{ cm}^{-2}$) at T_s = 800 °C. The wide tunability of doping and PL of 1L MoS₂ by the sulfurization temperature can be exploited to tailor material properties for different applications in electronics, optoelectronics, sensing, and catalysis. To this purpose, the main challenge for the scalability of this approach will be controlling the substrate temperature and sulfur flux on wafer scale, by designing of dedicated furnaces and adopting sulfur gas precursors instead of powders.

See the [supplementary material](#) for AFM morphology of sapphire, as-deposited MoO_x, and after sulfurization at 700 °C; tip-surface adhesion force map of MoS₂ grown at 700 °C and histogram of nano-domains size distribution; maps of Raman peaks frequency difference $\Delta\omega = \omega_{A1g} - \omega_{E2g}$ for MoS₂ obtained at 700 and 800 °C; Raman signatures for MoO_xS_y reported in the literature; measured Raman spectra inside and outside MoS₂ triangular domains obtained by sulfurization at 800 °C.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

Salvatore Ethan Panasci: Conceptualization (equal); Data curation (equal); Formal analysis (equal); Investigation (equal); Writing – original draft (equal). **Emanuela Schilirò:** Investigation (equal); Writing – review & editing (equal). **Antal Koos:** Formal analysis (equal); Investigation (equal); Writing – review & editing (equal). **Fabrizio Roccaforte:** Funding acquisition (equal); Project administration (equal); Writing – review & editing (equal). **Marco Cannas:** Methodology (equal); Writing – review & editing (equal). **Simonpietro Agnello:** Investigation (equal); Methodology (equal);

Writing – review & editing (equal). **Béla Pécz:** Funding acquisition (equal); Investigation (equal); Project administration (equal); Writing – review & editing (equal). **Filippo Giannazzo:** Conceptualization (lead); Funding acquisition (lead); Methodology (equal); Project administration (lead); Supervision (equal); Writing – review & editing (lead).

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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