

# High performance of nanostructured lead-acid batteries using rGO as additive for negative electrode

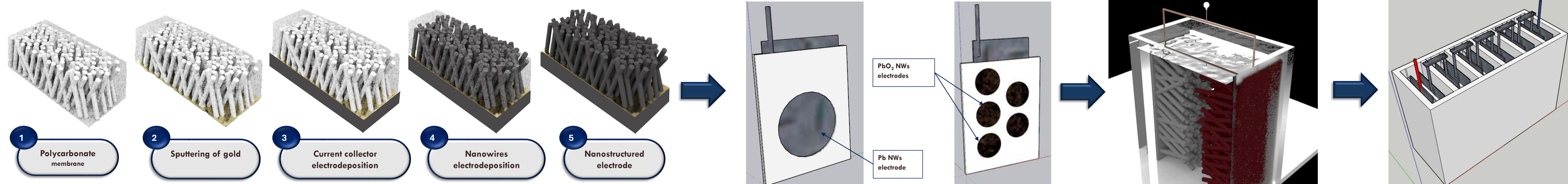
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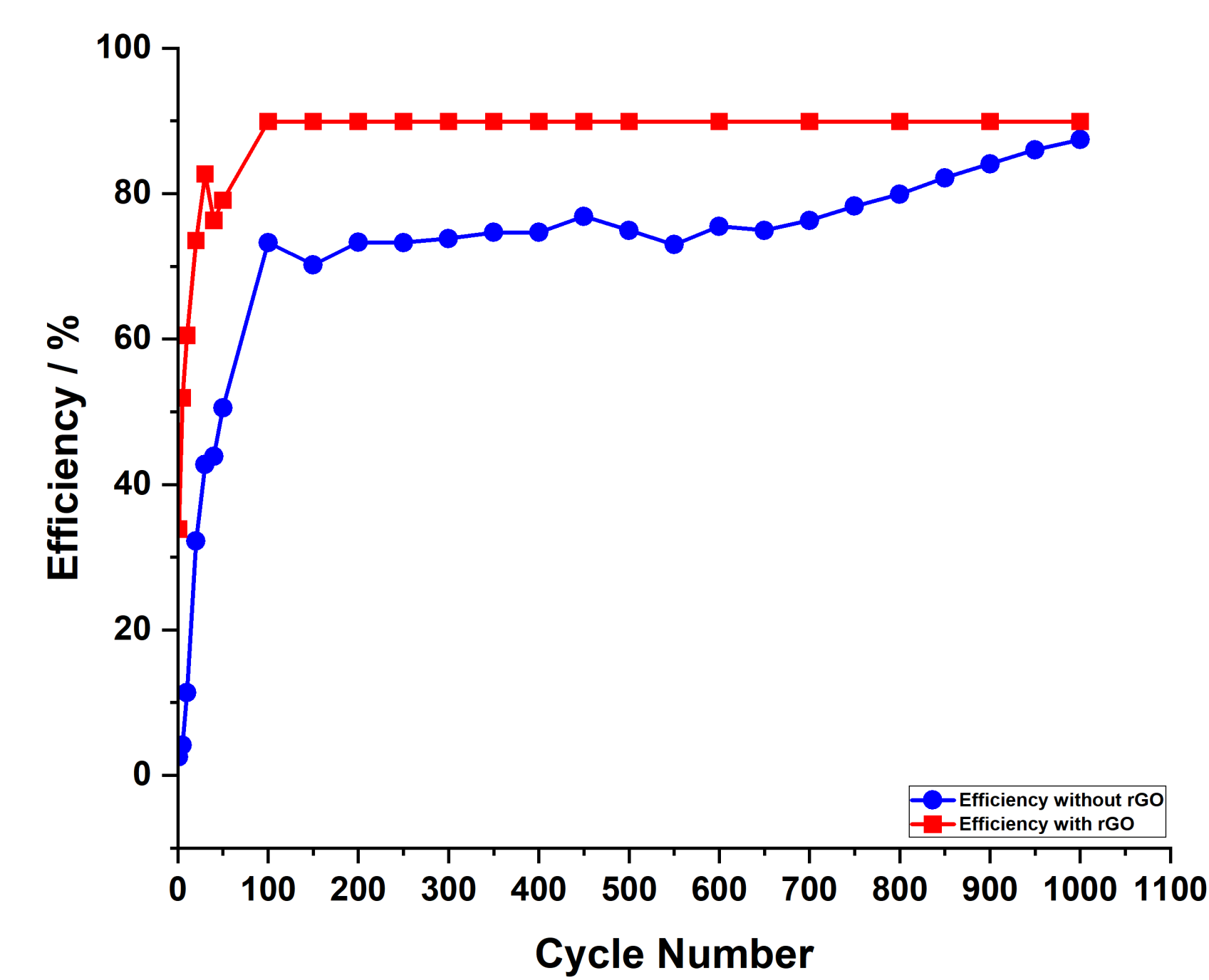
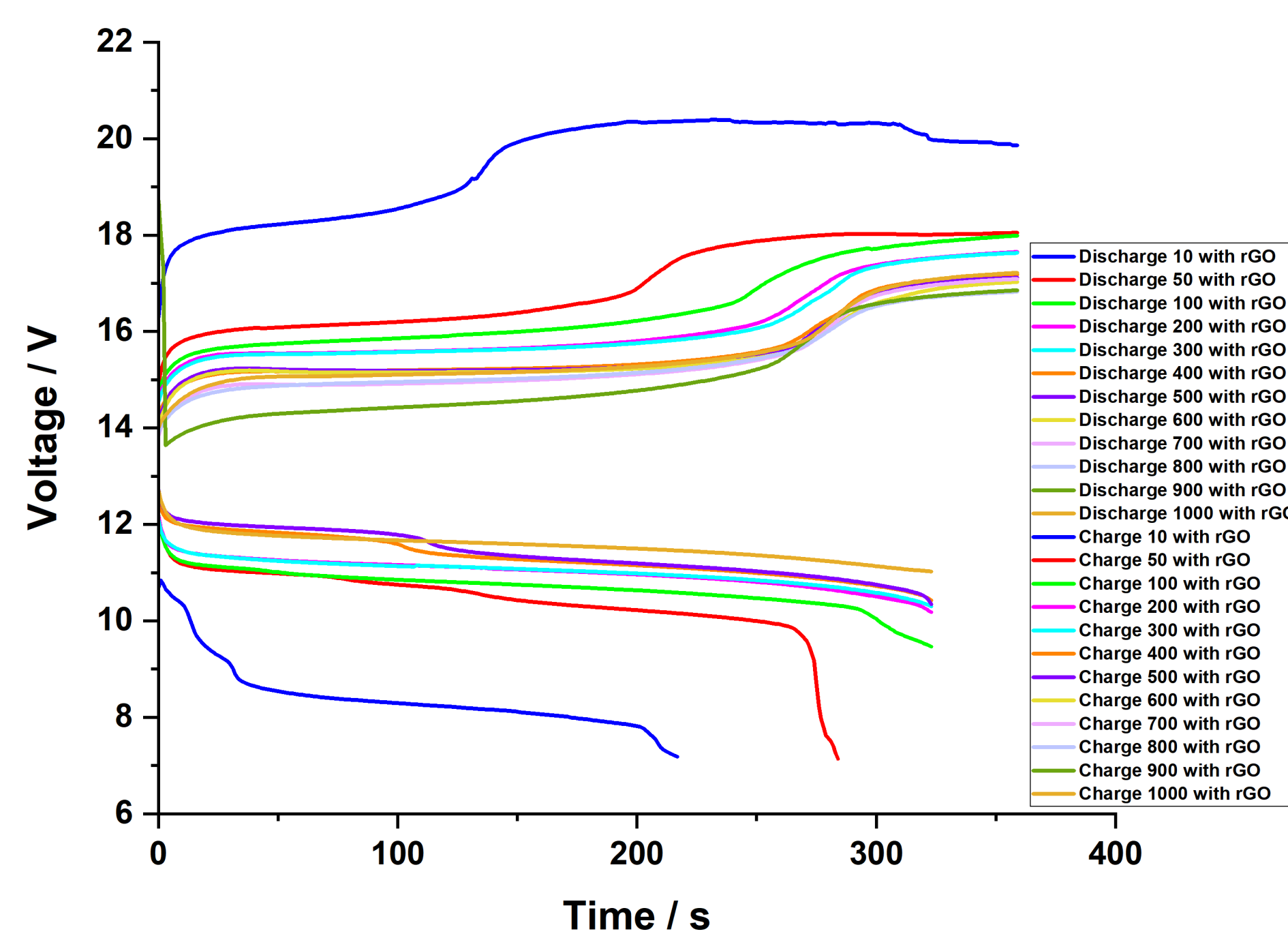
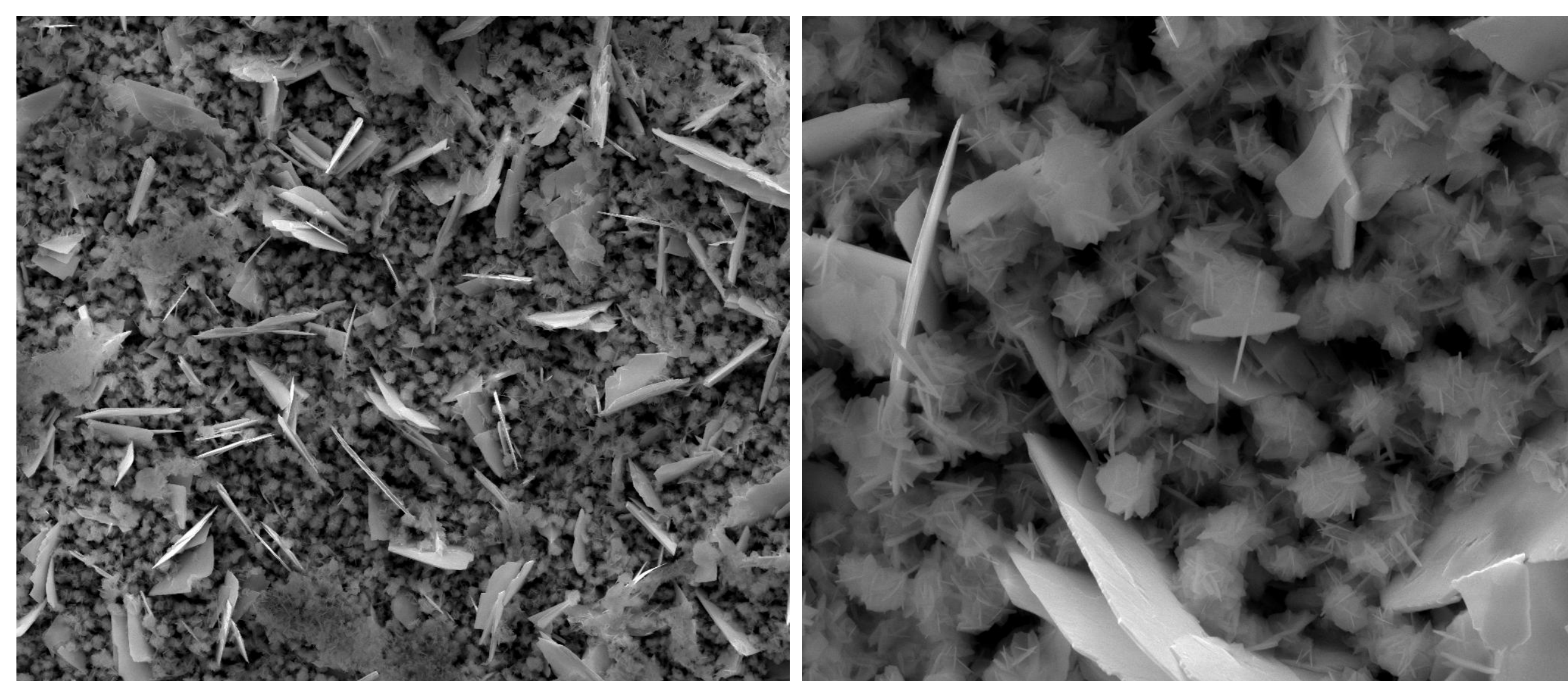
## Introduction

Lead-acid batteries (LABs) are among the most widely used and well-established energy storage technologies because of their dependability, affordability, simplicity of manufacture, and recyclability. Numerous studies and solutions have been put out over the years to enhance the performance of lead-acid batteries with the goal of lowering sulfurization occurrences while simultaneously extending life cycles and energy density. The use of nanostructured materials into energy storage devices stood out as a viable strategy to enhance LAB performance among all the suggested fixes. The creation of nanostructured electrodes, which have a large surface area and a high active material usage rate, is the main emphasis of this project's research activity.



## Experimental

In this work we present a 12V lead-acid batteries with nanostructured electrodes and using a Pb nanowires added with reduced graphene oxide (rGO) to improve their performance. This battery was cycled at high C-rate up to 10C (6 min to complete charge) and imposing a very deep discharge. These cycling conditions are much more stressful in terms of cut-off and charge/discharge rate in comparison to the usually functioning conditions of commercial batteries. The rGO contributes to better electrical conduction and decreases internal resistance, for higher efficiency during charging and discharging and a decrease in sulfation phenomena. Modified batteries exhibit higher specific capacity, longer cycle life, and better reaction to high currents than classic LABs.



## Results and discussion

The positive impact of rGO on nanostructured Pb electrodes was highlighted in this work. High-performance nanostructured Pb electrodes added with rGO were produced. Initially, Pb NW electrodes were electrodeposited subsequently rGO on their surfaces were deposited. Both the discharge potential and the charged-discharge rate of the electrodes were selected to simulate extremely stressful conditions. A C-rate equal to 10C and a depth of discharge equal to 90% were imposed. Excellent accomplishments were achieved in spite of these circumstances. The electrodes can actually operate at an efficiency of 90% for the duration of the test, which is quite close to the theoretical utilization of active mass. Additionally, when a constant current is provided, the electrode discharges at a voltage that is nearly constant. These improved results are ascribed to the increase on conductivity of the active material due to the presence of rGO, which inhibits the activity loss brought on by  $\text{PbSO}_4$  crystal formation.

## Acknowledgment

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